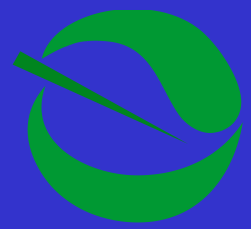




NACT 284
Volatile Organic Compound
Control Devices



Course Overview

- Volatile Organic Compound (VOC) Controls
- Examples of VOC Calculations
- Particulate Matter (PM) Options
- Inspection Strategies



Volatile Organic Compounds

Chemical definition of VOCs:

- Molecules which contain carbon
&
- High evaporative rate at low temperatures
- [$VP > 0.1\text{mm Hg}$]

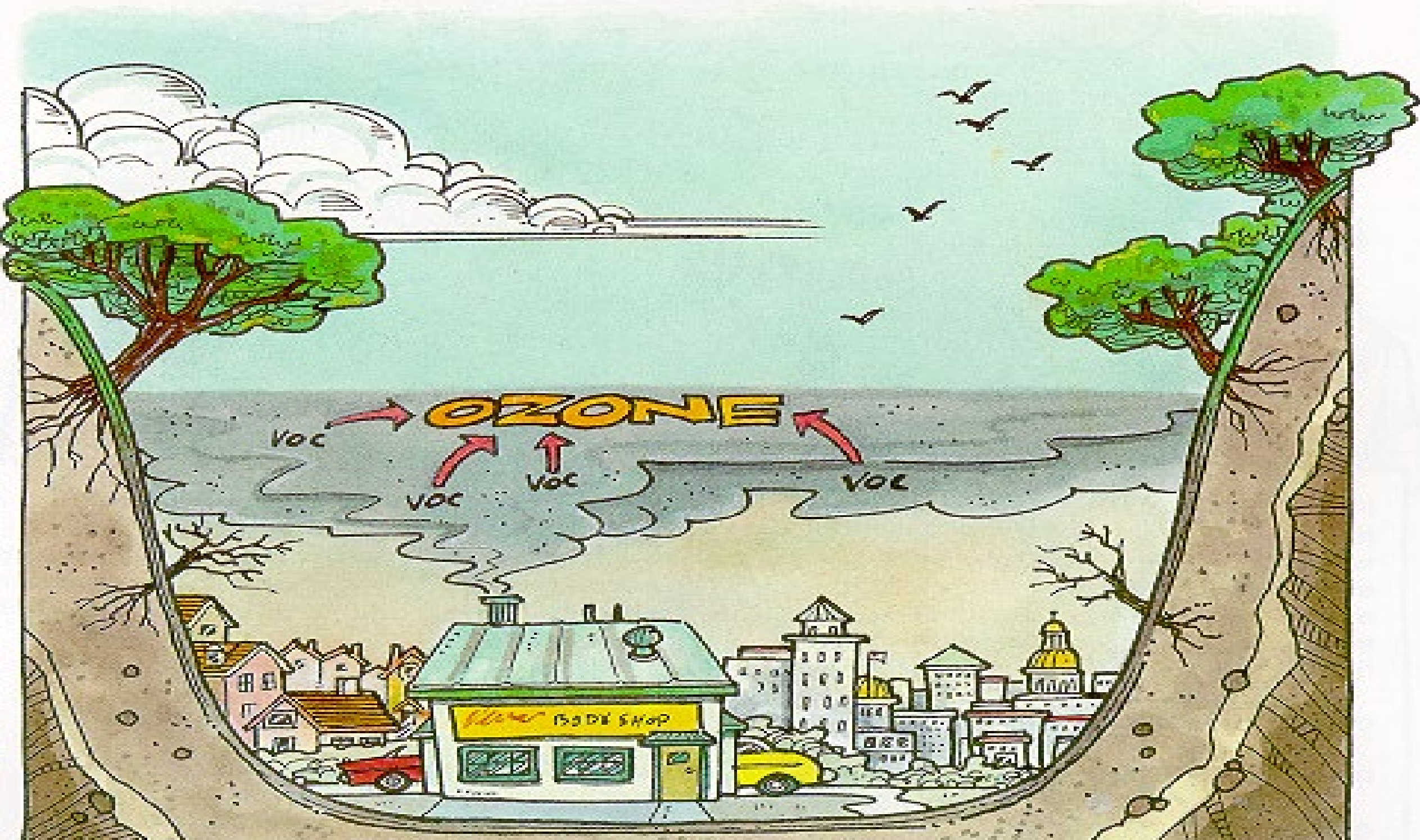


Legal Definition of VOCs

- Federal and State laws & regulations
 - * 40CFR51 § 51.100
 - * Latest Definitions of VOCs and ROGs as of...
- Total Organic Gases (TOGs)
- Reactive Organic Gases (ROGs)
- Fraction of Organic Gases (FROGS)
- Local Agency rules and permit conditions

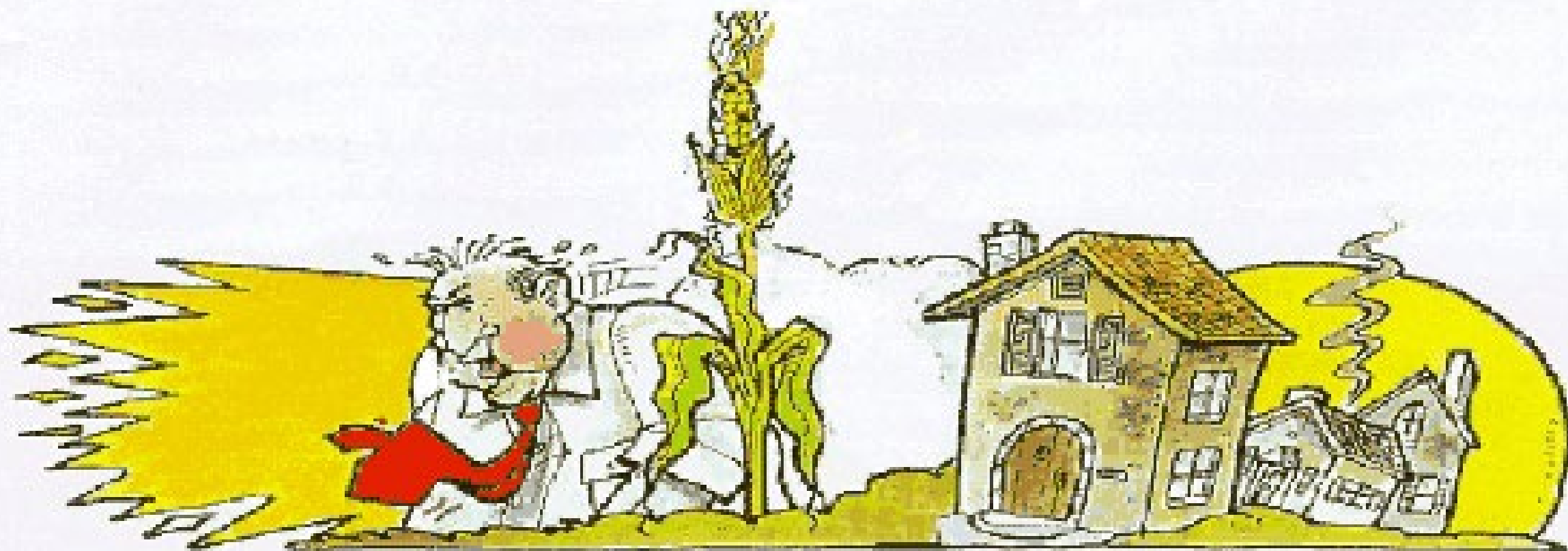


Why are VOCs Regulated?

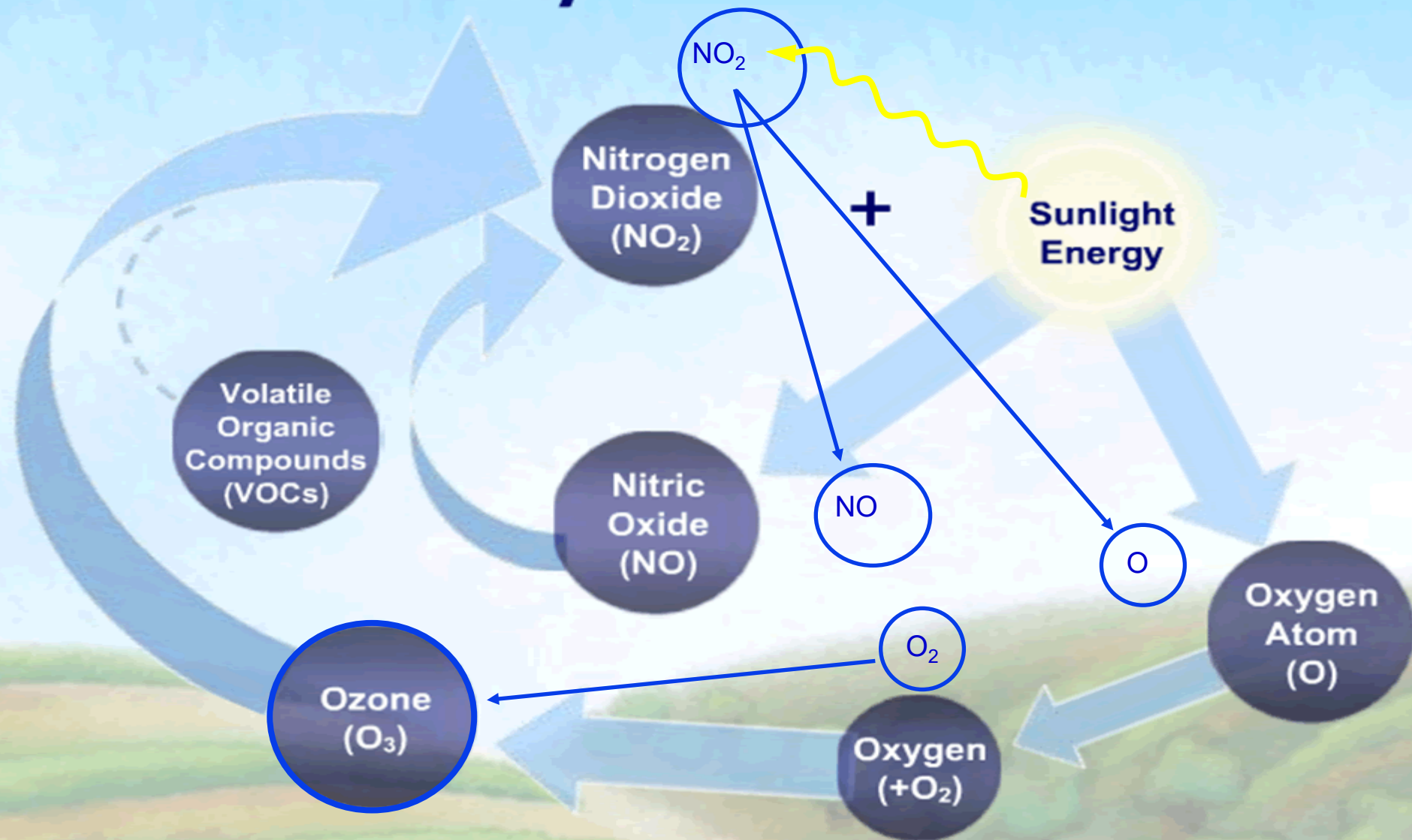


Why is Ozone Regulated?

OZONE CAUSES



Ozone Photochemistry



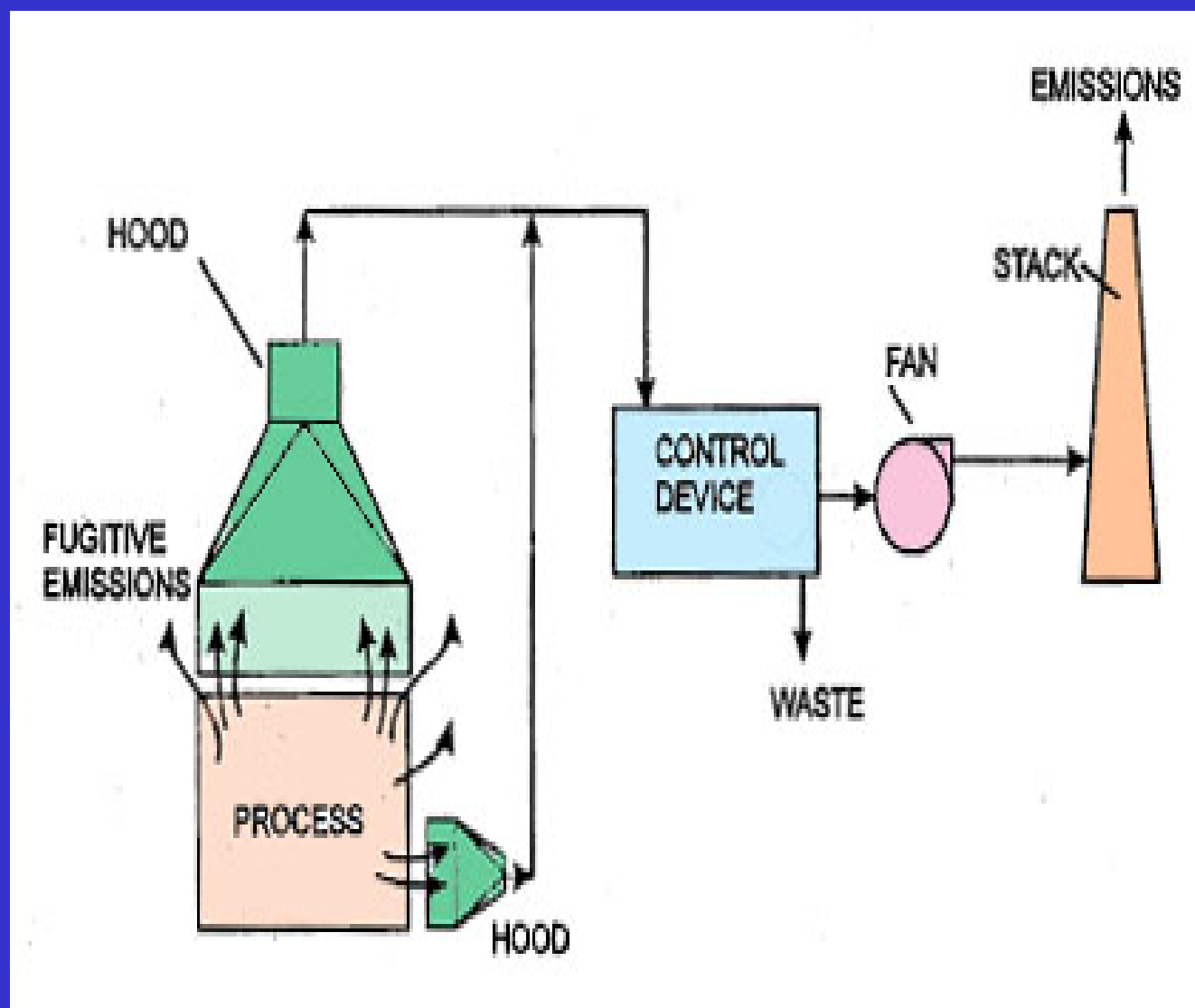


Ozone Formation

- $\text{VOCs} + \text{NO}_x + \text{sunlight} \rightarrow \text{O}_3$
- Ozone is formed when NO_x and Volatile Organic Compounds react in sunlight



VOC Control Process



- Capture
- Control
- Recovery,
- Disposal or
- Destruction



VOC Calculations : Capture & Control & Retention

- General Categories of VOC Emissions
 - * Fugitive (Not reasonably captured)
 - * Captured > Ducted to control device
 - * Consumed > Oxidized
 - * Retained > Retention factors vary



VOC Capture Efficiency *

$$\text{VOC Capture Efficiency} = \frac{\text{VOCs captured}}{\text{VOCs used}} \times 100$$

VOCs used (and therefore emitted) 100 lbs

VOCs captured (entering control device) 80 lbs

VOC capture efficiency (by calculation) ??????

** Capture Efficiency is the percentage of emissions captured and vented to a control device. -- EPA*



VOC Capture Efficiency *

$$\text{VOC Capture Efficiency} = \frac{\text{VOCs captured}}{\text{VOCs used}} \times 100$$

VOCs used (and therefore emitted) 100 lbs

VOCs captured (entering control device) 80 lbs

VOC capture efficiency (by calculation) 80%

** Capture Efficiency is the percentage of emissions captured and vented to a control device. -- EPA*

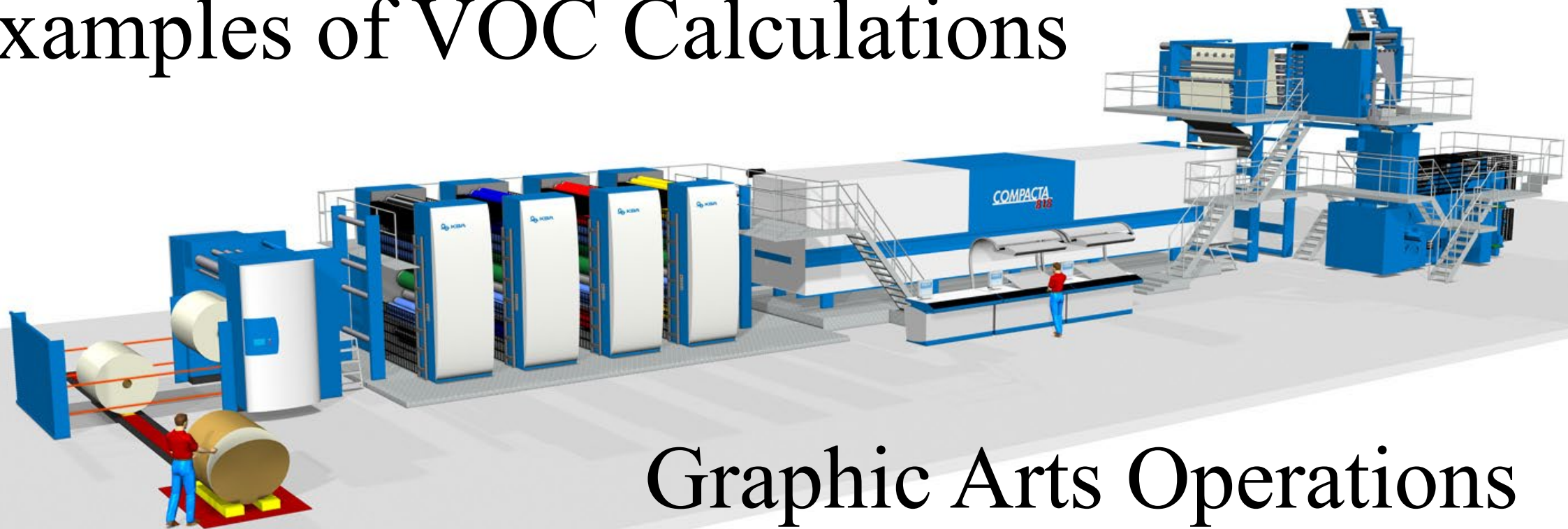


VOC Control Efficiency

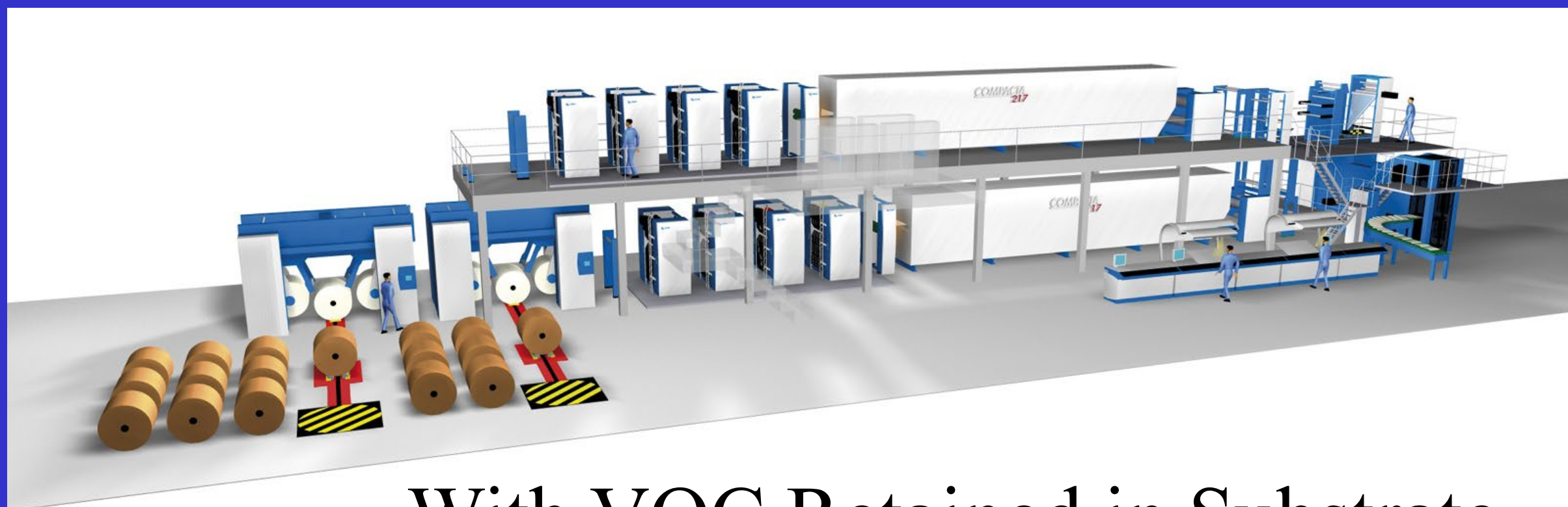
$$\% \text{ CE} = \left[1 - \frac{\text{outlet emission rate}}{\text{inlet emission rate}} \right] \times 100$$

$$\% \text{ CE} = \left[1 - \frac{2 \text{ lbs/hr}}{100 \text{ lbs/hr}} \right] \times 100 = 98$$

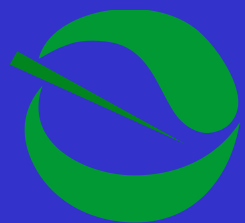
Examples of VOC Calculations



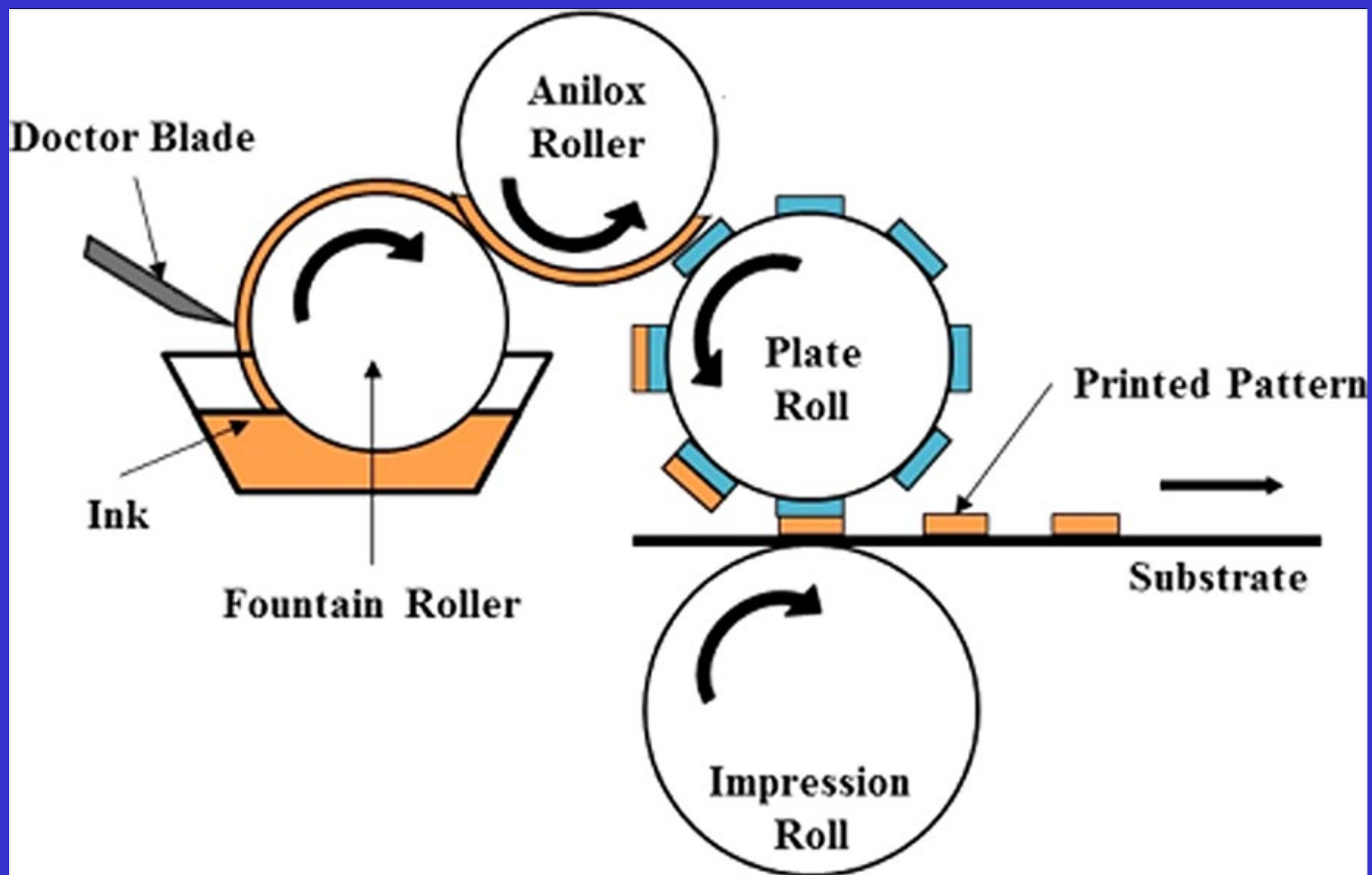
Graphic Arts Operations



With VOC Retained in Substrate



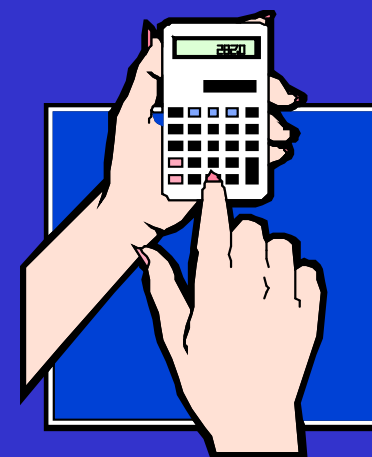
Graphic Arts Operation





VOC Calculations

- A facility uses 100 lbs/hr of ink that has a VOC content of 35% by weight.
- 20% of the VOC is retained in the substrate
- The incinerator has a 95% control efficiency



How many lbs/hr of VOC is emitted?

$$\text{VOC Emissions} = (100 \text{ lbs/hr}) (0.35) (1-0.20) (1-0.95) = 1.4 \text{ lbs/hr}$$

Let's Discuss Control of VOC

- **Containment**
- **Transfer Efficiency**
- **Absorption**
- **Adsorption**
- **Condensation**
- **Oxidation**



**Controlled Spraying
aka
Pollution Prevention**

**Reduces VOC emissions
Increases transfer efficiency
Low fluid tip pressure
Employee gun handling training**

DEVILBISS.



**High Volume Low Pressure
(HVLP) Spray Guns**



**(HVLP) Spray Gun :
Polyester Resin Operations**



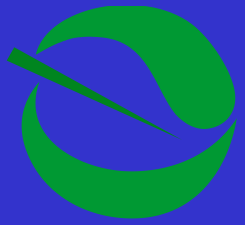
**Fluid Impingement Technology :
Polyester Resin Operations**



Gel Coat Application in a Spray Booth

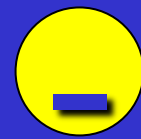
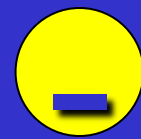
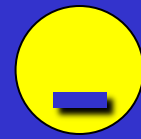
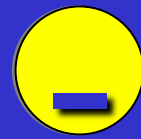


**Let's Discuss
Adsorption
Systems**



Adsorption Mechanism

Gas

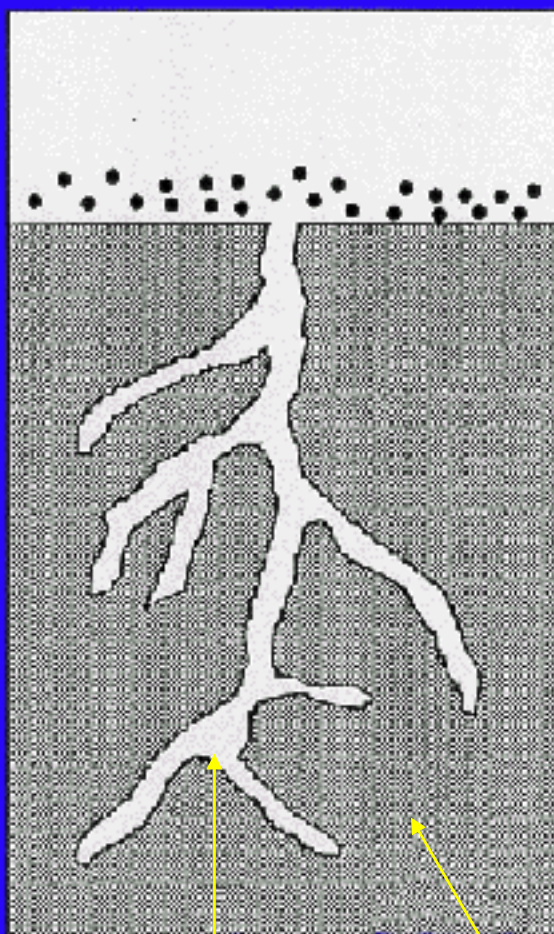


Solid
surface

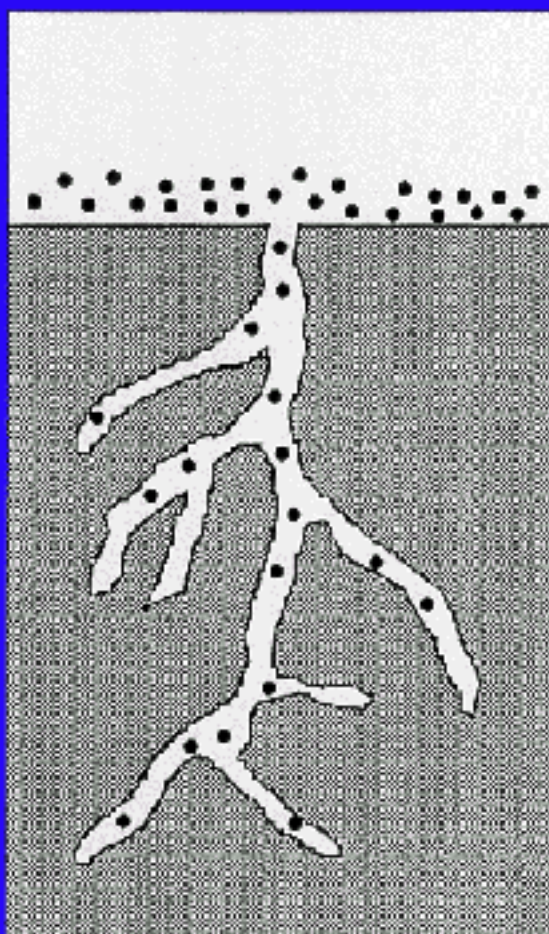


Adsorption Mechanism

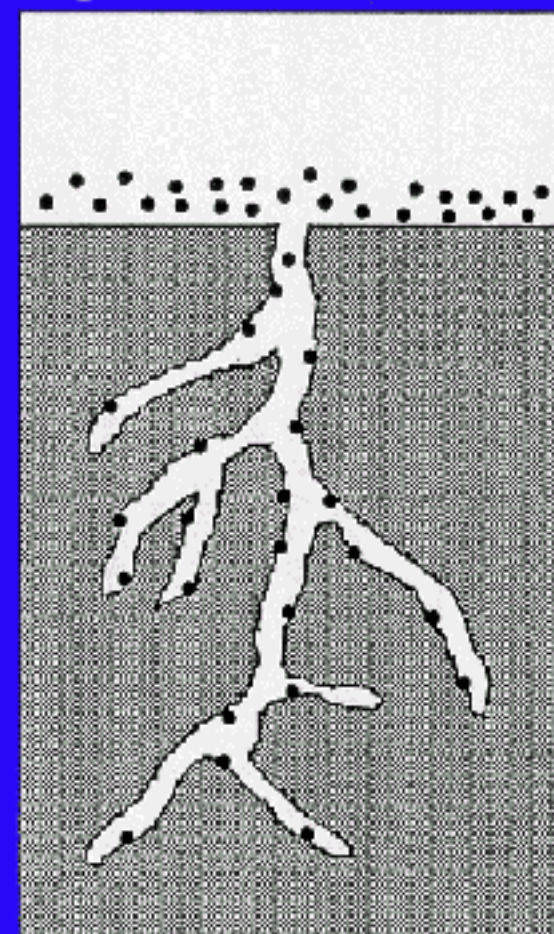
Step 1. VOC diffuses to adsorbent surface



Step 2. VOC migrates into pores



Step 3. VOC adsorbed and builds up on adsorbent



Pore

Carbon

• VOC molecule



Adsorption Mechanism

- Chemically unchanged
- Desorbed and recovered
- Polar and non-polar adsorbates
- Mixed adsorbates separated by distillation

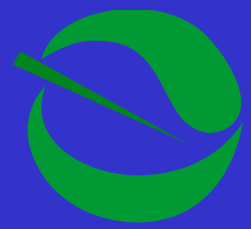


Adsorption

- Adsorption materials (adsorbents)
 - * Activated carbon
 - * Hydrous oxides
 - Silica gel
 - Aluminum oxide
 - Magnesium silicate
 - * Zeolites (molecular sieves)
 - * Naturals
 - Clays
 - Bauxite
 - Fuller's Earth
 - * Metals

**Carbon
Adsorbers
at a Soil
Remediation
Site**





Factors Affecting Adsorption

- Temperature
- Pressure
- Gas velocity
- Particulate matter

Adsorber Design Considerations

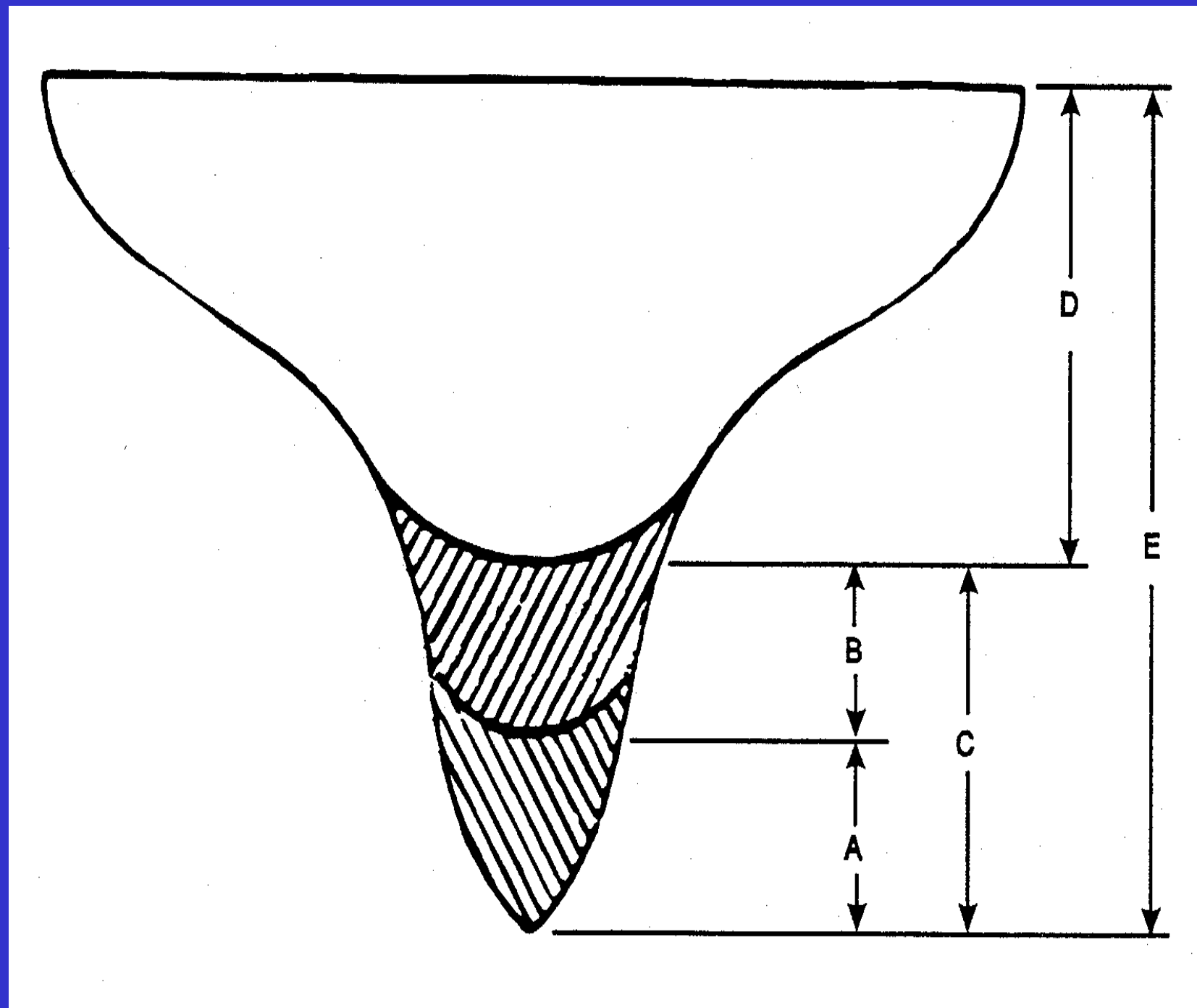


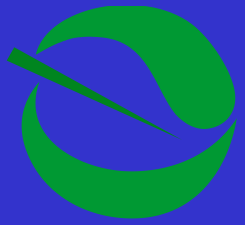
- ◆ Porosity of Adsorbent
- ◆ Bed Cross-Sectional Area
- ◆ Bed Length
- ◆ Multiple Organic Compounds
- ◆ Steaming Requirements
- ◆ Fouling
- ◆ Timers/Monitors
- ◆ Channeling



Pore Space Representation

- A = Residual VOCs or heel
- B = Working capacity
- C = Equilibrium Capacity
- D = Empty pore space
- E = Total pore space (total capacity)





Carbon Adsorption Keywords

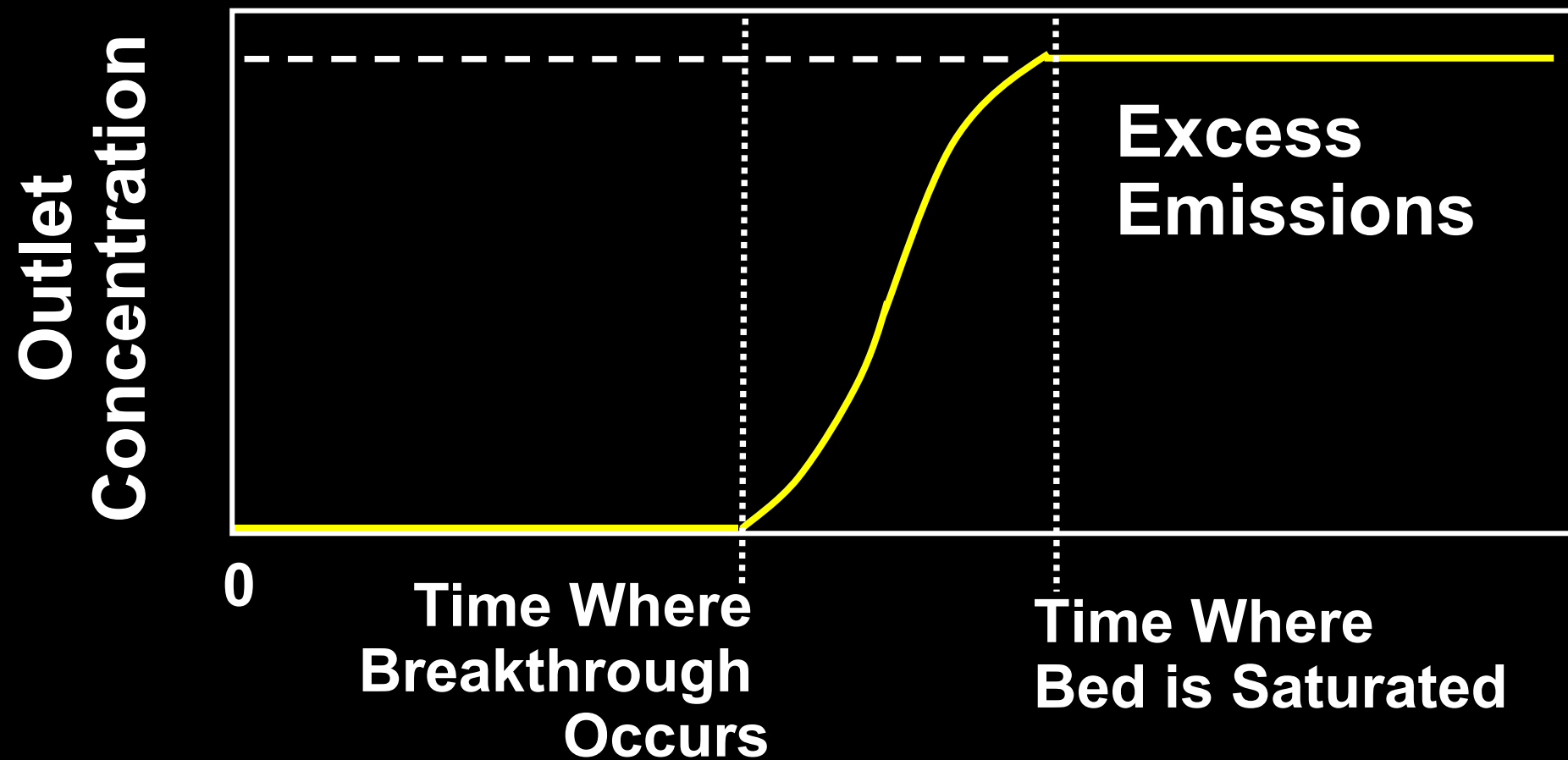
- Fresh zone
 - * Area where adsorption will occur
- Mass transfer zone
 - * Where adsorption occurs
- Saturated zone
 - * Area where adsorption has already occurred



Keywords (continued)

- Heel
 - * Amount of VOCs left in the carbon after regeneration
- Breakthrough
 - * VOCs that do not get captured

Adsorber Breakthrough





Types of Adsorption Systems

*Non-regenerative systems

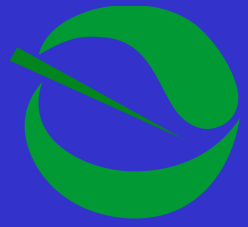
*Regenerative systems

- on site
- off site



Characteristics of Activated Carbon

- Sources
 - * Wood, coal, peat, nut shells
- Porosity
 - * 600-1600 m²/g (2-3 football fields per 1/28 ounce)
- Preparation
 - * Anaerobic heat then steam or CO₂,
- Degree of adsorption depends on adsorbate
 - * MW, BP, polarity, surfactive index, solubility



Examples of Activated Carbon





Finely Granulated Carbon



Types of Carbon

Adsorption Systems

- Open
- Closed
- Rotary
- Fluidized bed
- Bulk plant adsorber and absorber



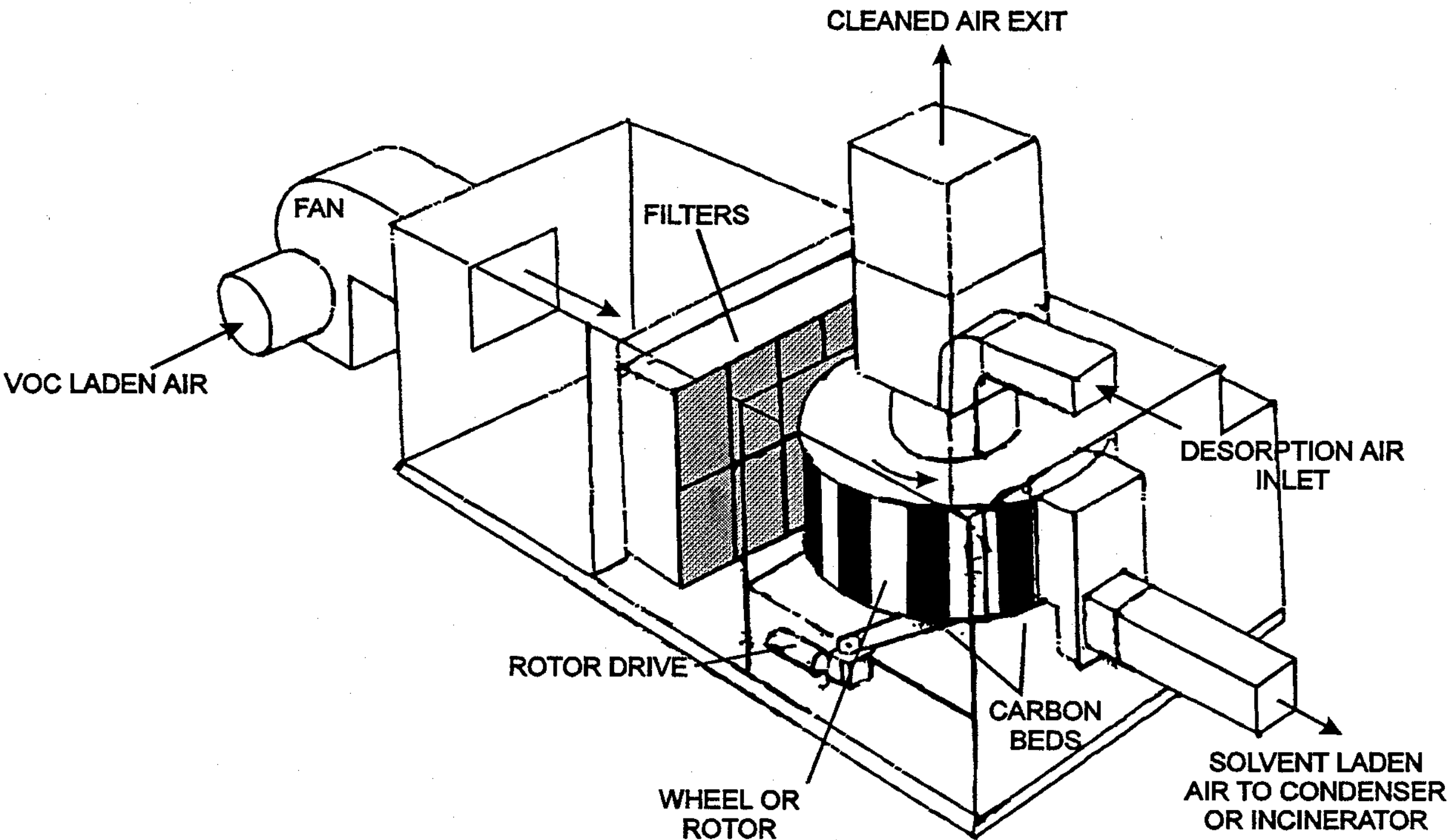
Bulk plant adsorber & absorber



10.10.1999

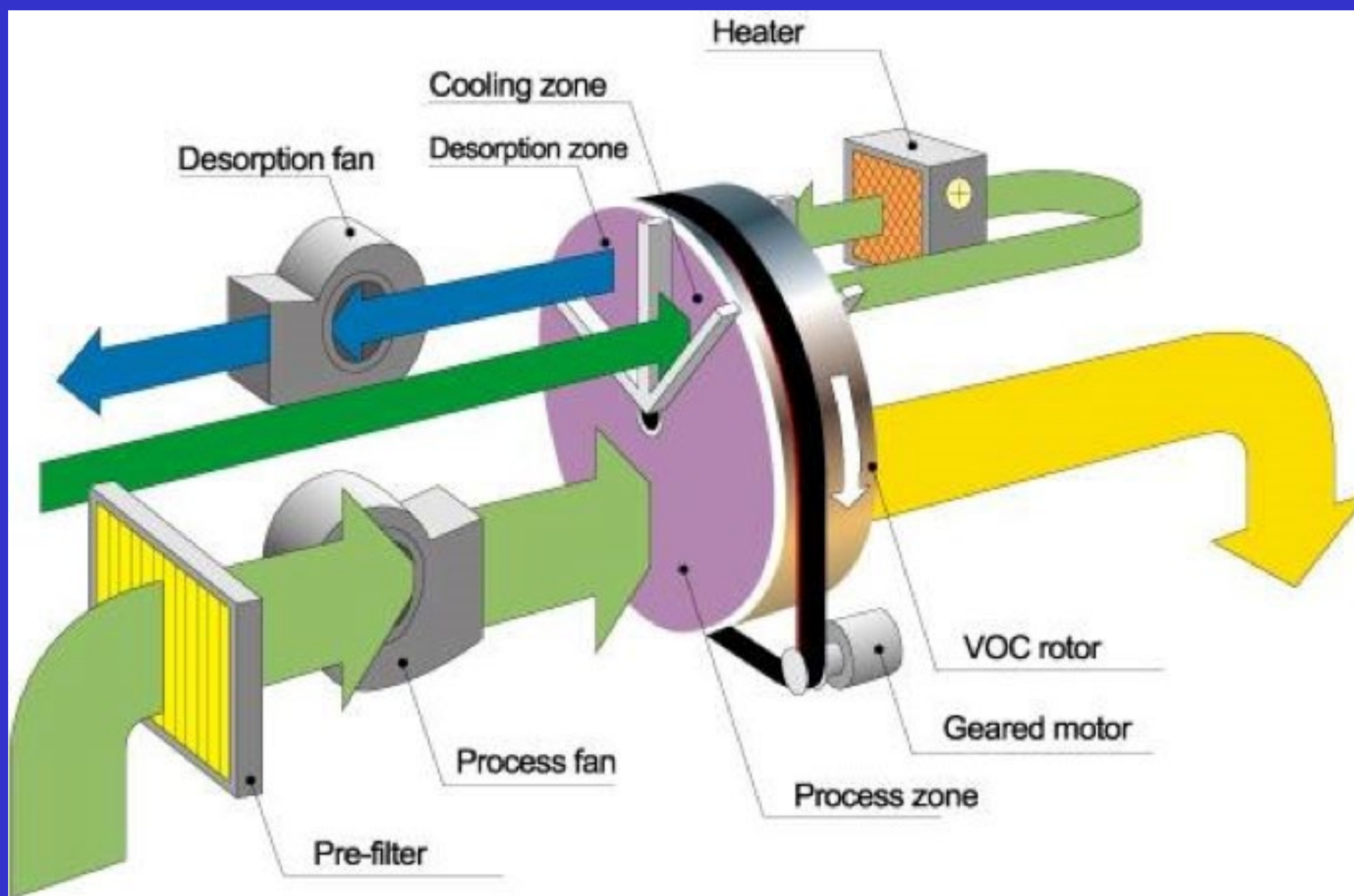


Rotary Concentrator Adsorption System





Rotary Concentrator Adsorption System





Adsorber Inspections

- Hood static pressures
- Inlet VOC concentrations
- Inlet temperatures
- Inlet VOC concentration not $> 25\%$ LEL
- Outlet VOC concentrations
- Fan motor current
- Solvent recovery rates



**Let's Discuss
Absorbers**



Absorbers



- Pollutants dissolved in liquid
- Absorbate dissolves in absorbent



Factors Favoring Absorption

- Pollutant solubility in liquid
- Adequate diffusion at liquid / gas interface
- Maximized contact between gas and liquid

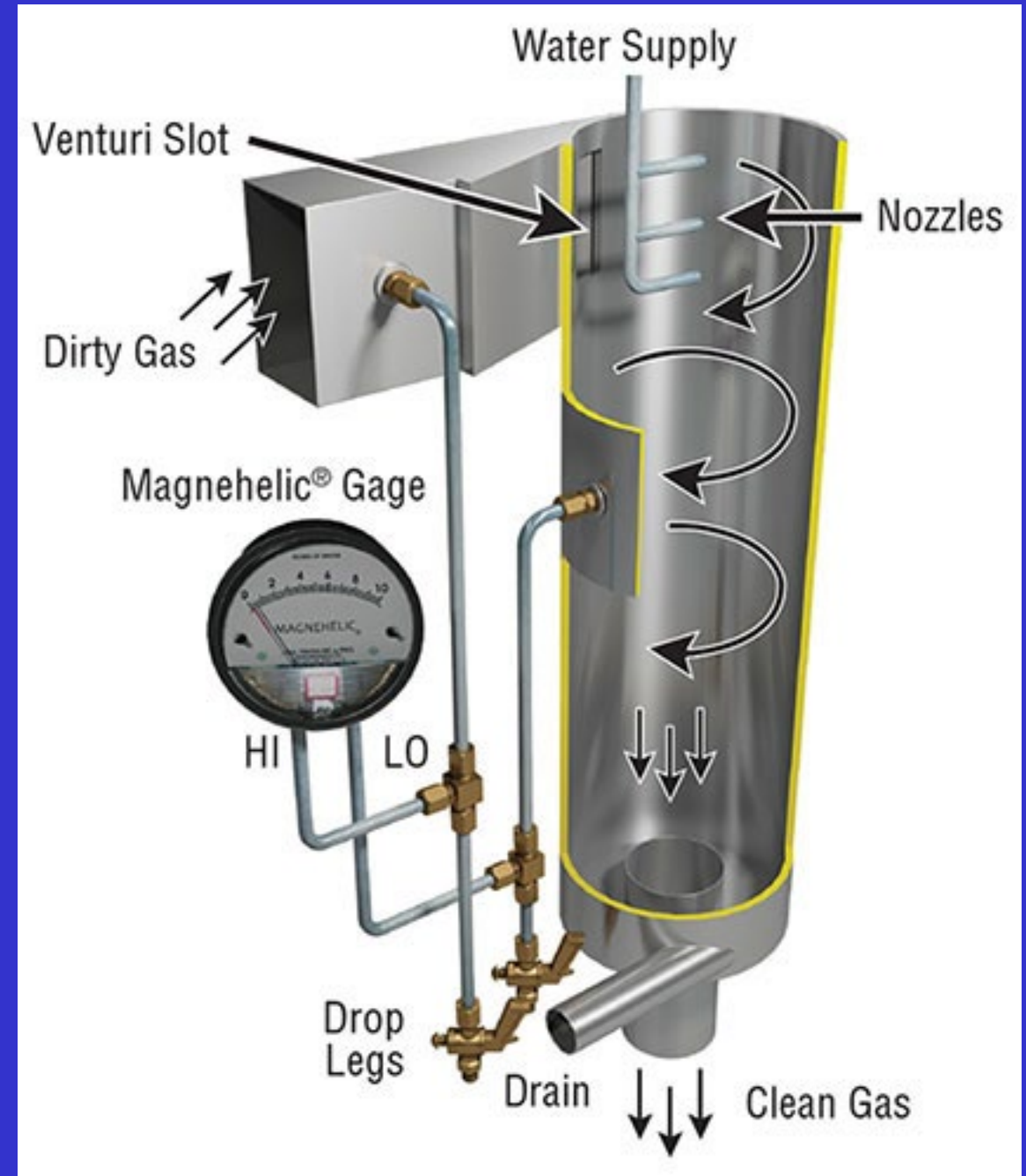
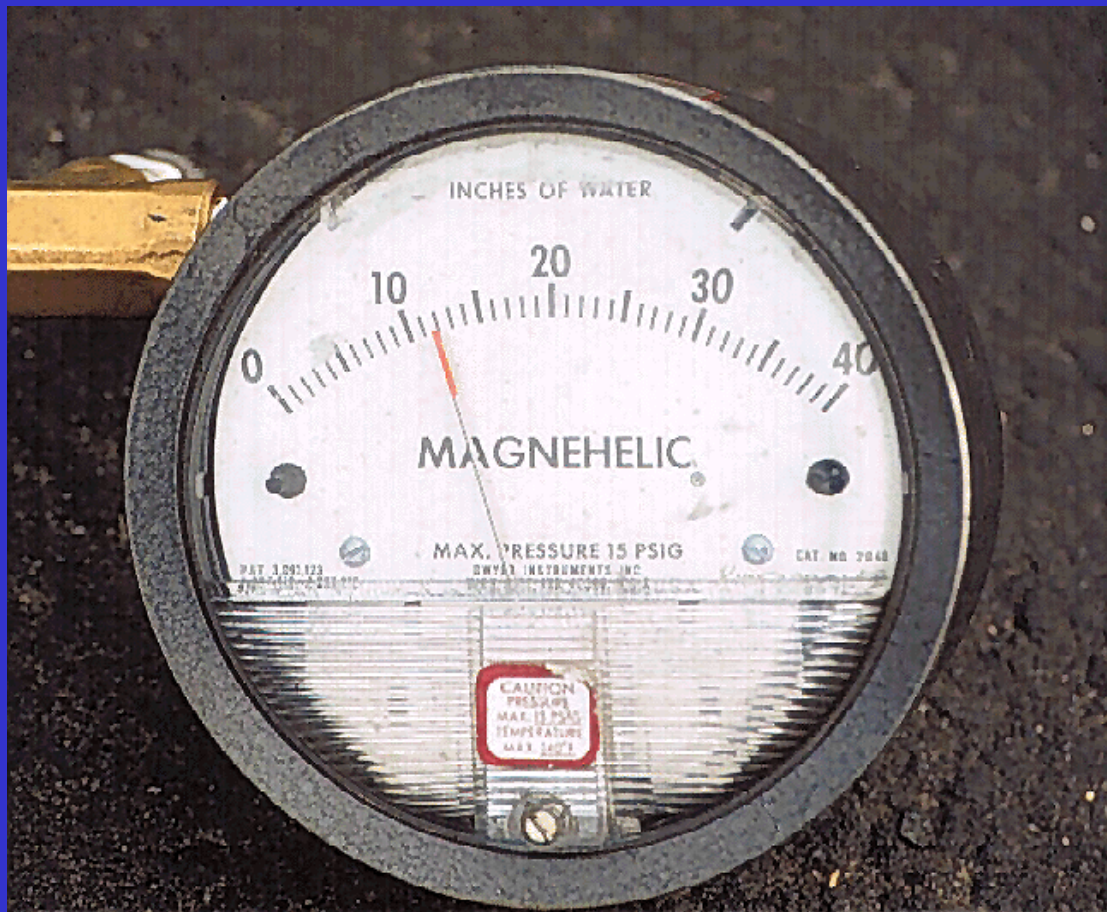


Absorber Design

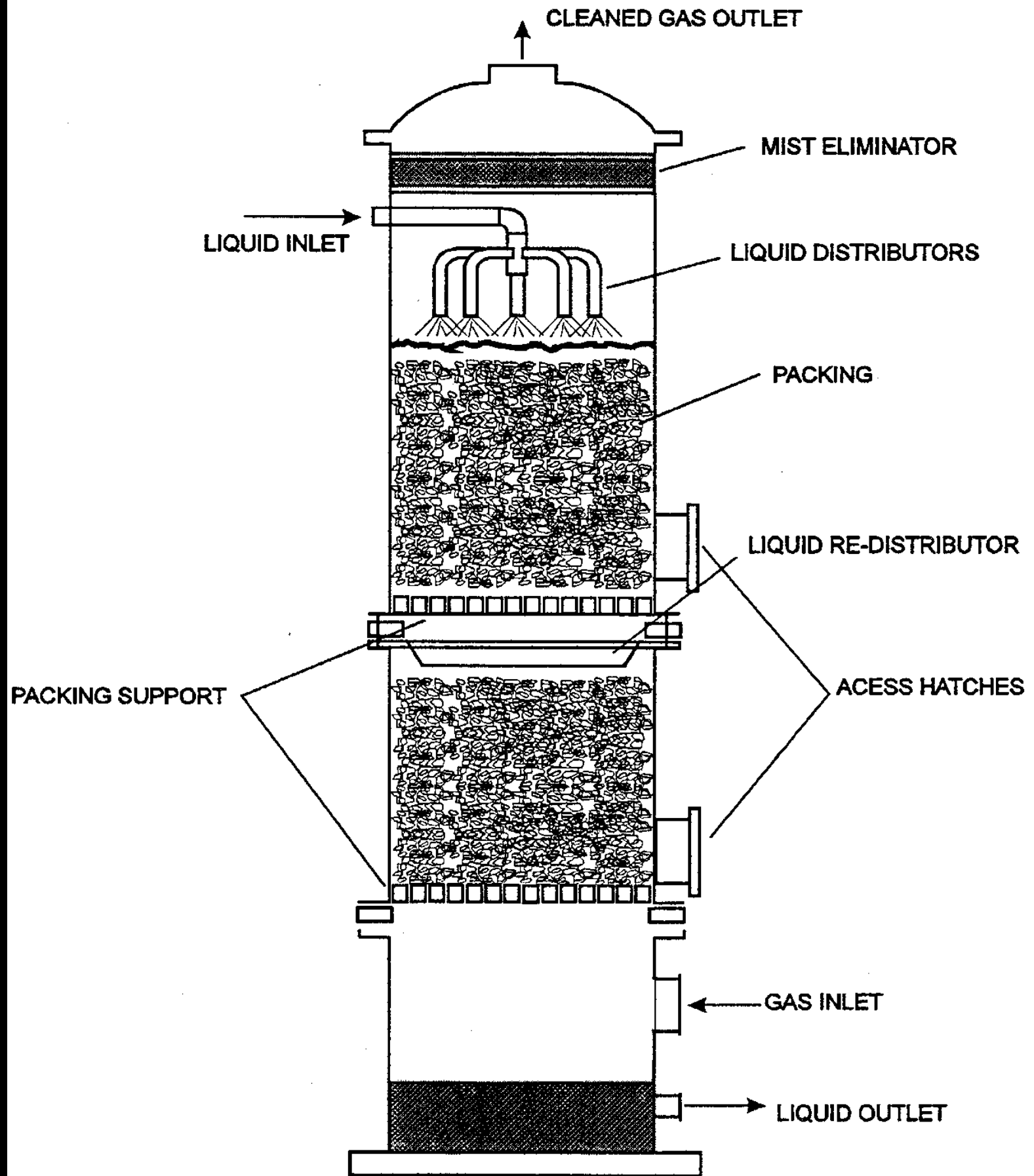
- Produce large surface area
- Minimize air flow resistance to reduce pressure drop
- Inlet pressure - outlet pressure = pressure drop



Pressure Drop : Magnehelic



Packed Bed Wet Scrubber





Packed-Bed Wet Scrubber



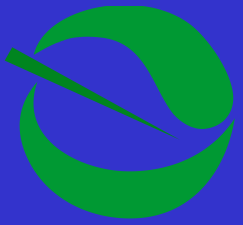
Absorber Design Factors

- Select liquid solvent
- Column material
- Column size
- Column height
- Number of plates
- Pressure drop

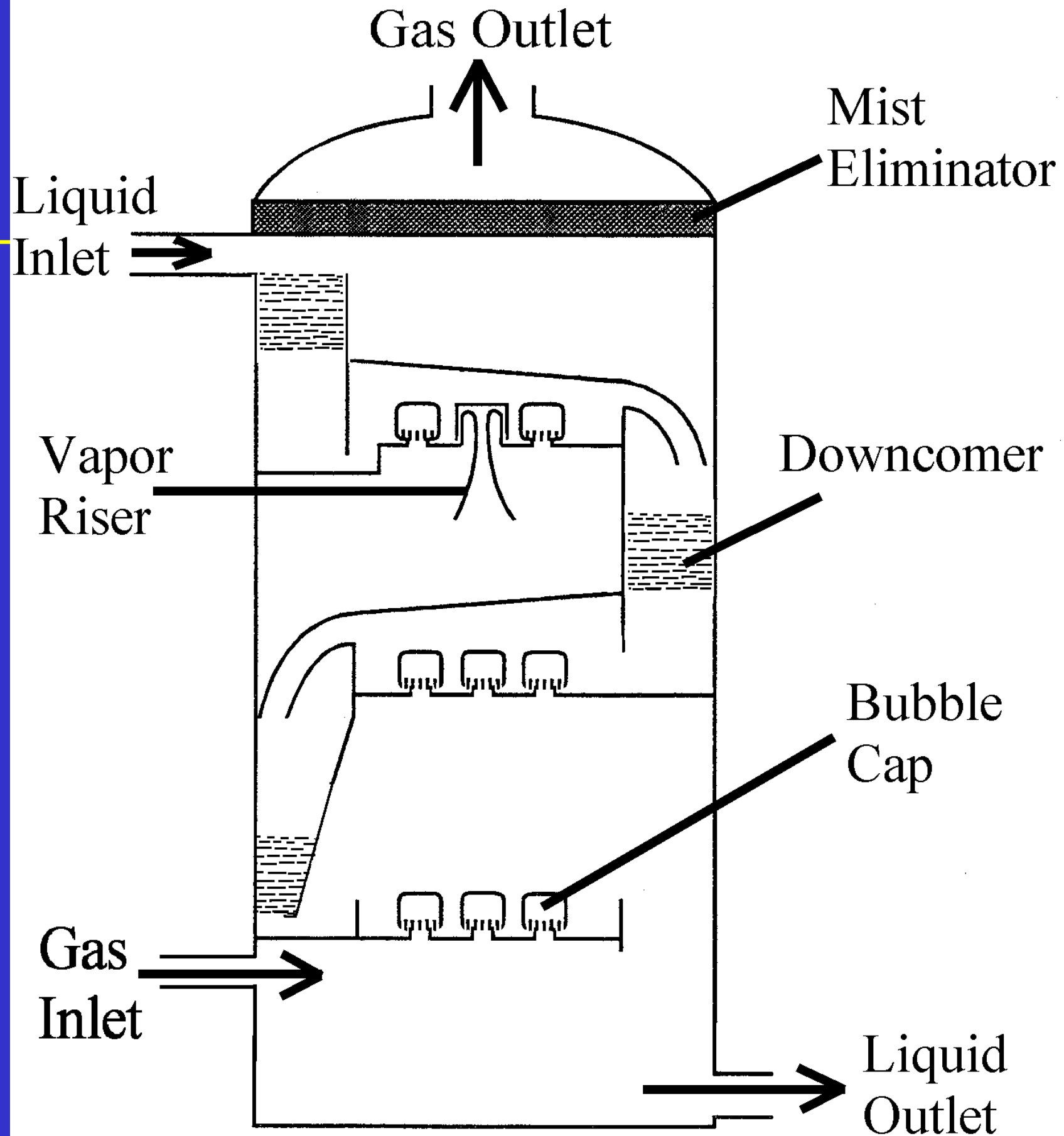


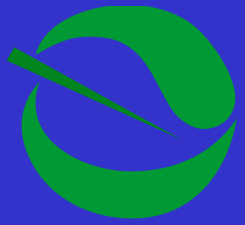
Absorbers: Packed Columns

- Flow patterns
- Liquid reuse and treatment
- Packing material
- Packing quality



Absorbers: Plate Columns





Absorbers: Plate Columns

- Maximize contact between liquid & gas
- Diameter of column
- Plates
 - * Number
 - * Type
 - * Layout



Packed vs Plate Columns

- Packed columns
 - +More common
 - Plugged by particles
 - +Better for corrosive pollutants
 - +Lighter than plate



Packed versus Plate Columns

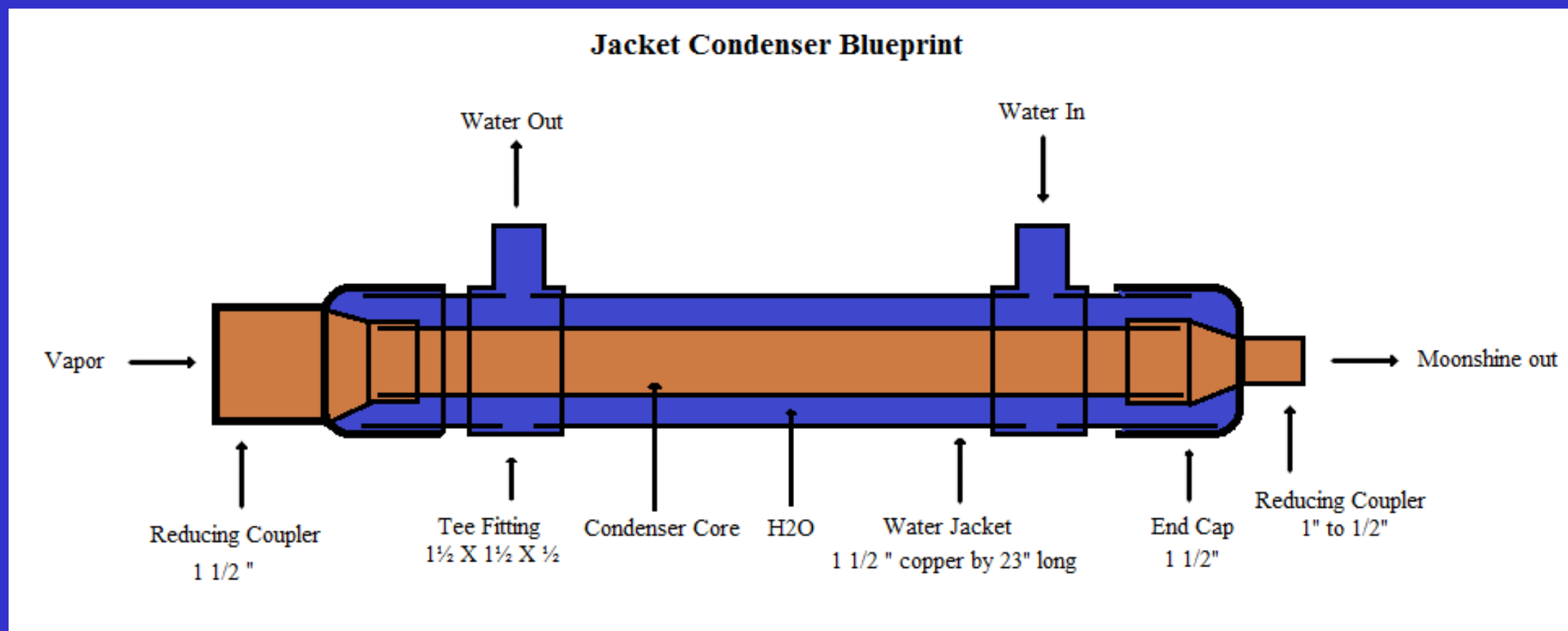
- Plate columns are better for:
 - + Large temperature changes
 - + Lower liquid flow rates
 - + Higher gas flow rates
 - + Foaming liquids
 - + Chemical reactions
 - + Large systems

A winter scene in a forest. Two large, reddish-brown tree trunks are prominent in the foreground. The background is filled with numerous evergreen trees heavily laden with snow. A wooden fence runs across the middle ground. The overall atmosphere is serene and cold.

Let's Discuss Condensers



Condensers : Surface & Contact

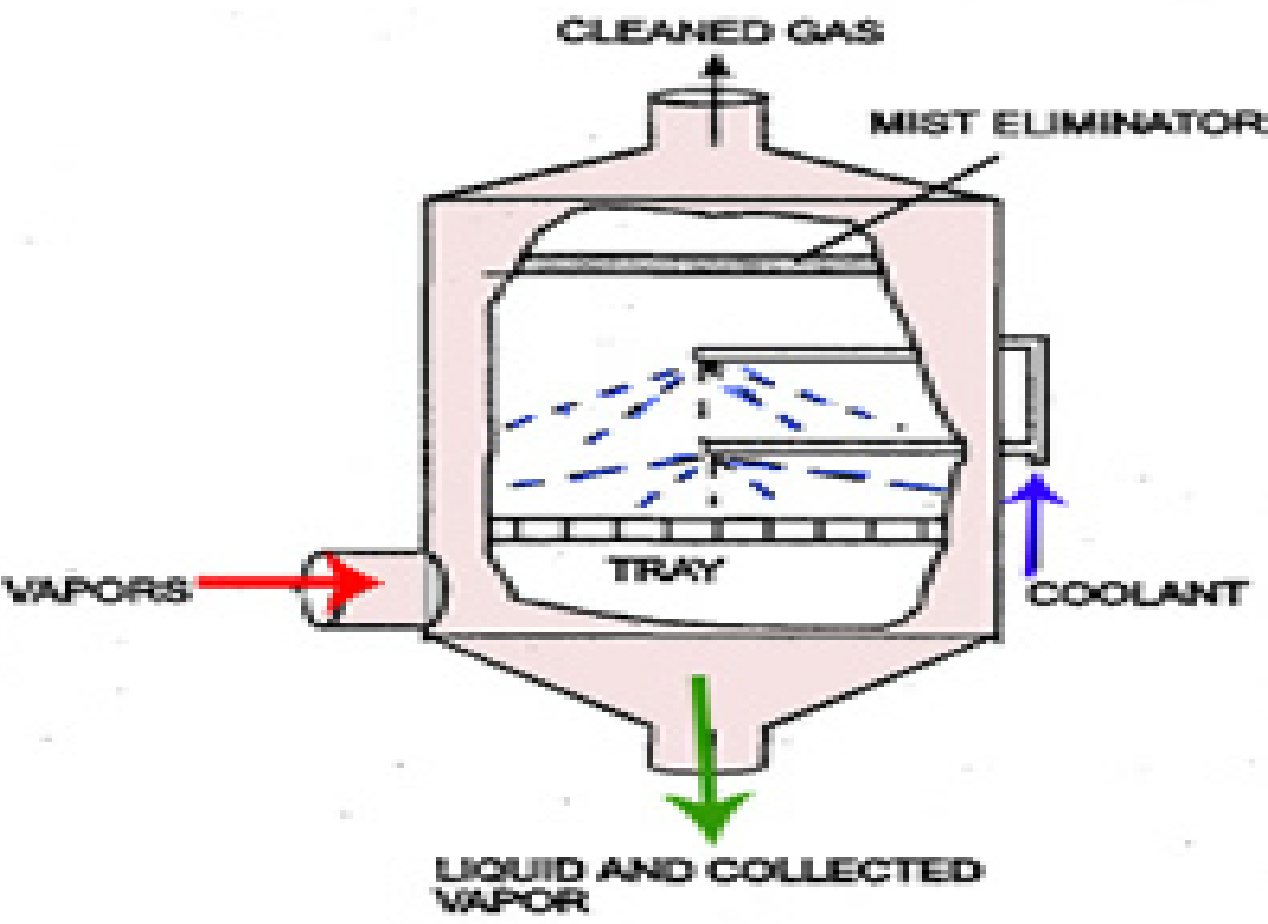


- Condensation = Process of changing a gas to a liquid.
- Condensation allows recovery of solvents and air pollution control

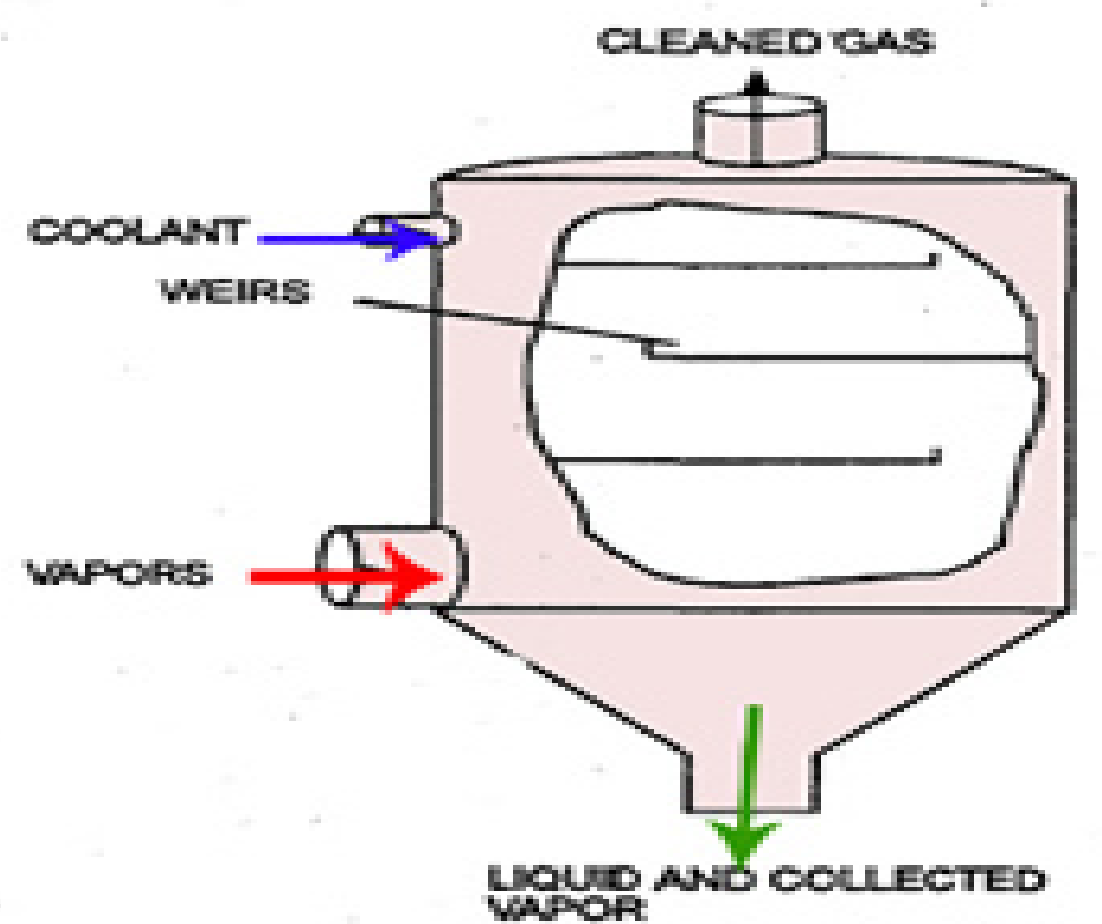


Contact Condensers

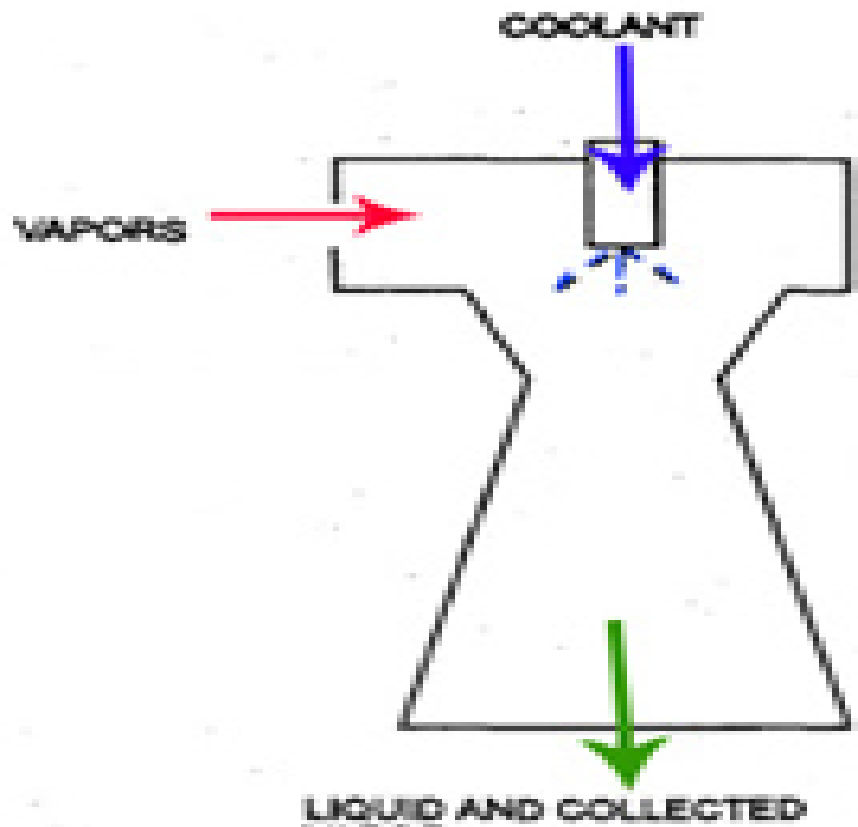
- Contact condensers +/-
 - + Cheaper
 - + More flexible
 - + Less repair time
 - Wet waste disposal problem



SPRAY



BAROMETRIC



JET

Contact Condensers



Surface Condensers

- * Shell and tube

 - (most common)

- * Fin Fan

- * Tubular

- * Double pipe

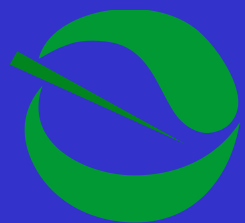
- * Spiral plate

- * Flat plate

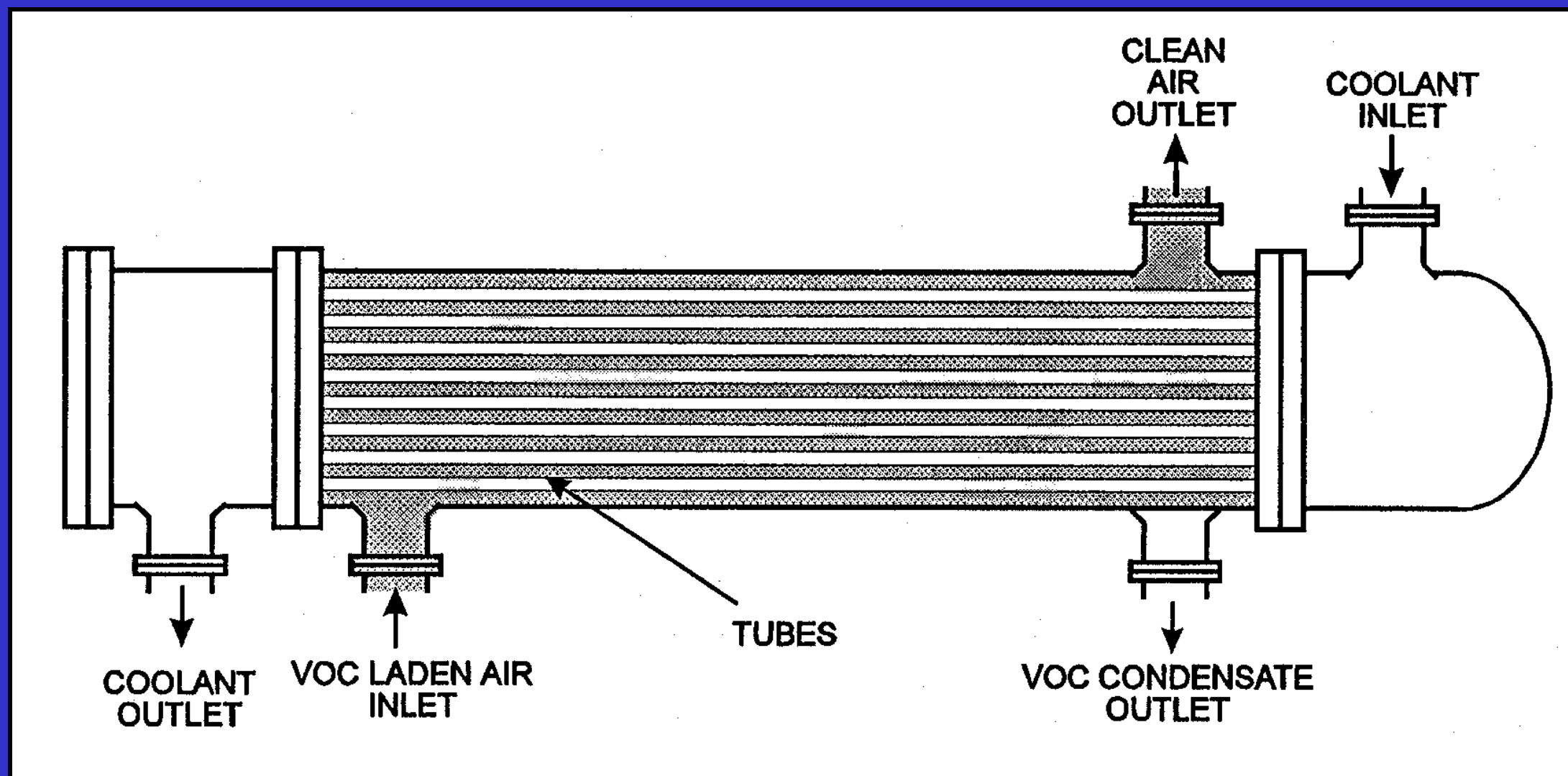


Condensers

- Surface condensers +/-
 - + Better recovery
 - + Commonly used for air pollutants
 - + Reduced waste disposal problems
 - More costly



Shell and Tube

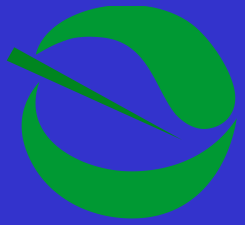




**Dry Air-Cooled Heat Exchanger :
Steam Condenser**



Dry Air-Cooled Condenser Fans



Condenser Concerns

- Freezing
- Fouling
- Cleaning
- Pressure drop



Condenser Inspection

- Look for
 - Excessive corrosion and rusting
 - Leaking coolant or VOC
 - Excessive odors
 - Continuous emissions monitor



Condenser Inspection

- Record
 - VOC outlet concentration
 - Waste stream flow rate
 - Condenser pressure drop
 - Coolant pressure
 - Coolant flow rate



Let's Discuss Oxidizers



Oxidation

- Destruction of VOCs by Combustion

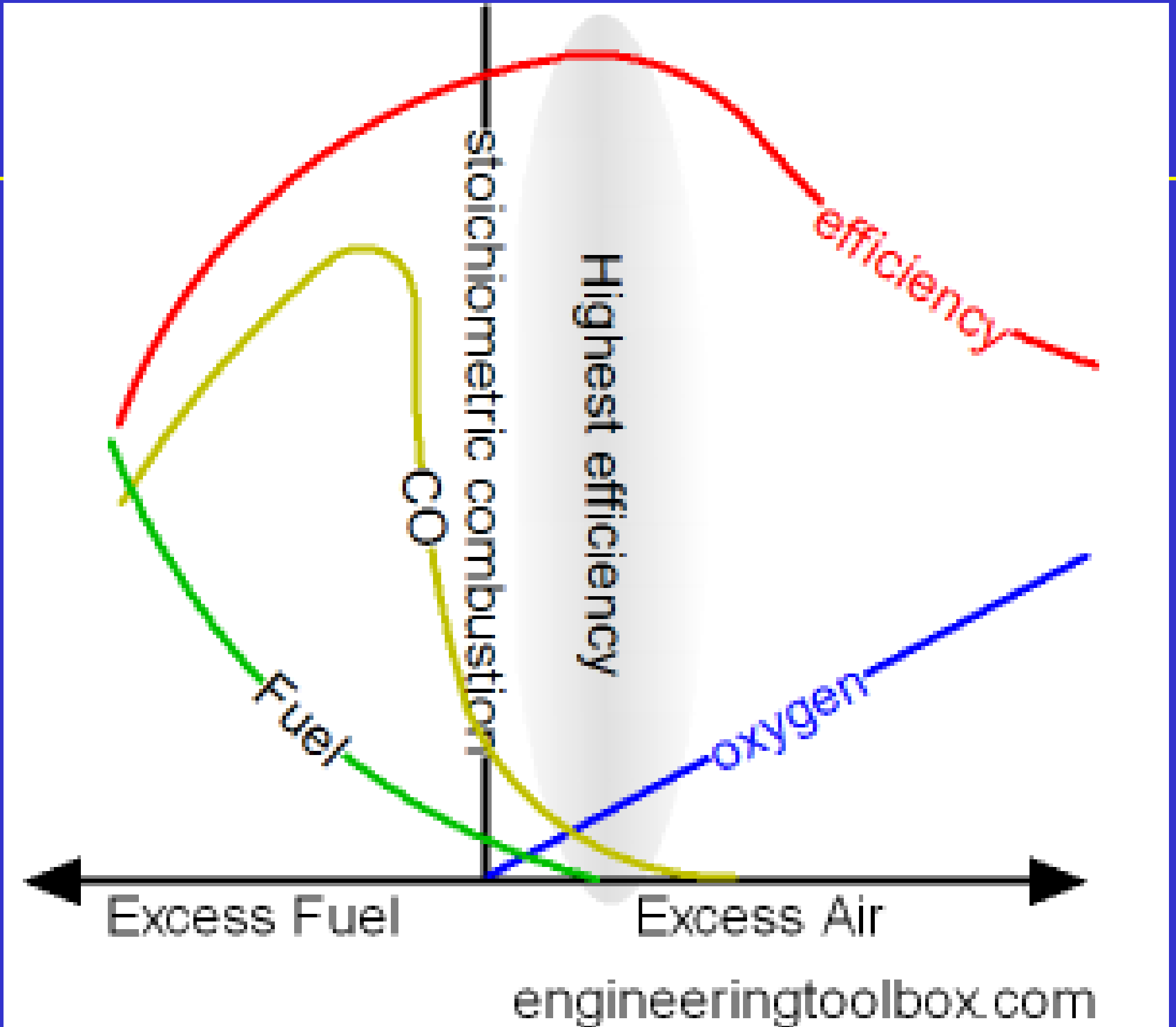
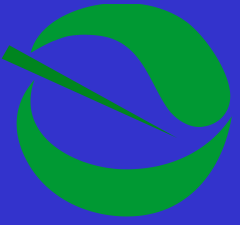
Reactions with oxygen



Toluene + Oxygen = Carbon Dioxide + Water

- 
- ◆ **Time**
 - ◆ **Temperature**
 - ◆ **Turbulence (mixing)**
 - ◆ **Oxygen (air)**
 - ◆ **Nitrogen (air)**

Combustion Considerations

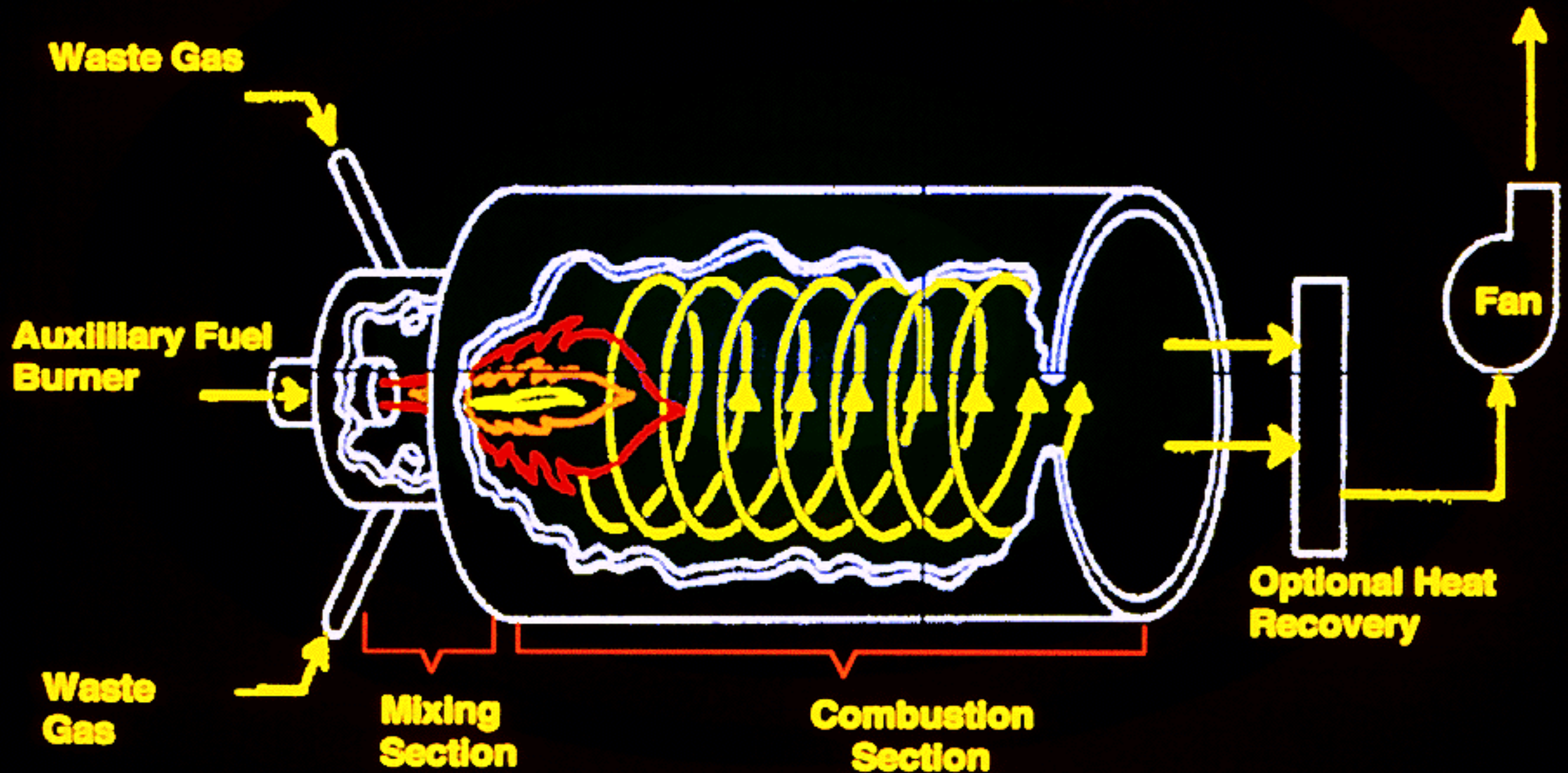




Combustion Devices

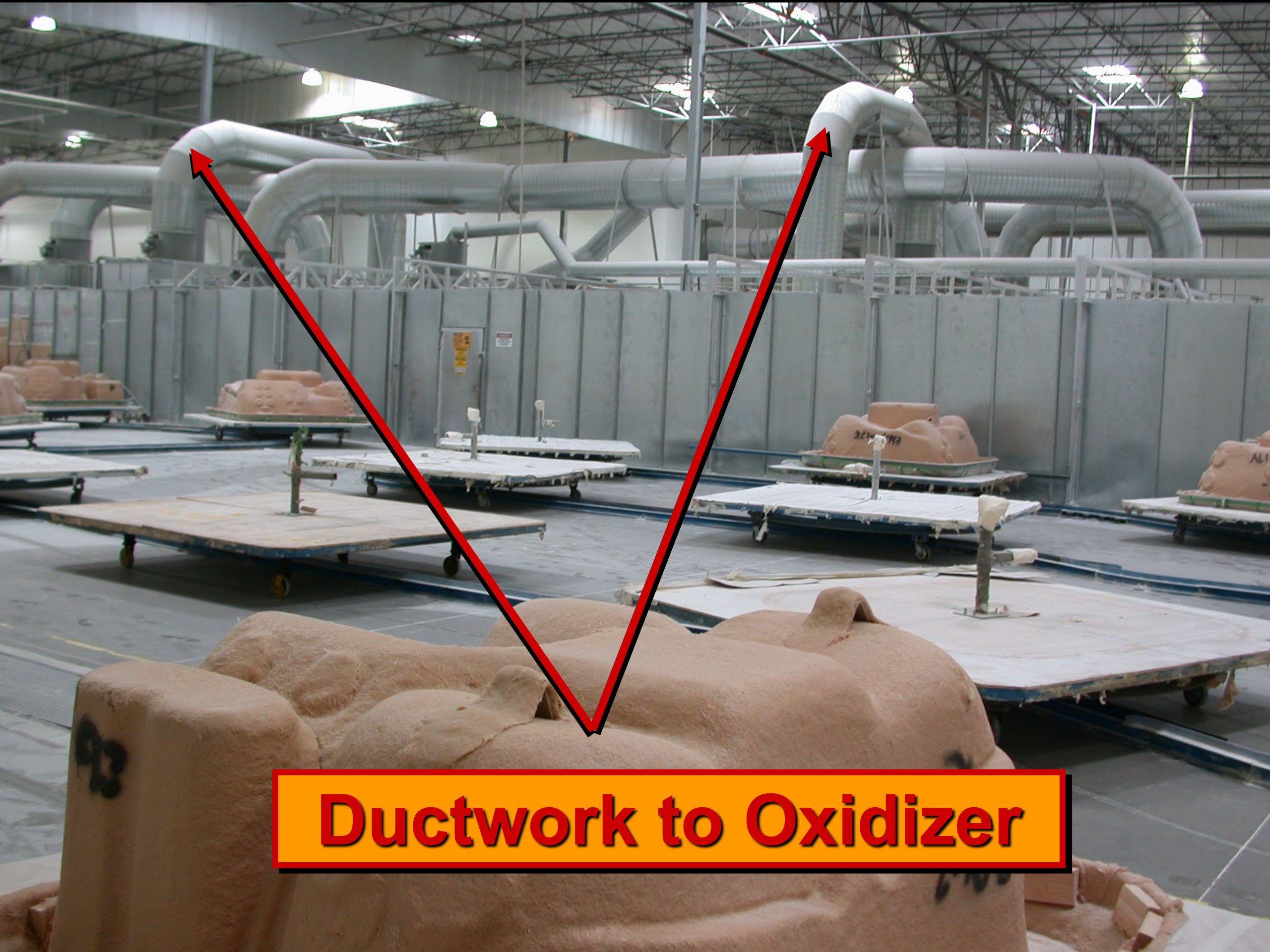
- Thermal incinerator (uses a flame)
- Catalytic incinerators (uses a catalyst)
- Boilers (burn VOCs to make steam)
- Process heaters (burn VOCs to add heat in chemical plants and refineries)
- Flares (simple flame)

Thermal Oxidizer/Afterburner

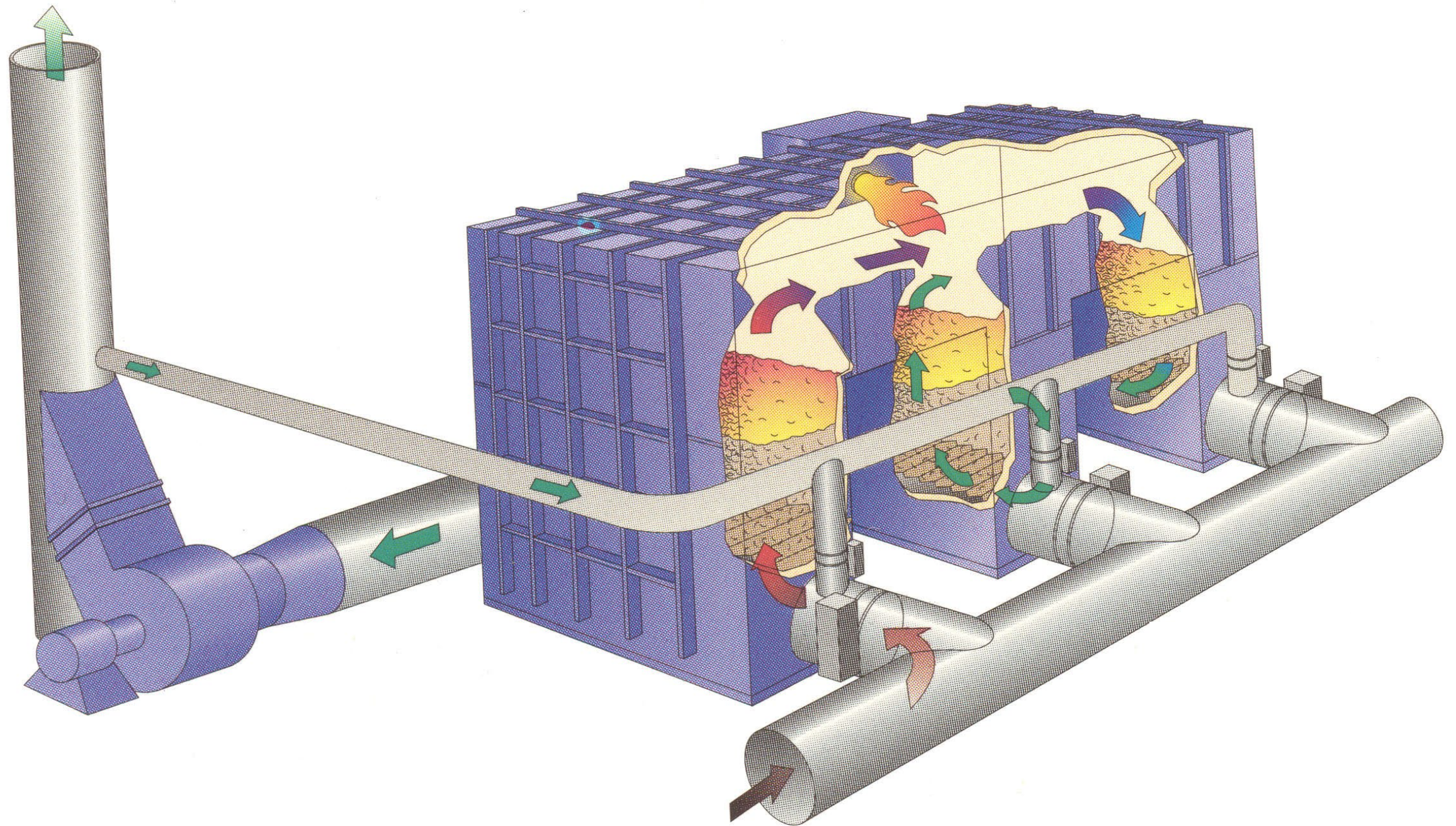


Thermal Incinerator





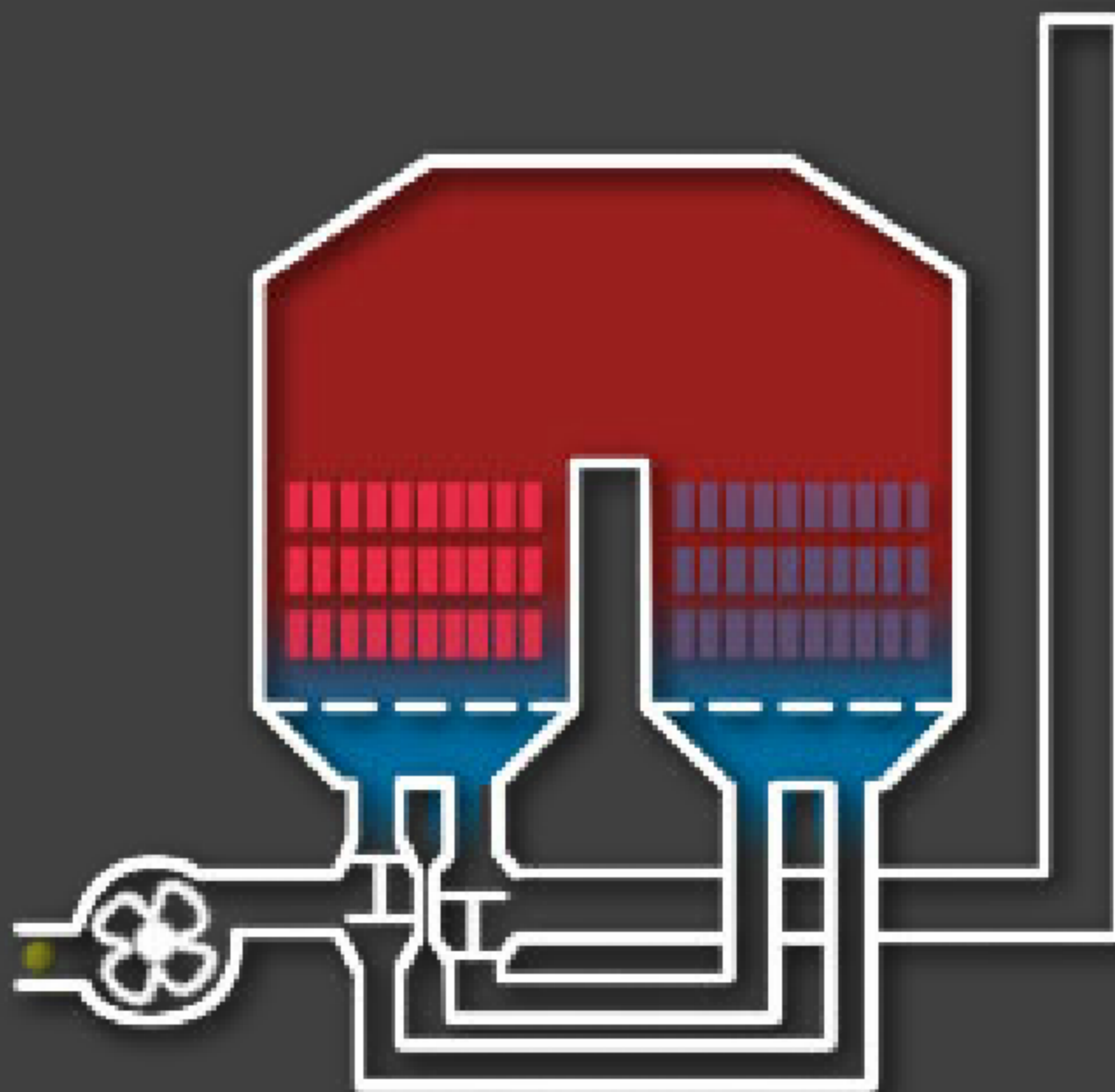
Ductwork to Oxidizer



Regenerative Thermal Oxidizer



RTO Operation





DIVISION PLATE

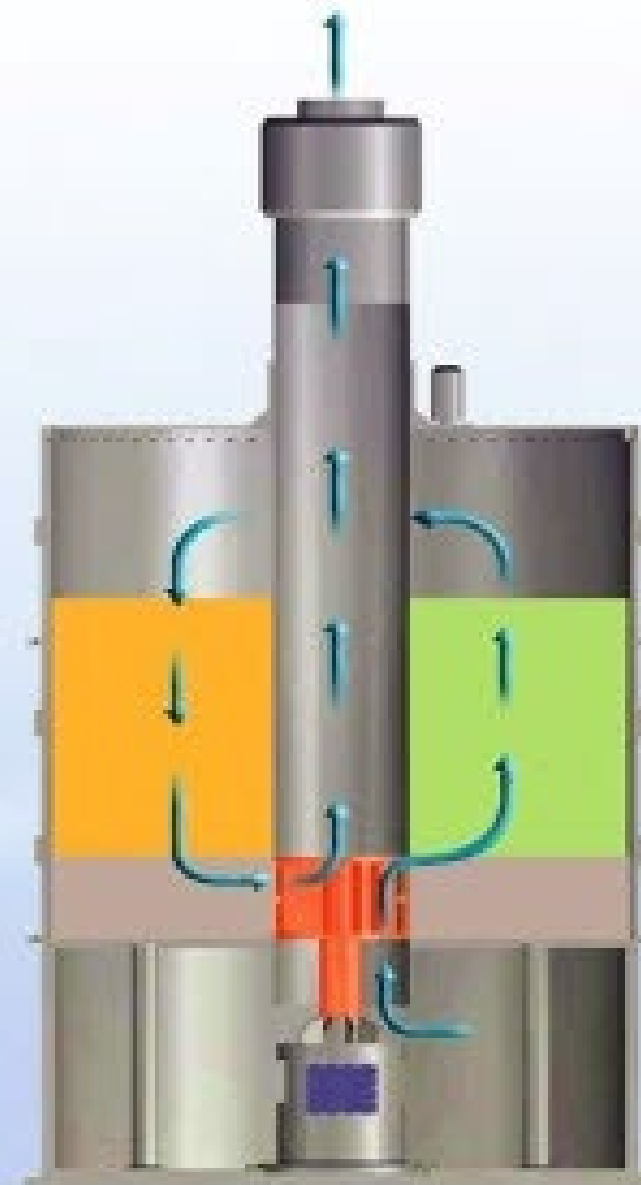
PURGE ZONE

HEATING ZONE

COOLING ZONE

DEAD ZONE

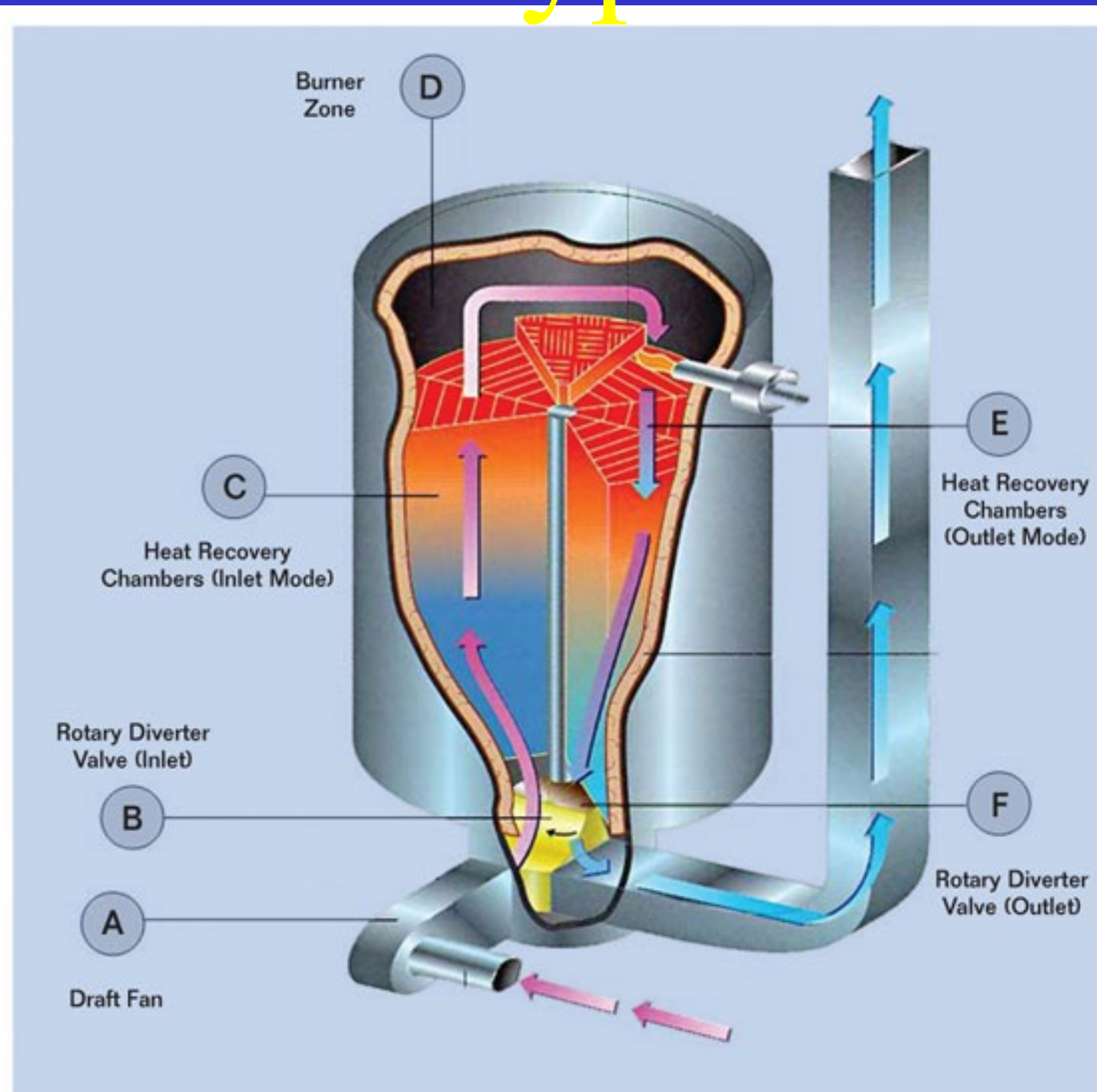
CERAMIC
Regenerative Material



ONE CAN TYPE RTO

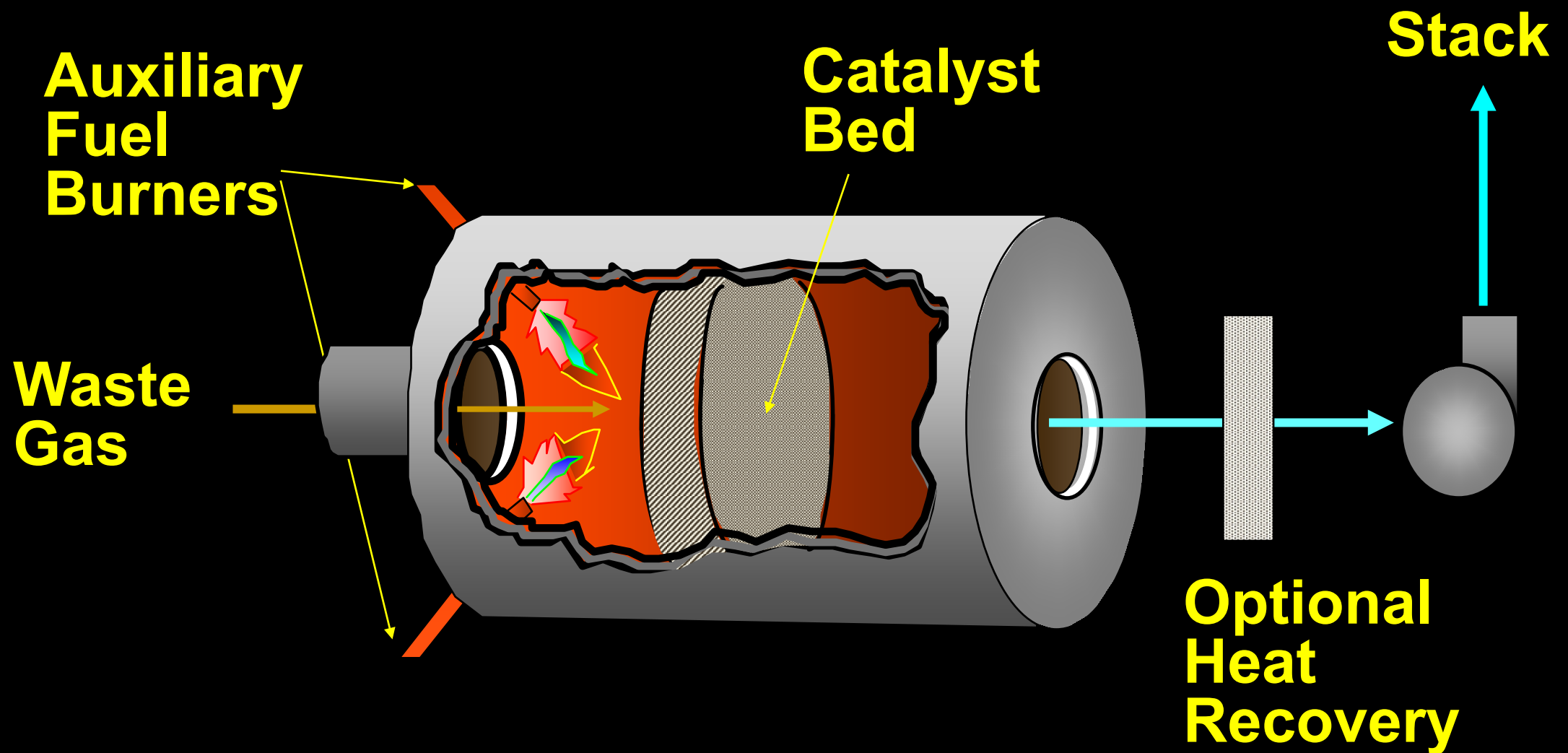


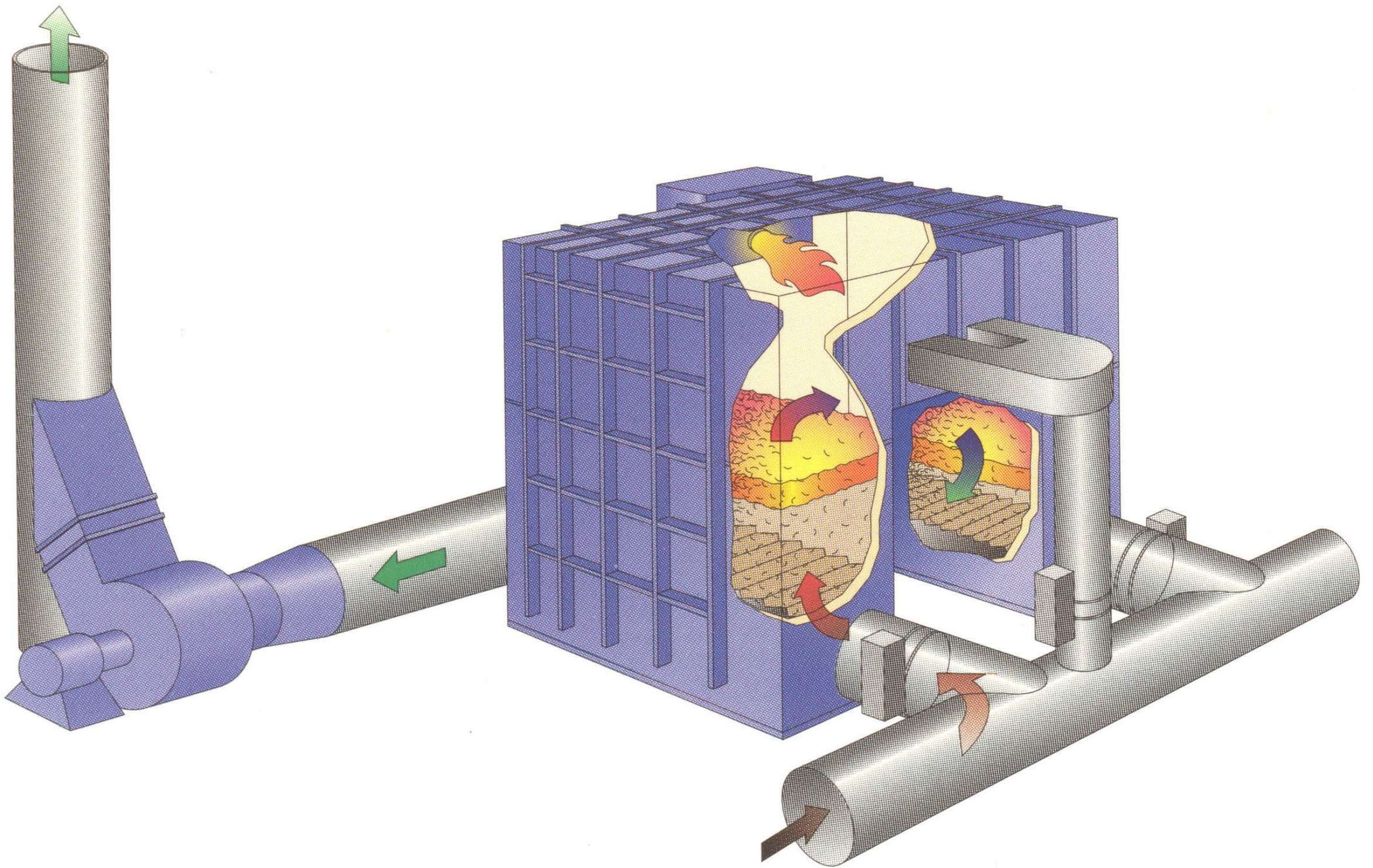
Can Type RTO



In the regenerative thermal oxidizer, the single rotary valve indexes across an open air path and methodically seals it off by reaching the next set position. A continuous air purge captures any scavenging dirty air in the switch and returns it for treatment in the oxidizer.

Catalytic Oxidizer/Incinerator





Regenerative Catalytic Oxidizer



Selection Criteria

- Type of VOCs
- Concentration of VOCs
- Process flow rate
- Economics

Catalytic vs. Thermal for VOC Control

Catalytic

Lower Operating
Temp. & Lower
Fuel Usage

Higher Capital &
Maintenance
Costs

Catalyst Fouling
& Poisoning

Thermal

Higher Operating
Temp. & Higher
Fuel Usage

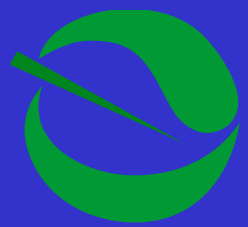
Lower Capital &
Maintenance
Costs

No Catalyst
Involved Here



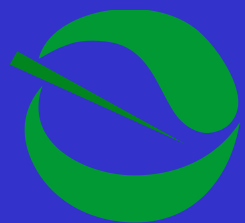
Catalyst Problems

- Scouring
- Thermal burnout
- Thermal aging
- Masking
- Catalyst fouling and poisoning



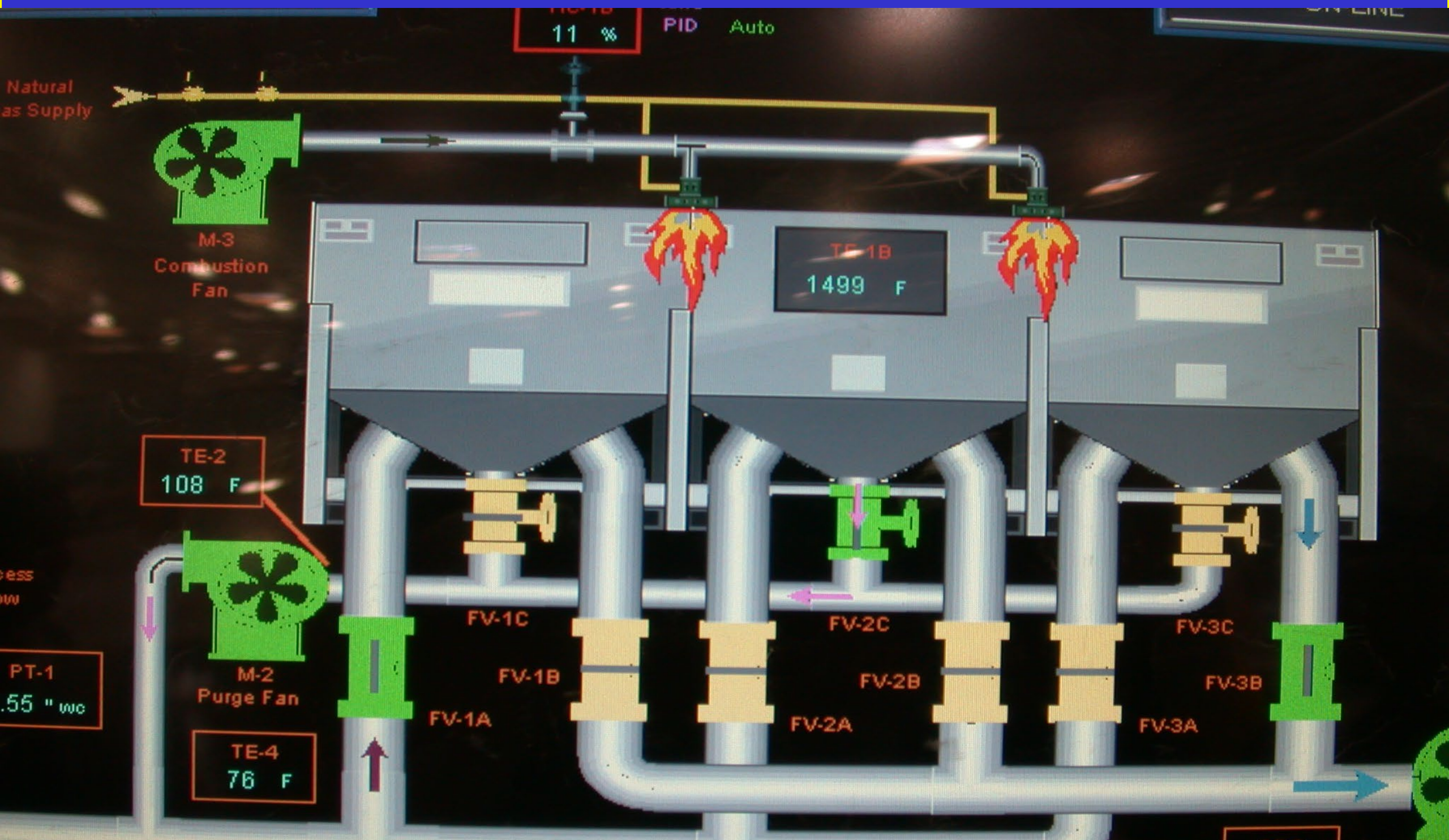
Catalytic Poisons

- Fast acting poisons
 - * phosphorus P, bismuth Bi, lead Pb, arsenic As, antimony Sb, mercury Hg
- Slow acting
 - * iron Fe, tin Sn, silica Si
- Reversible
 - * sulfur S, zinc Zn, chlorine, bromine, fluorine etc. halogens



Catalyst Efficiency

- Operating temperature
- Space velocity
- VOC composition
- VOC concentration
- Catalyst properties
- Poisons and inhibitors



Heat Exchangers

MAIN MENU

tune PID
PIC-1
71

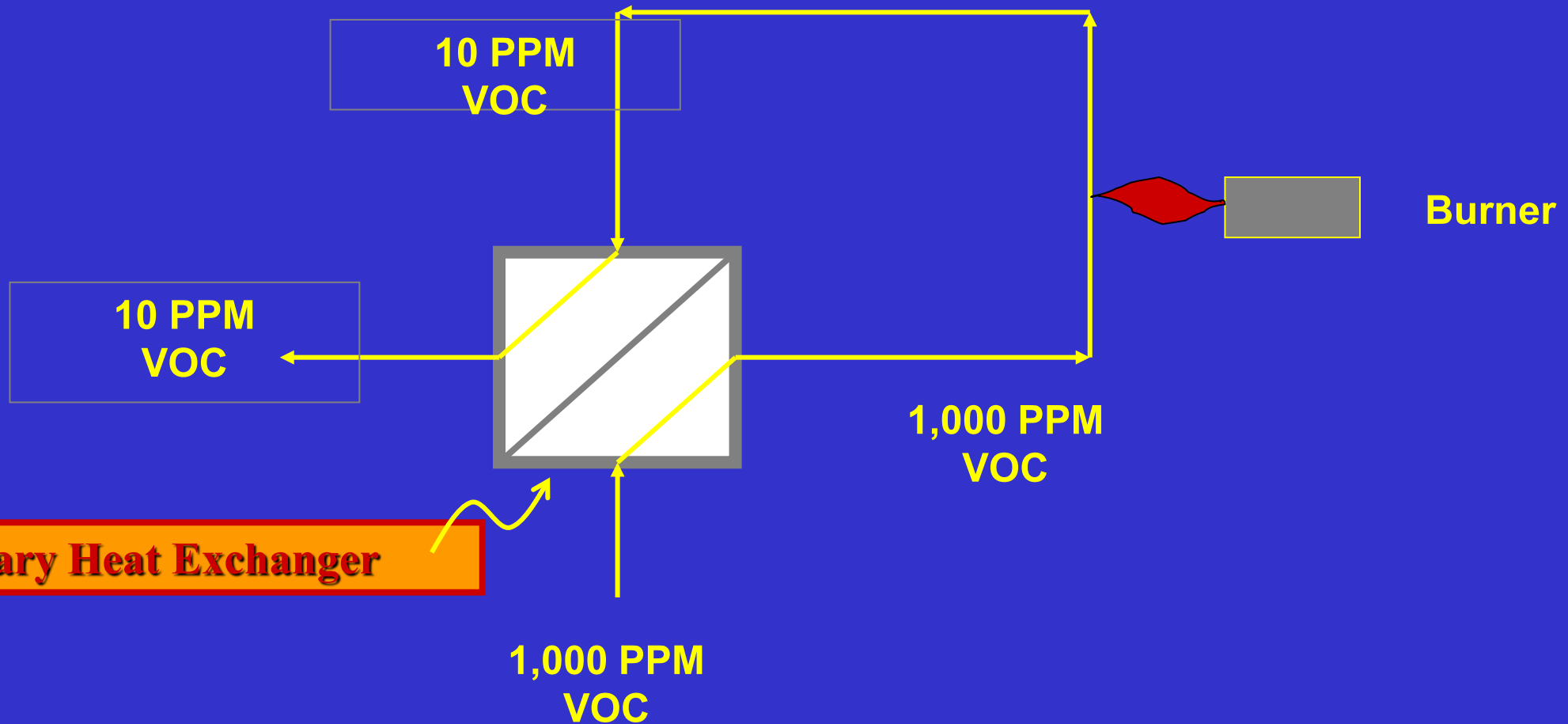
Thermal & Catalytic Oxidizer Heat Exchangers

There are two basic types of heat exchangers used for thermal or catalytic oxidizers

- Metal Heat Exchangers or “recuperative heat exchangers”
- Ceramic Bed Heat Exchangers or “regenerative heat exchangers”

Thermal & Catalytic Oxidizer Heat Exchangers

$$\text{DRE} = \frac{1,000 - 10}{1,000} = \underline{99\%}$$

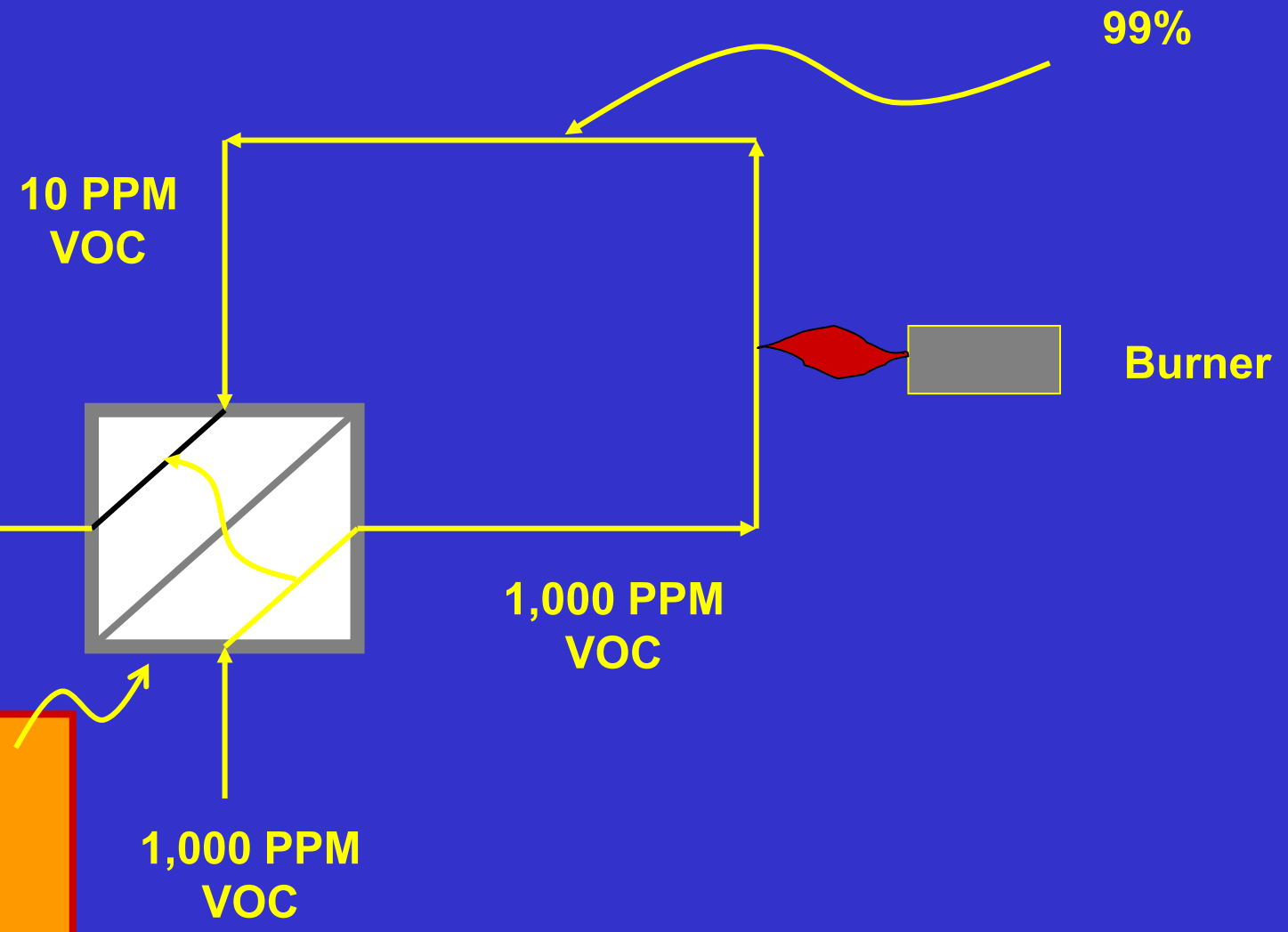


Thermal & Catalytic Oxidizer Heat Exchangers

$$\text{DRE} = \frac{1,000 - 30}{1,000} = \underline{97\%}$$

10 + 20 = 30
PPM VOC

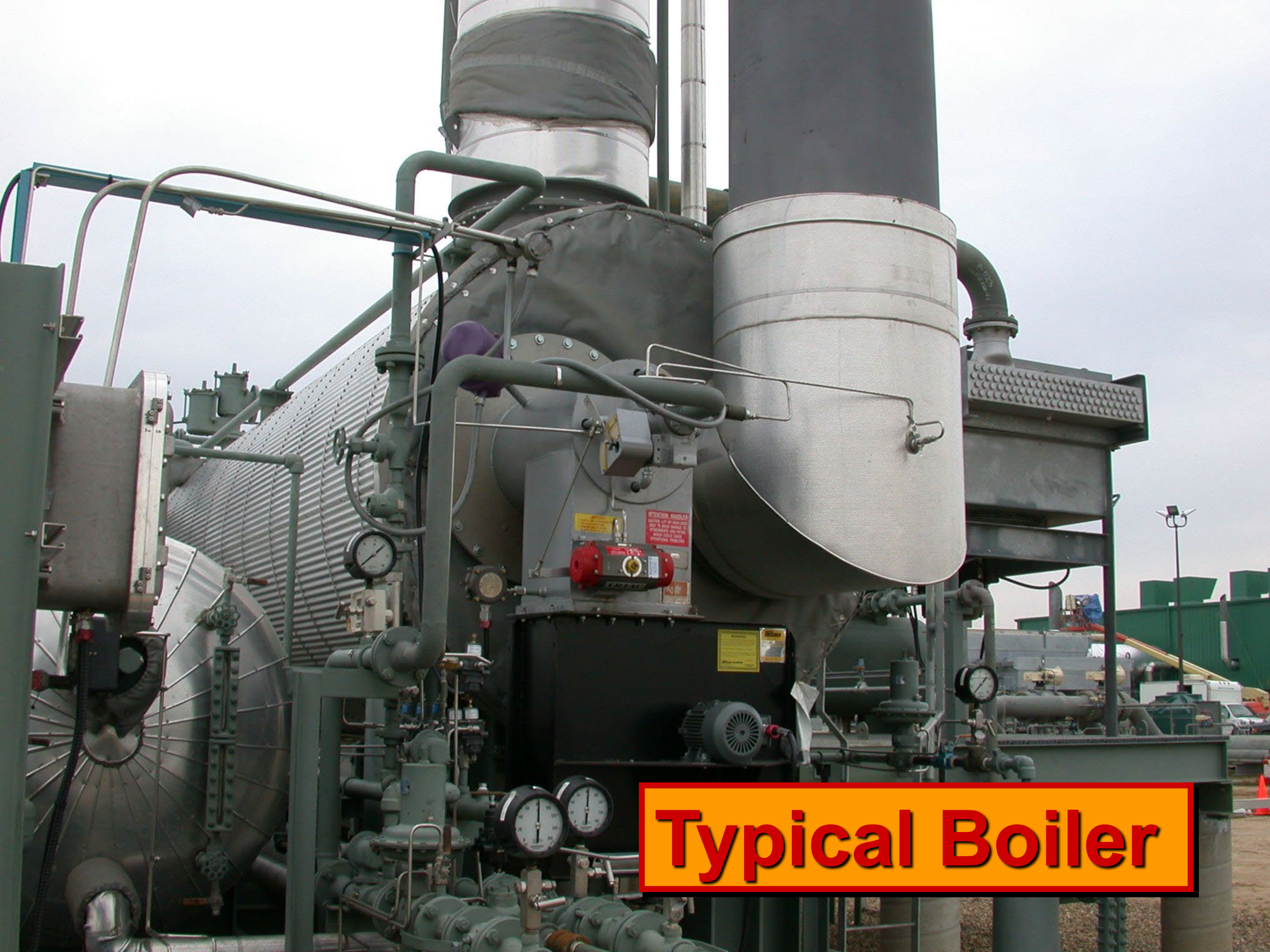
Primary Heat Exchanger
(with 2% leakage)
2% of 1000 ppm = 20 ppm





Boilers, Process Heaters & Flares

- Boilers make steam
- Process heaters add heat to material
- Flares are thermal incinerators without a combustion chamber



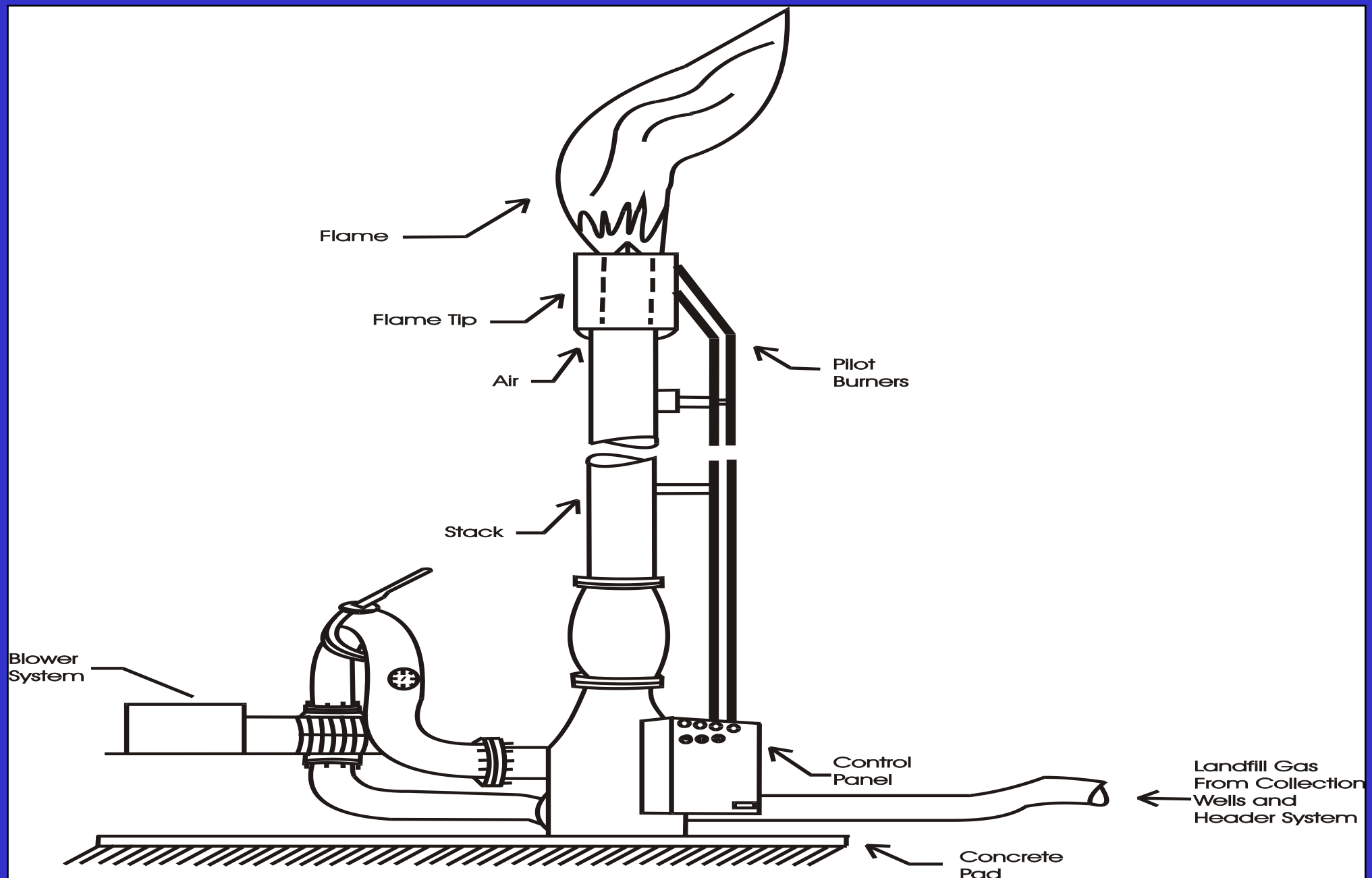
Typical Boiler

Let's Discuss Flares



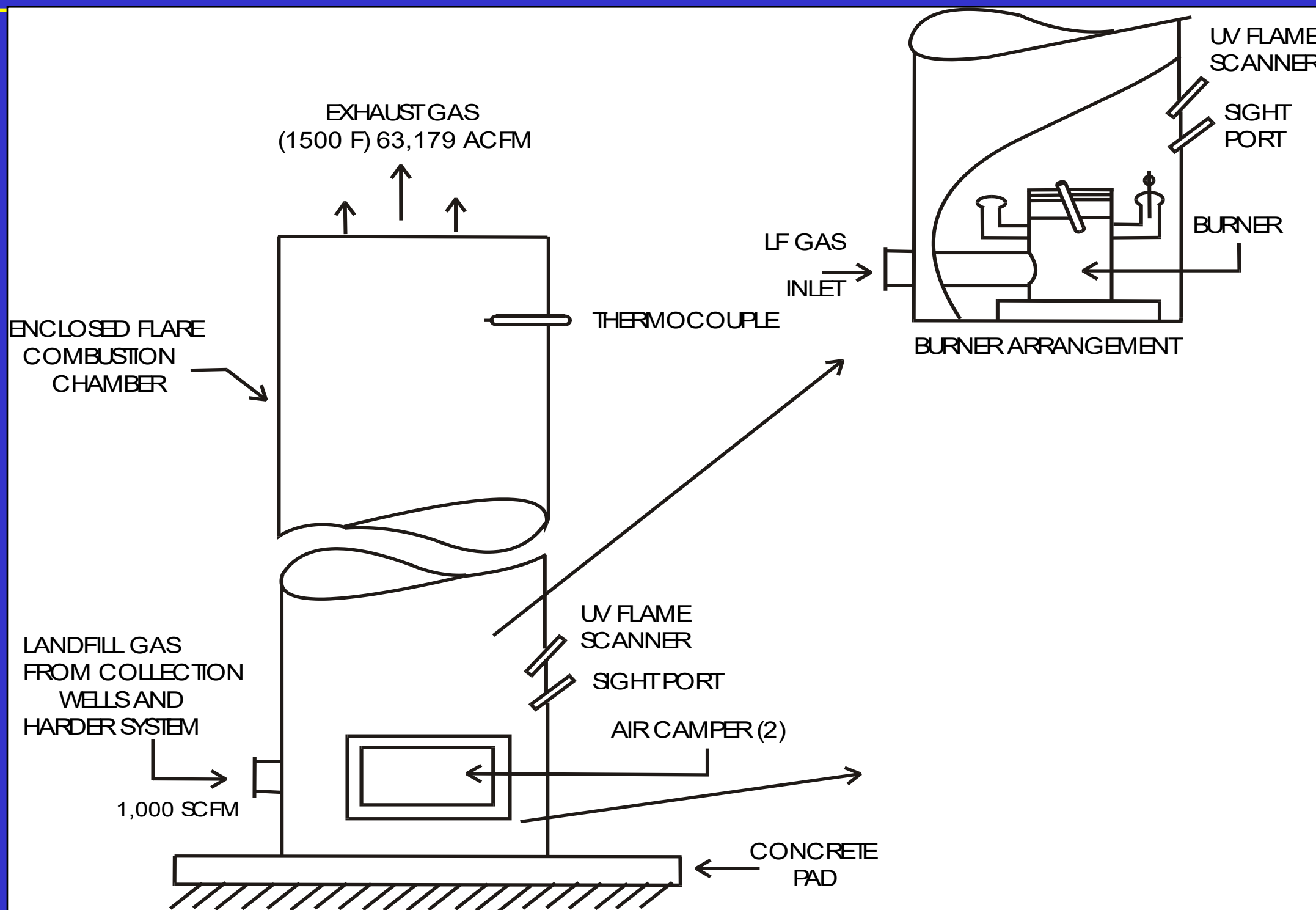


Flare Types – Open or Elevated





Flare Types – Enclosed or Ground





Flares

Gasoline
Marketing:
Bulk Terminal

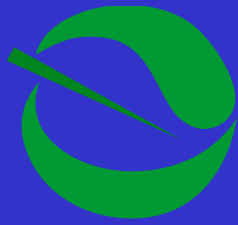


Bluff Road Municipal
Solid Waste Landfill
Lincoln, NE



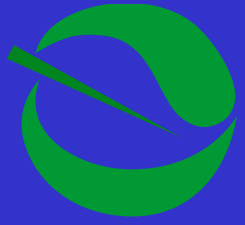


Waste Gas Collection & Flare



Shell Deer Park
Refinery in
Texas on the
Houston Ship
Channel.





Flaring gases from an oil platform.





Incinerator Inspection

- Look for
 - * Excessive corrosion and rust
 - * Holes in incinerator shell or ducts
 - * Visible emissions
 - * Excessive odors
 - * Last time catalyst was replaced



Incinerator Inspection

- Record
 - * VOC outlet concentration
 - * Incinerator inlet temperature
 - * Incinerator outlet temperature
 - * Pressure drop



VOC Control : Three-Way Catalyst

INSPECTIONS





Three Stages

- Pre-Inspection
 - * file review, rule review, inspection forms, copy of permit, safety equipment check
- Inspection
 - * facility safety indoctrination, pre-inspection meeting
- Post-Inspection Interview



Pre-Inspection Guidelines

- Regulation review
- Equipment check
- Pre-entry and entry
- Pre-inspection meeting
- Permit check



Pre-Inspection Meeting

- Facility name and ownership
- Address including city and zip
- Contact name and title
- Phone number including area code
- Production rate



Pre-Inspection Meeting

- Operating schedule
- Operation season
- Date of last source test
- Fuel usage and sulfur content



Inspection Report

- Description of facility & processes
- Flowchart with equipment location & emission points
- Process diagram (materials handled, flow rates, temperatures, pressures)
- Statement as to compliance or non-compliance
- Enforcement action recommendation



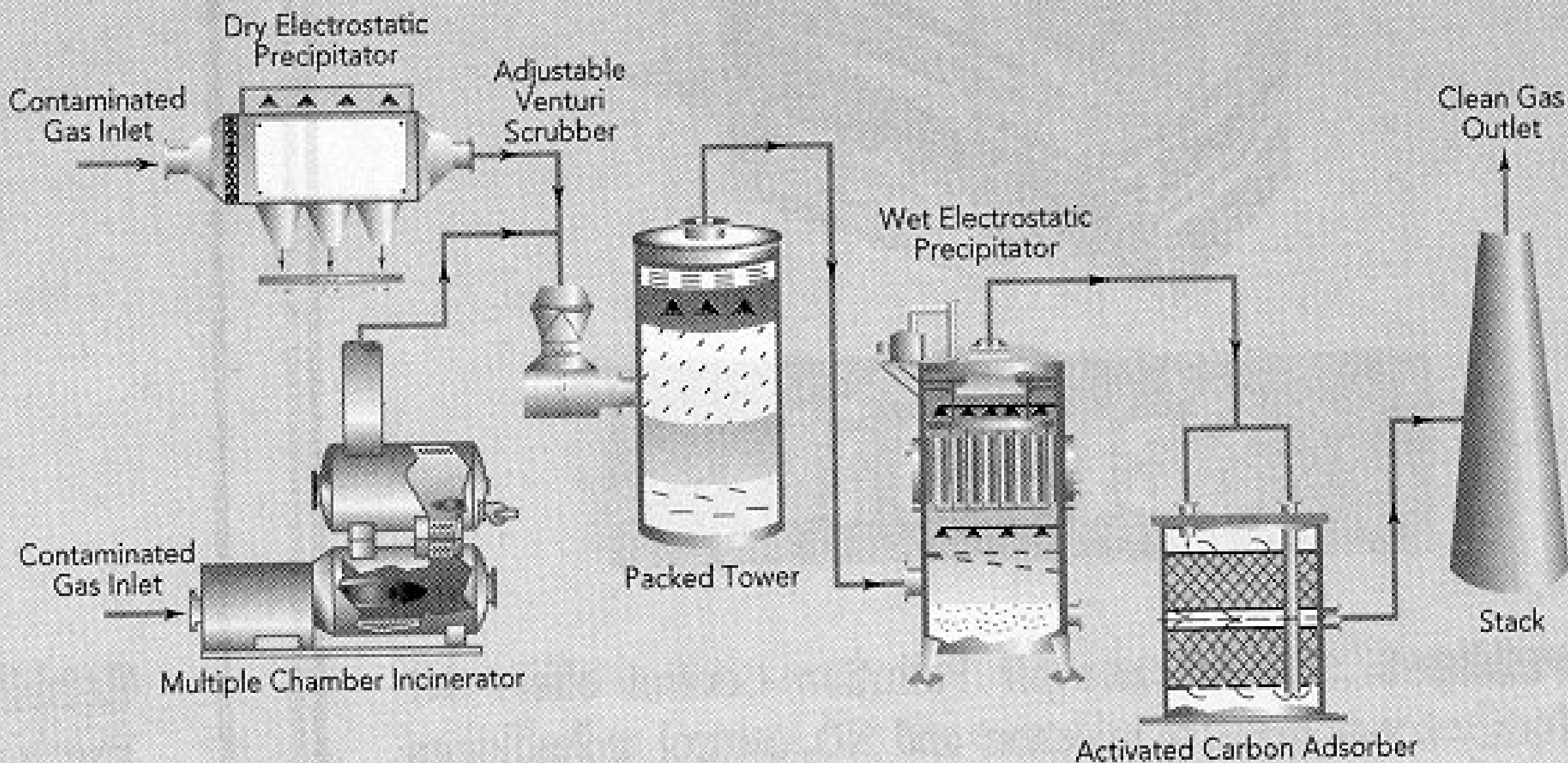
Usage Records

- Review usage records
- Obtain necessary copies



Six points of Inspection

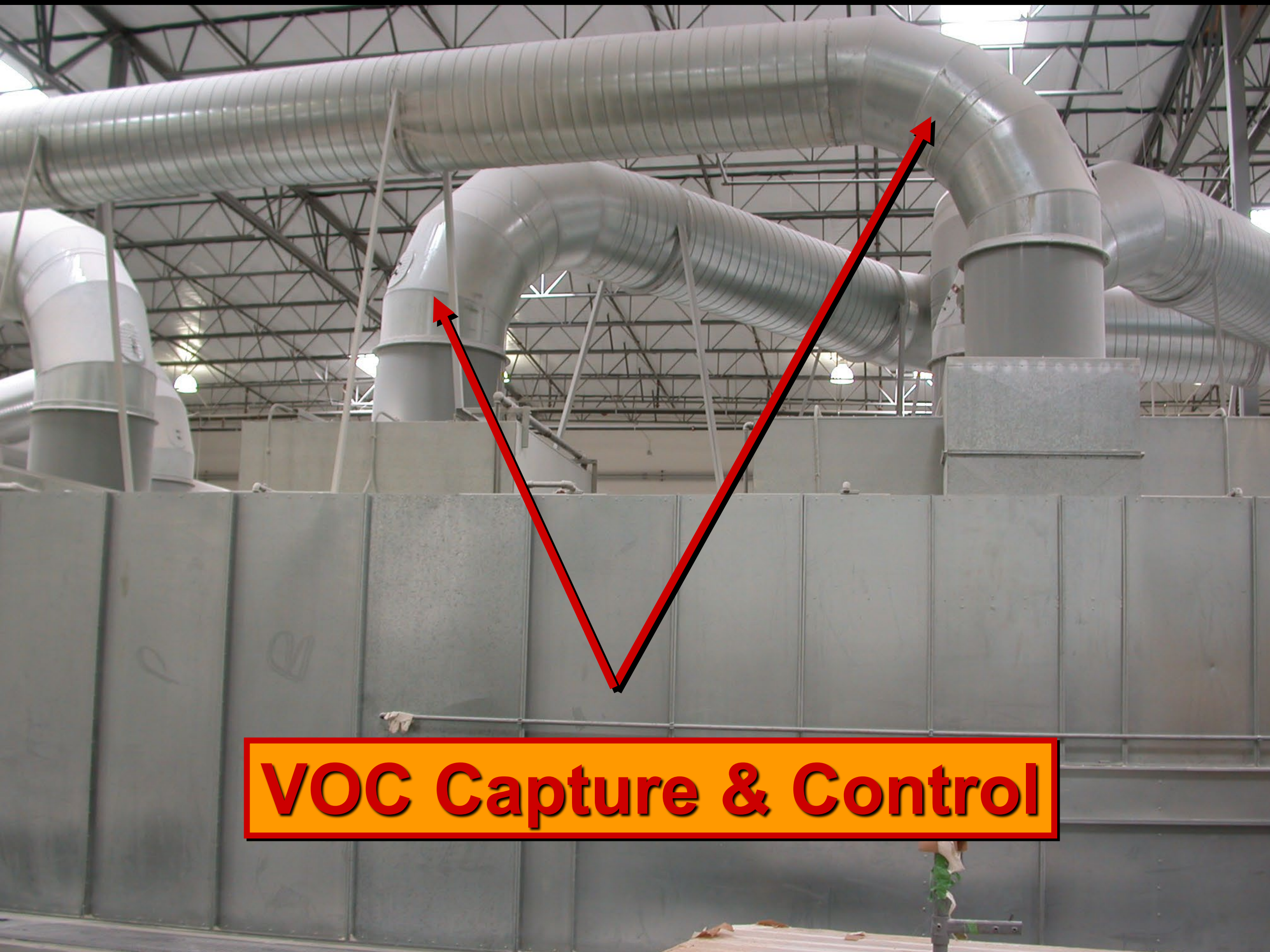
Capture, Transport, Air Mover,
Instrumentation, Control, Subsystem





Capture

- Are process emissions drawn into a control device at the point of release?
- Are they drawn into a collection device?

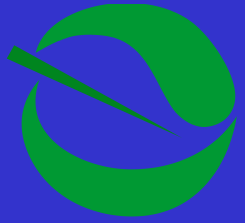


VOC Capture & Control



Transport

- Are the emissions moved to the control device without loss?
- Are there any leaks?



Air Mover

- Is the fan big enough for the job?
- Is it operating as designed and permitted?

Air Movers





Instrumentation

- Are the proper instruments present?
- Are they functioning?
- Are they calibrated regularly?
- Are they showing the proper units?

Instrumentation

OXIDIZER HIGH TEMPERATURE LIMIT
(NOT TO EXCEED 1900° F)

TEMPERATURE CONTROLLER

PREHEAT TEMPERATURE CONTROLLER

Barber-Colman 560 temperature controller panel. The digital display shows 1603 and 1700. The control panel includes buttons for HOLD, RUN, REM, RESET, DISPLAY, SET POINT 1, and SET POINT 2. A vertical scale on the right ranges from 0 to 40.

Barber-Colman 560 temperature controller panel. The digital display shows 1606 and -32-. The control panel includes buttons for HOLD, RUN, REM, REM LOCAL, MAN AUTO, DISPLAY, SET POINT 1, and SET POINT 2. A vertical scale on the right ranges from 0 to 40.

Barber-Colman 560 temperature controller panel. The digital display shows 1603 and 1600. The control panel includes buttons for HOLD, RUN, REM, REM LOCAL, MAN AUTO, DISPLAY, SET POINT 1, and SET POINT 2. A vertical scale on the right ranges from 0 to 40.

NEW - START UP - LOGIC

- 1. PUSH COMBUSTION FAN START BUTTON
- 3. PUSH OXIDIZER ID FAN START BUTTON
- 2. PUSH PURGE AIR FAN START BUTTON

OXIDIZER INLET STATIC
PRESSURE DAMPER
POSITION INDICATOR



OXIDIZER FAN BLEED-IN
AIR DAMPER POSITION



Control Device

- Is it functioning?
- Are there any visible leaks?
- Can the device handle the job?



Subsystem

- What is the ultimate fate of captured or concentrated emissions?
- Pressure gauges for accuracy & change
- Fines system for leaks & proper discharge
- Motor for proper operation





The End