
EPA's APTI
Course #450/#468
Monitoring Compliance
Test And
Source Test Observation

Reference Materials



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Notice

This is not an official policy and standards document. The opinions and selections are those of the author and not necessarily those of the Environmental Protection Agency. Every attempt has been made to represent the present state of the art as well as subject areas still under evaluation. Any mention of products or organizations does not constitute endorsement by the United States Environmental Protection Agency (USEPA).

Usage of This Manual

The Central States Air Resource Agencies Association (CenSARA) is one of several multi- jurisdictional organizations (MJOs) operating for the U.S. Environmental Protection Agency (USEPA), through the Air Pollution Training Institute (APTI), to update more of the frequently used APTI courses. The primary objectives of the MJOs are to:

- Promote the exchange of information between the States;
- Serve as a forum to discuss regional air quality issues of common concern;
- Share resources for the common benefit of the member states; and
- Provide training services to their member air pollution control agencies.

APTI provides courses on air pollution control technology, ambient air and source monitoring, and air quality management. Historically, APTI designed courses that meet the job training needs of governmental agency personnel and others in the field of air pollution. This requires a thorough examination of both the materials for instruction and the characteristics of the student audience. Based on studies conducted by APTI of those who have participated in the various training courses, courses were developed and revised to provide training that enables every student to achieve specific course objectives. A basic goal of APTI was to provide training that will enable a student to do specific jobs in his or her home environmental agency. However, recently APTI has taken a new direction and has given money to the various MJOs, of which CenSARA is one, to update needed training course for their member states.

CenSARA meets these training needs of its member states by identifying, designing, developing and delivering needed, cost-effective, responsive, and focused educational opportunities for state and local air agency staff. Agenda and course materials are obtained from a variety of sources including EPA, colleges and universities, regional training consortia, and individual instructors. Yet, due to changes in environmental regulations, the implementation of new policies, and the advancement of technologies, agendas and course materials become out-of-date. When this happens, staffs' ability to enhance skills, knowledge and abilities are constrained, limiting their ability to excel in the dynamic field of air pollution control. So by providing up-to-date, high quality educational opportunities for staff, their chances to greatly enhance their skills, knowledge and abilities is significantly improved.

Consequently, CenSARA announced a Request for Proposals (RFP) to the environmental training community to solicit technical proposals and cost bids to review current compliance test and observation programs within the USEPA and to **update** as necessary the content title, agenda topics, course length, instructor and student manuals, lectures involving presentation slides, classroom and homework exercises, and other handouts and materials for EPA's APTI Course #450/#468 entitled: "*Monitoring Compliance Test and Source Test Observation*." In response to CenSARA's RFP, EnviroTech Solutions, William T. "Jerry" Winberry, 1502 Laughridge Drive, Cary, North Carolina 27511, jwinberry@mindspring.com, 919-467-2785, was awarded the contract to update EPA's APTI Course #450/#468. Mr. Winberry is the author of this Student Workbook (SW) and every attempt has been made to represent the most recent advances in sampling and analytical methodology

Disclaimer

This material has been developed and assembled to provide training associated with EPA's APTI Course #450/#468 entitled: "*Monitoring Compliance Test and Source Test Observation*." It is not intended to be used for regulatory purposes, or to be a substitute for, nor interpreted as official Agency policy. Every attempt was made to reflect the technical state of art and regulatory information as of the date of this publication. This is not an official policy and standards document. The opinions and selections are those of the author and not necessarily those of the EPA. Any mention of products or organizations does not constitute endorsement by the EPA.

U.S. Environmental Protection Agency Air Pollution Training Institute Course #450/#468

Monitoring Compliance Test and Source Test Observation

Student Workbook (SW)

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agendas and course materials become out-of-date. When this happens, staffs' ability to enhance skills, knowledge and abilities are constrained, limiting their ability to excel in the dynamic field of air pollution control. So, by providing up-to-date, high quality educational opportunities for staff, their chances to greatly enhance their skills, knowledge and abilities is significantly improved. Consequently, CenSARA has taken the initiative to update EPA's APT! Course #450/#468.

The updated Course #450/#468 is targeted primarily at agency personnel who have the responsibility to monitor compliance test and source test observation of various FRMs and SW- 846 methodologies. The revised course materials provide the best available and most current information as well as relevant example exercises to improve the knowledge and expertise of agency observer personnel. The desired outcome of this effort is that agency staff be able to properly observe compliance test and perform source test observations to protect public health and the environment by developing an understanding of EPA's stack testing programs and strategy, various stack test methodologies for quantifying HAPs from industrial processes, and associated quality assurance/quality control activities and requirements.

The specific objectives of updating APTI Course #450/#468 materials include:

- Explain why sampling of source emissions for PM and HAPs is important in air pollution control agency programs;
- Define symbols and common terms used in the application of FRMs 1-5 and SW-846 source sampling;
- Recognize, interpret and apply sections of the Code of Federal Regulations, Appendix A, pertinent to source sampling for particulate pollutants;
- Understand the construction, operation and calibration of the component parts of the FRM 5 particulate matter sampling train;
- Define "Isokinetic sampling" and illustrate why it is important in sampling for PM;
- Understanding the "working" isokinetic rate equation and its derivation
- Learn how to use specific software for determining isokinetic sampling rates;
- Apply FRMs 1 through 4 in preparation for applying FRM 5 at a source to characterize PM emissions;
- Understand how the "S-type" pitot tube is constructed and standardized and how it is applied in source sampling;
- Understand the difference between systematic error and precision as applied to source test measurements;
- Properly assemble and leak check a FRM 5 sampling train;
- Describe the operation of the VOST, Semi-VOST, Acid Gas and FRM 201/201A sampling trains;
- Describe what are "condensibles" and how the FRM 202 sampling train operates to measure condensibles; and
- Calculate the "% Isokinetic" value for a source test, and interpret the effect of over or under- isokinetic values on the source test results.

One of the products of the #450/#468 update is the Student Workbook (SW). The SW provides copies of visuals presented in each technical lecture. Contained in this SW are the visuals for each of the Sessions and lessons presented in APT! Course #450/#468.

I. Background

Over the last several years, public awareness of air pollution and its sources has increased steadily. Specifically, since the passage of the Clean Air Act Amendments (the Act) of 1990 and the implementation of EPA's maximum achievable control technology (MACT) program, more emphasis has been placed on local, State, and Federal agencies to keep the public informed on health effects of various air pollutants and their levels of exposure. This awareness has required industry to become accountable for air pollution emissions as part of their source's compliance strategy and regulatory agencies to become knowledgeable of the methodology used to quantify reported emissions.

Traditionally, air pollution control agencies require all sources to quantitate their emissions in order to demonstrate compliance. To insure that the source test methods utilized to demonstrate compliance are performed according to EPA guidelines, regulatory agencies observe stack test to document compliance with the methodology. This has meant that sources must know how to properly sample and report emissions using EPA reference methods and that regulatory agencies must know how to properly observe these test to insure that reliable and accurate data is gathered and reported.

In addition, with the passage of the Act and Title IV, Acid Rain, those fossil-fueled-fired-steam generators (FFFSG) covered must report emissions in units of pollutant mass rate (pmr) as part of EPA's emission trading program. This required the installation and certification of velocity monitors if the continuous emission monitoring system (CEMS) data was to be used as part of EPA's market base-trading program. These new regulatory requirements have now required Agency personnel to know how to observe and verify the proper use and implementation of Federal Reference Methods 2F, 2G, and 2H in certification of the velocity CEMS as part of a source's market base-trading emission program.

Consequently, the Act has required regional, state, and local air pollution control agencies to better understand the stack test methodologies that are used to quantitate air pollution emissions from industrial sources. Agencies see a need to provide training to it's personnel on how to properly observe stack test used to meet regulatory compliance for criteria and hazardous air pollutants (HAPs).

II. Course Objective

The objective of EPA's APTI Course #450/#468 is to provide Agency personnel the needed background information, checklist, and guidance associated with EPA methodologies involving stack test for characterizing and quantifying criteria and HAPs from industrial sources. This course will address Federal Reference Methods (FRMs)1-5 (isokinetic stack testing and stack test basics), FRMs 2F, 2G, and 2H (velocity), stack testing for volatile organic compounds (FRMs 18, 25, 25A/B), SO₂ (FRMs 6, 6A, and 8), NO_x (FRMs 7, 7C), dioxin/furans and PCBs (FRM 23), heavy metals (FRM 29), and instrumental (FRMs 3A, 6C, and 7E).

The course will present standardized stack test methodology for sampling and analysis of air pollutants as identified in the Code of Federal Regulations (CFR), Part 60, Appendix A. In

addition, information will be presented on EPA's stack test monitoring programs associated with PM-10 (Methods 201 A/B) and condensable particulate (Method 202) monitoring from CFR 51 Subpart M. Specific source test checklists will be demonstrated during the presentation for each test methodology as part of the course. Participants will learn how to use the checklist in performing source test observations, how to observe mandated QA/QC requirements associated with each methodology, and how to use source test databases available to the air pollution scientist.

At the end of this course, the agency personnel will gain an understanding of the unique source test methods used to accurately characterize air pollutant emissions from industrial sources. The agency personnel will learn about EPA's programs that require characterization of air pollutants from industrial sources utilizing FRMs.

III. Course Manuals

Course presentations and videos will be supported by four manuals. They are:

- **#468 Student Workbook (Hard Copy):** The updated Student Workbook (SW) contains copies of selected slides from each of the lecture presentations, and **Agency Inspector Checklist.**
- **#468 Stack Test Manual (Flash Drive):** The Stack Test Manual contains full copies of selected stack test sampling and analytical methods found in the CFR, Part 60, Appendix A, and SW-846, Standard Methods.
- **Apex Instruments Federal Reference Method 5 Operational Manual:** The Federal Reference Method 5 Operation Manual is a document describing specifically the construction and operation of Federal Reference Method 5 stack testing sampling train.
<http://www.apexinst.com/cms/wp-content/uploads/2015/03/Model-500-Series-Manual-Method-5.pdf>.

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Appendix A

Equations

General

$$\Delta H = K (\Delta p)$$

$$\overline{pmr}_s = c_s Q_s$$

$$A_s = \pi \left(\frac{D_s}{2} \right)^2$$

$$A_s = LW$$

$$F_d = \frac{20.9 - \%O_2}{\%CO_2}$$

$$\%EA = \frac{(\%O_2) + 0.5(\%CO)}{0.0264(\%N_2) - (\%O_2) + 0.5(\%CO)}$$

$$\%N_2 = 100 - \%CO_2 - \%O_2 - \%CO$$

Method 1

Equation 1-1

$$D_e = \frac{2LW}{L+W}$$

Equation 1-2

$$R_i = \arccosine[(\cosine Y_i)(\cosine P_i)]$$

Equation 1-3

$$\bar{R} = \frac{\sum R_i}{n}$$

Equation 1-4

$$S_d = \sqrt{\frac{\sum_{i=1}^n (R_i - \bar{R})^2}{(n-1)}}$$

Method 2

Equation 2-1

$$D_e = \frac{2LW}{L+W}$$

Equation 2-2

$$C_{p(s)} = C_{p(std)} \sqrt{\frac{\Delta p_{std}}{\Delta p_s}}$$

Equation 2-3

$$Deviation = C_{p(s)} - \bar{C}_p (A \text{ or } B)$$

Equation 2-4

$$\sigma(\text{side A or B}) = \frac{\sum_{i=1}^3 |C_{p(s)} - \bar{C}_p (A \text{ or } B)|}{3}$$

Equation 2-5

$$=M_d(1-B_{ws})+18.0B_{ws}$$

Equation 2-6

$$=P_{bar} + P_g$$

Equation 2-7

$$=273+t_s \text{ for metric.}$$

Equation 2-8

$$=460+t_s \text{ for English.}$$

Equation 2-9

$$v_s = K_p C_p (\sqrt{\Delta p})_{avg} \sqrt{\frac{T_{s(avg)}}{P_s M_s}}$$

Equation 2-10

$$Q_{sd} = 3,600(1-B_{ws}) v_s A \frac{T_{std}}{T_{s(avg)}} \frac{P_s}{P_{std}}$$

Method 2F

Equation 2F-1

$$F_1 = \frac{(P_4 - P_5)}{(P_1 - P_2)}$$

Equation 2F-2

$$F_2 = C_p \sqrt{\frac{\Delta P_{std}}{(P_1 - P_2)}}$$

Equation 2F-3

$$\% \text{ Diff} = \frac{F_2^{\max} - F_2^{\min}}{F_2^{\min}} \times 100\%$$

Equation 2F-4

$$M_s = M_d(1-B_{ws})+18.0B_{ws}$$

Equation 2F-5

$$P_s = P_{bar} + \frac{P_g}{13.6}$$

Equation 2F-6

$$T_{s(i)} = 273+t_{s(i)}$$

Equation 2F-7

$$T_{s(i)} = 460+t_{s(i)}$$

Equation 2F-8

$$v_{a(i)} = K_p F_{2(i)} \sqrt{\frac{(P_1 - P_2)_i T_{s(i)}}{P_s M_s}} (\cos \theta_{y(i)}) (\cos \theta_{p(i)})$$

Equation 2F-9

$$v_{a(avg)} = \frac{\sum_{i=1}^n v_{a(i)}}{n}$$

Equation 2F-10

$$Q_{sw} = 3,600 (v_{a(avg)}) (A) \left(\frac{T_{std}}{T_{s(avg)}} \right) \left(\frac{P_s}{P_{std}} \right)$$

Equation 2F-11

$$Q_{sw} = 3,600 (1 - B_{ws}) (v_{a(avg)}) (A) \left(\frac{T_{std}}{T_{s(avg)}} \right) \left(\frac{P_s}{P_{std}} \right)$$

Method 3

Equation 3-1

$$M_d = 0.440 (\% CO_2) + 0.320 (\% O_2) + 0.280 (\% N_2 + \% CO)$$

Method 3A

Equation 3A-1

$$C_{gas} = \frac{C_{ma} - C_{oa}}{C_m - C_o} (\bar{C} - C_m) + C_{ma}$$

Method 4

Equation 4-1

$$V_{wc(std)} = \frac{(V_f - V_i) \rho_w RT_{std}}{P_{std} M_w} = K_1 (V_f - V_i)$$

Equation 4-2

$$V_{wsg(std)} = \frac{(W_f - W_i) RT_{std}}{P_{std} M_w} = K_2 (W_f - W_i)$$

Equation 4-3

$$V_{m(std)} = V_m Y \frac{(P_m)(T_{std})}{(P_{std})(T_m)} = K_3 Y \frac{V_m P_m}{T_m}$$

Equation 4-4

$$B_{ws} = \frac{V_{wc(std)} + V_{wsg(std)}}{V_{wc(std)} + V_{wsg(std)} + V_{m(std)}}$$

Equation 4-5

$$V_{wc} = \frac{(V_f - V_i) \rho_w RT_{std}}{P_{std} M_w} = K_1 (V_f - V_i)$$

Equation 4-6

$$V_{m(std)} = V_m \left(\frac{P_m}{P_{std}} \right) \left(\frac{T_{std}}{T_m} \right) = K_2 \frac{V_m P_m}{T_m}$$

Equation 4-7

$$B_{ws} = \frac{V_{wc}}{V_{wc} + V_{m(std)}} + B_{wm}$$

Method 5

Equation 5-9a

$$\Delta H_{@} = 0.0319 \Delta H \frac{T_m}{P_{bar}} \frac{\theta^2}{Y^2 V_m^2}$$

Equation 5-1

$$V_{m(std)} = V_m Y \left(\frac{T_{std}}{T_m} \right) \left[\frac{P_{bar} + \frac{\Delta H}{13.6}}{P_{std}} \right]$$

$$= K_1 V_m Y \frac{P_{bar} + \left(\frac{\Delta H}{13.6} \right)}{T_m}$$

Equation 5-1a

$$[V_m - (L_p - L_a)]$$

$$\left[V_m - (L_1 - L_a) \theta_1 - \sum_{i=2}^n (L_i - L_a) \theta_i - (L_p - L_a) \theta_p \right]$$

Equation 5-2

$$V_{w(std)} = \frac{V_{lc} \rho_w R T_{std}}{M_w P_{std}}$$

$$= K_2 V_{lc}$$

Equation 5-3

$$B_{ws} = \frac{V_{w(std)}}{V_{m(std)} + V_{w(std)}}$$

Equation 5-4

$$C_a = \frac{m_a}{V_a \rho_a}$$

Equation 5-5

$$W_a = C_a V_{aw} \rho_a$$

Equation 5-6

$$c_s = (0.001 \text{ g/mg}) \left(\frac{m_n}{V_{m(std)}} \right)$$

Equation 5-7

$$I = \frac{100 T_s \left[K_3 V_{lc} + \left(\frac{V_m Y}{T_m} \right) \left(P_{bar} + \frac{\Delta H}{13.6} \right) \right]}{60 \theta V_s P_s A_n}$$

Equation 5-8

$$I = \frac{100 T_s V_{m(std)} P_{std}}{60 T_{std} v_s \theta A_n P_s (1 - B_{ws})}$$

$$= \frac{K_4 T_s V_{m(std)}}{P_s v_s A_n \theta (1 - B_{ws})}$$

Equation 5-8a

$$Q = K_1 \frac{P_{bar}}{t_w + t_{std}} \frac{V_w}{\theta}$$

Equation 5-9

$$K' = \frac{K_1 V_m Y \left(P_{bar} + \frac{\Delta H}{13.6} \right) \sqrt{T_{amb}}}{P_{bar} T_m \theta}$$

Equation 5-10

$$V_{m(std)} = K_1 V_m \frac{P_{bar} + \frac{\Delta H}{13.6}}{T_m}$$

Equation 5-11

$$V_{cr}(std) = K' \frac{P_{bar} \theta}{T_{amb}}$$

Equation 5-12

$$Y = \frac{V_{cr}(std)}{V_{m}(std)}$$

Method 6

Equation 6-1

$$\begin{aligned} V_{m}(std) &= V_m Y \left(\frac{T_{std}}{T_m} \right) \left(\frac{P_{bar}}{P_{std}} \right) \\ &= K_1 Y \left(\frac{V_m P_{bar}}{T_m} \right) \end{aligned}$$

Equation 6-2

$$C_{SO_2} = K_3 \frac{(V_t - V_{tb}) N \left(\frac{V_{so \ln}}{V_a} \right)}{V_{m}(std)}$$

Equation 6-3

$$RE = \frac{C_d - C_a}{C_a} (100)$$

Equation 6-4

$$V_{sb}(std) = V_{sb} \left(\frac{T_{std}}{T_{amb}} \right) \left(\frac{P_{bar}}{P_{std}} \right)$$

Equation 6-5

$$Q_{std} = \frac{V_{sb}(std)}{\theta}$$

Equation 6-6

$$V_{m}(std) = \bar{Q}_{std} \theta_s (1 - B_{wa}) \left(\frac{P_{bar} + P_{sr}}{P_{bar} + P_c} \right)$$

Equation 6-7

$$V_{m}(std) = \bar{Q}_{std} \theta_s (1 - B_{wa}) \sqrt{\frac{M_a}{M_s}} \left(\frac{P_{bar} + P_{sr}}{P_{bar} + P_c} \right)$$

Method 6A

Equation 6A-1

$$V_{CO_2}(std) = 5.467 \times 10^{-4} (m_{af} - m_{ai})$$

Equation 6A-2

$$V_w(std) = 1.336 \times 10^{-3} (m_{wf} - m_{wi})$$

Equation 6A-3

$$C_{SO_2} = 32.03 \frac{(V_t - V_{tb}) N \left(\frac{V_{so \ln}}{V_a} \right)}{V_{m}(std) + V_{CO_2}(std)}$$

Equation 6A-4

$$C_{CO_2} = \frac{V_{CO_2}(std)}{V_{m}(std) + V_{CO_2}(std)} \times 100$$

Equation 6A-5

$$C_w = \frac{V_{w(std)}}{V_{m(std)} + V_{w(std)} + V_{CO_2(std)}}$$

Equation 6A-7

$$m_{SO_2} = 32.03(V_t - V_{tb})N \left(\frac{V_{so ln}}{V_s} \right)$$

Equation 6A-8

$$E_{SO_2} = F_c (1.829 \times 10^9) \frac{m_{SO_2}}{(m_{af} - m_{ai})}$$

Method 6C

Equation 6C-1

$$C_{gas} = (\bar{C} - C_o) \frac{C_{ma}}{C_m - C_o}$$

Method 7

Equation 7-1

$$K_c = 100 \frac{A_1 + 2A_2 + 3A_3 + 4A_4}{A_1^2 + A_2^2 + A_3^2 + A_4^2}$$

Equation 7-2

$$V_{sc} = \left(\frac{T_{std}}{P_{std}} \right) (V_f - V_a) \left(\frac{P_f}{T_f} - \frac{P_i}{T_i} \right)$$

$$= K_1 (V_f - 25 ml) \left(\frac{P_f}{T_f} - \frac{P_i}{T_i} \right)$$

Equation 7-3

$$m = 2K_c A F$$

Equation 7-4

$$C = K_2 \frac{m}{V_{sc}}$$

Equation 7-5

$$RE = \frac{C_d - C_a}{C_a} (100)$$

Method 8

Equation 8-1

$$V_{m(std)} = V_m Y \left(\frac{T_{std}}{T_m} \right) \left(\frac{P_{bar} + \left(\frac{\Delta H}{13.6} \right)}{P_{std}} \right)$$

$$= K_1 V_m Y \left(\frac{P_{bar} + \left(\frac{\Delta H}{13.6} \right)}{T_m} \right)$$

Equation 8-2

$$C_{H_2SO_4} = K_2 \frac{N(V_t - V_{tb}) \left(\frac{V_{so ln}}{V_a} \right)}{V_{m(std)}}$$

Equation 8-3

$$C_{SO_2} = K_3 \frac{N(V_t - V_{tb}) \left(\frac{V_{so ln}}{V_a} \right)}{V_{m(std)}}$$

Equation 8-4

$$I = \frac{100T_s \left[K_3 V_{lc} + \left(\frac{V_m Y}{T_m} \right) \left(P_{bar} + \frac{\Delta H}{13.6} \right) \right]}{60 \theta V_s P_s A_n}$$

Equation 8-5

$$I = \frac{100 T_s V_{m(std)} P_{std}}{60 T_{std} v_s \theta A_n P_s (1 - B_{ws})}$$

$$= \frac{K_4 T_s V_{m(std)}}{P_s v_s A_n \theta (1 - B_{ws})}$$

Method 11

Equation 11-1

$$N_s = 2.039 \frac{W}{V_s}$$

Equation 11-2

$$N_A = 0.2039 \frac{W}{V_s}$$

Equation 11-3

$$N_I = N_T \frac{V_T}{V_I}$$

Equation 11-4

$$V_{m(std)} = V_m Y \left[\left(\frac{T_{std}}{T_m} \right) \left(\frac{P_{bar}}{P_{std}} \right) \right]$$

Equation 11-5

$$C_{H2S} = \frac{K[(V_{IT} N_I - V_{IT} N_T)_{sample} - (V_{IT} N_I - V_{IT} N_T)]}{V_{m(std)}}$$

Method 15

Equation 15-1

$$C = \frac{(K)(P_r)}{ML}$$

Equation 15-2

$$SO_2 \text{ equivalent} = \sum (H_2S, COS, 2CS_2) d$$

Equation 15-3

$$\text{Average } SO_2 \text{ equivalent} = \frac{\sum_{i=1}^N SO_2 \text{ equiv}_i}{N}$$

Method 15A

Equation 15A-1

$$V_{ms(std)} = \frac{V_{ms} Y (T_{std}) (P_{bar})}{(T_m) (P_{std})} = \frac{K_1 Y (V_m) (P_{bar})}{T_m}$$

Equation 15A-2

$$V_{mc(std)} = \frac{k_1 Y_c (V_{mc}) (P_{bar})}{T_m}$$

Equation 15A-3

$$C_{TRS} = \frac{K_2 (V_t - V_{tb}) N \left(\frac{V_{soln}}{V_a} \right)}{V_{ms(std)} - V_{mc(std)}}$$

Equation 15A-4

$$C_{RG} = \frac{(C_{COS})(Q_{COS})}{Q_{COS} + Q_{N2}}$$

Equation 15A-5

$$R = \frac{C_{TRS}}{C_{RG}} \times 100$$

Method 16

Equation 16-1

$$C = K \frac{P_r}{ML}$$

Equation 16-2

$$TRS = \sum (H_2S, MeSH, DMS, 2DMDS) d$$

Equation 16-3

$$Average TRS = \frac{\sum_{i=1}^N TRS_i}{N(1 - B_{WO})}$$

Equation 16-4

$$C = \frac{\sum_{i=1}^N S_i}{N}$$

Method 16A

Equation 16A-1

$$V_{m(std)} = V_m Y \frac{T_{std}}{T_m} \frac{P_{bar}}{P_{std}} = K_1 Y \frac{V_m - P_{bar}}{T_m}$$

Equation 16A-2

$$C_{TRS(ppm)} = \frac{K_2 (V_t - V_{tb}) N \left(\frac{V_{soln}}{V_a} \right)}{V_{m(std)}}$$

Equation 16A-3

$$C_{RG} = \frac{(Q_{H2S})(C_{H2S})}{Q_{H2S} + Q_{CG}}$$

Equation 16A-4

$$R = \frac{C_{TRS}}{C_{RG}} \times 100$$

Equation 16A-5

$$N_T = \frac{1}{ml \text{ Na}_2 \text{ S}_2 \text{ O}_3 \text{ Consumed}}$$

Equation 16A-6

$$N_1 = \frac{N_T V_T}{V_1}$$

Equation 16A-7

$$V_{m(std)} = (\bar{Q}_{std})(\theta_s)(1 - B_{wa}) \frac{M_a}{M_b}$$

Equation 16A-8

$$C_{H2S} = \frac{K N_T (V_{TB} - V_T)}{V_{m(std)}}$$

Equation 18-3

$$C_s = \frac{G_V \times 10^6 \frac{293}{T_s} \frac{P_s}{760}}{V_m Y \frac{293}{T_m} \frac{P_m}{760} 1000}$$

$$= \frac{G_V \times 10^3 \frac{P_s}{T_s} \frac{T_m}{P_m}}{V_m Y}$$

Method 16B

Equation 16B-1

$$C_{TRS} = (C_{SO2})(d)$$

Equation 16B-2

$$C_{TRS} = \frac{\sum_{i=1}^n C_{TRS}}{N}$$

Method 18

Equation 18-1

$$C_s = \frac{10^6 (\bar{x} q_c)}{q_c + q_d}$$

Equation 18-2

$$C_s = 106 \bar{x} \left(\frac{q_{c1}}{q_{c1} + q_{d1}} \right) \left(\frac{q_{c2}}{q_{c2} + q_{d2}} \right)$$

Equation 18-4

$$C_s = \frac{\frac{L_V}{M} \rho (24.055 \times 10^6)}{V_m Y \frac{293}{T_m} \frac{P_m}{760} 1000} = 6.24 \times 10^4 \frac{L_V \rho T_m}{M V_m Y P_m}$$

Equation 18-5

$$C_c = \frac{C_s P_r T_i F_r}{P_i T_r (1 - B_{ws})}$$

Method 20

Equation 20-1

$$C_d = \frac{C_w}{1 - B_{ws}}$$

Equation 20-2

$$F_o = \frac{0.209 F_d}{F_c}$$

Equation 20-3

$$X_{CO2} = \frac{5.9}{F_o}$$

Equation 20-4

$$C_{adj} = C_d \frac{5.9}{20.0 - \%O_2}$$

Equation 20-5

$$C_{adj} = C_d \frac{X_{CO_2}}{\%CO_2}$$

Equation 20-6

$$E = C_d F_d \frac{20.9}{20.9 - \%O_2}$$

Equation 20-7

$$E = C_d F_c \frac{100}{\%CO_2}$$

Equation 20-8

$$E = C_w F_c \frac{100}{\%CO_{2w}}$$

Method 23

Equation 23-1

$$RRF_i = \frac{1}{n} \sum_{j=1}^n \frac{A_{cij} m_{ci}^*}{A_{cij}^* m_{ci}}$$

Equation 23-2

$$C_i = \frac{m_i^* A_i}{A_i^* RRF_i V_{mstd}}$$

Equation 23-3

$$RRF_{rs} = \frac{A_{ci}^* m_{rs}}{A_{rs} m_{ci}^*}$$

Equation 23-4

$$R^* = \frac{A_i^* m_{rs}}{A_{rs} RRF_{rs} m_i^*} \times 100\%$$

Equation 23-5

$$RRF_s = \frac{A_{ci}^* m_s}{A_{cis} m_{ci}^*}$$

Equation 23-6

$$R_s = \frac{A_s m_i^*}{A_i^* RRF_s m_s} \times 100\%$$

Equation 23-7

$$MDL = \frac{2.5 A_{ai} m_i^*}{A_{ci}^* RRF_i}$$

Equation 23-8

$$C_{Tr} = \sum_{i=1}^n C_i$$

Method 25

Equation 25-1

$$\Delta P = 0.01 \frac{FP_b \theta}{V_t}$$

Equation 25-2

$$V_s = 0.3857V \left[\frac{P_t}{T_t} - \frac{P_{ii}}{T_{ii}} \right]$$

Equation 25-3

$$C_t = \left[\frac{\frac{P_f}{T_f}}{\frac{P_t}{T_t} - \frac{P_{ii}}{T_{ii}}} \right] \left[\frac{1}{r} \sum_{j=1}^r C_{tm_j} \right]$$

Equation 25-4

$$C_c = 0.3857 \frac{V_v P_f}{V_s T_f} \left[\frac{1}{q} \sum_{k=1}^q C_{cm_k} \right]$$

Equation 25-5

$$C = C_t + C_c$$

Equation 25-6

$$m_c = 0.4993C$$

Equation 25-7

$$\text{Percent recovery} = 1.604 \frac{M V_v P_t C_{cm}}{L P T_f N}$$

Equation 25-8

$$RSD = \frac{100}{\bar{x}} \sqrt{\frac{\sum (x_i - \bar{x})^2}{n-1}}$$

Method 25A

Equation 25A-1

$$C_c = K C_{meas}$$

Method 26

Equation 26-1

$$\mu\text{g} \frac{\text{Cl}^-}{\text{ml}} = \text{g of NaCl} \times 10^3 \times \frac{35.453}{58.44}$$

Equation 26-2

$$\mu\text{g} \frac{\text{Br}^-}{\text{ml}} = \text{g of NaBr} \times 10^3 \times \frac{79.904}{102.90}$$

Equation 26-3

$$\mu\text{g} \frac{\text{F}^-}{\text{ml}} = \text{g of NaF} \times 10^3 \times \frac{18.998}{41.99}$$

Equation 26-4

$$m_{HX} = K V_s (S_X^- - B_X^-)$$

Equation 26-5

$$m_{X2} = V_s (S_X^- - B_X^-)$$

Equation 26-6

$$C = K \frac{m_{HX.X2}}{V_{m(std)}}$$

Method 26A

Equation 26A-1

$$\mu\text{g} \frac{\text{Cl}^-}{\text{ml}} = \text{g of NaCl} \times 10^3 \times \frac{35.453}{58.44}$$

Equation 26A-2

$$\mu\text{g} \frac{\text{Br}^-}{\text{ml}} = \text{g of NaBr} \times 10^3 \times \frac{79.904}{102.90}$$

Equation 26A-3

$$\mu\text{g} \frac{\text{F}^-}{\text{ml}} = \text{g of NaF} \times 10^3 \times \frac{18.998}{41.99}$$

Equation 26A-4

$$m_{HX} = K V_s (S_{X-} - B_{X-})$$

Equation 26A-5

$$m_{X2} = V_s (S_{X-} - B_{X-})$$

Equation 26A-6

$$C = K \frac{m_{HX, X2}}{V_{m(std)}}$$

Method 29

Equation 29-1

$$M_{fh} = C_{a1} F_d V_{so\ln,1}$$

Equation 29-2

$$M_{bh} = C_{a2} F_a V_a -$$

Equation 29-3

$$M_t = (M_{fh} - M_{fbb}) + (M_{bh} - M_{bbb})$$

Equation 29-4

$$Hg_{fh} = \frac{Q_{fh}}{V_{f1B}} (V_{so\ln,1})$$

Equation 29-5

$$Hg_{bh2} = \frac{Q_{bh2}}{V_{f2B}} (V_{so\ln,2})$$

Equation 29-6

$$Hg_{bh3(A,B,C)} = \frac{Q_{bh3(A,B,C)}}{V_{f3(A,B,C)}} (V_{so\ln,3(A,B,C)})$$

Equation 29-7

$$Hg_{bh} = Hg_{bh2} + Hg_{bh3A} + Hg_{bh3B} + Hg_{bh3C}$$

Equation 29-8

$$Hg_t = (Hg_{fh} - Hg_{fbb}) + (Hg_{bh} - Hg_{bbb})$$

Equation 29-9

$$C_s = \frac{K_4 M_t}{V_{m(std)}}$$

Method 0010

$$m_{Cl_2} = S \times V_2$$

$$\frac{\text{Max POHC}_i \text{ Mass}}{DV_{\text{eff}(std)}} = \text{Max POHC}_i \text{ conc}$$

$$C = K \times \frac{m}{V_{m(std)}}$$

$$\frac{LDL_{POHC} \times 10}{POHC_{i \text{ conc}}} = V_{TBC}$$

Method 0060

$$C_{POHC} (\mu\text{g/ml}) \times \text{sample volume (ml)} \\ = \text{amount } (\mu\text{g}) \text{ of POHC in sample}$$

$$M_{fh} = C_{a1} F_d V_{soln,1}$$

$$M_{bh} = C_{a2} F_a V_a$$

Method 0030

$$C_g = \frac{\text{Wt. of Analytes in Sample, } \mu\text{g (i.e. VOST tubes \& condensate)}}{\text{Volume of sample at standard conditions}}$$

$$M_t = (M_{fh} - M_{fhh}) + (M_{bh} - M_{bhh})$$

Method 0050

Equation 0050-4

$$m_{HCl} = S \times V_s \times \frac{36.46}{35.45}$$

Method 0061

$$m = (S - B) \times V_{ls} \times d$$

Equation 0050-5

$$m_{Cl_2} = S \times V_2$$

Equation 0050-6

$$C = K \times \frac{m}{V_{m(std)}}$$

Method 0051

$$m_{HCl} = S \times V_s \times \frac{36.46}{35.45}$$

Appendix B

Abbreviations and Terminology

%CO	Percent CO by volume, dry basis
%CO ₂	Percent CO ₂ by volume, dry basis
%N ₂	Percent N ₂ by volume, dry basis
%O ₂	Percent O ₂ by volume, dry basis
0.280	Molecular weight of N ₂ or CO, divided by 100
0.320	Molecular weight of O ₂ divided by 100
0.440	Molecular weight of CO ₂ divided by 100
100	Conversion to percent.
13.6	Specific gravity of mercury.
18.0	Molecular weight of water, g/g mole (lb/lb mole).
3600	Conversion Factor, sec/hr.
60	Sec/min.
A	Cross sectional area of stack, m ² (ft ²).
A	Absorbance of sample.
A	Analytical detection limit, µg/ml.
A1	Absorbance of the 100µg NO ₂ standard.
A2	Absorbance of the 200µg NO ₂ standard.
A3	Absorbance of the 300µg NO ₂ standard.
A4	Absorbance of the 400µg NO ₂ standard.
A _n	Cross-sectional area of nozzle, m ² (ft ²).
B	Liquid volume of digested sample prior to aliquotting for analysis, ml.
B _{ws}	Water vapor in the gas stream [from Method 4 (reference method) or Method 5], proportion by volume.
C	Corrected to standard conditions, mg/dsm ³ (lb/dscf).
C	TGNMO concentration of the effluent, ppm C equivalent.
Ca	Acetone blank residue concentration, mg/mg.

Ca	Actual concentration of SO ₂ in audit sample, mg/dscm.
Ca	Actual audit sample concentration, mg/dscm.
Ca1	Concentration of metal in Analytical Fraction 1A as read from the standard curve, µg/ml.
Ca2	Concentration of metal in Analytical Fraction 2A as read from the standard curve, (µg/ml).
Cc	Calculated condensible organic (condensate trap) concentration of the effluent, ppm C equivalent.
Ccm	Measured concentration (NMO analyzer) for the condensate trap ICV, ppm CO ₂ .
Cd	Determined concentration of SO ₂ in audit sample, mg/dscm.
Cd	Determined audit sample concentration, mg/dscm.
CH ₂ S	Concentration of H ₂ S at standard conditions, mg/dscm.
Cp	Pitot tube coefficient, dimensionless.
Cp(s)	Type S pitot tube coefficient, dimensionless.
Cp(std)	Standard pitot tube coefficient; use 0.99 if the coefficient is unknown and the tube is designed according to the criteria of Sections 6.7.1 to 6.7.5 of this method.
cs	Concentration of particulate matter in stack gas, dry basis, corrected to standard conditions, g/dscm (gr/dscf).
Cs	Concentration of a metal in the stack gas, mg/dscm.
CSO ₂	Concentration of SO ₂ , dry basis, corrected to standard conditions, mg/dscm (lb/dscf).
Ct	Calculated noncondensable organic concentration (sample tank) of the effluent, ppm C equivalent.
Ctm	Measured concentration (NMO analyzer) for the sample tank, ppm NMO.
D	Diameter of stack, in.
De	Equivalent diameter.
F	Dilution factor. Required only if sample dilution was needed to reduce the absorbance into the range of the calibration).
F	Sampling flow rate, cc/min.
Fa	Aliquot factor, volume of Sample Fraction 2 divided by volume of Sample Fraction 2A (see Section 8.3.4.)

Fd	Dilution factor (Fd = the inverse of the fractional portion of the concentrated sample in the solution actually used in the instrument to produce the reading Ca1. For example, if a 2 mL aliquot of Analytical Fraction 1A is diluted to 10 mL to place it in the calibration range, Fd = 5).
ΔH	Average pressure differential across the orifice meter, mm H2O (in. H2O).
Hgbh	Total mass of Hg collected in the back
Hgbh2	Total mass of Hg collected in Sample Fraction 2, μg .
Hgbh3(A,B,C)	Total mass of Hg collected separately in Fraction 3A, 3B, or 3C, μg .
Hgbhb	Blank correction value for mass of Hg detected in back
Hgfh	Total mass of Hg collected in the front
Hgfhb	Blank correction value for mass of Hg detected in front
Hgt	Total mass of Hg collected in the sampling train, μg .
I	Percent of isokinetic sampling.
K	0.127 mm H2O (metric units); 0.005 in. H2O (English units).
K4	103 mg/ μg .
Kc	Spectrophotometer calibration factor.
Kp	Velocity equation constant.
L	Length.
L	Volume of liquid injected, μl .
L1	Individual leakage rate observed during the leak-check conducted prior to the first component change, m3/min (ft3/min)
La	pretest leak-check or for a leak-check following a component change; equal to 0.00057 m3/min (0.020 cfm) or 4 percent of the average sampling rate, whichever is less.
Li	Individual leakage rate observed during the leak-check conducted prior to the "ith" component change (i = 1, 2, 3...n), m3/min cfm).
Lp	Leakage rate observed during the post-test leak-check, m3/min (cfm).
m	Mass of NOx as NO2 in gas sample, μg .
M	Molecular weight of the liquid injected, g/g mole.
ma	Mass of residue of acetone after evaporation,mg.

Mbh	Total mass of each metal (except Hg) collected in the back
Mbhb	Blank correction value for mass of metal detected in back
Mc	TGNMO mass concentration of the effluent, mg C/dsm ³ .
Md	Dry molecular weight, g/g mole (lb/lb mole)
Md	Molecular weight of stack gas, dry basis (see Section 8.6), g/g mole (lb/lb mole).
Mfh	Total mass of each metal (except Hg) collected in the front half of the sampling train (Sample Fraction 1), µg.
Mfhb	Blank correction value for mass of metal detected in front
mn	Total amount of particulate matter collected, mg.
Ms	Molec Molecular weight of stack gas, wet basis, g/g mole (lb/lb mole).
Mt	Total mass of each metal (separately stated for each metal) collected in the sampling train, µg.
Mw	Molecular weight of water, 18.0 g/g-mole (18.0 lb/lb-mole).
N	Total number of traverse points.
n	total number of traverse points.
N	Normality of barium standard titrant, meq/ml.
N	Carbon number of the liquid compound injected (N = 12 for decane, N = 6 for hexane).
NA	Normality of standard C ₆ H ₅ AsO solution, g eq/liter.
NI	Normality of standard I ₂ solution, g eq/liter.
NS	Normality of standard (0.1 N) Na ₂ S ₂ O ₃ solution, g eq/liter.
NT	Normality of standard (0.01 N) Na ₂ S ₂ O ₃ solution, assumed to be 0.1 NS, g eq/liter.
p	Velocity head of stack gas, mm H ₂ O (in. H ₂ O).
P	Allowable pressure change, cm Hg.
Pb	Barometric pressure, cm Hg.
Pbar	Barometric pressure at measurement site, mm Hg (in. Hg).
Pf	Final pressure of the intermediate collection vessel, mm Hg absolute.
Pg	Stack static pressure, mm Hg (in. Hg).

pi	Individual velocity head reading at traverse point "i", mm (in.) H ₂ O.
Pi	pitch angle at traverse point i, degree.
PI	Initial absolute pressure of flask, mm Hg (in. Hg).
Ps	Absolute stack pressure (P _{bar} + P _g), mm Hg (in. Hg),
Δps	Velocity head measured by the Type S pitot tube, cm (in.) H ₂ O.
Pstd	Standard absolute pressure, 760 mm Hg (29.92 in. Hg).
Δpstd	Velocity head measured by the standard pitot tube, cm (in.) H ₂ O.
Pt	Gas sample tank pressure after sampling, but before pressurizing, mm Hg absolute.
Ptf	Final gas sample tank pressure after pressurizing, mm Hg absolute.
Pti	Gas sample tank pressure before sampling, mm Hg absolute.
q	Total number of analyzer injections of intermediate collection vessel during analysis (where k = injection number, 1 ... q).
Qbh2	Quantity of Hg, μg, TOTAL in the ALIQUOT of Analytical Fraction 2B selected for digestion and analysis . NOTE: For example, if a 10 mL aliquot of Analytical Fraction 2B is taken and digested and analyzed (according to Section 11.1.3 and its NOTES Nos. 1 and 2), then calculate and use the total amount of Hg in the 10 mL aliquot for Qbh2.
Qbh3(A,B,C)	Quantity of Hg, μg, TOTAL, separately, in the ALIQUOT of Analytical Fraction 3A, 3B, or 3C selected for digestion and analysis (see NOTES in Sections 12.7.1 and 12.7.2 describing the quantity "Q" and calculate similarly).
Qfh	Quantity of Hg, μg, TOTAL in the ALIQUOT of Analytical Fraction 1B selected for digestion and analysis . NOTE: For example, if a 10 mL aliquot of Analytical Fraction 1B is taken and digested and analyzed (according to Section 11.1.3 and its NOTES Nos. 1 and 2), then calculate and use the total amount of Hg in the 10 mL aliquot for Qfh.
Qsd	Dry volumetric stack gas flow rate corrected to standard conditions, dscm/hr (dscf/hr).
R	$[(\text{mm Hg})(\text{m}^3)]/[(\text{K})(\text{g-mole})] \{21.85 [(\text{in. Hg})(\text{ft}^3)]/[(\text{R})(\text{lb-mole})]\}$.
r	Total number of analyzer injections of sample tank during analysis (where j = injection number, 1 ... r).
Ravg	average resultant angle, degree.
RE	Relative error for QA audit samples, percent.

Ri	resultant angle at traverse point i, degree.
Sd	standard deviation, degree.
T	Sensitivity factor for differential pressure gauges.
Tf	Final absolute temperature of flask, K (R).
Tf	Final temperature of intermediate collection vessel, K.
Ti	Initial absolute temperature of flask, K (R).
Tm	Absolute DGM temperature
Ts	Stack temperature, C (F).
Ts(abs)	Absolute stack temperature, K (R).=273 + Ts for metric units, =460 + Ts for English units.
Tstd	Standard absolute temperature, 293 K (528 R).
Tt	Sample tank temperature at completion of sampling, K.
Ttf	Sample tank temperature after pressurizing, K.
Tti	Sample tank temperature before sampling, K.
V	Sample tank volume, m3.
Va	Volume of acetone blank, ml.
Va	Volume of sample aliquot titrated, ml.
Va	Volume of absorbing solution, 25 mL.
VA	Volume of C6H5AsO solution used for standardization, ml.
Va	Total volume of digested sample solution (Analytical Fraction 2A), mL
VAI	Volume of standard C6H5AsO solution used for titration analysis, mL.
Vaw	Volume of acetone used in wash, ml.
Vf	Volume of flask and valve, ml.
Vf1B	Volume of aliquot of Analytical Fraction 1B analyzed, ml. NOTE: For example, if a 1 mL aliquot of Analytical Fraction 1B was diluted to 50 mL with 0.15 percent HNO3 as described in Section 11.1.3 to bring it into the proper analytical range, and then 1 mL of that 50
Vf2B	Volume of Analytical Fraction 2B analyzed, ml. NOTE: For example, if 1 mL of Analytical Fraction 2B was diluted to 10 mL with 0.15 percent HNO3 as described in Section 11.1.3 to bring it into the proper analytical range, and then 5 mL of that 10 mL was analyzed, Vf2B would be 0.5 ml.

Vf3(A,B,C)	Volume, separately, of Analytical Fraction 3A, 3B, or 3C analyzed, mL (see previous notes in Sections 12.7.1 and 12.7.2, describing the quantity "V" and calculate similarly).
VI	Volume of standard I2 solution used for standardization, ml.
VIT	Volume of standard I2 solution used for titration analysis, normally 50 ml.
Vlc	Total volume of liquid collected in impingers and silica gel (see Figure 5-6), ml.
Vm	Volume of gas sample as measured by dry gas meter, dcm (dcf).
Vm	Dry gas volume as measured by the DGM, dcm (dcf).
Vm	Volume of gas sample at meter conditions, liters.
Vm(std)	Volume of gas sample measured by the dry gas meter, corrected to standard conditions, dscm(dscf).
vs	Stack gas velocity, calculated by Method 2,
vs	Average stack gas velocity, m/sec (ft/sec).
Vs	Gas volume sampled, dsm ³ .
Vsc	Sample volume(dry basis), ml. at standard conditions
VSI	Volume of 0.1 N Na ₂ S ₂ O ₃ solution used for standardization, ml.
Vsoln	Total volume of solution in which the SO ₂ sample is contained, 100 ml.
Vsoln,1	Total volume of digested sample solution (Analytical Fraction 1), ml.
Vsoln,2	Total volume of Sample Fraction 2, ml.
Vsoln,3(A,B,C)	Total volume, separately, of Analytical Fraction 3A, 3B, or 3C, ml.
Vt	Volume of barium standard titrant used for the sample (average of replicate titration), ml.
VT	Volume of standard (0.01 N) Na ₂ S ₂ O ₃ solution used in standardizing iodine solution (see Section 10.2.1), ml.
Vt	Sample train volume, cc.
Vtb	Volume of barium standard titrant used for the blank, ml.
VTT	Volume of standard (0.01 N) Na ₂ S ₂ O ₃ solution used for titration analysis, ml.
Vv	Intermediate collection vessel volume, m ³ .

Vw(std)	Volume of gas,corrected to standard conditions, scm (scf).
W	width.
W	Weight of K ₂ Cr ₂ O ₇ used to standardize Na ₂ S ₂ O ₃ or C ₆ H ₅ AsO solutions, as applicable (see Sections 10.2.2 and 10.2.3), g.
Wa	Weight of residue in acetone wash, mg.
xi	Individual measurements.
Y	Dry gas meter calibration factor.
Yi	yaw angle at traverse point i, degree.

STACK TESTING FOR HAZARDOUS AIR POLLUTANTS (HAPs)

FIELD OBSERVATION CHECKLIST

PRETEST CHECKLIST

PRETEST MEETING CHECKLIST

Plant Name _____ Date _____

Plant Address _____

Source to be Tested _____

Plant Representative _____ Phone _____

Plant Manager _____ Phone _____

Test Team Company Name/Address _____

Team Representative _____ Phone _____

Regulatory Agency(s)/Address _____

Agency _____

Representative _____ Phone _____

I. FACILITY OPERATION

A. Process

- a. Process Description _____
- b. Method of process weight rate or fuel feed determination _____
- c. Process parameters (list below or use operation data sheet for process)

Process Parameters to be Monitored	Design/Normal Values	Acceptable Limits

See operations data sheet for _____

- d. Acceptable values of raw material and/or fuel composition _____
- e. Normal operating cycle and its effect on test plan _____
- f. Test Operating Cycle _____

B. Control Devices

- a. Control Device(s) _____
- b. Operating Parameters (list below or use operating data sheet for process)

n. Identify Feed Streams:

No. 1: _____ No. 2: _____ No. 3: _____ No. 4: _____

II. TRIAL BURN PARAMETERS

Parameter	Item	Test Run #1	Test Run #2	Test Run #3
Feed Rate				
Heat Content of Feed				
Auxiliary Fuel Rates	Gas Oil			
Feed Composition	Water, % Ash, %			
POHC (s), % or ppm				
Other Organics				
Sulfur Content, % or ppm				
Nitrogen Cont., % or ppm				
Phosphorus Cont., % or ppm				
Metals, % or ppm				
Liquid	Viscosity Specific Gravity			
Sludges	Moisture, %			
Gases	Density			

III. INCINERATOR PARAMETERS

Parameter	Item	Test Run #1	Test Run #2	Test Run #3
Temperature	Primary Combustion Chamber			
	Secondary Combustion Chamber			
	Heat Recovery Device Inlet			
	Heat Recovery Device Outlet			
	Air Pollution Control Device Inlet			
	Air Pollution Control Device Outlet			
	Stack Gases			
Draft (in. Water)	Primary Combustion Chamber			
	Secondary Combustion Chamber			
	Heat Recovery Inlet			
	Heat Recovery Outlet			
-ESP	Unit Voltage, Kv			
	Unit Current, ma			

-Baghouse	Pressure Drop, inches			
	Air Flow Rate, acfm			
-Dry Scrubber	Lime Feed Rate, gpm			
	Air Flow Rate, acfm			
Air Flow Rates	Primary, acfm			
	Secondary, acfm			
Stack Gas Monitoring				
	Oxygen, %			
	CO ₂ , %			
	CO, ppm			

Parameter	Item	Test Run #1	Test Run #2	Test Run #3
	NO _x , ppm			
	HC, ppm			
	Opacity, %			
	SO ₂ , ppm			
	Other, ppm/%			
Feed Sample Taken?	Feed No.			
	Feed No.			
	Feed No.			
Auxiliary Fuel Samples Taken?				
Scrubber pH				
Type of Alkaline Added?				
Rate of Addition, lb/hr				
Ash Removal Rate, lb/hr				
Automatic Feed Cutoffs	Low Temperature, F			
	High CO, ppm			
	Low Oxygen, %			

IV. TESTING METHODOLOGY TO BE DISCUSSED

Sampling Location _____

Sampling Points _____

Cyclonic Flow Check _____

Pollutant Measurement Methods _____

Mandatory Calibrations Sheets to be Provided (If not provided immediately prior to starting test program, testing should be delayed until observer receives them or authorization to proceed by Task Manager)

Orifice _____ Dry Gas Meter (Y) _____ Nozzle _____ Pitot Tube _____
Delta H@ _____ Thermocouples _____ Barometer _____
Fuel, Raw Material/produce Sampling _____

Emission Rate Calculation _____

Destruction Removal Efficiency Calculation _____

Minimum Sample Volume Calculation _____

Sample Recovery Area _____

Dry Molecular Weight/dilution Determination _____
Integrated Bag _____ Grab Sample _____ Orsat _____
Fyrite _____ CEM _____

Stack or Duct Condition _____

Safety Requirements _____

Test Reporting Requirements _____

Test Schedule _____

Plant Tour _____

V. SAMPLING METHODOLOGIES TO BE USED

Pollutant Parameter	Methodology	Remarks, Additional Quality Assurance, and/or Modifications

VI. GENERAL COMMENTS

 Signature of Agency Representative Affiliation of Representative Date

 Signature of Plant Representative Date

STACK TESTING FOR HAZARDOUS AIR POLLUTANTS (HAPs)

FIELD OBSERVATION CHECKLIST

FEDERAL REFERENCE METHODS 1, 2, 3, 4, AND 5

SOURCE INFORMATION

Facility _____ Location _____
Source _____ Sampling Location _____
Test Team _____ Team Leader _____
Test Date Interval: ____ / ____ / ____ to ____ / ____ / ____
Observer _____ Affiliation _____

Sketch of sample location (if different from pretest report).

(Note: Include: duct dimensions, location of disturbances and ports, X-section showing sampling points.)

Stack or duct condition:

Material of construction: Brick _____ Concrete _____ Stainless Steel _____
Carbon steel _____ Other _____

External corrosion visible? _____

Leaks? _____

Internal corrosion? _____

Cake or pile up of particulate matter in sampling port area? _____

Remove before test? _____

Visible water droplets in stack? _____

Particulate matter appears acidic? _____

Insulation Thickness Stack under negative or positive pressure? _____

All ports accessible? _____

Nipple:

ID _____ Length _____ Capped _____

Flush with inside wall _____ Corrosion evident _____

Port dimensions:

Round diameter _____

Rectangular: L _____ W _____

Equivalent diameter _____

Facility for movement of sampling train:

Monorail _____ Skateboard and rail _____

Other _____

Electricity available? _____ No. of receptors _____

Distance to receptors _____

Ground fault circuit protection? _____ Adequate? _____

Facility for getting equipment to port _____

FEDERAL REFERENCE METHOD 1

Stack diameter properly determined? _____ Method used _____

[Note: If stack is between 4"-12" then Method 1a must be employed]

Stack Diameter? _____ Measured on site? _____

Port distance from upstream disturbance (A) _____ Upstream Diameters(A) _____

Port distance from downstream disturbance (B) _____ Downstream Diameters (B) _____

Number of Sampling Points Required? _____ (Draw a line vertically from the

ADistance A \cong axis down to the step chart, and from ADistance B \cong axis up to the step chart.

The maximum number of points marked on the chart yields the minimum number of points to

be sampled.) Actual sampling points _____

Ports and points at sampling location: _____

Circular: No. of Ports _____ Points/port _____

Rectangular: Grid configuration _____

No. of Ports _____ Points/port _____

Points located centroid of area? _____

First and last traverse points from ducts: _____

24 in. Diameter: 1.0 inch from wall? Yes _____ No _____

12-24 in. Diameter: > 0.5 inch from wall? Yes _____ No _____

Method 1 calculated correctly? _____

Points properly marked on pitot tube? _____

Port length/interior stack accounted for in calculation? _____

Cyclonic flow check verified (<20 degrees) and completed? _____

Average null point angle _____

Describe velocity profile of sampling plane _____

Draw picture of stack with port location and upstream/downstream disturbances marked and notated _____

Remarks _____

FEDERAL REFERENCE METHOD 2

Equipment identified matches pretest calibration list? _____

Pitot tube: Type S _____ Standard _____

Pitot tube coefficient _____ Calibration date _____

Design to EPA specifications? _____

Properly attached to sampling probe, if applicable? _____

Pitot tube, lines, and manometer assembled correctly? _____

Differential pressure gauge has correct sensitivity? _____

Measurement gauge: Manometer _____ Magnehelic _____ Range(s) _____

Gauge proper sensitivity for delta p range being measures? _____

Magnehelic gauges calibrated against manometer? _____ Date _____
Differential pressure gauge leveled and zeroed? _____
Visual check of pitot tube heads-good condition and meet EPA design specification? _____

Pitot tube pretest leak check? _____
Cyclonic flow checked prior to test series? _____
Static pressure measured during the test series? _____
Orientation of pitot tube correct during traverse? _____
Sampling port adequately sealed around pitot tube? _____
Process operating at correct conditions? _____
Stable readings taken at each traverse point? _____
Barometric pressure recorded and adjusted for elevation? _____

(Note: Barometric pressure must be adjusted minus 0.1 A per 100 ft elevation increase or visa versa for elevation decrease. Elevation at which barometric pressure is measured compared to elevation at the test platform.)

Airport: _____ Location _____ P_b at airport _____
Sampling location elevation _____ Change in ft. _____
delta in. Hg +/- _____ (0.10 in. Hg/100 ft) _____

Stack temperature measured? _____
Stack gas moisture content determined? _____
Orsat samples taken? _____ If not, explain why _____
Posttest leak check performed? _____ Leak rate results _____
Data recorded properly _____ Calculations correct? _____
Remarks _____

FEDERAL REFERENCE METHOD 3

Method used to acquire gas sample
Single point grab _____ Single point integrated _____ Multi-point integrated _____
Type of bag _____ Type of probe _____
Pump: One way squeeze bulb _____ Diaphragm _____ Other _____
Instrument used to determine concentration? Orsat _____ Fyrite _____ CEM _____
Is a filter used to remove particulate matter? _____
Sample train leak checked? _____
Sample train purged with stack gas prior to collecting the sample? _____
Sampling port properly sealed? _____ Sampling rate constant? _____
What rate? _____
Orsat performed in triplicate? _____ Analysis consistent? _____
All readings averaged and reported to nearest 0.1%? _____
Orsat:
Reagents at the proper level? _____ Analyzer level? _____

Reagents fresh? _____ Date charged to orsat? _____
 Analyzer leak checked? _____ Analyzed within 8 hours? _____
 Sample line purged? _____ Complete absorption of gases? _____
 Analysis repeated until following analysis criteria met?

- CO₂-any three analyses differ by
 A) $\leq 0.3\%$ when CO₂ $\geq 4.0\%$
 B) $\leq 0.2\%$ when CO₂ $< 4.0\%$
 O₂- any three analyses differ by
 A) $\leq 0.3\%$ when O₂ $\geq 15.0\%$
 B) $\leq 0.2\%$ when O₂ $< 15\%$

Method 3A for O₂/CO₂:

System leak checked during test day? _____
 Sample conditioner/dryer used? _____
 Heated sample lines used prior to sample conditioner? _____
 Moisture visible in sample lines? _____
 Filter used: In-stack or out-of-stack? _____
 Sample system flow rates within 10 % of calibration gas flow rates? _____
 All calibration gases measured below the span value of the analyzer? _____
 Data recording performed by strip chart, digital, manual or other? _____
 Fuel _____ F_o Range _____
 Calculate F_o: $F_o = (20.9 - \%O_2 / \%CO_2) =$ _____

Expected F_o Range:

Anthracite/Lignite	1.015-1.130	Natural Gas	1.600-1.836
Bituminous	1.083-1.230	Wood Bark	1.000-1.120
Distillate Oil	1.260-1.413	Municipal Gargage	1.043-1.177
Residual Oil	1.210-1.370	Other	_____

Remarks _____

FEDERAL REFERENCE METHOD 4

Method used to acquire moisture concentration in stack? _____
 Impinger method _____ Partial pressure method _____ Wet bulb/dry Bulb _____
 Nomograph _____ Other? _____
 Estimate of moisture: _____
 Method conducted in conjunction with pollutant emission test? _____
 Preliminary run: Conducted _____ Estimated _____
 Impinger method:
 Impingers properly placed? _____
 Impinger contents: 1st _____ 2nd _____ 3rd _____ 4th _____ 5th _____

Modifications? _____
 Cooling system: Crushed ice? _____ Other _____
 Sampling time per point _____ Number of points _____
 Probe heater on? _____ Temperature _____ Stable _____
 Crushed ice in ice bath around impingers? _____
 Pretest leaks check conducted? _____ Leak rate _____
 Sampling rate constant _____ Isokinetic? _____
 All data recorded properly? _____
 Post leak check conducted? _____ Leakage rate? _____
 Recovery-Impinger Contents
 Method: Volumetric? _____ Gravimetric? _____
 Trip balance calibrated (gravimetric)? _____
 Measurement of silica gel: Balance? _____ Other _____
 Color of silica gel? _____ Condition? _____
 (Note: Blue-good; pink-spent (unable to absorb water))
 All analytical data recorded properly? _____
 All readings averaged and reported to nearest 0.1 %? _____
 Moisture value consistent with historical process values and quick-check value? _____

Remarks _____

FEDERAL REFERENCE METHOD 5

Equipment identified matches pretest calibration list? _____

Appratus:

NOZZLE

Probe nozzle: Stainless steel _____ Glass _____ Other _____
 Probe design: Button-hook _____ Elbow _____ Straight tube _____
 Nozzle size: _____ Calculated _____ Measured three times _____
 Nozzle condition: Nicks _____ Bent opening _____

PROBE/PROBE LINER

Probe material: Borosilicate _____ Quartz _____
 Stainless steel _____ Inconel _____ Other _____
 Heating system: _____ Temperature _____
 Water cooled? _____ Length _____

PITOT TUBE

Type: Type S _____ Standard _____ Other _____

Properly attached to probe? _____
Meets EPA guidelines for attachment to probe? _____ Yes _____ No _____
Modifications _____
Pitot tube coefficient _____

FILTER HOLDER

Construction: Borosilicate _____ Quartz _____ Other _____
Frit construction: Glass frit? _____ Teflon frit? _____
Filter support _____ Silicone gasket _____
Other _____ Clean? _____
Cyclone and flask used (optional): _____

FILTER HEATING SYSTEM

Description _____
Temperature _____
Gas temperature at exit of box _____

CONDENSER

Number of impingers? _____ Clean? _____
Impinger contents: 1st _____ 2nd _____ 3rd _____
4th _____ 5th _____ 6th _____ 7th _____
Cooling system? _____
Proper connections? _____
Modifications? _____

METER BOX ASSEMBLY

Manufacturer _____ Model No. _____ Unit No. _____
Delta H@ value _____ Date _____
Dry gas meter AY≅ value _____ Date _____
Condition of sampling equipment? _____
Meter box leveled? _____ Timer/temperature readouts operational? _____
Manometer leveled? _____
Thermocouples calibrated? _____

Operation

Filter properly numbered and recorded on field test data sheet? _____
Filter checked visually for irregularities? _____
Filter media type? _____
Sampling site selected? _____ Stack temperature _____
Stack velocity head? _____ Stack dry molecular weight calculated? _____
Nozzle size properly selected? _____
Calibration data certified available: _____
Pitot tubes _____ Meter box Y factor _____
Meter box delta H@ Magnehelic gauge _____ Thermocouples _____
Probe nozzle _____

Note: If calibration sheets are not available, contact Task Manager for authorization to proceed. Authorization to proceed: Yes _____ No _____
 Date _____ Time _____ Task Manager _____

Selection of sampling points and time according to methodology? _____
 Number of sample points _____ Sample time/point _____
 Minimum sample volume required? _____
 Minimum run time required? _____
 Total sample time actual? _____

Sampling train assembled properly? _____
 Filter properly centered? _____ Impingers properly charged? _____
 Number of impingers required? _____ Which FRM? _____
 Volume in each impinger: 1st _____ 2nd _____ 3rd _____ 4th _____
 5th _____ 6th _____ 7th _____ 8th _____
 Drying medium: Silica gel? _____ Drierite _____
 Weight _____

Pitot tube lines checked for plugging or leaks? _____
 Stopcock grease used: Acetone-insoluble? _____ Heat-stable? _____
 Meter box leveled? _____ Manometers leveled? _____
 Probe heat uniform along length of probe? _____
 Dry gas meter field observation check: _____

A. Operating Meter Box Assembly Operate metering system for 10 minutes at the H@ value. Record the volume metered, DGM temperature and barometric pressure.

Final: DGM reading _____ ft³ Temp _____ °F P_b _____ in. Hg
 Initial: DGM reading _____ ft³ Temp _____ °F P_b _____ in. Hg
 Net: V_m = _____ ft³ Avg. T_m = _____ °F + 460 = _____

$$Y_c = 10/V_m [0.0319 (T_m = 460)/P_b]$$

$$Y_c = 0.97 Y = 1.03 Y = _____$$

Criteria: 0.97 Y < Y_c < 1.03 Y; System meets? _____ Exceeds? _____

- B. Utilizing EPA's Method 5 Dry Gas Meter Performance Test Device
- Remove the performance test device from its case and insert it into the gas inlet quick-connect coupling on the source sampling meter box.
 - Turn the power to the meter box on and start the pump.
 - Completely open the coarse flow rate control valve and close the fine flow rate control valve to give a maximum vacuum reading.
 - Allow the orifice and source sampling meter obox to warm up for 45 minutes with flow controls adjusted as described in step c before starting quality assurance runs.

- e. Make triplicate quality assurance runs. For each run, record initial and final dry gas meter volume, dry gas meter inlet and outlet temperatures, internal orifice pressure drop (Delta H), ambient temperature, and barometric pressure. Each run duration should be greater than 15 minutes. As the needle reaches the zero (12 o'clock) position, pump and stopwatch are stopped simultaneously. The dry gas volume and time are recorded.

QA Table For Evaluating DGM Orifice Value

Item	Run 1	Run 2	Run 3
DGM, final			
DGM, initial			
DGM, net			
DGM, T inlet (°F)			
DGM, T outlet (°F)			
DGM, T avg (°F)			
Ambient Temp. (°F)			
Pb (in. Hg)			
Orifice meter, Delta H			

- f. Calculate the corrected dry gas volume for each run using the following equation:

$$Y_c = 10/V_m [0.0319 (T_m = 460)/P_b]^2 = \underline{\hspace{4cm}}$$

$$Y_c = 0.97 Y = 1.03 Y = \underline{\hspace{4cm}}$$

Criteria: $0.97 Y < Y_c < 1.03 Y$; System meets? Exceeds?

Nomograph/calculator check:

- a. Nomograph:

If $\Delta H @ = 1.80$, $T(\text{meter}) = 100 \text{ F}$, % moisture = 10 %, $P_s/P_m = 1.00$, calculate C.

$$C = \underline{\hspace{2cm}} (0.95)$$

If $C = 0.95$, $T_s = 200 \text{ F}$, $D_n = 0.375$, $\Delta p \text{ reference} = \underline{\hspace{2cm}} (0.118)$

Align Delta p = 1.0 with Delta H = 10; and lock nomograph. Set p = 0.01.
 Read Delta H _____ (0.01)
 Nomograph O.K.? _____

b. Calculator:

Calculate Delta H/delta p using the following equation:

$$\Delta H/\Delta p = \{846.72 D_n^4 [\Delta H@] C_p^2 (1-B_{ws})^2 (M_d T_m P_s / M_s T_s P_m)\}$$

= _____
 = _____
 = _____

Parameters for calculator check:

Item Dimensions Given Calculated

- A. Orifice meter coefficient, Delta H@ in. Water 1.80 _____
- B. Pitot tube coefficient, Cp dimensionless 0.84 _____
- C. Abs. stack pressure, Ps = Pb + Pst in. Hg 29.96 _____
- D. Abs. meter pressure, Pm +/- Pb in. Hg 29.92 _____
- E. Abs. meter temperature, Tm = tm + 460 degrees R 560 _____
- F. Abs. stack temperature, Ts = ts + 460 degrees R 760 _____
- G. Dry molecular weight, Mwd lb/lb-mole 30 _____
- H. Moisture content, Bws + % water/100 dimensionless 0.10 _____
- I. Exact nozzle diameter, Dn in. 0.375 _____
- J. Average velocity head, Delta p in. Water 0.1 _____
- K. K factor = Delta H/Delta p 13.24 (_____)
- L. Delta H calculated for delta p = 0.1 in. Water 1.324 (_____)

Calculator equation set-up O.K.? _____

Nomograph /calculator setup:

- A. Delta H@ = _____
- B. Cp = _____
- C. Ps = _____
- D. Pm = _____
- E. Tm = _____
- F. Ts = _____
- G. Md = _____
- H. % M = _____
- I. C Factor = _____
- J. Dn desired = _____ in. Actual _____ in.

K. Avg. Delta p = _____
L. Delta H calculated = _____

- Nomograph set up O.K.? _____
- Calculator set up O.K.? _____
- K-factor figured properly? _____
- Care taken to avoid scrapping nipple or stack wall? _____
- Effective seal made around probe when in-stack? _____
- Pretest leak check performed? _____
- Probe moved at proper time? _____
- Pitot tube/probe marking correct for correct sampling points? _____
- Nozzle and pitot tube parallel to stack wall at all times? _____
- Filter changed during run? _____
- Any particulate loss during changing? _____
- Leak check before changing components during run? _____
- Data sheets completed, in ink, and data properly recorded? _____
- Percent isokinetics calculated at each point? _____ Within 90-110 %? _____
- Perform % isokinetic rate check: _____

$\Delta H/\Delta p = \{846.72 D_n^4 [\Delta H@] C_p^2 (1-B_{ws})^2 (M_d T_m P_s / M_s T_s P_m)\}$
= K-factor changed during run? _____ New K-factor _____
Velocity pressures and orifice pressure readings recorded accurately? _____
Post-test leak check performed? (Mandatory) _____
Leakage rate _____ @ in. Hg _____
If data sheets cannot be copied, record approximate stack temperature, average stack velocity delta p, average stack delta H, approximate stack gas moisture, approximate dry stack gas molecular weight nozzle diameter, pitot tube coefficient.
Gas density determination: Temp. sensor type _____ Pressure gauge _____
Temperature sensor attached to probe? _____ Other _____

Recovery:

- Construction of probe recovery brushes: Nylon _____ Clean _____ Other _____
- Wash bottle: Clean? _____
- Storage containers: Borosilicate glass? _____ Other _____ Clean? _____
Leak free? _____
- Petri-dishes: Glass? _____ Polyethylene _____ Other _____ Clean? _____
- Graduated cylinder/or balance: Subdivisions <= 2 mL? _____ Other _____
Balance type? _____ Calibrated? _____
- Plastic storage containers: Air-tight? _____ Clean? _____
- Probe allowed to cool sufficiently? _____

(Note: Cap placed over nozzle tip to prevent loss of particulate matter?)

During sampling train disassembly, are all openings capped? _____
Clean-up area: Description _____

Clean-up area clean and protected from wind? _____

Filter handling: Tweezers used? _____ Material of construction? _____

Surgical gloves worn? _____ Other _____

Any particulate matter spilled? _____

Water: Distilled? _____

Stopcock grease: Acetone-insoluble? _____ Heat-stable Silicone? _____

Probe handling: Acetone rinse? _____ Distilled water rinse? _____

Particulate recovery: Probe nozzle? _____ Probe fitting? _____

Probe liner? _____ Front half of filter holder? _____

Blanks: Acetone _____ Distilled water _____

Acetone reagent <.001 % residue? _____

Glass bottles residue <.001 %? _____

Acetone blanks residue <.001 % _____

Any visible particles on filter holder inside probe? _____

Color of silica gel? _____ Condition? _____

Method of silica gel transfer? _____

All jars adequately labeled? _____

Jars sealed tightly? _____ Liquid level marked on jars? _____

Jars locked up and chain-of-custody completed? _____

All data forms signed and chain-of-custody completed? _____

Remarks _____

Container No. 6 (0.1 M nitric acid blank):

Equal volume placed in Container No. 6 used to rinse sample train?

(Note: Need only if total chromium is being measured and only one blank per field test.)

Fluid level properly marked? _____

All jars adequately labeled? _____

Jars sealed tightly? _____ Liquid level marked on jars? _____

Jars locked up and chain-of-custody completed? _____

Teflon tape used to secure seal of jars? _____

All data forms signed and chain-of-custody completed? _____

Gas sample volume corrected to standard temperature and pressure? _____

Remarks: _____

f. Calculate the corrected dry gas volume for each run using the following equation:

$$Y_c = 10/V_m [0.0319 (T_m = 460)/P_b]^2 = \underline{\hspace{4cm}}$$

$$Y_c = \underline{\hspace{4cm}}$$

$$0.97 Y = \underline{\hspace{4cm}}$$

$$1.03 Y = \underline{\hspace{4cm}}$$

Criteria: $0.97 Y < Y_c < 1.03 Y$; System meets? Exceeds?

Nomograph/calculator check:

a. Nomograph:

If $\Delta H@ = 1.80$, $T(\text{meter}) = 100 \text{ F}$, % moisture = 10 %, $P_s/P_m = 1.00$, calculate C.

$$C = \underline{\hspace{2cm}} (0.95)$$

If $C = 0.95$, $T_s = 200 \text{ F}$, $D_n = 0.375$, Δp reference = (0.118)

Align: $\Delta p = 1.0$ with $\Delta H = 10$; and lock nomograph. Set $p = 0.01$.

Read ΔH (0.01)

Nomograph O.K.?

b. Calculator:

Calculate $\Delta H/\Delta p$ using the following equation:

$$\Delta H/\Delta p = \{846.72 D_n^4 [\Delta H@] C_p^2 (1-B_{ws})^2 (M_d T_m P_s / M_s T_s P_m)\}$$

$$= \underline{\hspace{4cm}}$$

$$= \underline{\hspace{4cm}}$$

$$= \underline{\hspace{4cm}}$$

Parameters for calculator check:

Item Dimensions Given Calculated

- A. Orifice meter coefficient, $\Delta H@$ in. Water 1.80
- B. Pitot tube coefficient, C_p dimensionless 0.84
- C. Abs. stack pressure, $P_s = P_b + P_{st}$ in. Hg 29.96
- D. Abs. meter pressure, $P_m \pm P_b$ in. Hg 29.92
- E. Abs. meter temperature, $T_m = t_m + 460$ degrees R 560
- F. Abs. stack temperature, $T_s = t_s + 460$ degrees R 760
- G. Dry molecular weight, M_{wd} lb/lb-mole 30 H. Moisture content, $B_{ws} + \% \text{ water}/100$ dimensionless 0.10
- I. Exact nozzle diameter, D_n in 0.375

Criteria: $0.97 Y < Y_c < 1.03 Y$; System meets? _____ Exceeds _____

B. Utilizing EPA's Method 5 Dry Gas Meter Performance Test Device

- a. Remove the performance test device from its case and insert it into the gas inlet quick-connect coupling on the source sampling meter box.
- b. Turn the power to the meter box on and start the pump.
- c. Completely open the coarse flow rate control valve and close the fine flow rate control valve to give a maximum vacuum reading.
- d. Allow the orifice and source sampling meter box to warm up for 45 minutes with flow controls adjusted as described in step c before starting quality assurance runs.
- e. Make triplicate quality assurance runs. For each run, record initial and final dry gas meter volume, dry gas meter inlet and outlet temperatures, internal orifice pressure drop (Delta H), ambient temperature, and barometric pressure. Each run duration should be greater than 15 minutes. As the needle reaches the zero (12 o'clock) position, the pump and stopwatch are stopped simultaneously. The dry gas meter volume and time are recorded.

QA Table For Evaluating DGM Orifice Value

Item	Run 1	Run 2	Run 3
DGM, final			
DGM, initial			
DGM, net			
DGM, T inlet (°F)			
DGM, T outlet (°F)			
DGM, T avg (°F)			
Ambient Temp. (°F)			
Pb (in. Hg)			
Orifice meter, Delta H			

STACK TESTING FOR HAZARDOUS AIR POLLUTANTS (HAPs)

FIELD OBSERVATION CHECKLIST

SW-846, METHOD 0061

**DETERMINATION OF HEXAVALENT CHROMIUM EMISSIONS
FROM STATIONARY SOURCES**

SOURCE INFORMATION

Facility _____ Location _____
Source _____ Sampling _____
Location _____
Test Team _____ Team Leader _____
Test Date Interval: ____ / ____ / ____ to ____ / ____ / ____
Observer _____ Affiliation _____

PRETEST MEETING CHECKLIST

Please reference Pretest Checklist.

FEDERAL REFERENCE METHODS 1 THROUGH 5

Please reference Federal Reference Methods 1 through 5 Checklist

SW-846, METHOD 0061

**DETERMINATION OF HEXAVALENT CHROMIUM EMISSIONS
FROM STATIONARY SOURCES**

Equipment identified matches pretest calibration list? _____
Train Design: Recirculating train where the impinger reagent is continuously recirculated to the nozzle. Aspirator design? _____ Pump/Sprayer design? _____

Apparatus: ..

NOZZLE

Probe nozzle: Glass/Teflon (Required) _____ Other _____
Probe design: Button-hook _____ Elbow _____ Other _____
Nozzle size: _____ Calculated _____ Measured three times _____
Nozzle condition: Nicks/Chips _____

PROBE/PROBE LINER

Probe material: Glass/Teflon (Required) _____

300 mL of 0.1 M nitric acid used in sample recovery placed in
Container 8A? _____

Fluid level properly marked? _____

(Note: Need only one blank per field test.)

Container No. 8B (Reagent water blank):

100 mL of reagent water blank placed in Container 8B? _____

Fluid level properly marked? _____

(Note: Need only one blank per field test.)

Container No. 9 (Nitric acid impinger reagent blank):

200 mL of nitric acid impinger reagent placed in Container 9? _____

Fluid level properly marked? _____

(Note: Need only one blank per field test.)

Container No. 10 (Permanganate reagent blank):

100 mL of permanganate impinger reagent placed in Container 10? _____

Fluid level properly marked? _____

(Note: Need only one blank per field test.)

Container No. 11 (8M HCl blank):

200 mL of reagent water placed in Container No. 11? _____

25 mL of 8 M HCl placed in Container No. 11? _____

Fluid level properly marked? _____

(Note: Need only one blank per field test.)

Container No. 12 (Filter blank):

Unused filter from same lot of field filter placed in Container No. 12? _____

Petri dish sealed? _____

All jars adequately labeled? _____

Jars sealed tightly? _____

Liquid level marked on jars? _____

Jars locked up and chain-of-custody completed? _____

Teflon tape used to secure seal of jars? _____

All data forms signed and chain-of-custody completed? _____

Gas sample volume corrected to standard temperature and pressure? _____

Remarks _____

Container No. 2 (Probe rinse/acetone):

Acetone rinse? _____ Distilled water rinse? _____
Probe nozzle? _____ Probe fitting? _____
Probe liner? _____ Front half of filter holder? _____
Fluid levels properly marked? _____

Container No. 3 (Probe rinse/ 100 mL nitric acid)

Exactly 100 mL of nitric acid used? _____
Probe nozzle? _____ Probe fittings? _____
Probe liner? _____ Front half of filter holder? _____
Fluid levels properly marked? _____
Any visible particles or permanganate solution (brown color) on filter holder or inside probe? _____

Container No. 4 (Impingers 1 through 3):

Liquids measured volumetrically to within 0.5 mL? _____
Liquids in one or several containers? _____
Impingers, back half of filter housing, connecting glassware rinsed with exactly 100 mL of 0.1 M nitric acid? _____
Fluid levels properly marked? _____

Container No. 5A (Normally empty impinger 4):

Liquid measured volumetrically to within 0.5 mL? _____
Impinger rinsed with exactly 100 mL of 0.1 M nitric acid? _____
Fluid levels properly marked? _____

Container No. 5B (Impingers 5 and 6 containing permanganate solutions):

Permanganate solution color? _____
Spent (Complete bleaching of purple color)? _____
Liquid measured volumetrically to with 0.5 mL? _____
Impingers rinsed with freshly prepared 100 mL permanganate sol.? _____
Impingers rinsed with 100 mL reagent water? _____
Fluid levels properly marked? _____

(Note: Bottles should not be filled completely to allow venting.)

Container No. 5C (Rinsing of impingers 5 and 6 with 8 M HCl):

200 mL reagent water added to Container No. 5C? _____
Exactly 50 mL of 8 M HCl used to rinse impingers 5 and 6? _____
Fluid level properly marked? _____

Container No. 6 (Impinger 7 containing silica gel):

Color of recovered silica gel? _____
Method of silica gel transferred? _____
Silica gel weighted to nearest 0.5 g? _____
Container properly labeled/sealed? _____

Container No. 7 (Acetone blank):

100 mL placed in Container 7? _____
Fluid level properly marked? _____

(Note: If particulate matter is not being determined, then an action blank is not necessary.)

Container No. 8A (0.1 M nitric acid blank):

Care taken to avoid scrapping nipple or stack wall? _____
 Effective seal made around probe when in-stack? _____
 Pretest leak check performed? _____
 Probe moved at proper time? _____
 Pitot tube/probe marking correct for correct sampling points? _____
 Nozzle and pitot tube parallel to stack wall at all times? _____
 Filter changed during run? _____
 Any particulate loss during changing? _____
 Leak check before changing components during run? _____
 Data sheets completed, in ink, and data properly recorded? _____
 Percent isokinetics calculated at each point? _____ Within 90-110 %? _____
 Perform % isokinetic rate check:

$$\Delta H / \Delta p = \{846.72 D_n^4 [\Delta H @] C_p^2 (1 - B_{ws})^2 (M_d T_m P_s / M_s T_s P_m)\} =$$

K-factor changed during run? _____
 New K-factor _____
 Velocity pressures and orifice pressure readings recorded accurately? _____
 Post-test leak check performed?(Mandatory) _____
 Leakage rate _____ @ in. Hg _____
 If data sheets cannot be copied, record approximate stack temperature, average stack velocity Δp , average stack ΔH , approximate stack gas moisture, approximate dry stack gas molecular weight, nozzle diameter, pitot tube coefficient. Gas density determination: Temp. sensor type _____ Pressure gauge _____
 Temperature sensor attached to probe? _____ Other _____

Recovery:

Construction of probe recovery brushes: Nylon _____ Clean _____ Other _____
 Wash bottle: Clean/glass? _____
 Storage containers: Borosilicate glass? _____ Other _____ Clean? _____
 Leak free? _____ Teflon-lined caps? _____
 Petri-dishes: Glass? _____ Polyethylene _____ Other _____ Clean? _____
 Graduated cylinder/or balance: Subdivisions ≤ 2 mL? _____ Other _____
 Balance type? _____ Calibrated? _____
 Plastic storage containers: Air-tight? _____ Clean? _____
 Probe allowed to cool sufficiently? _____
 (Note: Cap placed over nozzle tip to prevent loss of particulate matter?(_____)
 During sampling train disassembly, are all openings capped? _____
 Clean-up area: Description _____

Clean-up area clean and protected from wind? _____

Sample recovery:

Container No. 1 (Filter):

Filter handling: Polypropylene tweezers used? _____
 Surgical gloves worn? _____ Other _____
 Any particulate matter spilled? _____

Calculate Delta H/delta p using the following equation:

$$\Delta H/\Delta p = \{846.72 D_n^4 [\Delta H@] C_p^2 (1-B_{ws})^2 (M_d T_m P_s / M_s T_s P_m)\}$$

= _____
 = _____
 = _____

Parameters for calculator check:

Item Dimensions Given Calculated

- A. Orifice meter coefficient, $\Delta H@$ in. Water 1.80 _____
- B. Pitot tube coefficient, C_p dimensionless 0.84 _____
- C. Abs. stack pressure, $P_s = P_b + P_{st}$ in. Hg 29.96 _____
- D. Abs. meter pressure, $P_m +/- P_b$ in. Hg 29.92 _____
- E. Abs. meter temperature, $T_m = t_m + 460$ degrees R 560 _____
- F. Abs. stack temperature, $T_s = t_s + 460$ degrees R 760 _____
- G. Dry molecular weight, M_{wd} lb/lb-mole 30 _____
- H. Moisture content, $B_{ws} + \% \text{ water}/100$ dimensionless 0.10 _____
- I. Exact nozzle diameter, D_n in. 0.375 _____
- J. Average velocity head, Δp in. Water 0.1 _____
- K. K factor = $\Delta H/\Delta p$ 13.24(_____)
- L. ΔH calculated for $\Delta p = 0.1$ in. Water 1.324(_____)

Calculator equation set-up O.K.? _____

Nomograph /calculator setup:

- A. $\Delta H@ =$ _____
- B. $C_p =$ _____
- C. $P_s =$ _____
- D. $P_m =$ _____
- E. $T_m =$ _____
- F. $T_s =$ _____
- G. $M_d =$ _____
- H. $\% M =$ _____
- I. C Factor = _____
- J. D_n desired = _____ in. Actual _____ in..
- K. Avg. $\Delta p =$ _____
- L. ΔH calculated = _____

Nomograph set up O.K.? _____

Calculator set up O.K.? _____

K-factor figured properly? _____

A. Operating Meter Box Assembly

Operate metering system for 10 minutes at the H@ value. Record the volume metered, DGM temperature and barometric pressure.

Final: DGM reading _____ ft³ Temp _____ °F P_b _____ in. Hg
 Initial: DGM reading _____ ft³ Temp _____ °F P_b _____ in. Hg
 Net: V_m = _____ ft³ Avg. T_m = _____ °F + 460 = _____

$$Y_c = 10/V_m [0.0319 (T_m = 460)/P_b]^2 = \underline{\hspace{10em}}$$

$$Y_c = 0.97 Y = 1.03 Y = \underline{\hspace{10em}}$$

Criteria: $0.97 Y < Y_c < 1.03 Y$; System meets? _____ Exceeds? _____

B. Utilizing EPA's Method 5 Dry Gas Meter Performance Test Device

- a. Remove the performance test device from its case and insert it into the gas inlet quick-connect coupling on the source sampling meter box..
- b. Turn the power to the meter box on and start the pump.
- c. Completely open the coarse flow rate control valve and close the fine flow rate control valve to give a maximum vacuum reading.
- d. Allow the orifice and source sampling meter box to warm up for 45 minutes with flow controls adjusted as described in step c before starting quality assurance runs.
- e. Make triplicate quality assurance runs. For each run, record initial and final dry gas meter volume, dry gas meter inlet and outlet temperatures, internal orifice pressure drop (Delta H), ambient temperature, and barometric pressure. Each run duration should be greater than 15 minutes. As the needle reaches the zero (12 o'clock) position, the pump and stopwatch are stopped simultaneously. The dry gas meter volume and time are recorded.

(Note: If particulate emissions are not to be determined, the filter need not be desiccated or weighted.)

Filter checked visually for irregularities? _____
Filter media type? _____
Sampling site selected? _____ Stack temperature _____
Stack velocity head? _____ Stack dry molecular weight calculated? _____
Nozzle size properly selected? _____
Calibration data certified available: Pitot tubes _____ Meter box Y factor _____
Meter box delta H@ _____ Magnehelic gauge _____
Thermocouples _____ Probe nozzle _____
(Note: If calibration sheets are not available, contact Task Manager for authorization to proceed. Authorization to proceed:
Yes _____ No _____
Date _____ Time _____ Task Manager _____)

Selection of sampling points and time according to methodology? _____
Number of sample points _____ Sample time/point _____
Minimum sample volume required? _____
Minimum run time required? _____
Total sample time actual? _____

Sampling train assembled properly? _____
Filter properly centered? _____ Impingers properly charged? _____
Number of impingers required? _____

Sample train preparation (Volume added to each impinger):
1st (normally empty) _____
2nd/3rd (100 mL of HNO₃/H₂O₂) _____
4th (normally empty) _____
5th/6th (100 mL of permanganate solution) _____
(Note: Reagent prepared fresh daily? _____)
7th (200-300 g of preweighted silica gel) _____
Weight (Initial) _____

Pitot tube lines checked for plugging or leaks? _____
Stopcock grease used: Acetone-insoluble? _____ Heat-stable? _____
Meter box leveled? _____ Manometers leveled? _____
Probe heat uniform along length of probe? _____
Dry gas meter field observation check: _____

PITOT TUBE

Type: Type S _____ Standard _____ Other _____
Properly attached to probe? _____
Meets EPA guidelines for attachment to probe? _____ Yes _____ No _____
Modifications _____
Pitot tube coefficient _____

FILTER HOLDER

Construction: Borosilicate _____ Quartz _____ Other _____
Frit construction: Glass frit? _____ Teflon frit? _____
Filter support material? _____ Silicone gasket? _____
Without organic binder? _____ Clean? _____
Cyclone and flask used (optional): _____

FILTER HEATING SYSTEM

Description _____
Temperature _____
Gas temperature at exit of box _____

CONDENSER

Number of impingers? _____ Clean? _____
Teflon impingers used? _____
Impinger contents: 1st _____ 2nd _____ 3rd _____
4th _____ 5th _____ 6th _____ 7th _____
Cooling system? _____
Proper connections (leak free/ground glass/ or other material)? _____
Modifications? _____
All glassware properly cleaned as specified in methodology (Hot tap water, then hot soapy water, 3X with tap water, 3X reagent water, soaked in 10 % nitric acid for 4 hours, 3X reagent water, rinsed with acetone, then air dry, then opening covered with paraffin until use?) _____

METER BOX ASSEMBLY

Manufacturer _____ Model No. _____ Unit No. _____
Delta H@ value _____ Date _____
Dry gas meter "Y" value _____ Date _____
Condition of sampling equipment? _____
Meter box leveled? _____
Timer/temperature readouts operational? _____
Manometer leveled? _____
Thermocouples calibrated? _____

Operation

Filter properly numbered and recorded on field test data sheet? _____

STACK TESTING FOR HAZARDOUS AIR POLLUTANTS (HAPs)

**FIELD OBSERVATION CHECKLIST
SW-846, METHOD 0060**

DETERMINATION OF METALS IN STACK EMISSIONS

SOURCE INFORMATION

Facility _____ Location _____
Source _____ Sampling _____ Location _____
Test Team _____ Team Leader _____
Test Date Interval: ____ / ____ / ____ to ____ / ____ / ____
Observer _____ Affiliation _____

PRETEST MEETING CHECKLIST

Please reference Pretest Checklist

FEDERAL REFERENCE METHODS 1 THROUGH 5

Please reference Federal Reference Methods 1 through 5 Checklist.

SW-846, METHOD 0060

DETERMINATION OF METALS IN STACK EMISSIONS

Equipment identified matches pretest calibration list? _____
Particulate matter measured concurrently with metals? _____
Metals under evaluation: Sb ___ As ___ Ba ___ Be ___ Cd ___ Cr ___ Co ___ Cu ___ Pb ___ Mn ___
Hg ___ Ni ___ P ___ Se ___ Ag ___ Tl ___ Zn ___ Other ___ Other ___ Other ___ Other ___
Mercury emissions being measured (If mercury emissions are not being measured, then
impingers 4, 5, and 6 are not necessary and may be eliminated)? _____

Apparatus:

NOZZLE

Probe nozzle: Glass (Required) _____ Other _____
Probe design: Button-hook _____ Other _____
Nozzle size: _____ Calculated _____ Measured three times _____
Nozzle condition: Nicks/Chips _____

PROBE/PROBE LINER

Probe material: Borosilicate _____ Quartz _____
Heating system: _____ Temperature _____
Water cooled? _____ Length _____

SW-846, METHOD 0030/0031 TEST OBSERVATION SHEET (CON'T)

	Run 1	Run 2	Run 3	Run 4	Run 5	Run 6
7. BLANKS						
<p>a. Field Blank: For Every 6 Cartridge Sets, Obtain One Set, Place Near Sampling Train For 5 Minutes, Replace Caps, Store < 4° C.</p> <ul style="list-style-type: none"> - Field Blank Set Taken? - Labeled Properly? - Stored Properly < 4 °C? 						
<p>b. Trip Blank: For Every 6 Cartridge Sets, Obtain One Set, Do Not Uncap, Store < 4 °C.</p> <ul style="list-style-type: none"> - Trip Blank Set Taken? - Labeled Properly? - Stored Properly < 4 °C? 						
<p>c. Field Blank D.I. Water: Fill One Vial With Field D.I. Water to Overflow, Cap, Store.</p> <ul style="list-style-type: none"> - Field Blank D.I. Water Take? - Labeled Properly? - Stored Properly? 						
8. SHIPPING						
<p>a. Chain-of-Custody Sheets Completed?</p> <p>b. Samples Shipped at < 4 °C?</p> <p>c. Shipping Mode (UPS, Fed. Exp., Other)?</p>						

SW-846, METHOD 0030/0031 TEST OBSERVATION SHEET (CON'T)

Run 1 Run 2 Run 3 Run 4 Run 5 Run 6

5. TRAIN OPERATION

<ul style="list-style-type: none"> a. Probe in Stack At Sampling Point? b. Sample Port Sealed? c. Stack Gas Purged Through Probe via Isolation Valve Prior To Starting Run? d. Valve Opened for Condenser? e. Inlet to 1st Sorbent < 20 °C? f. Test Time: <ul style="list-style-type: none"> - Run Start - Run End - Net Sampling Time - Sampling Rate - Minimum Sampling Time Met? h. Dry Gas Meter <ul style="list-style-type: none"> - Final (L) - Initial (L) - Net (L) - Minimum Volume of Liters Met? 						
--	--	--	--	--	--	--

6. SAMPLE RECOVERY

<ul style="list-style-type: none"> a. Cartridges Removed With Gloved Hands and End Caps Replaced? b. Cartridges Properly Labeled? c. Cartridges Stored at < 4 °C? d. Condensate Recovered in Vial and Topped Off With DI Water? e. Condensate Vial Properly Labeled? f. All Samples Logged Into Master Field Log Book? g. Chain-of-Custody Completed and Signed? 						
--	--	--	--	--	--	--

SW-846, METHOD 0030/0031 TEST OBSERVATION SHEET

1. GENERAL							
a. Date b. Run No. C. Code No.							
2. CARTRIDGE INFORMATION		Run 1	Run 2	Run 3	Run 4	Run 5	Run 6
a. Cartridge ID - Cartridge No. 1 - Cartridge No. 2 - Cartridge No. 3 Certificate of Cleaniness? (Yes/No)							
3. TRAIN SET-UP							
a. Cartridges Placed Properly and Correctly Oriented? (Yes/No) b. Condensers Operating Properly? c. Condenser Trap in Place? d. Silica Gel in Place? e. Probe at Operating Temperature?							
4. TRAIN LEAK CHECK							
<p align="center">Initial</p> a. Checked @ - 10 " Hg? b. Leak Rate (<0.1 " Hg for 1 min.)? c. Vented through Charcoal Tube?							
<p align="center">Intermittent</p> a. Checked at Highest Vacuum? b. Leak Rate (<0.1 " Hg for 1 min.)? c. Vented through Charcoal Tube?.							

Gas temperature exiting sorbent bed at a maximum of 20 °C? _____
Data recorded on field data sheet properly? _____

Recovery:

Storage containers: Borosilicate
glass? _____ Other _____ Clean? _____
Leak free? _____ Teflon liners? _____

Clean-up area: Description

Clean-up area clean and protected from wind? _____
Recovery personnel cover hands with cotten gloves for sorbent tubes recovery? _____
Sorbent cartridges capped at both ends? _____
Sorbents properly marked for particular run? _____
Sorbents stored at < 4 °C until analyzed? _____
Knockout trap capped? _____
All data forms signed and chain-of-custody completed? _____

Station Pb _____
[Inches of Hg to be added or subtracted from Station Pb (0.1 " Hg X ft per
100 ft)

Describe condition of sampling equipment: _____

Operation

Sampling site selected? _____ Stack temperature _____
Stack velocity head? _____ Stack dry molecular weight calculated? _____
Nozzle size properly selected (if applicable)? _____
Calibration data certified available: Pitot tubes _____ Meter box Y factor _____
Rotameters _____ Magnehelic guage _____
Thermocouples _____ Probe nozzle _____
(Note: If calibration sheets are not available, contact Task Manager for authorization to
proceed. Authorization to proceed: Yes _____ No _____

Date _____ Time _____ Task Manager _____
Selection of sampling points and time according to methodology? _____
Number of sample points _____ Sample time/point _____
Minimum sample volume calculated per cartridge set: _____
Minimum run time required? _____
Total sample time actual? _____
Sampling point > 1 meter from side of wall? _____
Sampling point at average stack gas velocity? _____
Average stack gas temperature? _____
Sampling train assembled properly? _____

(Note: During preparation of sampling train, keep all openings where
contamination can occur covered with Teflon film or aluminum foil.)

Sorbent traps properly orientated in sampling train? _____
List orientation of sorbent cartridges: 1. _____ 2. _____ 3. _____
Drying medium: Silica gel? _____ Drierite _____
Pitot tube lines checked for plugging or leaks? _____
Meter box leveled? _____ Manometers leveled? _____
Probe heat uniform along length of probe? _____
Condenser water recirculating properly? _____
Care taken to avoid scrapping nipple or stack wall? _____
Effective seal made around probe when in-stack? _____
Pretest leak check performed (Required)? _____ Leak rate <0.1 " Hg? _____
Leak check performed through probe? _____
Pitot tube/probe marking correct for correct sampling point? _____
Pitot tube parallel to stack wall at all times? _____
Data sheets completed, in ink, and data properly recorded? _____
Probe maintained at 120 +/- 14 °C? _____

Isolation valve? _____ Quantity? _____ Stainless steel? _____

ORGANIC SAMPLING MODULES

Three sections: condenser _____ sorbent trap _____
condensate knockout trap _____

Condenser maintains stack gas not exceeding 20 °C _____

Condenser: water cooled? _____ coolant? _____

Sorbent trap jacketed to maintain the internal gas temperature not exceeding 20 °C (68 °F)? _____

Design of sorbent trap: Single unit? _____ Dual units? _____

Thermocouple well in bed of sorbent trap? _____

Sorbent in traps? 1. _____ 2. _____ 3. _____

Sorbent certified clean (< 10 ng per target analyte)? _____

Date of sorbent certification? _____ Within 4 weeks? _____

Sorbent stored < 4 °C before use? _____

Chain-of-custody with sorbent trap? _____

Sorbent trap received wrapped in aluminum foil? _____

Coolant? (Ice water?) _____

Knockout trap directly below sorbent trap? _____

Type of knockout trap? _____

Vacuum grease of any kind PROHIBITED! _____

METER BOX ASSEMBLY

Manufacturer _____ Model No. _____ Unit No. _____

Dry gas meter measures in liters? _____ Yes _____ No _____

Dry gas meter "Y" value _____ Date _____

Dry gas meter measure volume within 1 %? _____

Rotameters calibrated? _____ Yes _____ No _____

Calibration date? _____

Condition of sampling equipment? _____

Meter box leveled? _____

Timer/temperature readouts operational? _____

Manometer leveled? _____

Thermocouples calibrated (Read within 3 °C)? _____

Barometer: Mercury, Aneroid, Other _____

Barometer capable of measuring atmospheric pressure within 2.5 mm Hg? _____

Vacuum gauge: After silica gel trap? _____ Range (0-30 " Hg)? _____

Isolation valve: Located after vacuum gauge? _____

Leak free pump? _____

Temperature readouts: Digital? _____ Analog? _____ Dial Thermometer? _____

Barometer: Aneroid? _____ Mercury? _____

Calibrated against _____ Date: _____

Airport Location _____

Elevation _____ feet above sea level

Sampling location elevation _____ feet above sea level

Sampling rate: _____ Lpm

7. Total sample time per cartridge set:

Method 0030, VOST < 20 min? _____

Method 0031, SLO-VOST < 40 min? _____

8. Total number of cartridge sets per run:

Minimum of 4 sets? _____

Maximum of 6 sets? _____

9. Run time (2 hours actual):

Method 0030, VOST < 80 min ? _____

Method 0031, SLO-VOST < 160 min? _____

Appratus:

NOZZLE

Probe nozzle: Stainless steel _____ Glass _____ Other _____

Probe design: Button-hook _____ Elbow _____ Straight tube _____

Tapered (30 angle)? _____

Nozzle size: _____ Calculated _____ Measured three times _____

Nozzle condition: Nicks _____ Bent opening _____

PROBE/PROBE LINER

Probe material: Borosilicate _____ Quartz _____ Stainless steel _____

Inconel _____ Other _____

Heating system: _____ Temperature (120 °C +/- 14 °C) _____

Water cooled? _____ Length _____

Probe liner: Borosilicate/quartz (up to 480 °C) _____ Quartz(480-900 °C) _____

Temperature sensor attached to probe? _____

Temperature sensor placed in an interference-free arrangement with probe? _____

Temperature probe calibrated? _____ Yes _____ No _____ Date _____

Probe plug used? _____ Teflon? _____

PITOT TUBE (If Applicable)

Type: Type S _____ Standard _____ Other _____

Properly attached to probe? _____

Meets EPA guidelines for attachment to probe? _____ Yes _____ No _____

Modifications _____

Pitot tube coefficient _____

SAMPLE TRANSFER LINE

Construction: Borosilicate _____ Quartz _____ Teflon _____

Length (< 5ft)? _____ Heat trace? _____

Other _____ Clean? _____

-Calibration range of GC/MS, ng (0-3,000 ng)

2. Calculation: Quantity of POHC to be collected:

$$\text{Quantity collected (ng),} = [A \times (1-B) \times C \times D \times E] / (F \times G \times H)$$

Where:

- A = POHC feed rate, lb/hr
- B = DRE (as decimal)
- C = 453.59 g/lb
- D = Sample volume, L
- E = 1×10^9 ng/g
- F = 60 min/hr
- G = Stack gas flow rate, dscfm
- H = 28.316 L/ft³

3. Perform calculation for:

% DRE of 99.9; 99.99; 99.999 and 99.9999

Sample volumes: 5, 10, 15, and 20 liters

4. Construct matrix for % DREs and sample volumes:

% DRE	Total ng collected			
	5 L	10 L	15 L	20 L
99.9				
99.99				
99.999				
99.9999				

5. Selection of sample volume:

At the required DRE, select the total volume to be sampled so that the maximum ng collected will be within the mid-range of the calibration curve. Sample volume must not exceed 20 liters.

Sample volume: _____ Lpm

6. Sampling rate:

The sampling rate is dependent on the total sampling time and total volume to be collected and must be between 0.25 and 1.0 Lpm. At no time shall sample rate exceed 1.0 Lpm.

STACK TESTING FOR HAZARDOUS AIR POLLUTANTS

FIELD OBSERVATION CHECKLIST

SW-846, METHOD 0030 AND METHOD 0031

SAMPLING AND ANALYSIS FOR VOLATILE ORGANIC COMPOUNDS

SOURCE INFORMATION

Facility _____ Location _____
Source _____ Sampling Location _____
Test Team _____ Team Leader _____
Test Date Interval: _____ / _____ / _____ to _____ / _____ / _____
Observer _____ Affiliation _____

PRETEST MEETING CHECKLIST

Please reference Pretest Checklist

FEDERAL REFERENCE METHODS 1 THROUGH 5

Please reference Federal Reference Methods 1 through 5 Checklist.

SW-846, METHOD 0030

VOLATILE ORGANIC SAMPLING TRAIN (VOST)

Equipment identified matches pretest calibration list? _____
Principle organic hazardous constituents (POHCs) 1. _____ 2. _____
3. _____ 4. _____ 5. _____ 6. _____
7. _____ 8. _____ 9. _____ 10. _____

For each waste feed stream containing POHCs, calculate POHC feed rate:

1. POHC concentration in waste feed: _____ %

2. Waste feed rate _____ lb/hr

If liquid feed rate is gal/hr: _____
gal/hr X _____ lb/gal (density) = _____ lb/hr

3. POHC feed rate in each waste feed stream:
_____ decimal wt (% POHC) X _____ waste feed rate (lb/hr) = _____ lb POHC/hr

Verification of Method 0030 (VOST) sample volume to be collected:

1. Basis: -% Destruction Removal Efficiency (DRE) expected, %
-Quantity of POHC to be collected, ng.

Container #4 (Condensate trap): Labeled/volume measured? _____

Container #5 (Train component rinse): From filter to 1st impinger: _____

Labeled/volume measured? _____

Container #6 (Silica gel): Color noted/capped/weighted. Containers #2-5 stored under "blue ice" (<4 C) and during shipment? Container #1 shipped unrefrigerated?

All data forms signed and chain-of-custody completed? _____

Remarks _____

approximate dry stack gas molecular weight _____, nozzle diam.
Gas density determination: Temp. sensor type _____ Pressure gauge _____
Temperature sensor attached to probe? _____ Other _____

Recovery:

Post-test leak check performed (Mandatory)? _____
Construction of probe recovery brushes: Nylon _____ Clean _____ Other _____
Wash bottle: Clean? _____ Three? _____ Teflon/Glass? _____
Storage containers: Borosilicate glass? _____ Other _____ Clean? _____
Leak free? _____ Teflon liners? _____
Petri-dishes (For storage/transport of filters):
Glass _____ Polyethylene _____ Other _____ Clean? _____
Graduated cylinder/or balance: Subdivisions \leq 2 mL? _____ Other _____
Balance type? _____ Calibrated? _____
Plastic storage containers (To store silica gel): Air-tight? _____ Clean? _____
Probe allowed to cool sufficiently? _____
(Note: Cap placed over nozzle tip to prevent loss of particulate matter?)
During sampling train disassembly, are all openings capped? _____
Clean-up area: Description _____
Clean-up area clean and protected from wind? _____
Filter handling: Tweezers used? _____ Material of construction? _____
Surgical gloves worn? _____ Other _____
Any particulate matter spilled? _____
Water: Distilled? _____
Stopcock grease: Acetone-insoluble? _____ Heat-stable Silicone? _____
Probe handling: Acetone rinse? _____ Distilled water rinse? _____
Particulate recovery: Probe nozzle? _____ Probe fitting? _____
Probe liner? _____ Front half of filter holder? _____
Blanks: Methanol _____ Methylene chloride _____
Any visible particles on filter holder inside probe? _____
Color of silica gel? _____ Condition? _____
Method of silica gel transfer? _____
All jars adequately labeled? _____
Jars sealed tightly? _____ Liquid level marked on jars? _____
Jars locked up and chain-of-custody completed? _____
Methylene chloride/methyl alcohol used in recovery? _____
Distilled-in-glass grade? _____
Resin bed capped at both ends? _____
Knockout trap capped? _____
Liquid in impinger train (first three impingers) measured/weighted? _____

Shipping Containers:

Container #1 (Petri dish): Filter inside? _____
Container #2 (Probe assembly rinse): Rinse with methanol/methylene chloride? _____
Container #3 (Sorbent trap): Labeled/ends sealed? _____

- J. Average velocity head, Delta p in. Water 0.1 _____
- K. K factor = Delta H/Delta p 13.24 (_____)
- L. Delta H calculated for delta p = 0.1 in. Water 1.324 (_____)
- Calculator equation set-up O.K.? _____

Nomograph /calculator setup:

- A. Delta H@ = _____
- B. Cp = _____
- C. Ps = _____
- D. Pm = _____
- E. Tm = _____
- F. Ts = _____
- G. Md = _____
- H. % M = _____
- I. C Factor = _____
- J. Dn desired = _____ in. Actual _____ in.
- K. Avg. Delta p = _____
- L. Delta H calculated = _____

- Nomograph set up O.K.? _____
- Calculator set up O.K.? _____
- K-factor figured properly? _____
- Care taken to avoid scrapping nipple or stack wall? _____
- Effective seal made around probe when in-stack? _____
- Pretest leak check performed? _____
- Probe moved at proper time? _____
- Pitot tube/probe marking correct for correct sampling points? _____
- Nozzle and pitot tube parallel to stack wall at all times? _____
- Filter changed during run? _____
- Any particulate loss during changing? _____
- Leak check before changing components during run? _____
- Data sheets completed, in ink, and data properly recorded? _____
- Percent isokinetics calculated at each point? _____ Within 90-110 %? _____
- Perform % isokinetic rate check:

$$\Delta H/\Delta p = \{846.72 D_n^4 [\Delta H@] C_p^2 (1-B_{ws})^2 (M_d T_m P_s / M_s T_s P_m)\} =$$

- K-factor changed during run? _____ New K-factor _____
- Velocity pressures and orifice pressure readings recorded accurately? _____
- Post-test leak check performed? _____ (Mandatory)
- Leakage rate _____ @ in. Hg _____
- If data sheets cannot be copied, record approximate stack temperature, average stack velocity delta p _____, average stack delta H _____, approximate stack gas moisture _____,

$Y_c = 0.97 Y =$ _____
 $1.03 Y =$ _____

Criteria: $0.97 Y < Y_c < 1.03 Y$; System meets? _____ Exceeds? _____

Nomograph/calculator check:

a. Nomograph:

If $\Delta H@ = 1.80$, $T(\text{meter}) = 100 \text{ F}$, % moisture = 10 %, $P_s/P_m = 1.00$, calculate C.

$C =$ _____ (0.95)

If $C = 0.95$, $T_s = 200 \text{ F}$, $D_n = 0.375$, Delta preference = (0.118)

Align $\Delta p = 1.0$ with $\Delta H = 10$; and lock nomograph. Set $p = 0.01$.

Read ΔH _____ (0.01)

Nomograph O.K.? _____

b. Calculator:

Calculate $\Delta H/\Delta p$ using the following equation:

$$\Delta H/\Delta p = \{846.72 D_n^4 [\Delta H@] C_p^2 (1-B_{ws})^2 (M_d T_m P_s / M_s T_s P_m)\}$$

= _____
 = _____
 = _____

Parameters for calculator check:

Item Dimensions Given Calculated

- A. Orifice meter coefficient, $\Delta H@$ in. Water 1.80 _____
- B. Pitot tube coefficient, C_p dimensionless 0.84 _____
- C. Abs. stack pressure, $P_s = P_b + P_{st}$ in. Hg 29.96 _____
- D. Abs. meter pressure, $P_m +/- P_b$ in. Hg 29.92 _____
- E. Abs. meter temperature, $T_m = t_m + 460$ degrees R 560 _____
- F. Abs. stack temperature, $T_s = t_s + 460$ degrees R 760 _____
- G. Dry molecular weight, M_{wd} lb/lb-mole 30 _____
- H. Moisture content, $B_{ws} + \% \text{ water}/100$ dimensionless 0.10 _____
- I. Exact nozzle diameter, D_n in. 0.375 _____

- a. Remove the performance test device from its case and insert it into the gas inlet quick-connect coupling on the source sampling meter box.
- b. Turn the power to the meter box on and start the pump.
- c. Completely open the coarse flow rate control valve and close the fine flow rate control valve to give a maximum vacuum reading.
- d. Allow the orifice and source sampling meter box to warm up for 45 minutes with flow controls adjusted as described in step c before starting quality assurance runs.
- e. Make triplicate quality assurance runs. For each run, record initial and final dry gas meter volume, dry gas meter inlet and outlet temperatures, internal orifice pressure drop (Delta H), ambient temperature, and barometric pressure. Each run duration should be greater than 15 minutes. As the needle reaches the zero (12 o'clock) position, the pump and stopwatch are stopped simultaneously. The dry gas meter volume and time are recorded.

QA Table For Evaluating DGM Orifice Value

Item	Run 1	Run 2	Run 3
DGM, final			
DGM, initial			
DGM, net			
DGM, T inlet (°F)			
DGM, T outlet (°F)			
DGM, T avg (°F)			
Ambient Temp. (°F)			
Pb (in. Hg)			
Orifice meter, Delta H			

- f. Calculate the corrected dry gas volume for each run using following equation:

$$Y_c = 10/V_m [0.0319 (T_m = 460)/P_b]^2 = \underline{\hspace{10em}}$$

(Note: If calibration sheets are not available, contact Task Manager for authorization to proceed. Authorization to proceed: Yes _____ No _____)
 Date _____ Time _____ Task Manager _____

Selection of sampling points and time according to methodology? _____
 Number of sample points _____ Sample time/point _____
 Minimum sample volume required (3 dscm or 105.9 scf)? _____
 Minimum run time required? _____
 Total sample time actual? _____
 Probe can reach all traverse points? _____

Sampling train assembled properly? _____
 (Note: During preparation of sampling train, keep all openings where _____
 Filter properly centered? _____ Impingers properly charged? _____

contam

Number of impingers required? _____ Which FRM? _____
 Volume in each impinger: 1st _____ 2nd _____ 3rd _____ 4th _____
 5th _____ 6th _____ 7th _____ 8th _____

Distilled organic-free water (Type II) used? _____
 Drying medium: Silica gel? _____ Drierite _____
 Weight (200-300 grams?) _____ 6-16 mesh? _____

Pitot tube lines checked for plugging or leaks? _____
 Stopcock grease used: Acetone-insoluble? _____ Heat-stable? _____
 (Note: Do not use stopcock grease upstream of sorbent bed!)

Meter box leveled? _____ Manometers leveled? _____
 Probe heat uniform along length of probe? _____
 If glass liner used, install nozzle using Viton-A O-ring when stack temperatures are < 260°C _____ Woven glass-fiber gasket > 260°C _____

Dry gas meter field observation check:

A. Operating Meter Box Assembly

Operate metering system for 10 minutes at the Delta H@ value. Record the volume metered, DGM temperature and barometric pressure.

Final: DGM reading _____ ft³ Temp _____ °F P_b _____ in. Hg
 Initial: DGM reading _____ ft³ Temp _____ °F P_b _____ in. Hg
 Net: V_m = _____ ft³ Avg. T_m = _____ °F + 460 = _____

$Y_c = 10/V_m [0.0319 (T_m = 460)/P_b]^2 =$ _____

Y_c = 0.97 Y = _____
 1.03 Y = _____

Criteria: 0.97 Y < Y_c < 1.03 Y; System meets? _____ Exceeds? _____

B. Utilizing EPA's Method 5 Dry Gas Meter Performance Test Device

Impinger stem in knockout trap one-half length of normal stem? _____

IMPINGER TRAIN

Number of impingers? _____ Clean? _____
Impinger contents: 1st (Water?) _____ 2nd (Water?) _____
3rd (Empty/ or caustic solution?) _____ 4th (Silica Gel?) _____
5th _____ 6th _____ 7th _____ 8th _____
Volume/weight of contents of impingers: 1st (100 mL?) _____ 2nd (100 mL?) _____
3rd (Empty?) _____ 4th (200-300 g silica gel?) _____
5th _____ 6th _____ 7th _____ 8th _____
Impinger types: Greenburg-Smith? _____ Standard? _____
Impinger tip design: Tapered tip? _____ Standard tip? _____
Cooling system? _____
Proper connections? _____
Modifications? _____

METER BOX ASSEMBLY

Manufacturer _____ Model No. _____ Unit No. _____
Delta H@ value _____ Date _____
Dry gas meter "Y" value _____ Date _____
Dry gas meter measure volume within 1 %? _____
Condition of sampling equipment? _____
Meter box leveled? _____
Timer/temperature readouts operational? _____
Manometer leveled? _____
Thermocouples calibrated (Read within 3 C)? _____
Barometer: Mercury _____ Aneroid _____ Other _____
Barometer capable of measuring atmospheric pressure within 2.5 mm Hg? _____

Operation

Filter properly numbered and recorded on field test data sheet? _____
Filter type: Glass? _____ Quartz-fiber? _____
Filter has organic binder (Method requires no organic binder)? _____
Filter manufacturer _____ Filter manuf. Number _____
Filter weighted to constant weight/recorded? _____
Filter checked visually for irregularities? _____
Filter collection efficiency meets EPA's specification of 99.95 %? _____
Stopcock grease used is solvent-insoluble, heat-stable silicone grease? _____
Sampling site selected? _____ Stack temperature _____
Stack velocity head? _____ Stack dry molecular weight calculated? _____
Nozzle size properly selected? _____
(Note: During sampling run, nozzle can not be changed!)
Calibration data certified available: Pitot tubes _____ Meter box Y factor _____
Meter box delta H@ _____ Magnehelic guage _____
Thermocouples _____ Probe nozzle _____

Nozzle condition: Nicks _____ Bent opening _____

PROBE/PROBE LINER

Probe material: Borosilicate _____ Quartz _____ Stainless steel _____

Inconel _____ Other _____

Heating system: _____ Temperature (120 C +/- 14 °C) _____

Water cooled? _____ Length _____

Probe liner: Borosilicate/quartz (up to 480 °C) _____ Quartz(480-900 °C) _____

Temperature sensor attached to probe? _____

Temperature sensor placed in an interference-free arrangement with probe? _____

PITOT TUBE

Type: Type S _____ Standard _____ Other _____

Properly attached to probe? _____

Meets EPA guidelines for attachment to probe? _____ Yes _____ No _____

Modifications _____

Pitot tube coefficient _____

FILTER HOLDER

Construction: Borosilicate _____ Quartz _____ Other _____

Frit construction: Glass frit? _____ Teflon frit? _____

Filter support _____ Silicone gasket _____

Other _____ Clean? _____

Cyclone and flask used(optional): _____

FILTER HEATING SYSTEM

Description _____

Temperature (120+/- 14 °C) _____

Gas temperature at exit of Box _____

ORGANIC SAMPLING MODULE

Three sections: Condenser _____ Sorbent trap _____

Condensate knockout trap _____

Condenser maintains stack gas not exceeding 20 °C (68 °F)? _____

Sorbent trap jacketed to maintain internal gas temperature not exceeding 20°C (68°F)? _____

Design of sorbent trap: Single unit? _____ Dual units? _____

Thermocouple well in bed of sorbent Trap? _____

Sorbent in trap? (XAD-2?) _____ 20 grams? _____

Sorbent certified clean (< 10 ng per target analyte)? _____

Date of sorbent certification? _____ Within 4 weeks? _____

Chain-of-custody with sorbent trap? _____

Sorbent trap received wrapped in aluminum foil? _____

Coolant? (Ice water?) _____

Knockout trap directly below sorbent trap? _____

Type of knockout trap? _____

SOURCE SAMPLING FOR HAZARDOUS AIR POLLUTANTS (HAPs)

FIELD OBSERVATION CHECKLIST

SW-846, METHOD 0010

**SAMPLING AND ANALYSIS
FOR
SEMI-VOLATILE ORGANIC COMPOUNDS**

SOURCE INFORMATION

Facility _____
Location _____
Source _____
Sampling Location _____
Test Team _____
Team Leader _____
Test Date Interval: ____ / ____ / ____ to ____ / ____ / ____
Observer _____
Affiliation _____

PRETEST MEETING CHECKLIST

Please reference Pretest Checklist

FEDERAL REFERENCE METHODS 1 THROUGH 5

Please reference Federal Reference Methods 1 through 5 Checklist.

SW-846, METHOD 0010

**SAMPLING AND ANALYSIS FOR
SEMI-VOLATILE ORGANIC COMPOUNDS**

Equipment identified matches pretest calibration list? _____

Apparatus:

NOZZLE

Probe nozzle: Stainless steel _____ Glass _____ Other _____
Probe design: Button-hook _____ Elbow _____ Straight tube _____
Tapered (30 angle)? _____
Nozzle size: _____ Calculated _____ Measured three times _____