



## Course Overview

- Volatile Organic Compound (VOC) Controls
- Examples of VOC Calculations
- Particulate Matter (PM) Options
- Inspection Strategies



## Volatile Organic Compounds

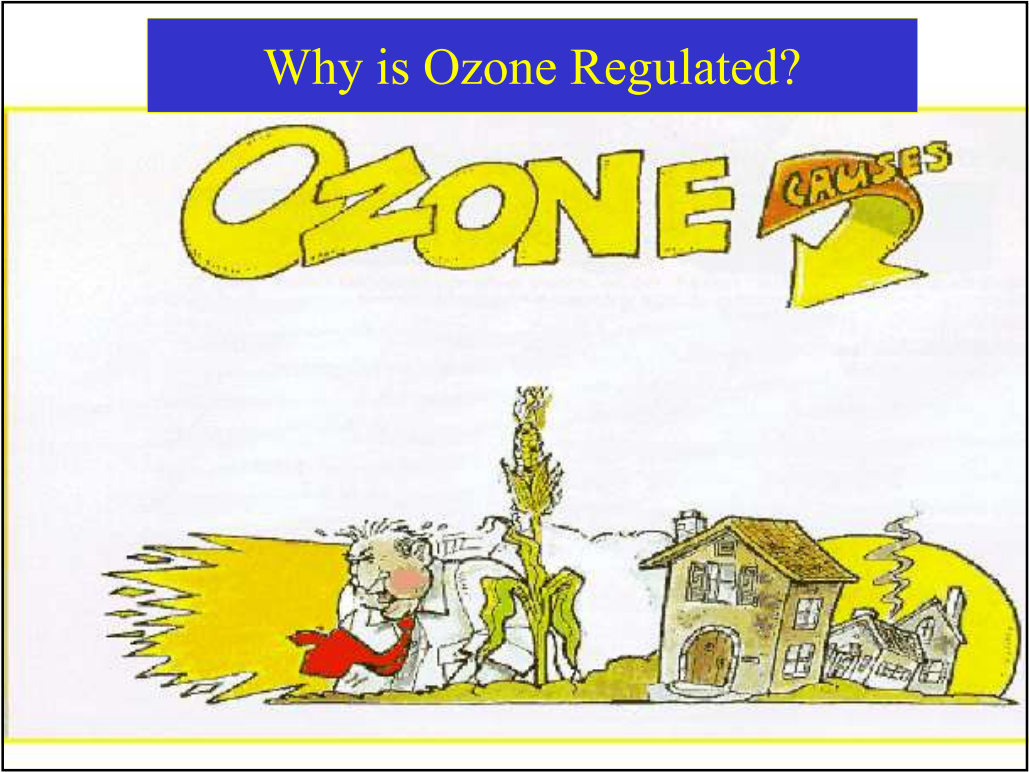
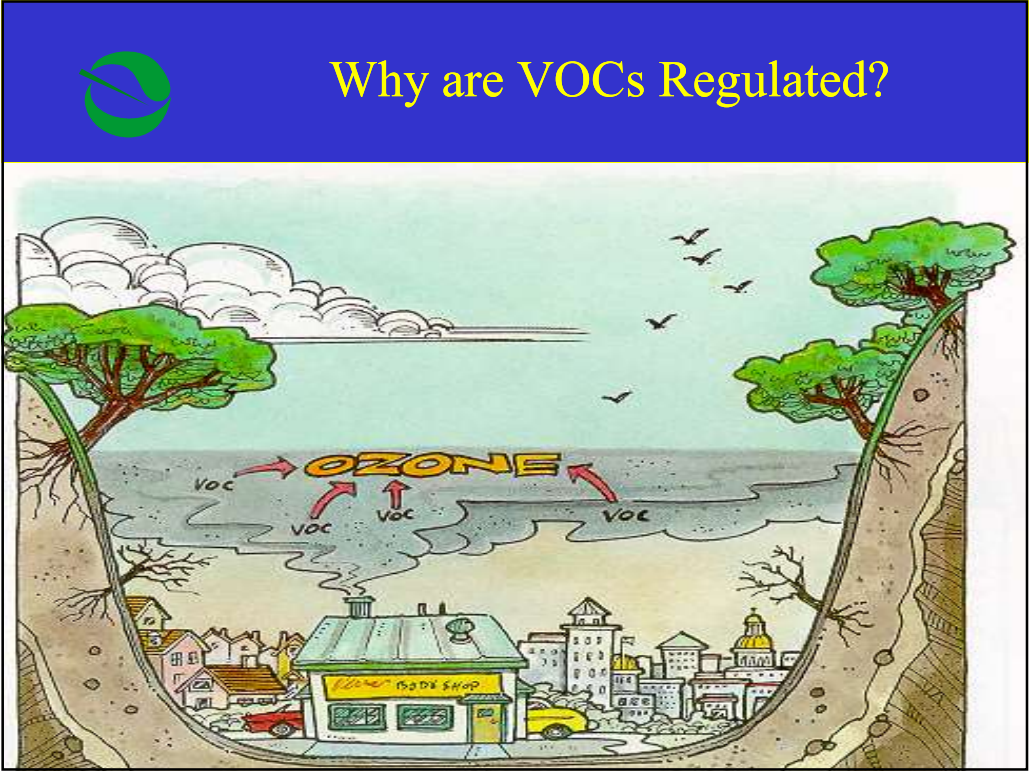
### Chemical definition of VOCs:

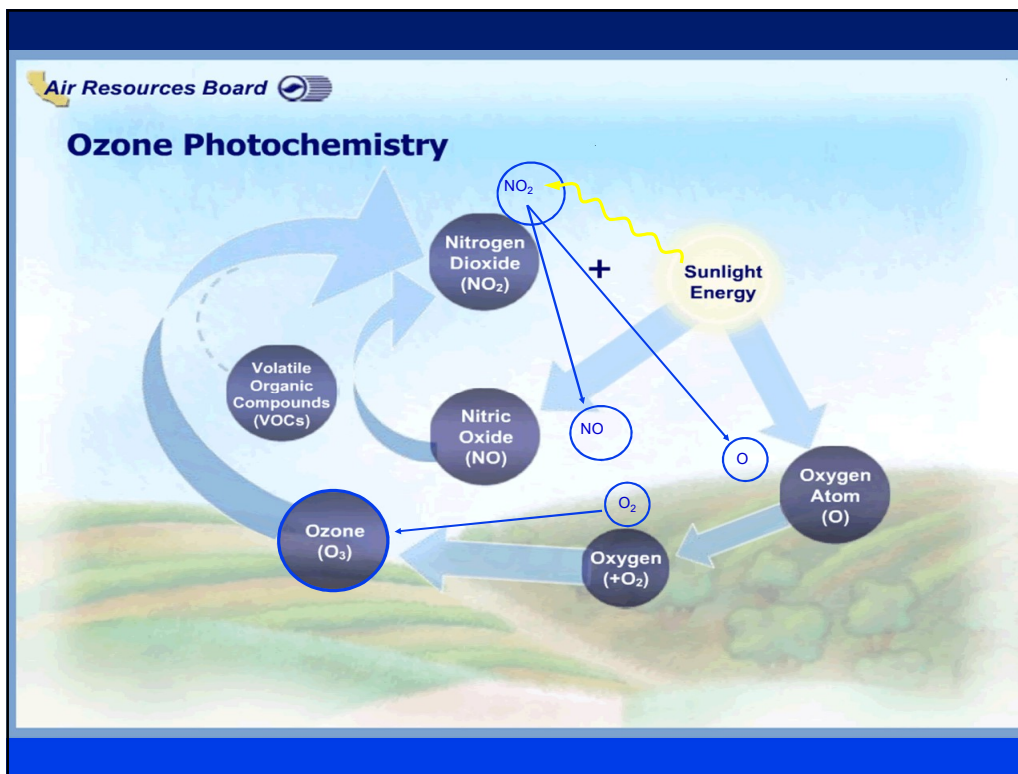
- Molecules which contain carbon  
&
- High evaporative rate at low temperatures
- [  $VP > 0.1\text{mm Hg}$  ]



## Legal Definition of VOCs

- Federal and State laws & regulations
  - \* 40CFR51 § 51.100
  - \* Latest Definitions of VOCs and ROGs as of...
- Total Organic Gases (TOGs)
- Reactive Organic Gases (ROGs)
- Fraction of Organic Gases (FROGS)
- Local Agency rules and permit conditions



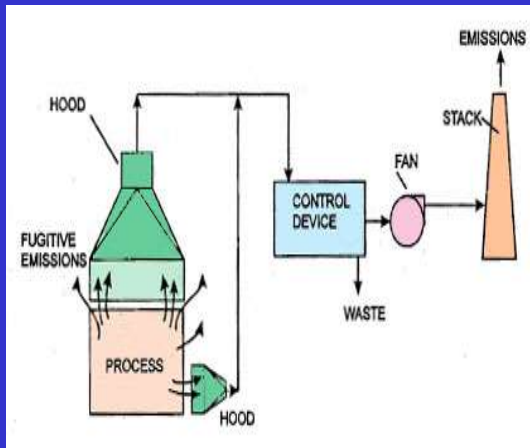


## Ozone Formation

- $\text{VOCs} + \text{NO}_x + \text{sunlight} > \text{O}_3$
- Ozone is formed when  $\text{NO}_x$  and Volatile Organic Compounds react in sunlight



## VOC Control Process



- Capture
  - Control
- Recovery,  
Disposal or  
Destruction



## VOC Calculations : Capture & Control & Retention

- General Categories of VOC Emissions
  - \* Fugitive (Not reasonably captured)
  - \* Captured > Ducted to control device
  - \* Consumed > Oxidized
  - \* Retained > Retention factors vary



## VOC Capture Efficiency \*

$$\text{VOC Capture Efficiency} = \frac{\text{VOCs captured}}{\text{VOCs used}} \times 100$$

**VOCs used (and therefore emitted) 100 lbs**

**VOCs captured (entering control device) 80 lbs**

**VOC capture efficiency (by calculation) ?????**

*\* Capture Efficiency is the percentage of emissions captured and vented to a control device. -- EPA*



## VOC Capture Efficiency \*

$$\text{VOC Capture Efficiency} = \frac{\text{VOCs captured}}{\text{VOCs used}} \times 100$$

**VOCs used (and therefore emitted) 100 lbs**

**VOCs captured (entering control device) 80 lbs**

**VOC capture efficiency (by calculation) 80%**

*\* Capture Efficiency is the percentage of emissions captured and vented to a control device. -- EPA*

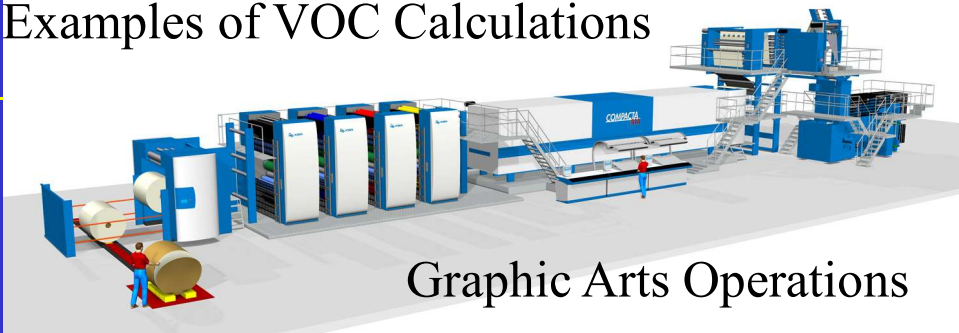


## VOC Control Efficiency

$$\% \text{ CE} = \left[ 1 - \frac{\text{outlet emission rate}}{\text{inlet emission rate}} \right] \times 100$$

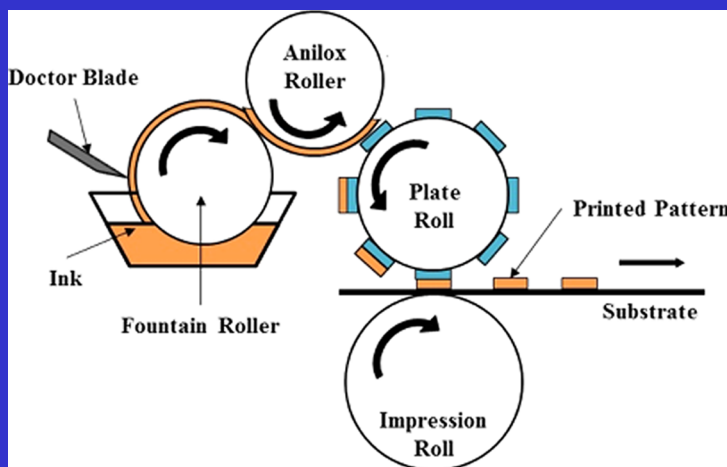
$$\% \text{ CE} = \left[ 1 - \frac{2 \text{ lbs/hr}}{100 \text{ lbs/hr}} \right] \times 100 = 98$$

### Examples of VOC Calculations





## Graphic Arts Operation



## VOC Calculations

- A facility uses 100 lbs/hr of ink that has a VOC content of 35% by weight.
- 20% of the VOC is retained in the substrate
- The incinerator has a 95% control efficiency



How many lbs/hr of VOC is emitted?

$$\text{VOC Emissions} = (100 \text{ lbs/hr}) (0.35) (1-0.20)(1-0.95) = 1.4 \text{ lbs/hr}$$





## VOC Calculations

### Including W/E

$$\text{VOC g/l} = \frac{W_m - W_w - W_{ec}}{V_m}$$

### Excluding W/E

$$\text{VOC g/l} = \frac{W_m - W_w - W_{ec}}{V_m - V_w - V_{ec}}$$

$W_m$  = Weight of material, as applied, in grams

$W_w$  = Weight of water, in grams

$W_{ec}$  = Weight of exempt compounds, in grams

$V_m$  = Volume of material applied, in liters

$V_w$  = Volume of water, in liters

$V_{ec}$  = Volume of exempt compounds, in liters

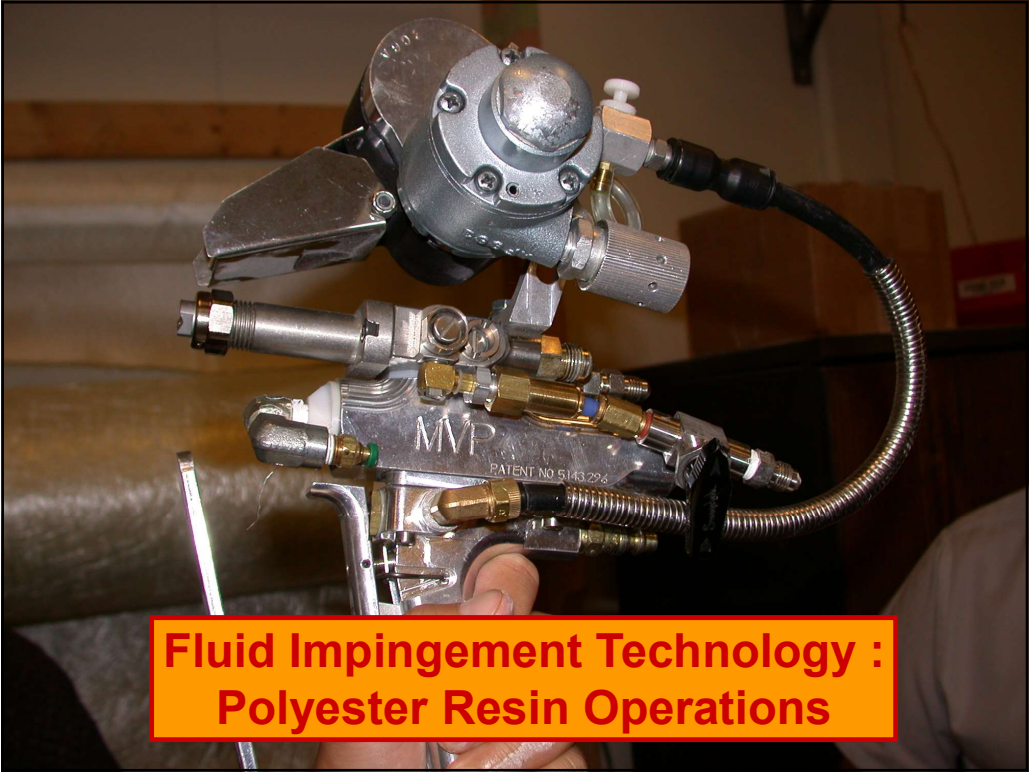
## Let's Discuss Control of VOC

- Containment
- Transfer Efficiency
- Absorption
- Adsorption
- Condensation
- Oxidation






**(HVLP) Spray Gun :  
Polyester Resin Operations**



**Fluid Impingement Technology :  
Polyester Resin Operations**

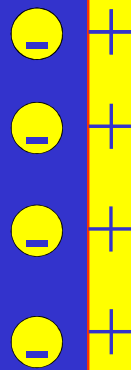



 **Adsorption Mechanism**

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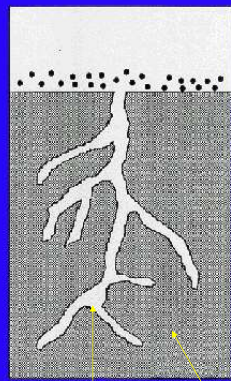
**Gas**

**Solid surface**

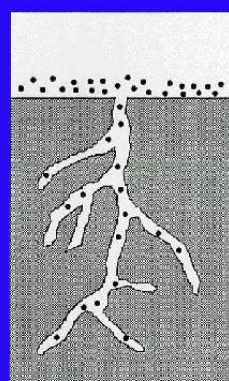


 **Adsorption Mechanism**

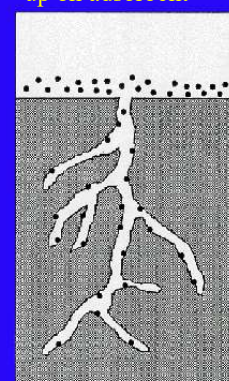
Step 1. VOC diffuses to adsorbent surface



Step 2. VOC migrates into pores



Step 3. VOC adsorbed and builds up on adsorbent



Pore     Carbon     • VOC molecule



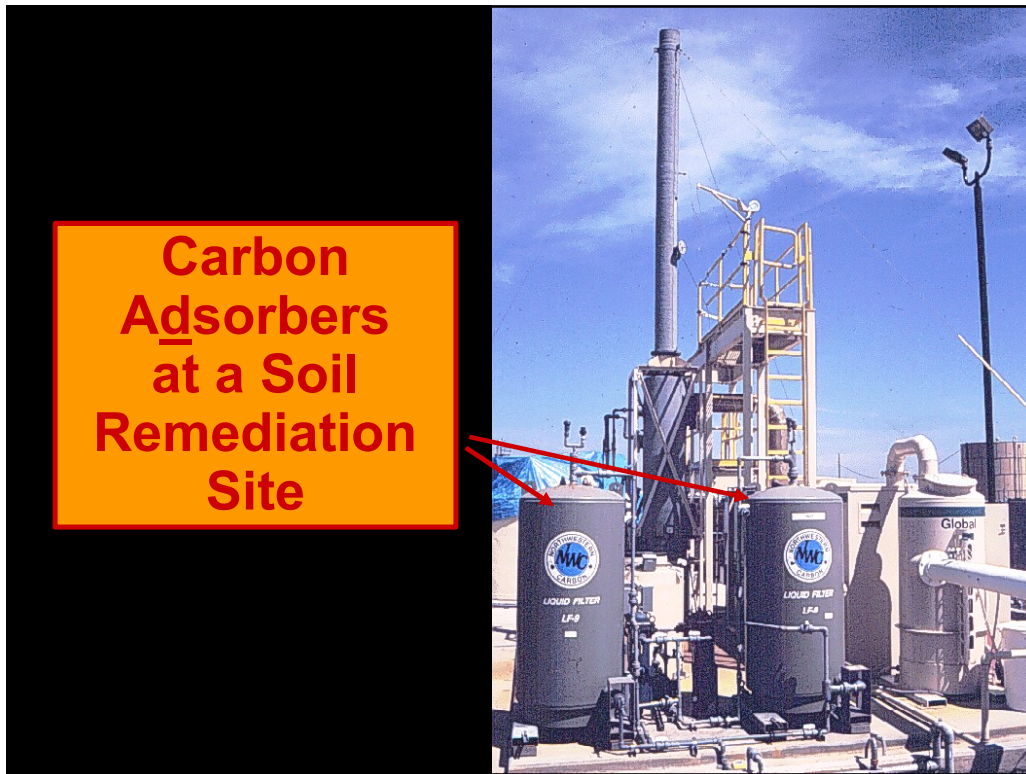
## Adsorption Mechanism

- Chemically unchanged
- Desorbed and recovered
- Polar and non-polar adsorbates
- Mixed adsorbates separated by distillation



## Adsorption

- Adsorption materials (adsorbents)
  - \* Activated carbon
  - \* Hydrous oxides
    - Silica gel
    - Aluminum oxide
    - Magnesium silicate
  - \* Zeolites (molecular sieves)
  - \* Naturals
    - Clays
    - Bauxite
    - Fuller's Earth
  - \* Metals



## Factors Affecting Adsorption

- Temperature
- Pressure
- Gas velocity
- Particulate matter

## Adsorber Design Considerations

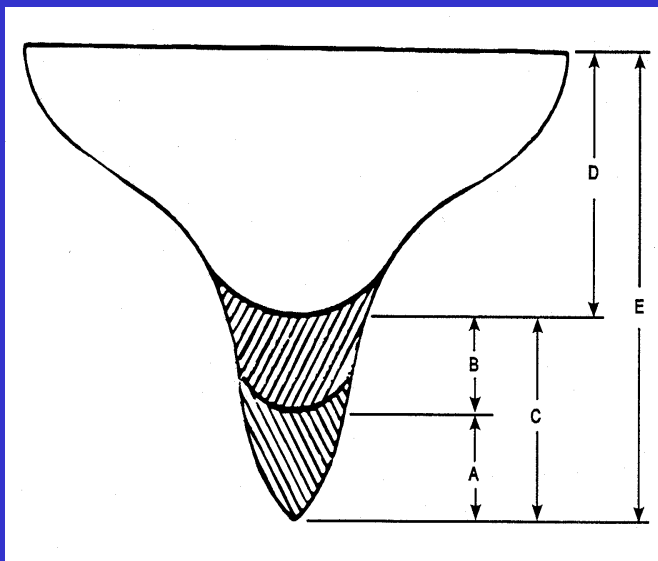


- ◆ Porosity of Adsorbent
- ◆ Bed Cross-Sectional Area
- ◆ Bed Length
- ◆ Multiple Organic Compounds
- ◆ Steaming Requirements
- ◆ Fouling
- ◆ Timers/Monitors
- ◆ Channeling



## Pore Space Representation

- A = Residual VOCs or heel
- B = Working capacity
- C = Equilibrium Capacity
- D = Empty pore space
- E = Total pore space (total capacity)







## Carbon Adsorption Keywords

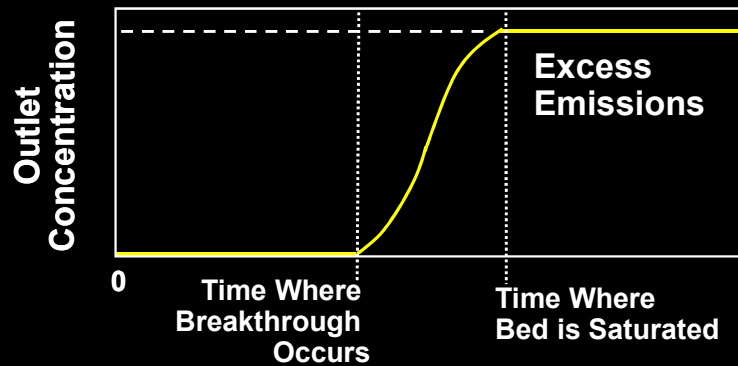
- Fresh zone
  - \* Area where adsorption will occur
- Mass transfer zone
  - \* Where adsorption occurs
- Saturated zone
  - \* Area where adsorption has already occurred



## Keywords (continued)

- Heel
  - \* Amount of VOCs left in the carbon after regeneration
- Breakthrough
  - \* VOCs that do not get captured

## Adsorber Breakthrough



## Types of Adsorption Systems

\*Non-regenerative systems

\*Regenerative systems

- on site
- off site



## Characteristics of Activated Carbon

- Sources
  - \* Wood, coal, peat, nut shells
- Porosity
  - \* 600-1600 m<sup>2</sup>/g (2-3 football fields per 1/28 ounce)
- Preparation
  - \* Anaerobic heat then steam or CO<sub>2</sub>,
- Degree of adsorption depends on adsorbate
  - \* MW, BP, polarity, surfactive index, solubility



## Examples of Activated Carbon





Finely Granulated Carbon



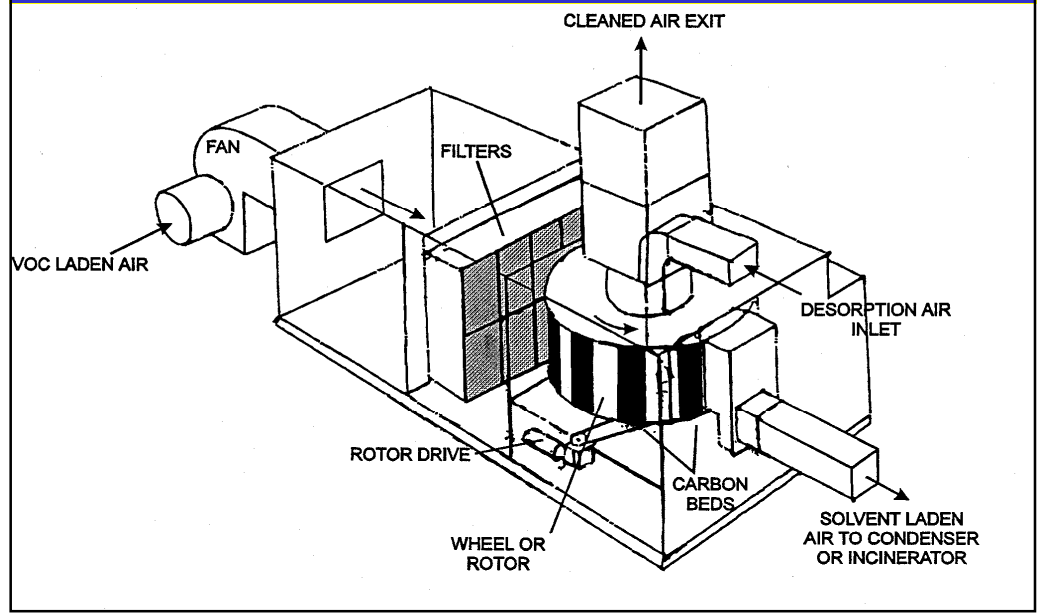
## Types of Carbon Adsorption Systems

- Open
- Closed
- Rotary
- Fluidized bed
- Bulk plant adsorber and absorber

# Bulk plant adsorber & absorber

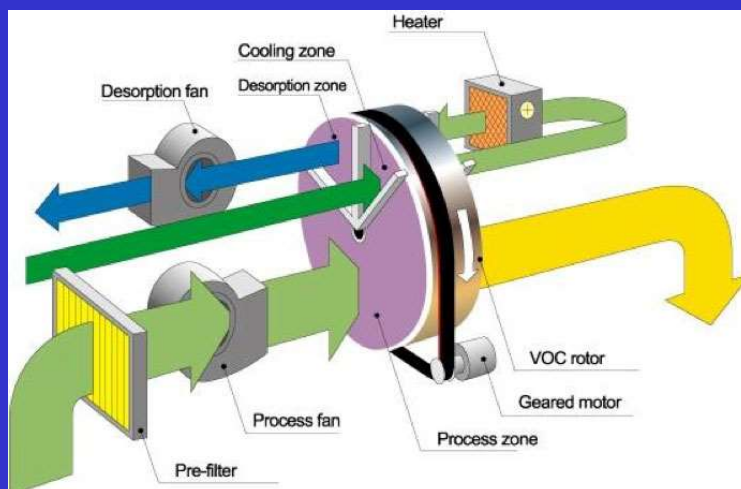


# Rotary Concentrator Adsorption System





## Rotary Concentrator Adsorption System



## Adsorber Inspections

- Hood static pressures
- Inlet VOC concentrations
- Inlet temperatures
- Inlet VOC concentration not  $> 25\%$  LEL
- Outlet VOC concentrations
- Fan motor current
- Solvent recovery rates



## Absorbers



- Pollutants dissolved in liquid
- Absorbate dissolves in absorbent



## Factors Favoring Absorption

- Pollutant solubility in liquid
- Adequate diffusion at liquid / gas interface
- Maximized contact between gas and liquid



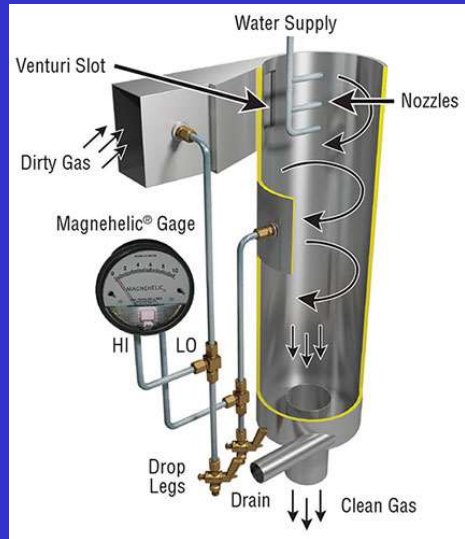
## Absorber Design

- Produce large surface area
- Minimize air flow resistance to reduce pressure drop
- Inlet pressure - outlet pressure = pressure drop

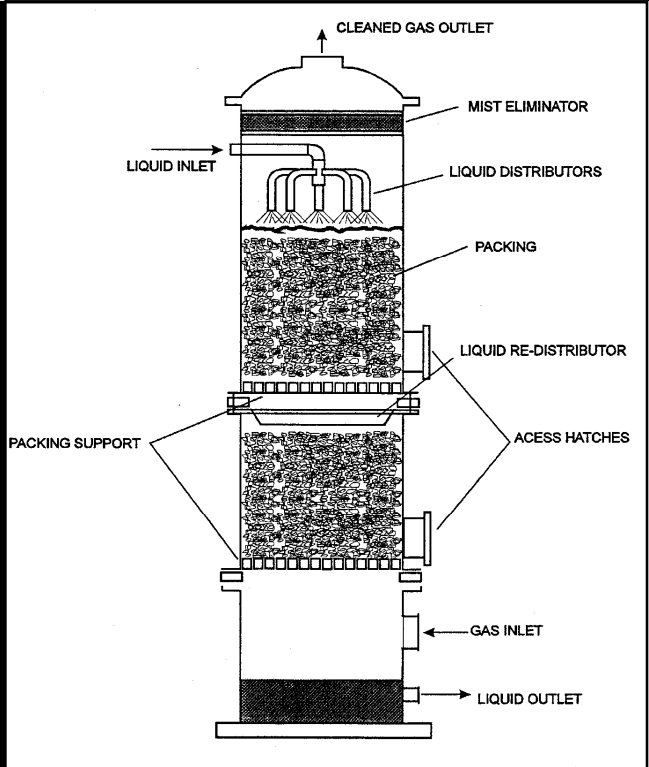




# Pressure Drop : Magnehelic



## Packed Bed Wet Scrubber





## Absorber Design Factors

- Select liquid solvent
- Column material
- Column size
- Column height
- Number of plates
- Pressure drop

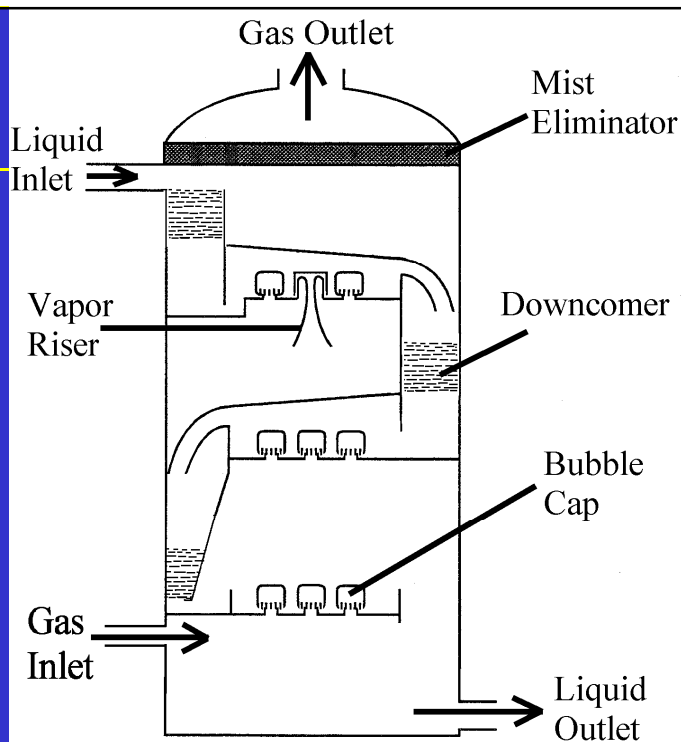


## Absorbers: Packed Columns

- Flow patterns
- Liquid reuse and treatment
- Packing material
- Packing quality



## Absorbers: Plate Columns





## Absorbers: Plate Columns

- Maximize contact between liquid & gas
- Diameter of column
- Plates
  - \* Number
  - \* Type
  - \* Layout



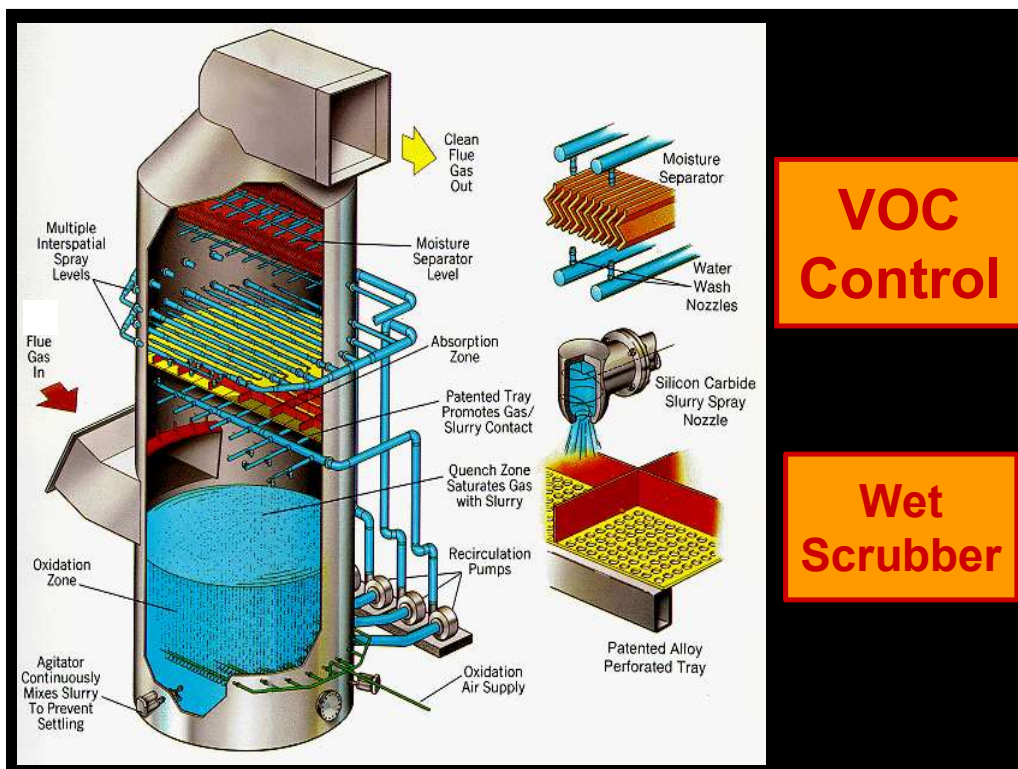
## Packed vs Plate Columns

- Packed columns
  - +More common
  - Plugged by particles
  - +Better for corrosive pollutants
  - +Lighter than plate



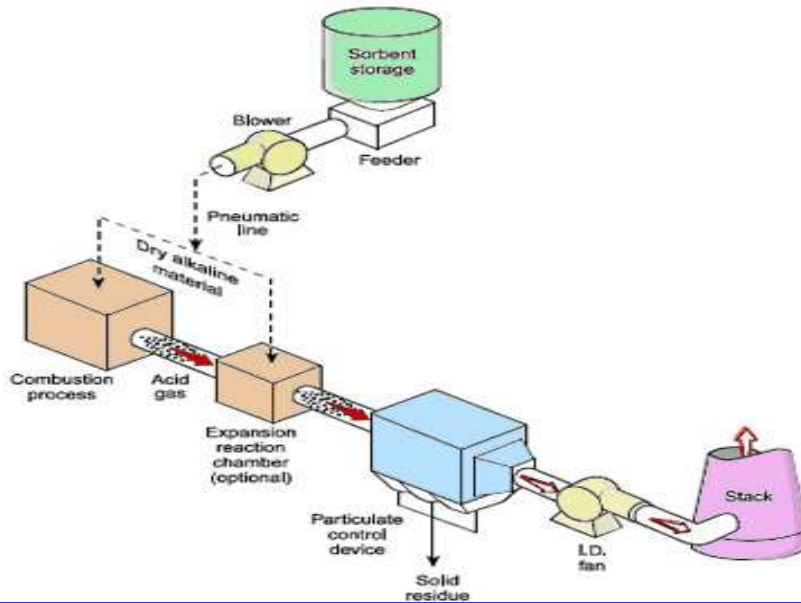
## Packed versus Plate Columns

- Plate columns are better for:
  - + Large temperature changes
  - + Lower liquid flow rates
  - + Higher gas flow rates
  - + Foaming liquids
  - + Chemical reactions
  - + Large systems

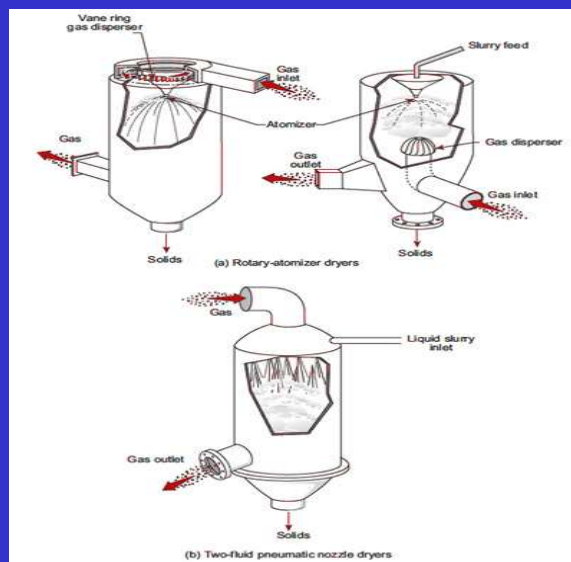




# Dry Sorbent Injection

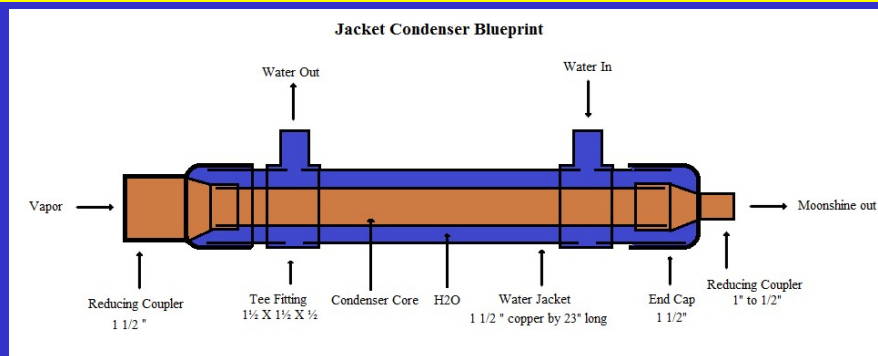


# Spray Dryer Absorbers





## Condensers : Surface & Contact



- Condensation = Process of changing a gas to a liquid.
- Condensation allows recovery of solvents and air pollution control



## Condensers

Vapor  $\xrightarrow{\text{Cold}}$  Liquid

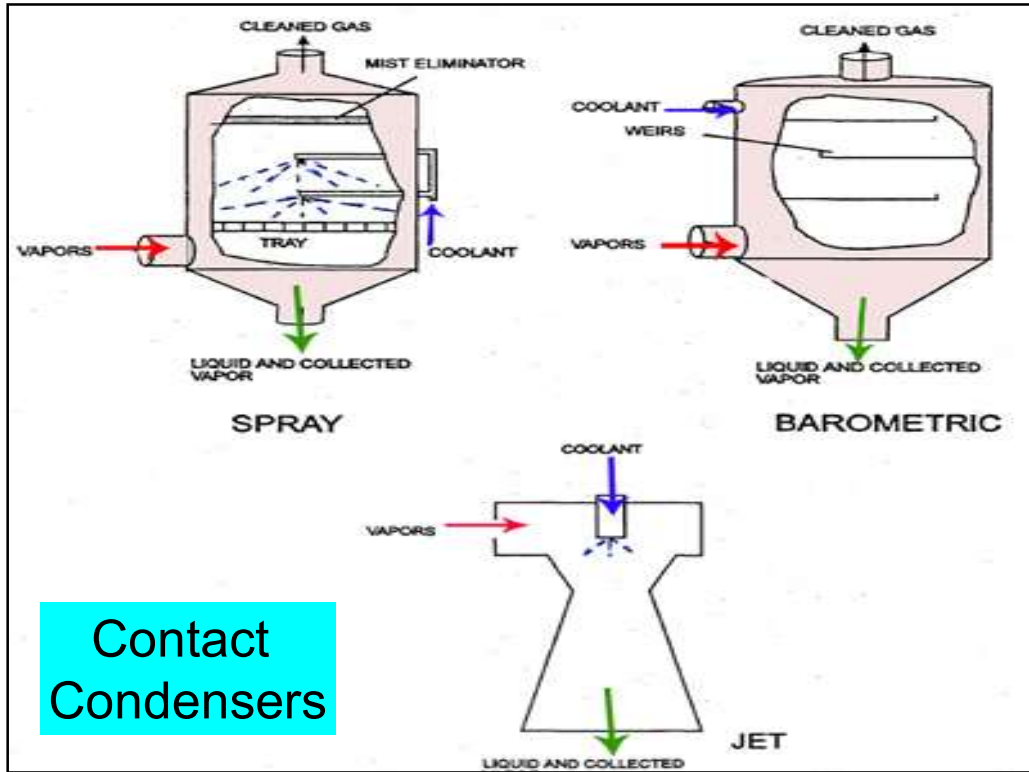
- Condensation = Process of changing a gas to a liquid.
- Condensation allows recovery of solvents and air pollution control
- Decrease temperature, increase pressure



## Contact Condensers

- Contact condensers +/-
  - + Cheaper
  - + More flexible
  - + Less repair time
  - Wet waste disposal problem





## Contact Condensers

\*Spray

\*Jet

\*Barometric



## Surface Condensers

- \* Shell and tube  
(most common)
- \* Double pipe
- \* Fin Fan
- \* Spiral plate
- \* Tubular
- \* Flat plate

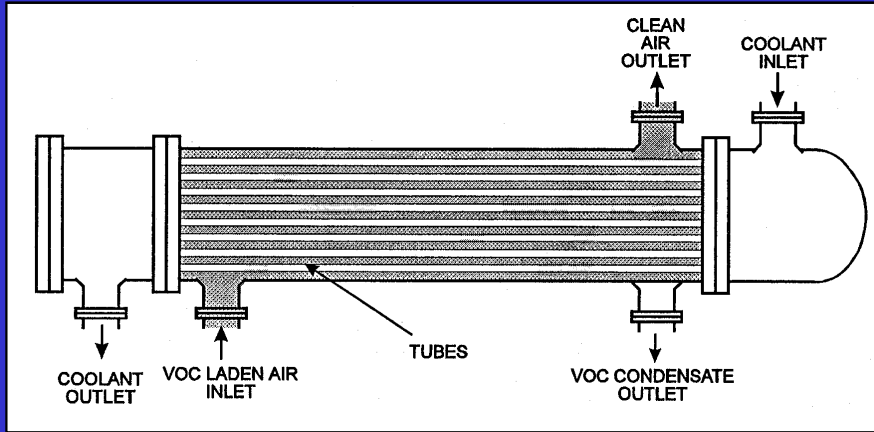


## Condensers

- Surface condensers +/-
  - + Better recovery
  - + Commonly used for air pollutants
  - + Reduced waste disposal problems
  - More costly



# Shell and Tube





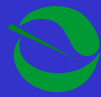
## Condenser Concerns

- Freezing
- Fouling
- Cleaning
- Pressure drop



## Condenser Concerns

- Vibration
- Leaking gaskets
- Leaking connections
- Start-up procedures
- Shut-down procedures



## Condenser Inspection

- Look for
  - Excessive corrosion and rusting
  - Leaking coolant or VOC
  - Excessive odors
  - Continuous emissions monitor



## Condenser Inspection

- Record
  - VOC outlet concentration
  - Waste stream flow rate
  - Condenser pressure drop
  - Coolant pressure
  - Coolant flow rate



**Let's Discuss Oxidizers**



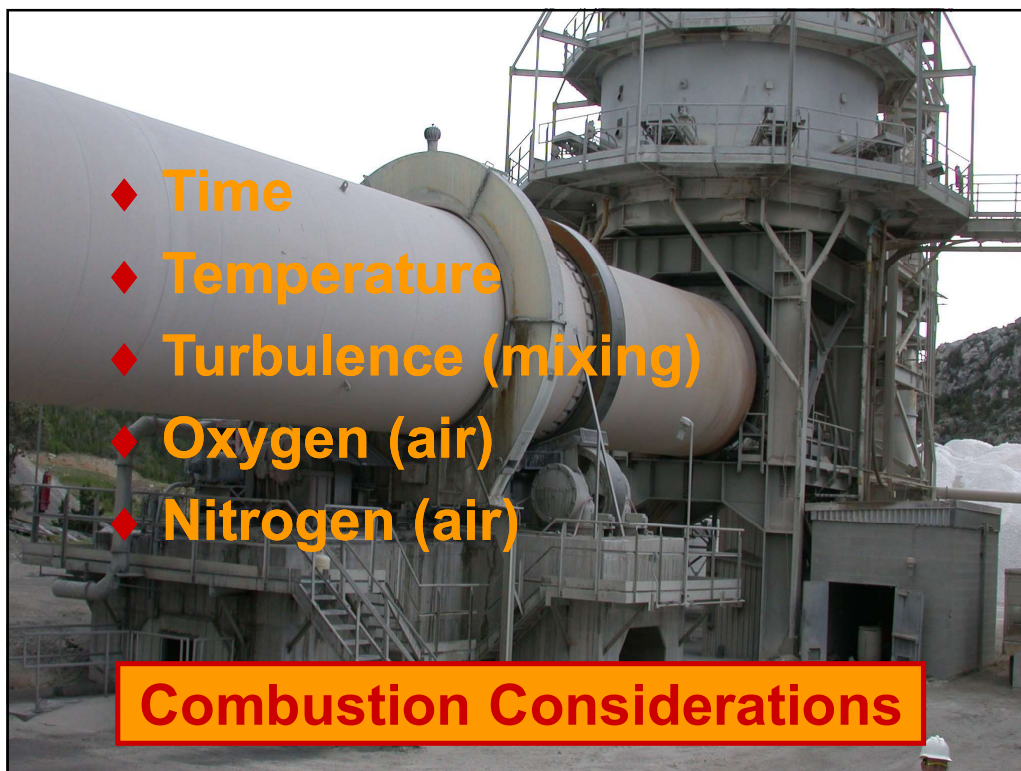
## Oxidation

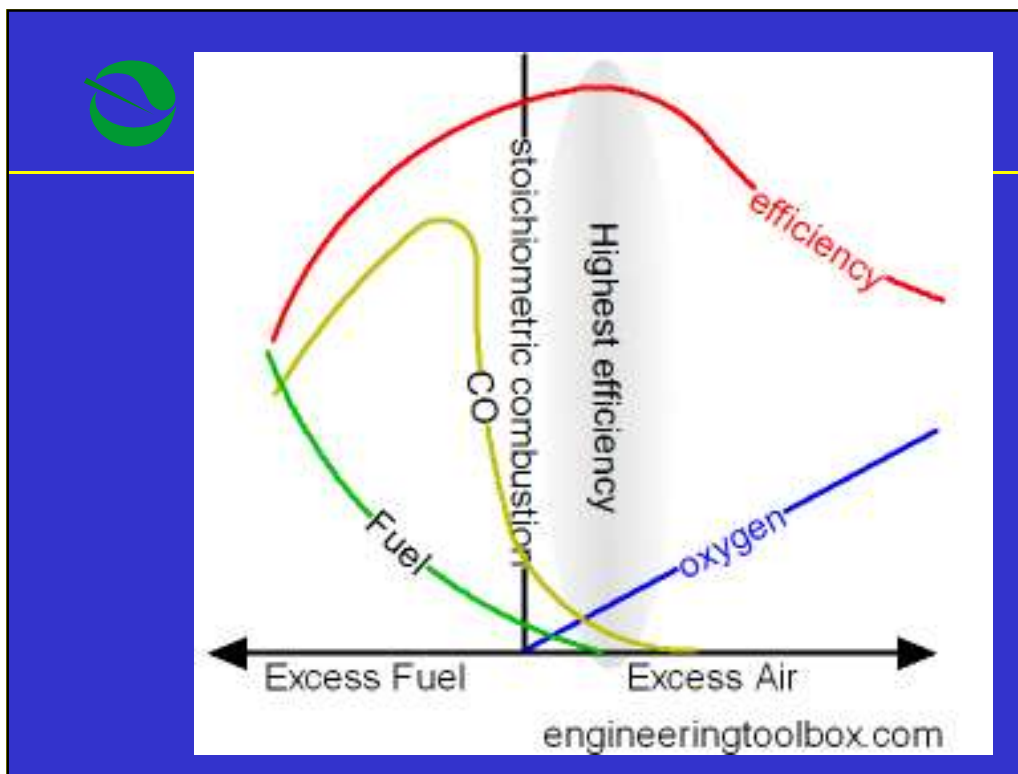
- Destruction of VOCs by Combustion

Reactions with oxygen



Toluene + Oxygen = Carbon Dioxide + Water

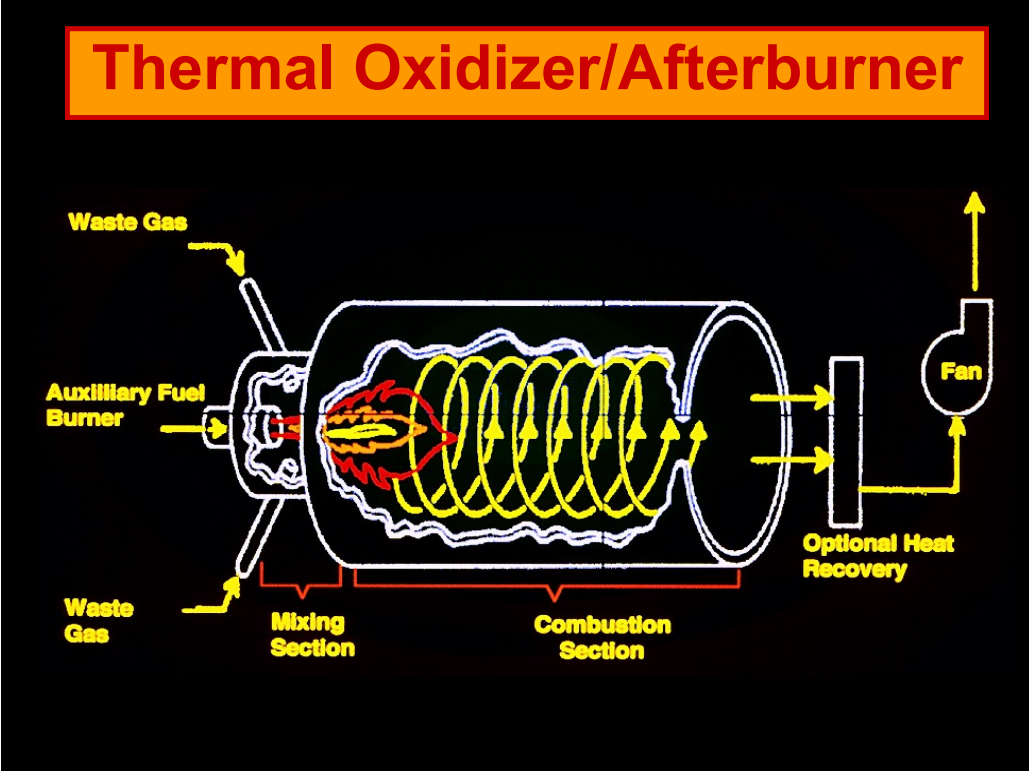


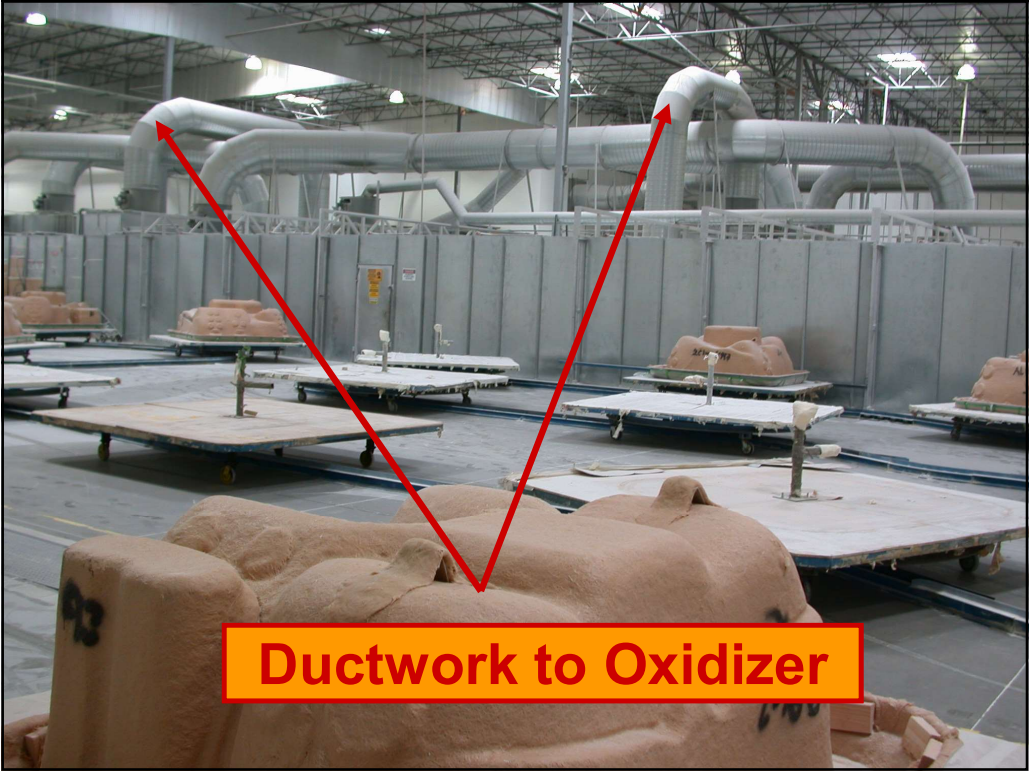


## Combustion Devices

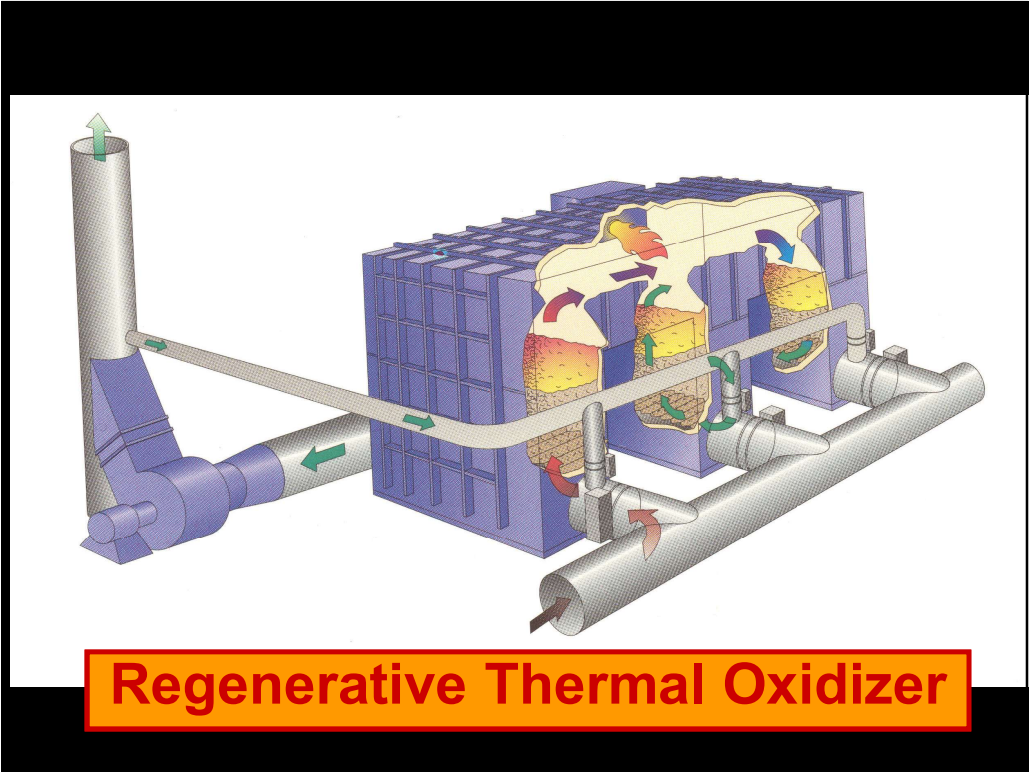
- Thermal incinerator (uses a flame)
- Catalytic incinerators (uses a catalyst)
- Boilers (burn VOCs to make steam)
- Process heaters (burn VOCs to add heat in chemical plants and refineries)
- Flares (simple flame)



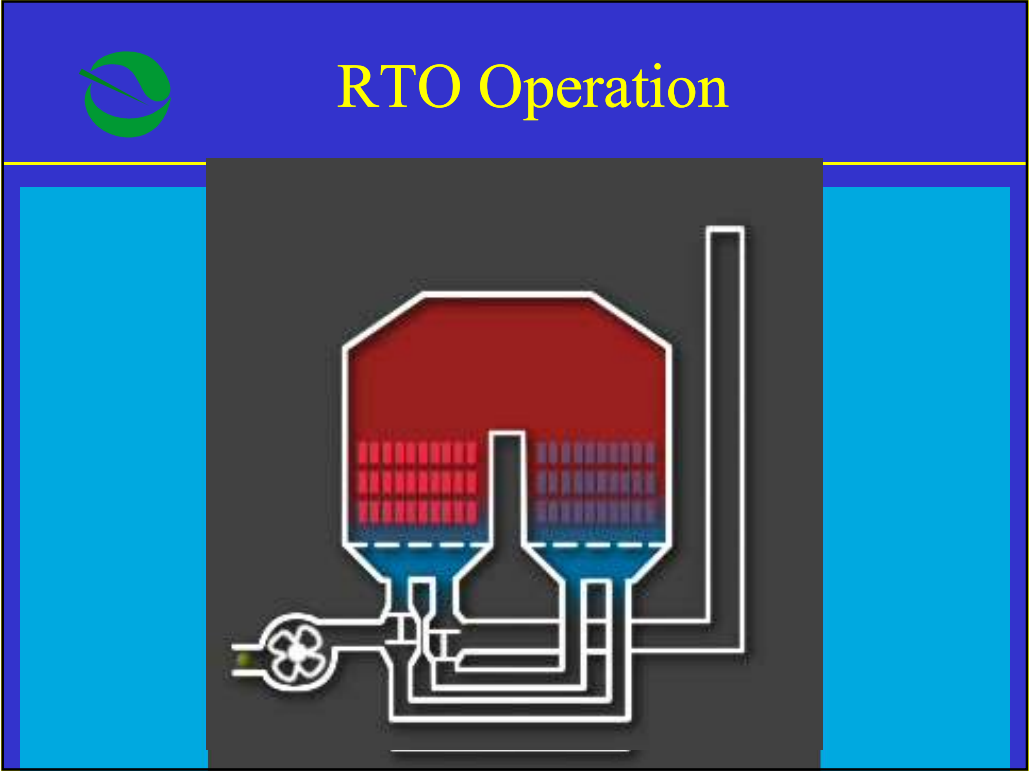


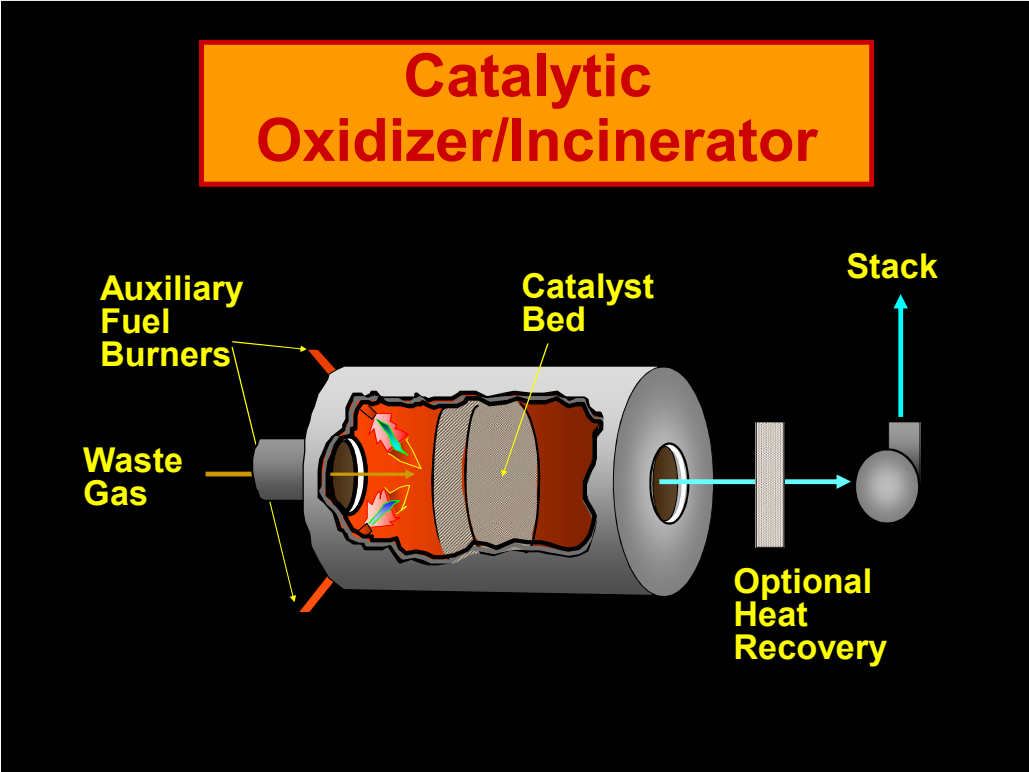
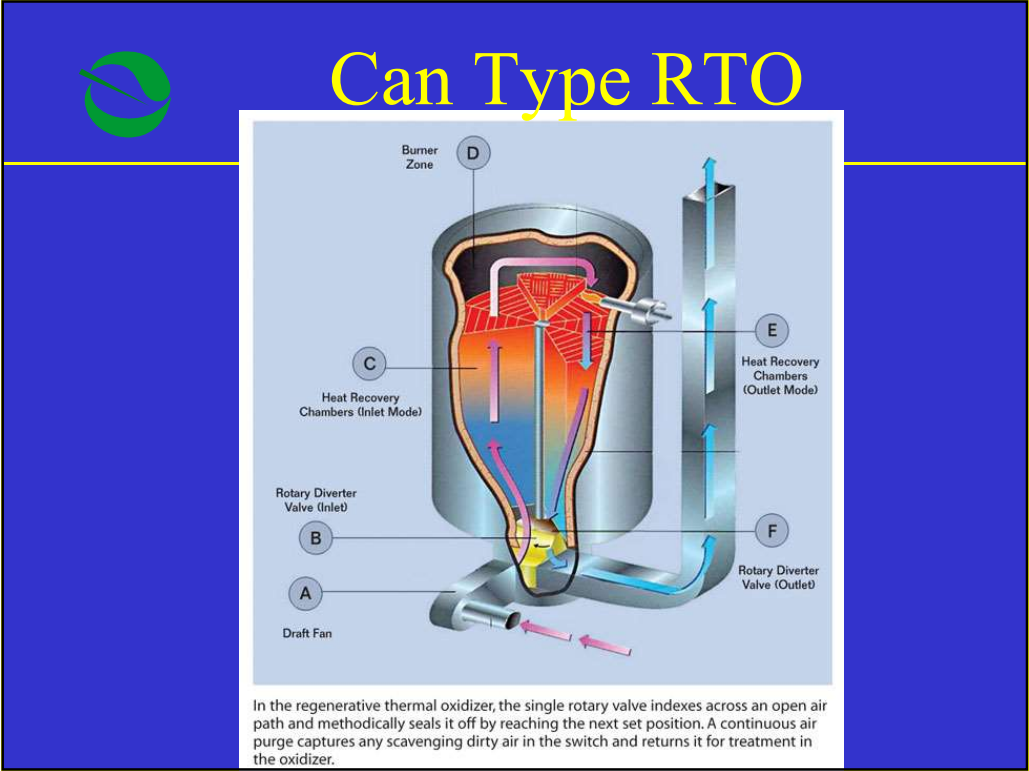


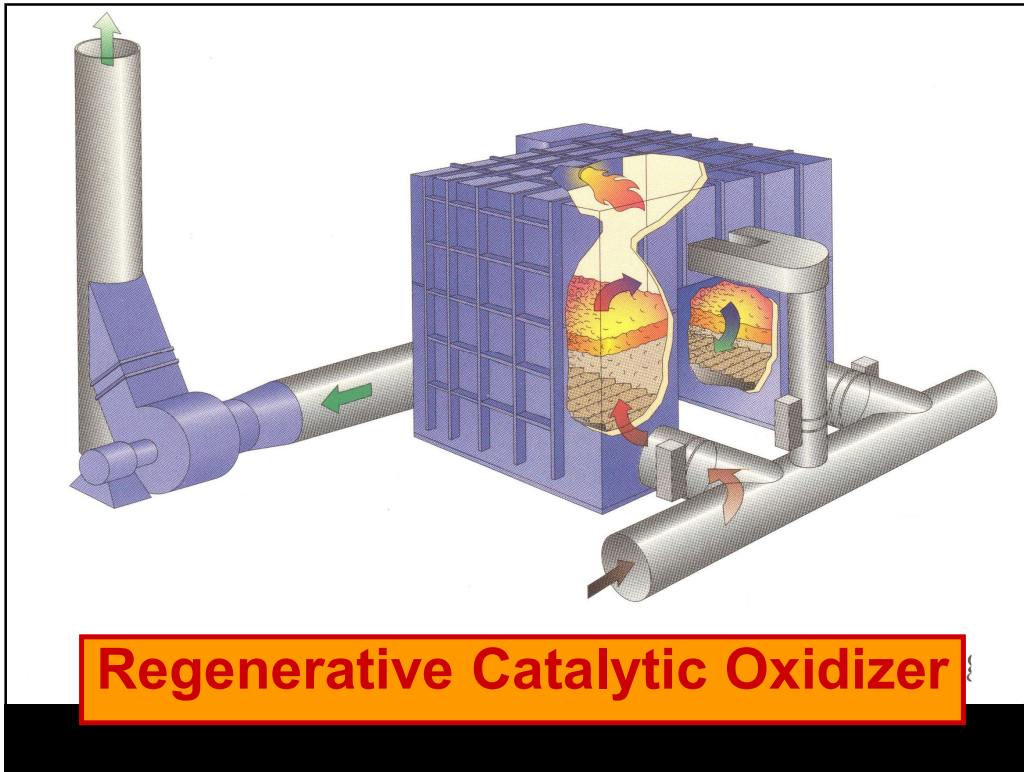
**Ductwork to Oxidizer**



**Regenerative Thermal Oxidizer**







## Selection Criteria

- Type of VOCs
- Concentration of VOCs
- Process flow rate
- Economics

## Catalytic vs. Thermal for VOC Control

Catalytic	Thermal
Lower Operating Temp. & Lower Fuel Usage	Higher Operating Temp. & Higher Fuel Usage
Higher Capital & Maintenance Costs	Lower Capital & Maintenance Costs
Catalyst Fouling & Poisoning	No Catalyst Involved Here



## Catalyst Problems

- Scouring
- Thermal burnout
- Thermal aging
- Masking
- Catalyst fouling and poisoning



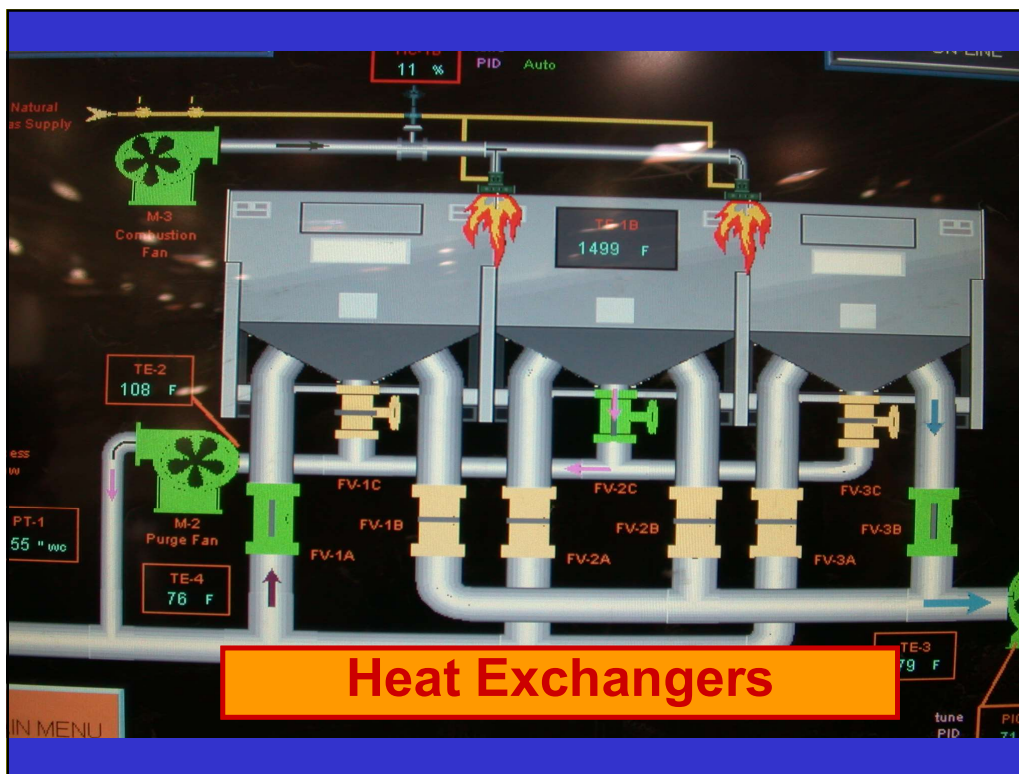
## Catalytic Poisons

- Fast acting poisons
  - \* phosphorus P, bismuth Bi, lead Pb, arsenic As, antimony Sb, mercury Hg
- Slow acting
  - \* iron Fe, tin Sn, silica Si
- Reversible
  - \* sulfur S, zinc Zn, chlorine, bromine, fluorine etc. halogens



## Catalyst Efficiency

- Operating temperature
- Space velocity
- VOC composition
- VOC concentration
- Catalyst properties
- Poisons and inhibitors

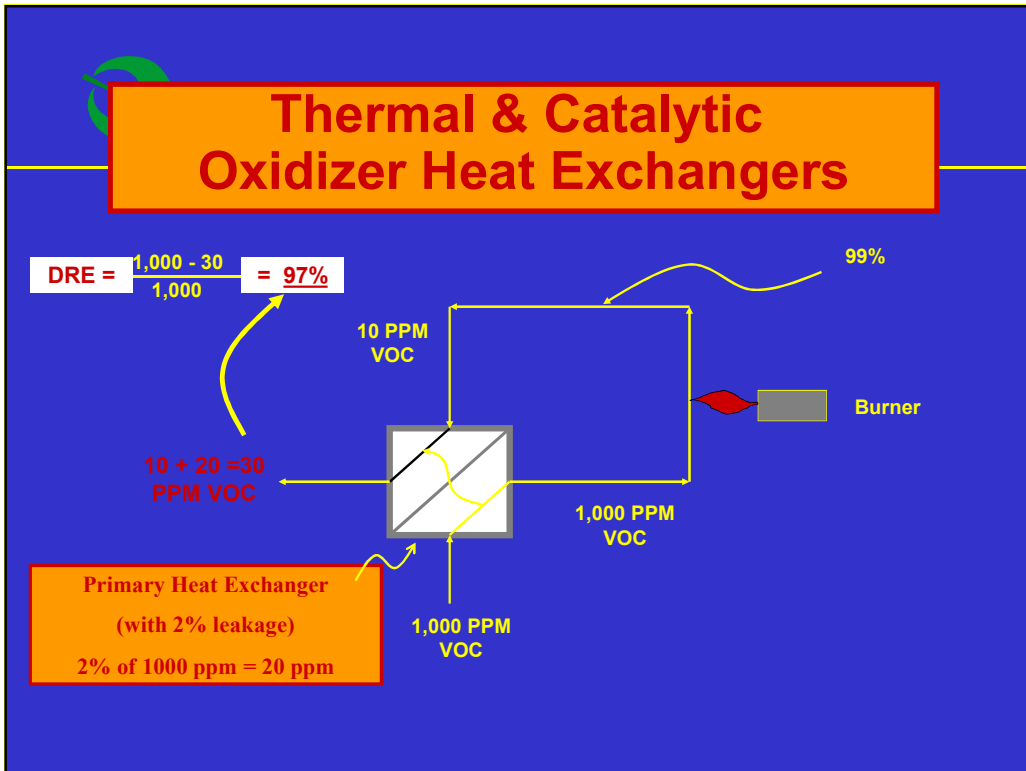
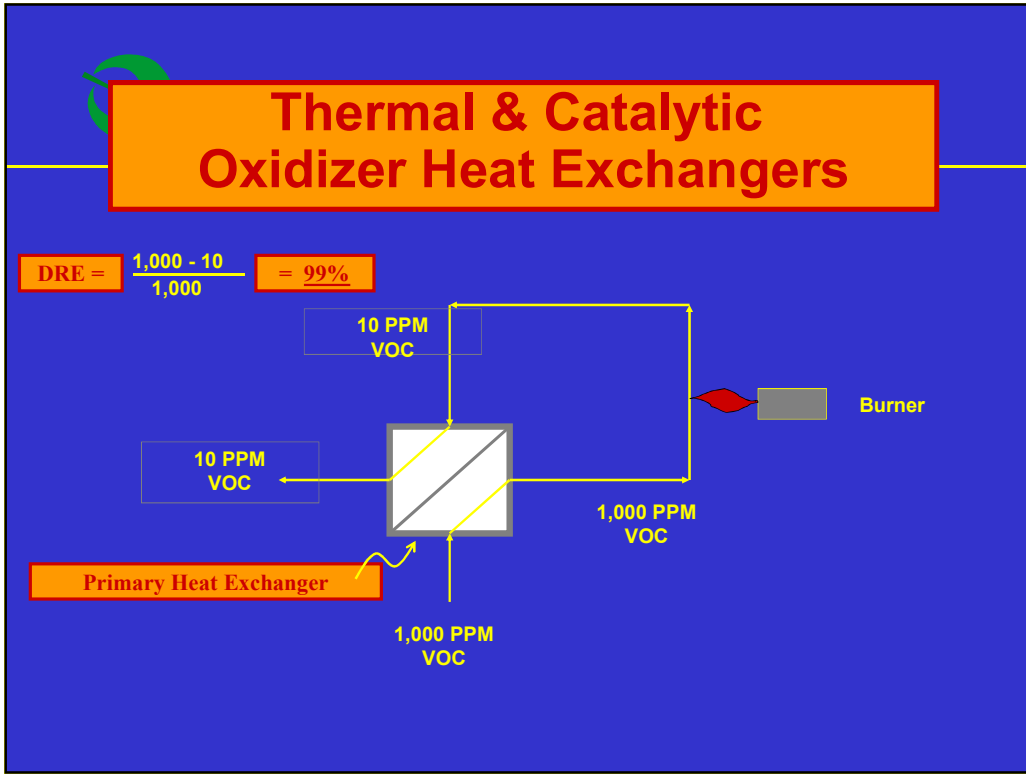


## Thermal & Catalytic Oxidizer Heat Exchangers

There are two basic types of heat exchangers used for thermal or catalytic oxidizers

- Metal Heat Exchangers or “recuperative heat exchangers”
- Ceramic Bed Heat Exchangers or “regenerative heat exchangers”

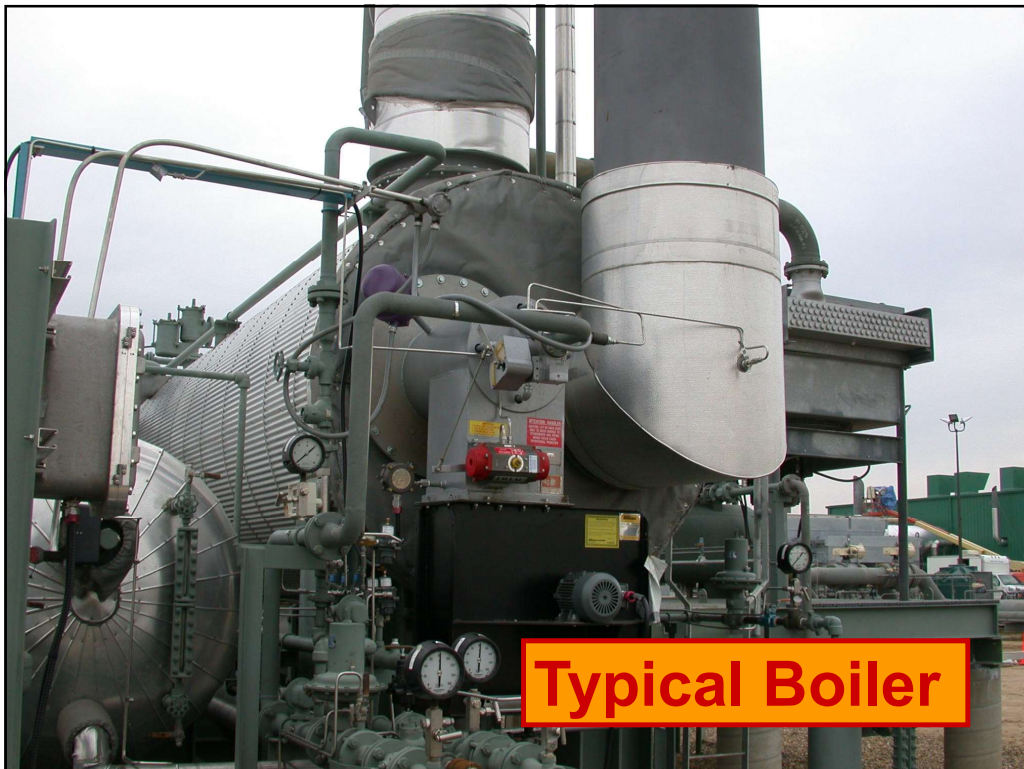






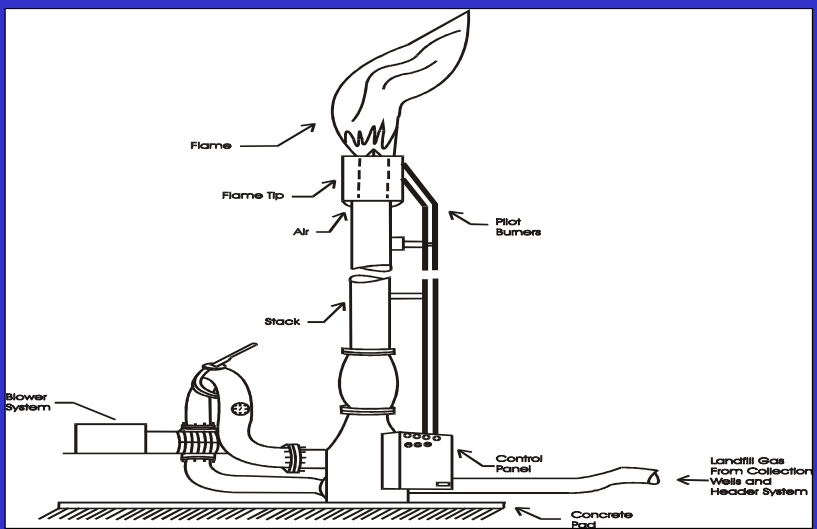
## Boilers, Process Heaters & Flares

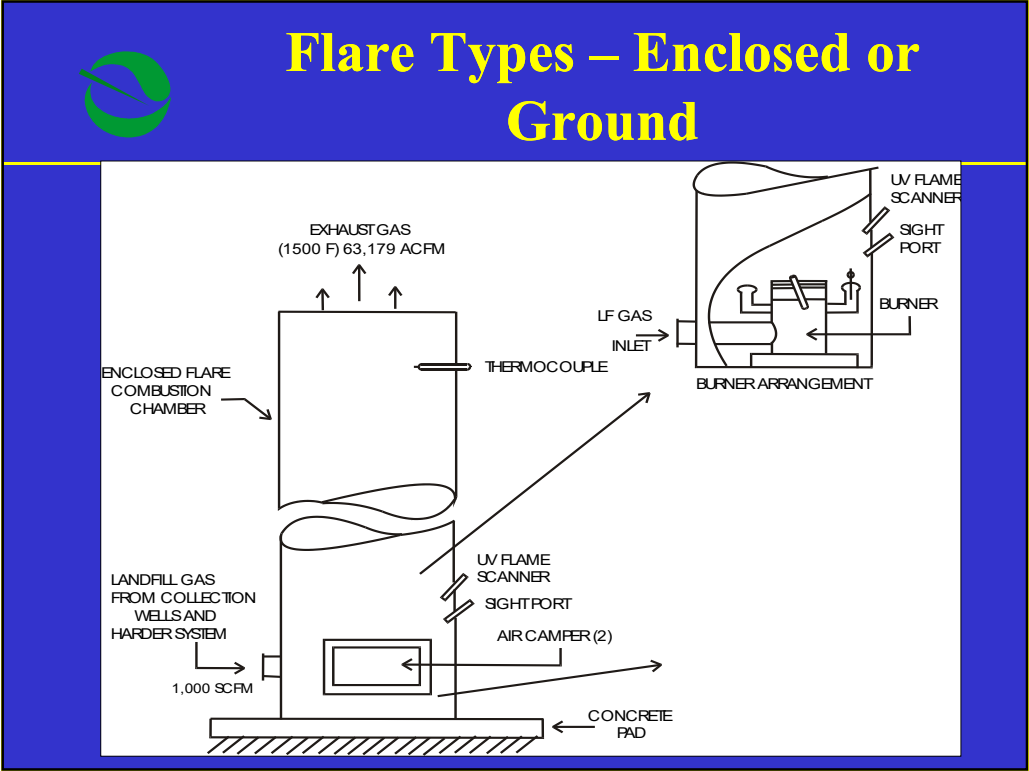
- Boilers make steam
- Process heaters add heat to material
- Flares are thermal incinerators without a combustion chamber





## Flare Types – Open or Elevated









Shell Deer Park Refinery in Texas on the Houston Ship Channel.



Flaring gases from an oil platform.





## Incinerator Inspection

- Look for
  - \* Excessive corrosion and rust
  - \* Holes in incinerator shell or ducts
  - \* Visible emissions
  - \* Excessive odors
  - \* Last time catalyst was replaced



## Incinerator Inspection

- Record
  - \* VOC outlet concentration
  - \* Incinerator inlet temperature
  - \* Incinerator outlet temperature
  - \* Pressure drop



# Venturi Scrubbers

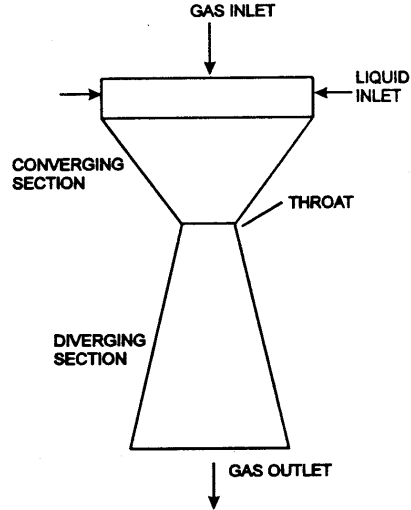


Figure 605.10 Venturi



# Venturi Scrubbers

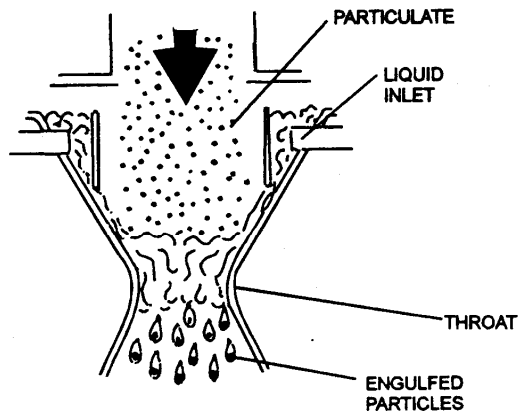
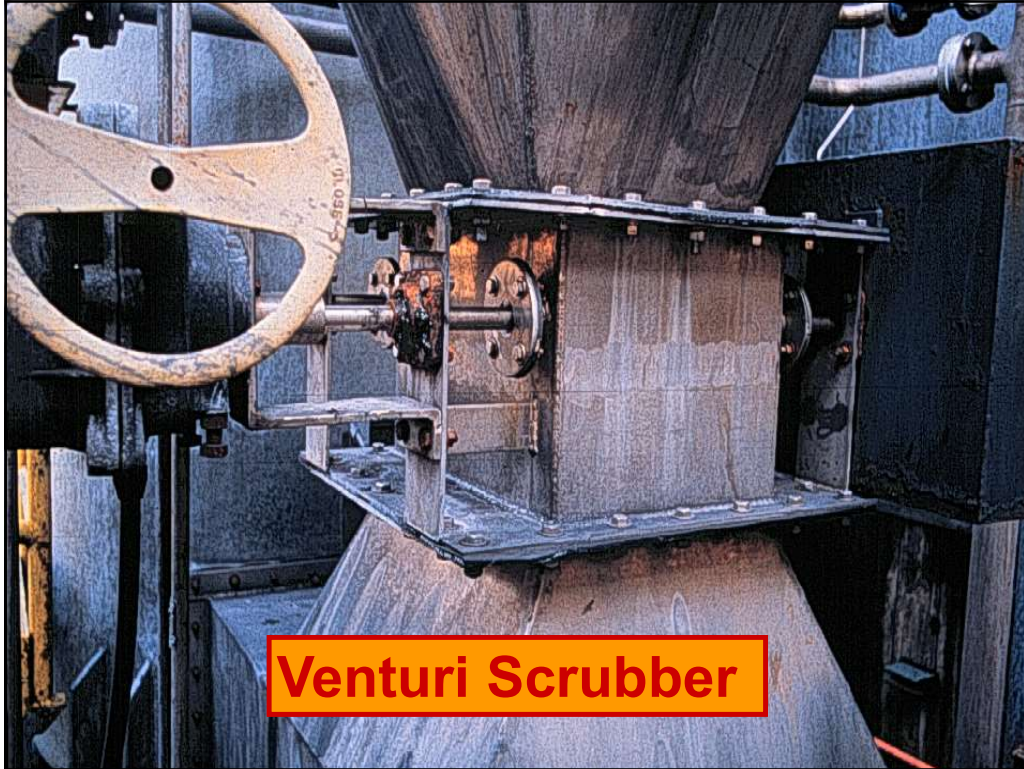


Figure 605.11 Impaction of Particles and Target Droplets In Venturi





## Venturi Scrubber Types

- Venturi with weir
- Venturi with jets
- “Dentist sink” venturi
- Rod deck scrubber (with cyclone demister)
- Collision scrubber
- Orifice scrubber
- Ejector venturi scrubber
- Gas ejector scrubber





## Three Stages

- Pre-Inspection
  - \* file review, rule review, inspection forms, copy of permit, safety equipment check
- Inspection
  - \* facility safety indoctrination, pre-inspection meeting
- Post-Inspection Interview



## Pre-Inspection Guidelines

- Regulation review
- Equipment check
- Pre-entry and entry
- Pre-inspection meeting
- Permit check



## Pre-Inspection Meeting

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- Facility name and ownership
- Address including city and zip
- Contact name and title
- Phone number including area code
- Production rate



## Pre-Inspection Meeting

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- Operating schedule
- Operation season
- Date of last source test
- Fuel usage and sulfur content



## Inspection Report

- Description of facility & processes
- Flowchart with equipment location & emission points
- Process diagram (materials handled, flow rates, temperatures, pressures)
- Statement as to compliance or non-compliance
- Enforcement action recommendation



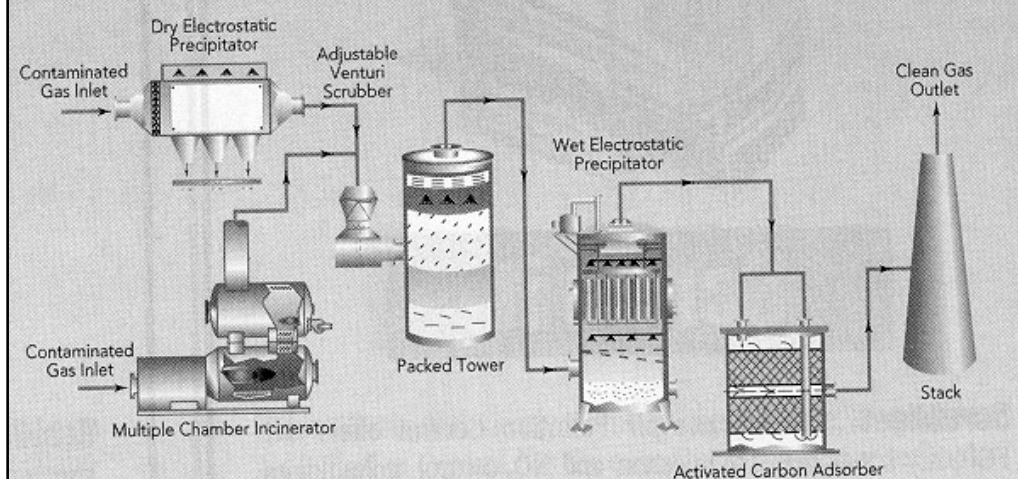
## Usage Records

- Review usage records
- Obtain necessary copies



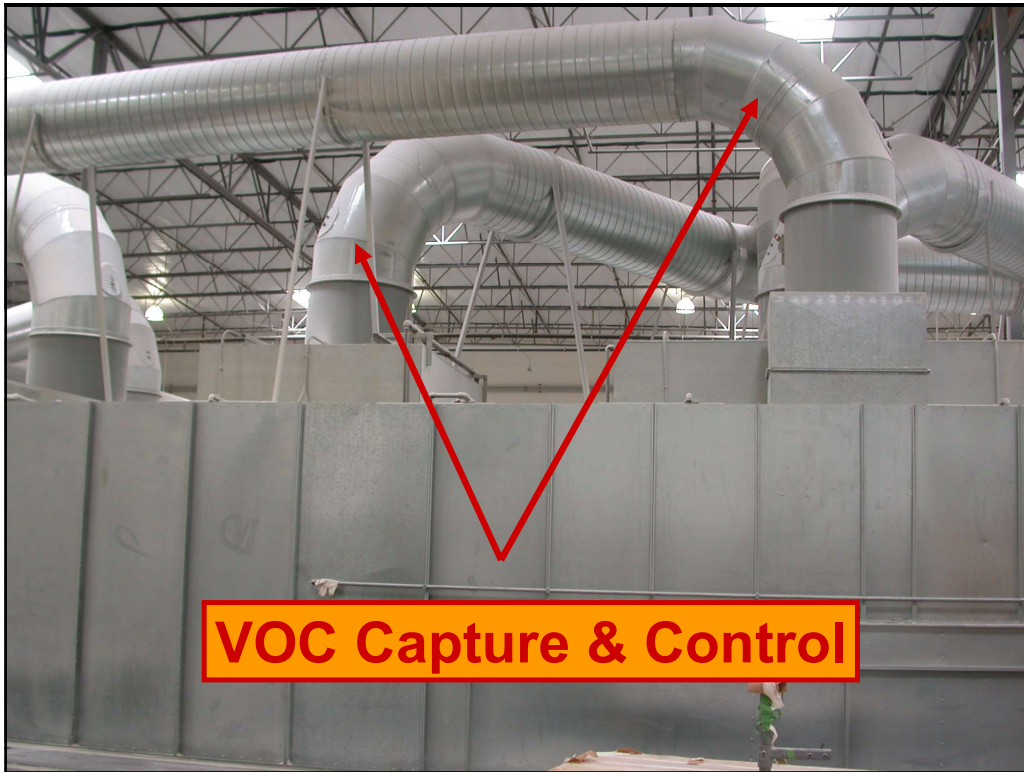
## Six points of Inspection

Capture, Transport, Air Mover,  
Instrumentation, Control, Subsystem



## Capture

- Are process emissions drawn into a control device at the point of release?
- Are they drawn into a collection device?



## Transport

- Are the emissions moved to the control device without loss?
- Are there any leaks?



## Air Mover

- Is the fan big enough for the job?
- Is it operating as designed and permitted?







# Instrumentation

- Are the proper instruments present?
- Are they functioning?
- Are they calibrated regularly?
- Are they showing the proper units?





## Control Device

- Is it functioning?
- Are there any visible leaks?
- Can the device handle the job?



## Subsystem

- What is the ultimate fate of captured or concentrated emissions?
- Pressure gauges for accuracy & change
- Fines system for leaks & proper discharge
- Motor for proper operation



**The End**