

# Introduction to Control Devices



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## Combustion Considerations

### 3 T's of Combustion

- Time (residence time)
- Temperature
- Turbulence (mixing)
- Increase 3T's > more NOx
- Decrease 3T's > more CO & PICs

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## Let's Discuss CO Catalyst

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## CO Catalyst

- $2\text{CO} + \text{O}_2 \Rightarrow 2\text{CO}_2$
- 700 to 1000 °F operating temp
- 90% plus efficiency
- Pressure drop 1-2 in. H<sub>2</sub>O
- Problems
  - Expensive
  - High maintenance
  - Catalyst replacement

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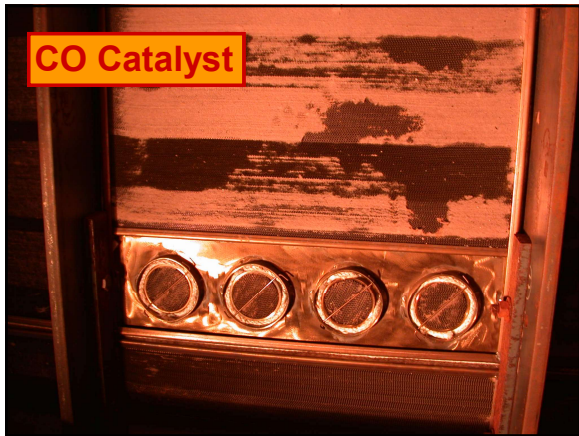
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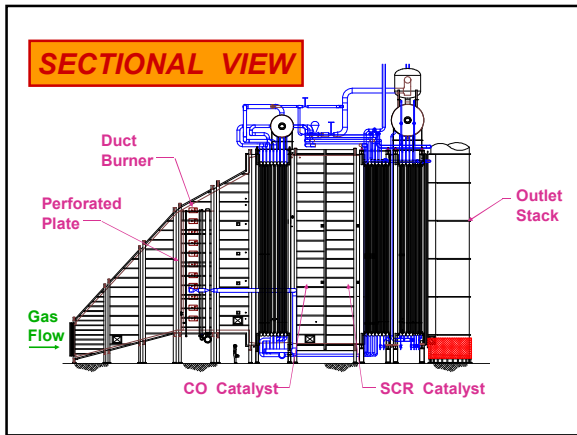
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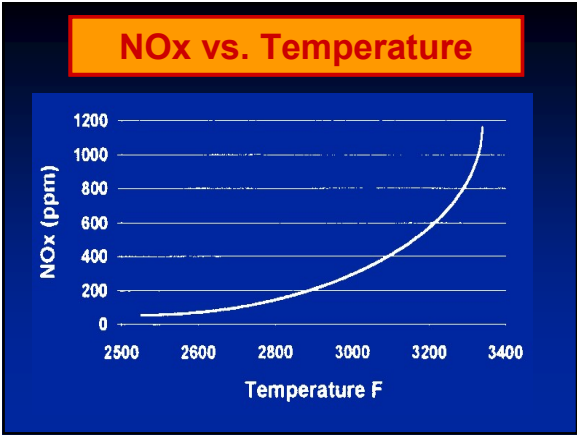
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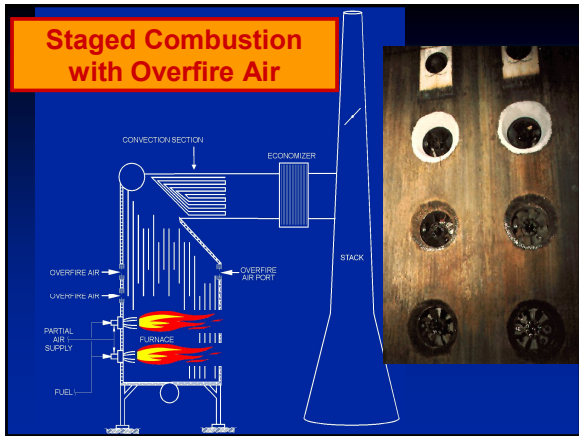
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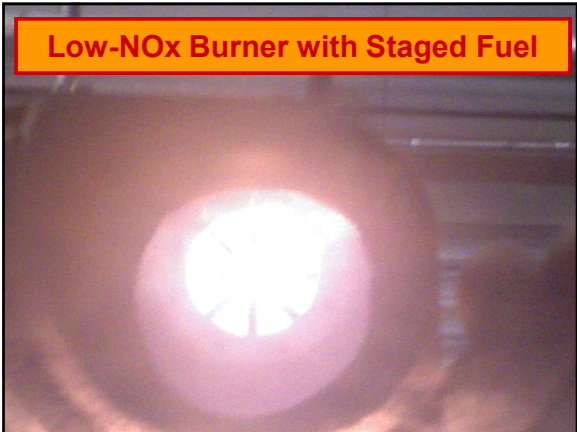
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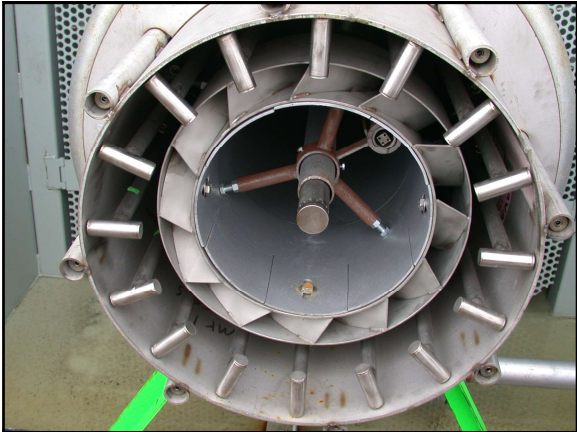
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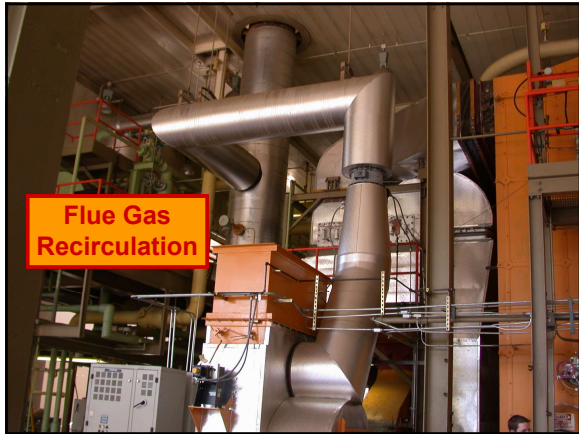
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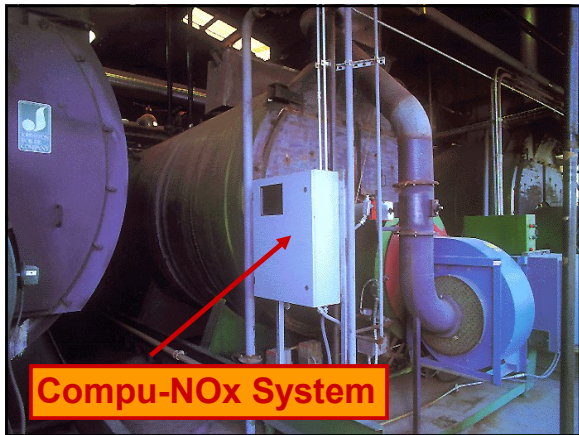
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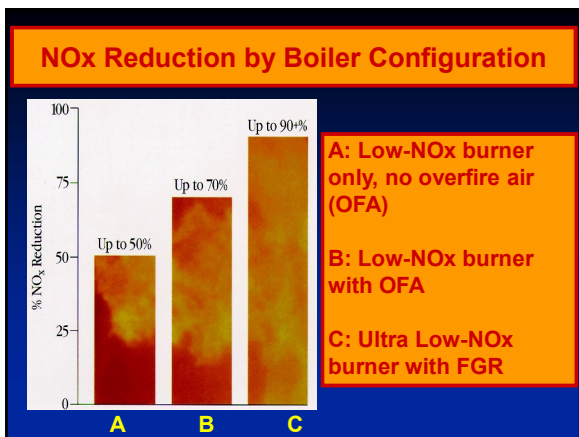
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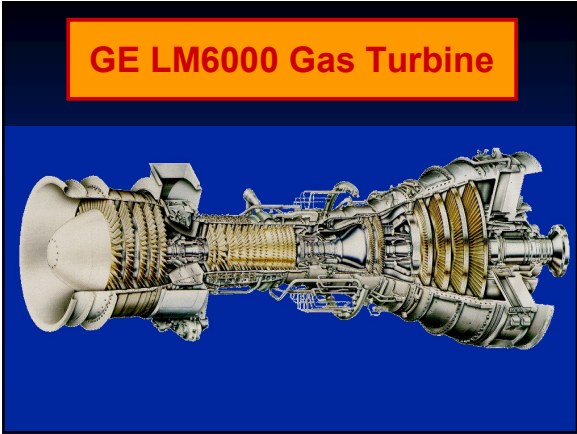
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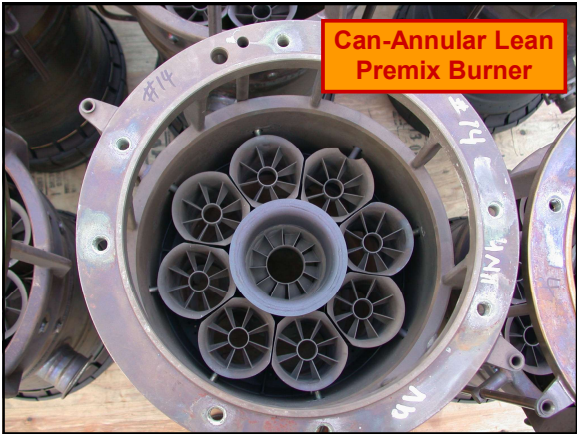
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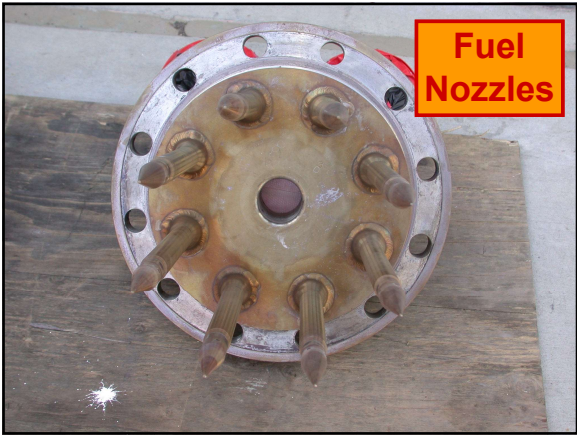
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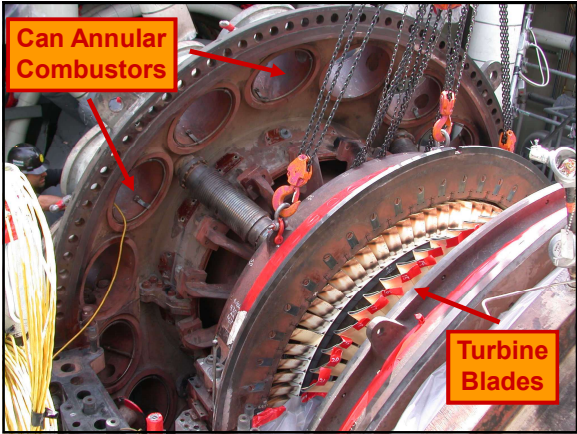
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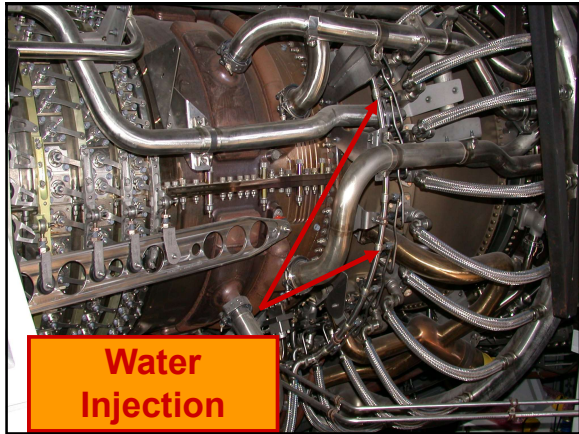
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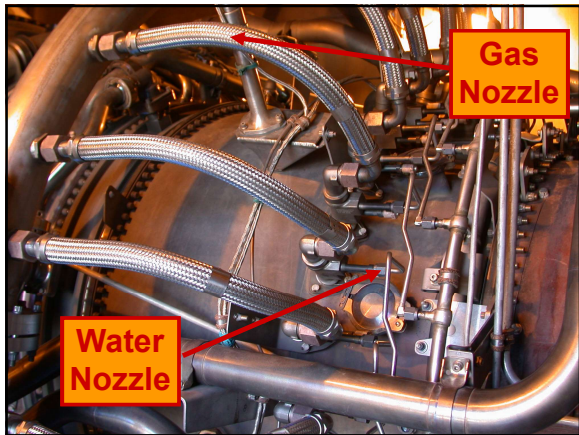
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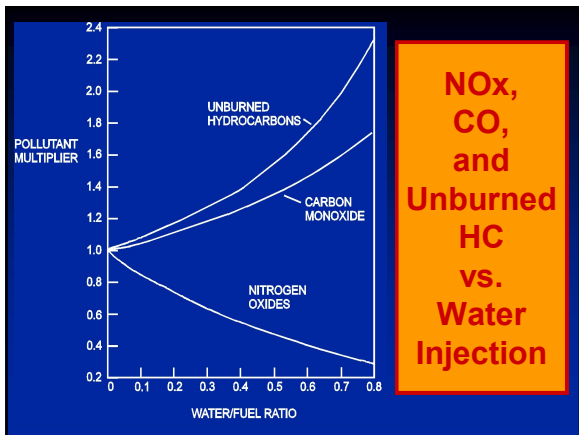
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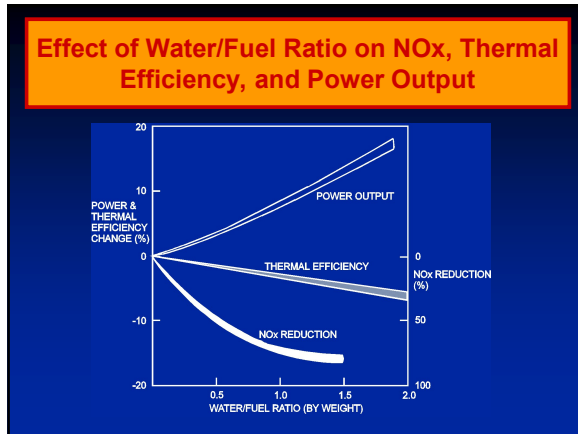
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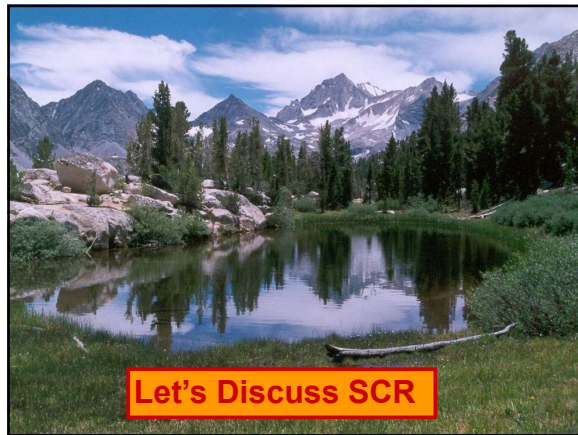
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**SCR - Introduction**

Overview of the SCR Process

$\text{NO} + \text{NH}_3 + \frac{1}{4} \text{O}_2 \rightarrow \text{N}_2 + 1.5\text{H}_2\text{O}$  (1)\*  
 $6\text{NO}_2 + 8\text{NH}_3 \rightarrow 7\text{N}_2 + 6\text{H}_2\text{O}$  (2)  
 $2\text{NO}_2 + 4\text{NH}_3 + \text{O}_2 \rightarrow 3\text{N}_2 + 6\text{H}_2\text{O}$  (3)

\* The vast majority of NO<sub>x</sub> is in the form of NO, so reaction (1) dominates

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# Introduction to Control Devices

**SCRT®**  
**Johnson Matthey**

SCRT® – Selective Catalytic Reduction Technology

Partikel-Reduzierung | CRT-System

Rußpartikel frei

& Stickoxide frei

Stickoxid-Reduzierung | SCR-System

$$2 \text{NH}_3 + \text{NO} + \text{NO}_2 \rightarrow 2 \text{N}_2 + 3 \text{H}_2\text{O}$$

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**Selective Catalytic Reduction (SCR)**

- **65-90% control**
- **Problems**
  - Expensive
  - High maintenance
  - Ammonia “slip”
  - Catalyst replacement & disposal

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**SCR – Where is it Used?**

- **Widespread Use**
  - Coal and Gas Fired Utility Boilers
  - Gas Turbine Electric Generators (Simple and Combined Cycle)
- **More Recently**
  - Refinery Combustion Systems
  - Smaller Industrial Boilers (Gas, Biomass Fired)
  - Mobile Diesel Engines

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# Introduction to Control Devices



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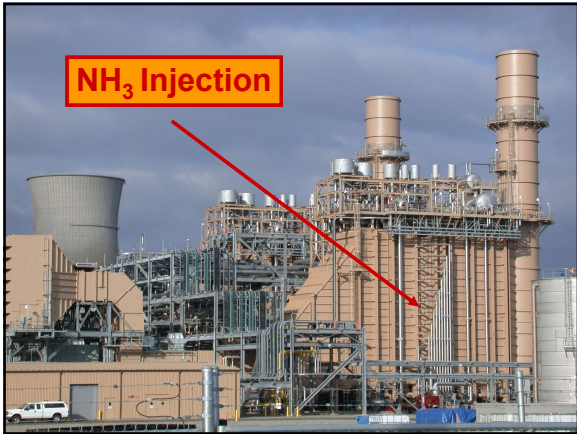
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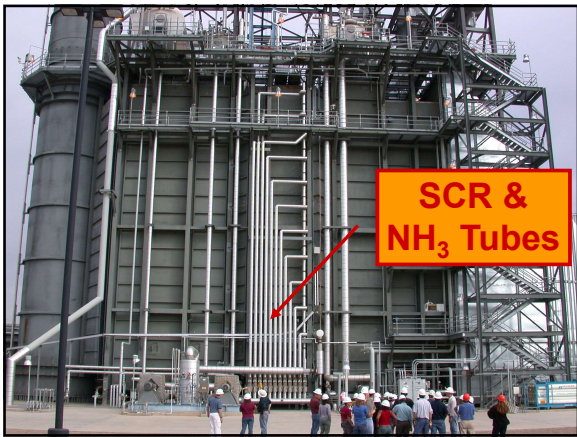
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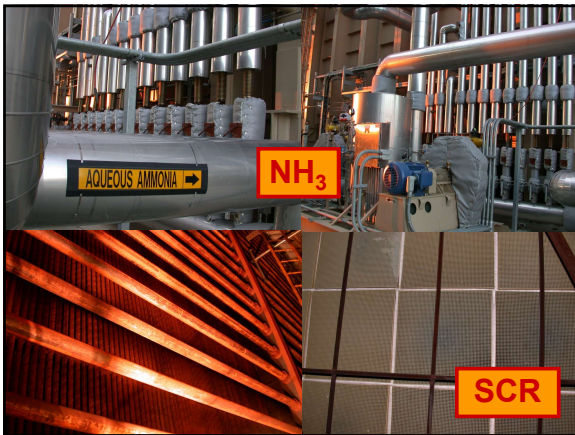
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# Introduction to Control Devices



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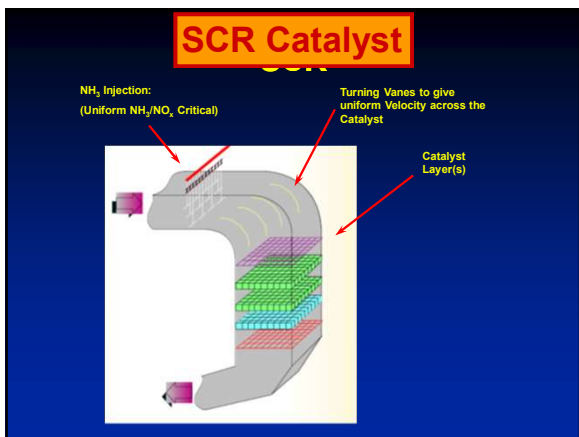
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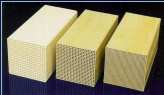
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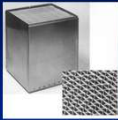
# Introduction to Control Devices

## SCR Catalyst Types

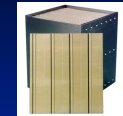
**Extruded Ceramic Honeycomb**



**Corrugated (Haldor-Topsoe)**

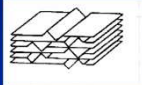


**Plate**



**Composition**

- Vanadium Pentoxide (V2O5)
- Titanium Dioxide (TiO2)
- Molybdenum
- Tungsten



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## Utility Boiler: ID Fans for SCR



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## Catalyst Degrade with Time

Reason for Degradation Fuel Dependent

- Bituminous Coal-Arsenic Poisoning
- Other Coal- Calcium sulfate blinding
- Potassium & Chlorine Poisoning



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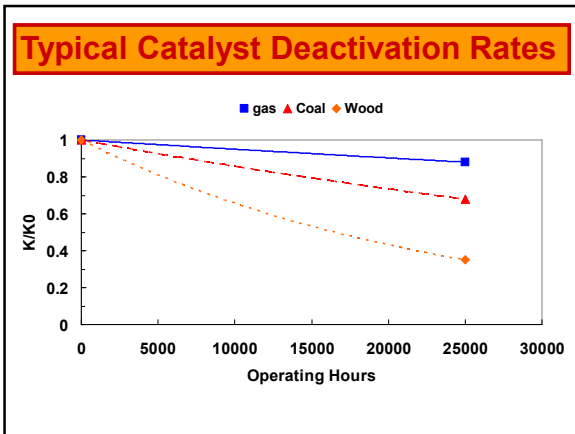
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# Introduction to Control Devices



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### NO<sub>x</sub> Control Techniques – Selective Catalytic Reduction

- Factors affecting efficiency
  - Catalyst activity
  - Masking or poisoning
  - Space velocity (gas flow rate divided by bed volume)
  - Excess ammonia or urea slip

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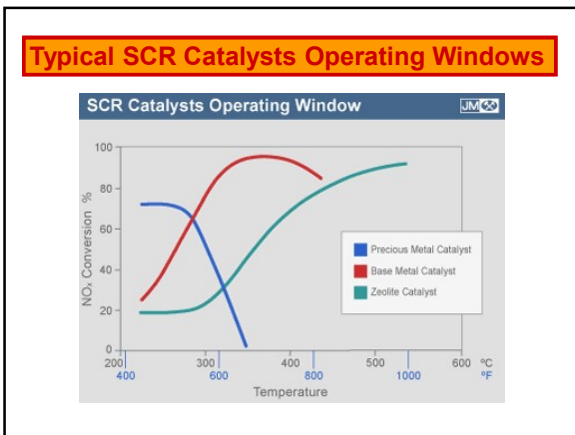
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# Introduction to Control Devices

## NOx Control Techniques – Selective Catalytic Reduction

- Performance indicators
  - Inlet and Outlet NOx concentration
  - Ammonia / urea injection rate
  - Catalyst bed inlet temperature
  - Catalyst activity (coupon)
  - Outlet ammonia concentration
  - Inlet gas flow rate
  - Fuel sulfur content
  - Pressure differential across catalyst bed

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Let's Discuss SNCR

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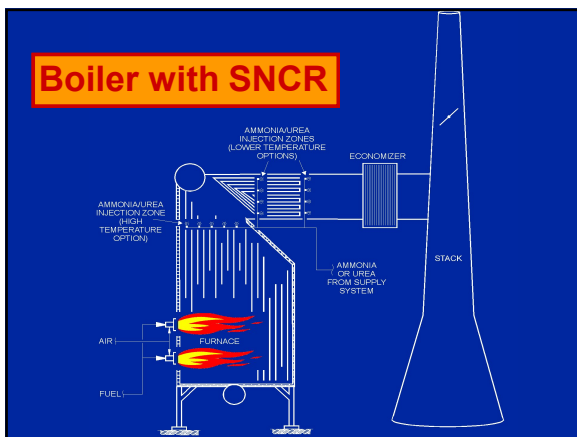
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## Boiler with SNCR

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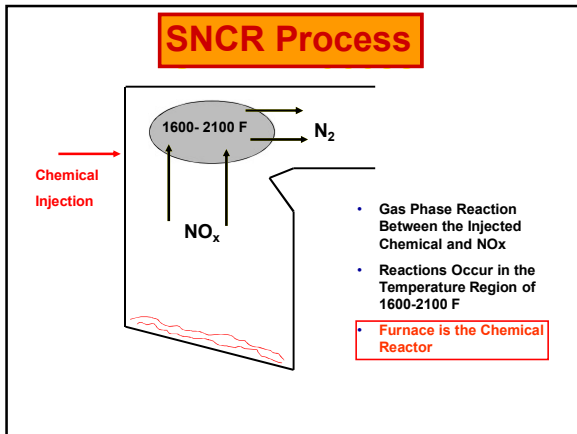
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**Selective Non-Catalytic Reduction**

- $\text{NO}_x$  control through ammonia injection
- No catalyst necessary
- Temperature range 1600 °F – 2100 °F
- Injected upstream of convection section
- 20% - 50% control under normal conditions
- Problems:
  - Changing flue temperatures with changing load
  - Formation of ammonium salts
  - Ammonia slip

**Catalyst** (with a red slash through it)

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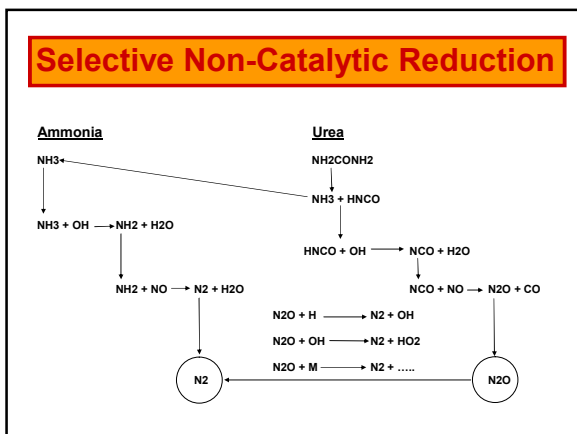
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# Introduction to Control Devices

### Detailed SNCR Mechanism

The diagram illustrates the chemical pathways for NOx reduction using ammonia. It shows the reaction of NO with NH3 to form NH2OH and H2O, followed by the reaction of NH2OH with NO and NO2 to produce N2 and H2O. The reactions are presented in three columns, with the first column showing the initial reaction and the second and third columns showing the subsequent reactions.

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### Ammonia vs. Urea

Parameter	Ammonia	Urea
Form	High Vapor Pressure Liquid Ammonia/Water Solution	Liquid Solution
Safety	Anhydrous/29.4% Aqueous – Safety Issue 19% Aqueous – Fewer Safety Issues	No Safety Issues
Storage	Anhydrous – Pressure Vessel Aqueous – Atmospheric Pressure	Atmospheric Pressure Crystallization at Low Temps.
Injectors	Needs Carrier Gas	Atomizer (Pressure or Twin Fluid)
Temperature	Peak Removal @ 1750° F	Peak Removal @ 1850° F Large Dilute Drops Shield Urea
System Complexity	Relatively Simple	Relatively Simple

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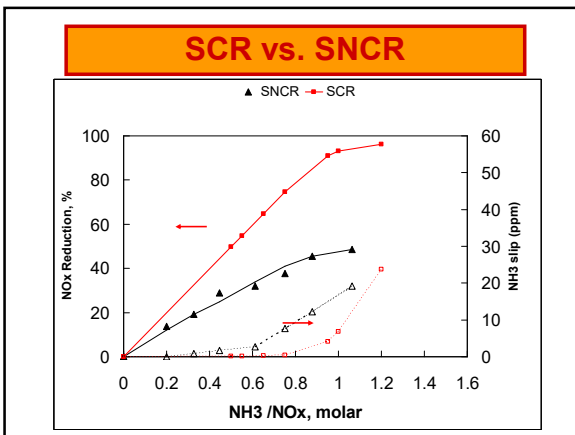
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# Introduction to Control Devices

SCR vs. SNCR		
	SNCR	SCR
NOx Reductiuon	20-50%	50-95%
Hardware	Simple	More Complex
Capital Cost	Low (1)	High (5-10)
Reagent Utilization	Typ. 30%	Almost 100%
O&M	Reagent	Reagent/Catalyst
Designability	Poor	Good
NH3 slip	5-20 ppm	<10 ppm

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NH <sub>3</sub> Emissions Limits	
• Regulatory Limit	
• NH <sub>3</sub> /SO <sub>3</sub> Reactions	
– Ammonium Bisulfate: $\text{NH}_3 + \text{SO}_3 \rightarrow \text{NH}_4\text{HSO}_4$	
– Ammonium Sulfate: $2\text{NH}_3 + \text{SO}_3 \rightarrow (\text{NH}_4)_2\text{SO}_4$	
• NH <sub>3</sub> /Ash Absorption (issue for coal-fired utility units that sell their ash for making cement)	
• NH <sub>3</sub> /HCl Reactions (detached plume)	
– NH <sub>3</sub> /HCl $\text{NH}_4\text{Cl(s)}$	

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Comparison of NOx Control Technologies – Gas-Fired Boilers			
Technology	Approx. Reduction	Approx. lbs/MMBTU	Approx. ppmv @ 3% O <sub>2</sub>
Standard burners	Base case	0.14	120
Low NOx burners	60%	0.06	45
Ultra Low NOx Burners – 1 <sup>st</sup> gen.	80%	0.03	25
Ultra Low NOx Burners – 2 <sup>nd</sup> gen.	95%	0.007	6
FGR	55%	0.025	20
Compu- NOx w/ FGR	90%	0.015	12
SNCR	40%	0.033 - 0.085	27 - 70
Catalytic Scrubbing	70%	0.017 - 0.044	14 - 36
SCR	90 – 95%	0.006 - 0.015	5 - 12

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# Introduction to Control Devices



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**Objectives**

- Define "particulate matter or PM"
- Identify sources of particulates
- Analyze opacity issues
  - Potassium plumes
  - Ammonium-chloride plumes

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**What is Particulate Matter??**

- It is what the test measurement says it is
- Meaning:
  - Solid particles that are captured on a filter
  - Condensable matter collected in a set of impingers
- What eventually condenses in the atmosphere is also considered as particulate matter along with "solid" particulate in the gas stream

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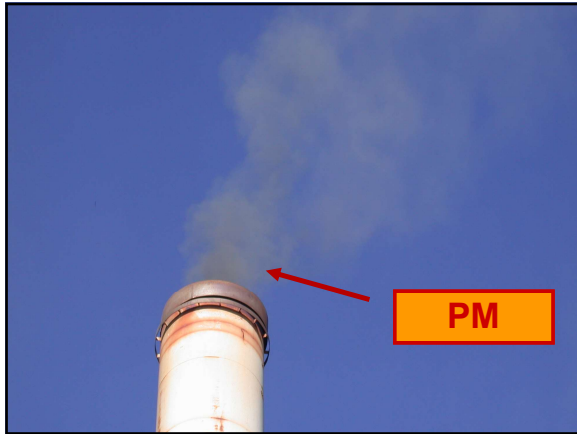
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# Introduction to Control Devices



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**Sources of "Particulate Matter"**

- Ash in the fuel
  - Silica and Alumina - generally large particles that are retained or collected in the boiler/precipitator
  - Intrinsic ash - generates the small particles that are more troublesome to control
  - Alkalis - potassium, sodium and calcium
- Condensables (HCl, SO<sub>3</sub>, NH<sub>4</sub>Cl) which are also considered as "particulates"

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**Ammonia Slip**

- $\text{NH}_3 + \text{OH} \Rightarrow \text{NH}_2 + \text{H}_2\text{O}$
- $\text{NH}_2 + \text{NO} \Rightarrow \text{N}_2 + \text{H}_2\text{O}$
- $2\text{NH}_3 + \text{OH} + \text{NO} \Rightarrow 2\text{H}_2\text{O} + \text{N}_2 + \text{NH}_3$
- 10 to 25 ppm NH<sub>3</sub> Slip
- Could be higher
- Always have Some NH<sub>3</sub> slip

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# Introduction to Control Devices



- $\text{NH}_3$  and HCl released as gases
- Combine and condense into aerosol particles
- Two parallel processes taking place
  - Rate of formation reaction controlled by concentrations
  - Rate of condensation control by temperature
- Both affected by air dilution in the plume

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## $\text{NH}_4\text{Cl}$ Formation

- Function of the concentrations of  $\text{NH}_3$  and HCl
- Concentrations decrease as air is mixed into the plume
- Lower concentrations  $\Rightarrow$  less  $\text{NH}_4\text{Cl}$  formed
- Therefore: air dilution is good

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## What Can Be Done??

- Minimize (eliminate Cl) in fuel
- Install acid gas controls
- Minimize  $\text{NH}_3$  slip  $\Leftarrow$  monitor
- High stack gas temperatures
- High ambient air temperatures (winter time a problem??)
- Promote rapid gas/air mixing ??
- Install high gas temperature concentric stack annulus ??

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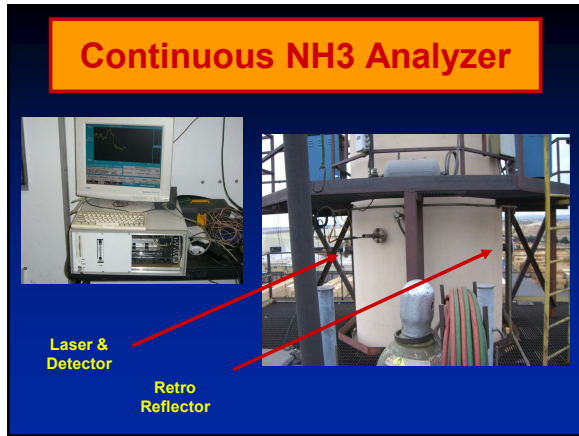
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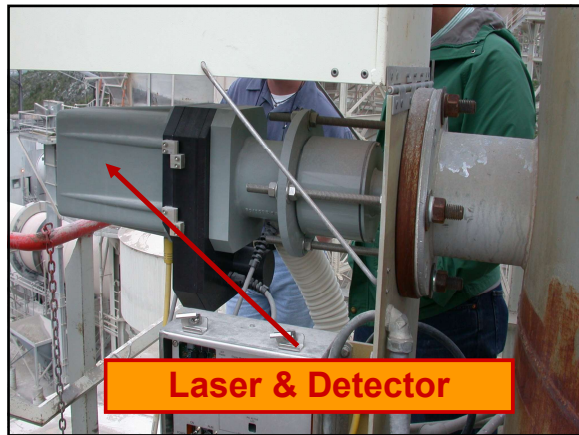
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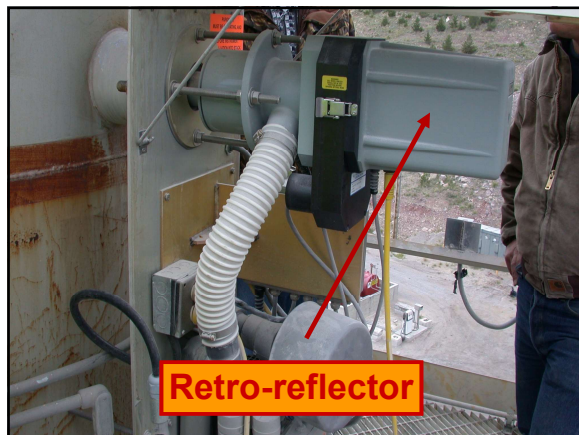
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# Introduction to Control Devices



**Let's Discuss VOC Control**

- ◆ Material Usage Minimization
- ◆ Containment
- ◆ Absorption
- ◆ Adsorption
- ◆ Oxidation

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**Material Usage Minimization**

**Basic Strategy:**

If we optimize the efficiency of the amount of VOC-containing material we use, we also limit VOC emissions

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**Materials Minimization**

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# Introduction to Control Devices

We must consider real-world demands  
e.g. Spray painted cars look better



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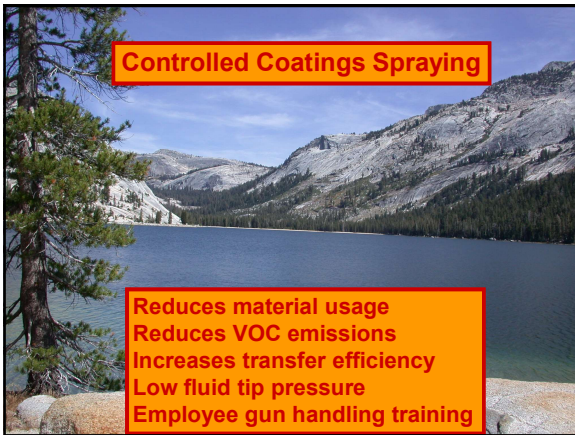
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Controlled Coatings Spraying



Reduces material usage  
Reduces VOC emissions  
Increases transfer efficiency  
Low fluid tip pressure  
Employee gun handling training

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Motor Vehicle Coating –  
High Volume Low Pressure (HVLP)  
Spray Gun

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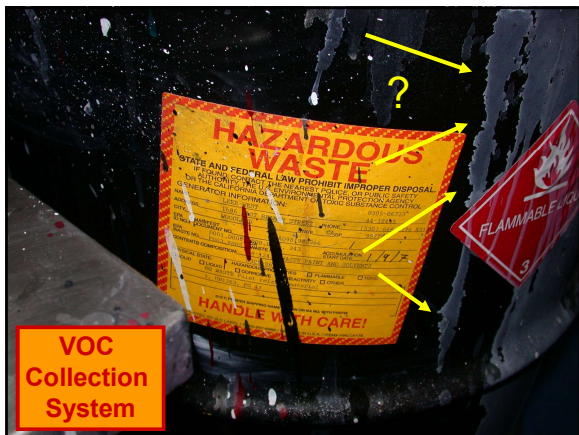
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# Introduction to Control Devices



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# Introduction to Control Devices



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**VOC Control Techniques – Capture System**

- **General description**
  - Total efficiency is product of capture and control device efficiencies
  - Two types of systems
    - Enclosures and local exhausts (hoods)

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# Introduction to Control Devices

## VOC Control Techniques – Capture System

- General description
  - Two types of enclosures
    - Permanent total (M204) – 100% capture efficiency
    - Nontotal or partial – must measure capture efficiency

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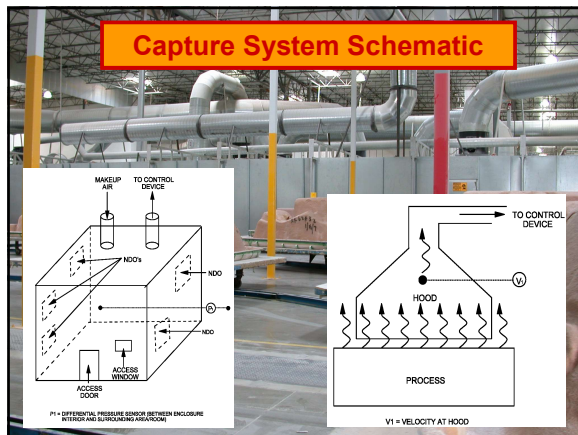
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## VOC Control Techniques – Capture System

- Performance indicators
  - Enclosures
    - Face velocity
    - Differential pressure
    - Average face velocity and daily inspections

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# Introduction to Control Devices

## VOC Control Techniques – Capture System

- Performance indicators (cont.)  
Exhaust Ventilation
  - Face velocity
  - Exhaust flow rate in duct near hood
  - Hood static pressure

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## Any Concerns Here?



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## Let's Discuss Packed Column Absorbers (a.k.a Scrubbers)



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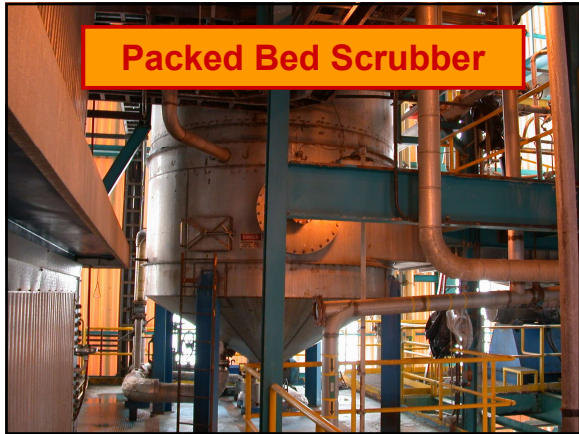
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# Introduction to Control Devices



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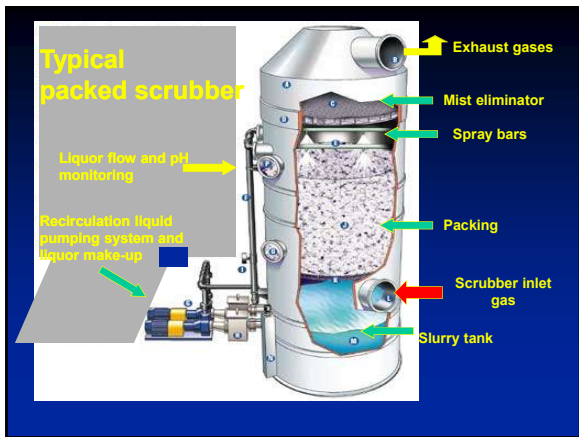
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# Introduction to Control Devices



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## Scrubbers

- Used for a variety of pollutants
  - Both particulates and VOCs
  - Acid gases
  - Odors (e.g. rendering operations)
- Primary indicators
  - Water (liquor) flow rate
  - pH
  - Outlet temperature
- Secondary (longer term) indicators
  - Inlet & water temperatures
  - Gas pressure drop

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## Monitoring Approach – SO<sub>2</sub>

Indicator	Slurry pH	Slurry flow rate
Indicator range	<9.0 - corrective action, reporting	<175 – corrective action, reporting
Measurement location	Recirculation line	Recirculation line
QA/QC	Annual cal.	Annual cal.
Frequency	1/15 minutes	1/15 minutes
Averaging time	hourly	hourly

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# Introduction to Control Devices



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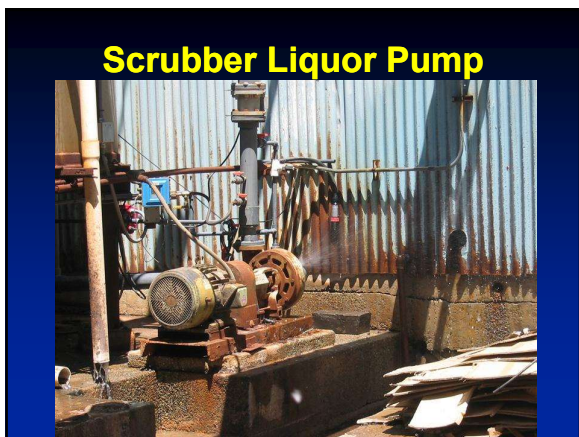
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# Introduction to Control Devices



**Let's Discuss FGD : Flue Gas Desulphurization**

**SO<sub>2</sub> Scrubbers**

- ◆ Wet
- ◆ Spray Dry (Semi-Dry)
- ◆ Dry (DSI : Dry Sorbent Injection)

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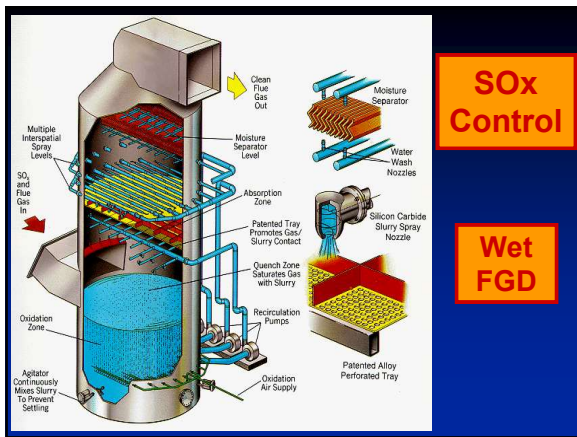
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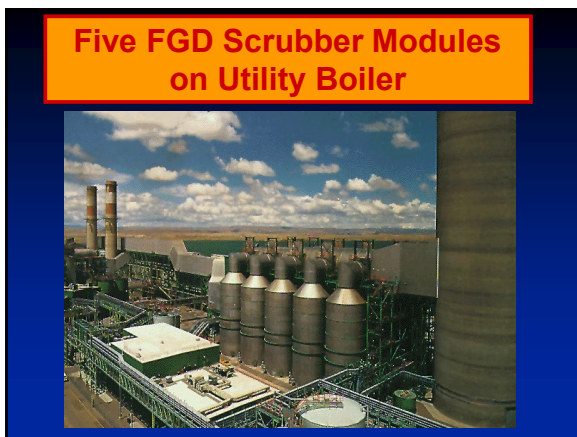
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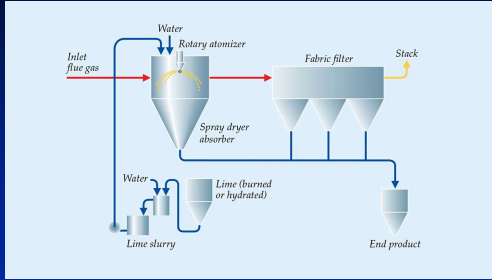
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# Introduction to Control Devices

## Spray Dryer Absorber



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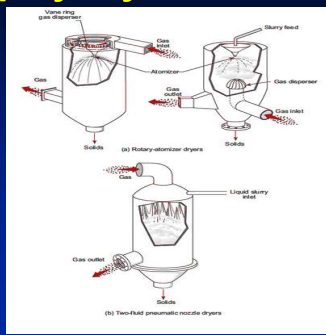
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## Spray Dryer Absorbers



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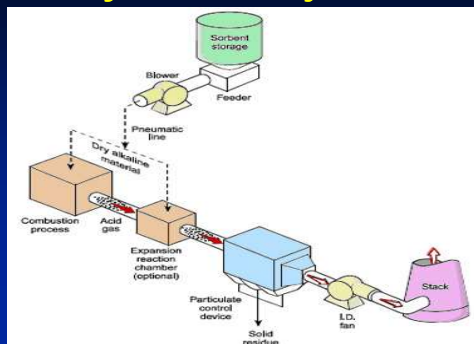
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## Dry Sorbent Injection



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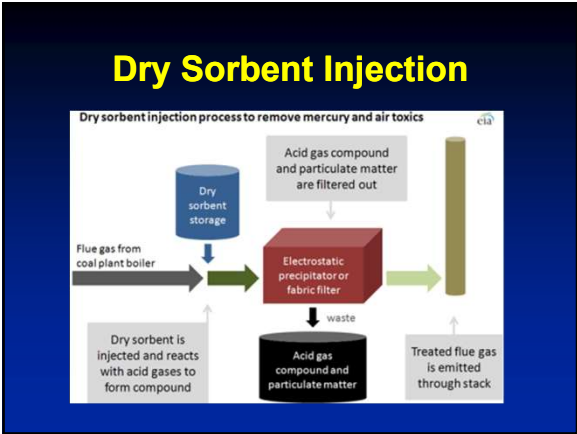
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# Introduction to Control Devices



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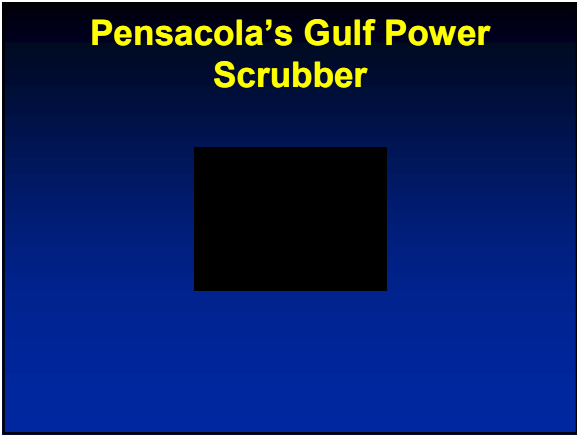
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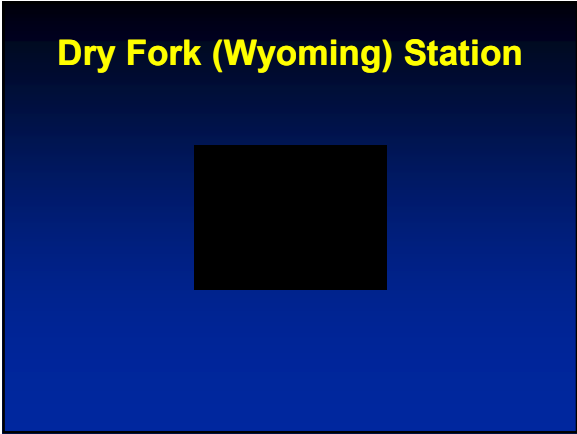
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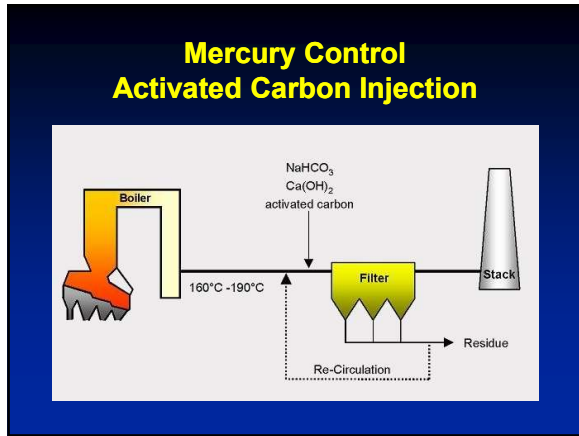
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# Introduction to Control Devices



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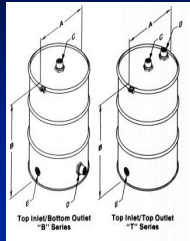
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# Introduction to Control Devices

## Carbon Adsorber – Fixed Bed Example



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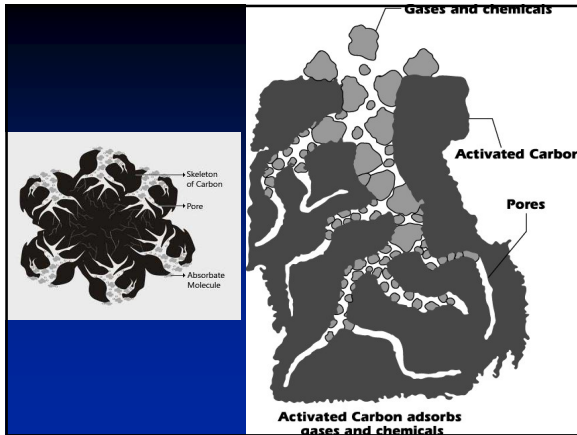
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## Carbon Adsorber

- **General description**
  - Gas molecules stick to the surface of a solid
  - Activated carbon often used as it
    - Has a strong attraction for organics
    - Has a large capacity for adsorption (many pores)
    - Relatively inexpensive

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# Introduction to Control Devices

**Carbon Adsorber**

- **Activated Carbon is typically made of charcoal**
  - Wood
  - Coal
  - Nutshells
  - Coconut shells
- **Other Common Types of Adsorbers**
  - Silica gel
  - Activated alumina
  - Zeolites

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**Carbon Adsorber**

- **3 types – fixed bed (most common), moving bed, and fluidized bed**
  - Typically appear in pairs – prevent carbon breakthrough
  - Used for control as well as recovery

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**Carbon Adsorber**

- **General description (continued)**
  - **Regeneration process**
    - Steam
    - Hot gas
    - Vacuum
  - Work best if molecular weight of compound between 50 & 200 (depends on source of carbon raw material)

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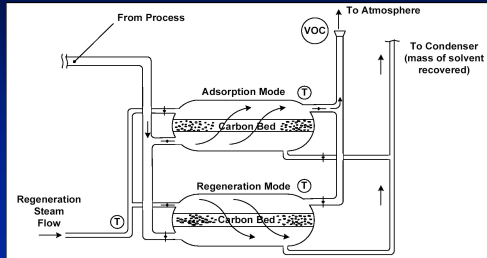
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# Introduction to Control Devices

## Carbon Adsorber – Fixed Bed Schematic



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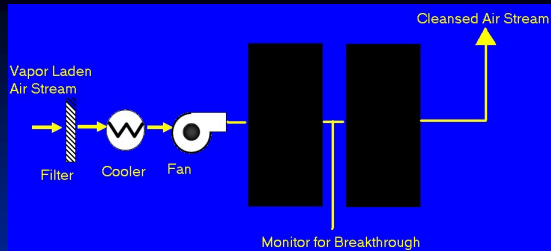
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## Carbon Adsorption System

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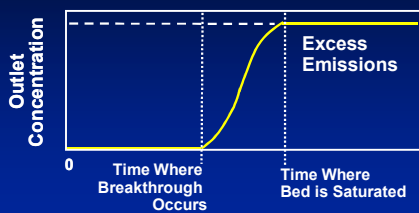
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## Adsorber Breakthrough



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# Introduction to Control Devices

**Carbon Adsorber**

- **Factors affecting efficiency**
  - Presence, polarity, and concentration of specific compounds
  - Flow rate & channeling
  - Temperature & fouling
  - Relative humidity

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**Carbon Adsorber**

- **Performance indicators**
  - Outlet VOC concentration
  - Regeneration cycle timing or bed replacement frequency
  - Total regeneration stream flow or vacuum profile during regeneration cycle
  - Bed operating and regeneration temperature

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**Carbon Adsorber**

- **Performance indicators**
  - Inlet gas temperature
  - Gas flow rate
  - Inlet VOC concentration
  - Pressure differential
  - Inlet gas moisture content
  - Leaks

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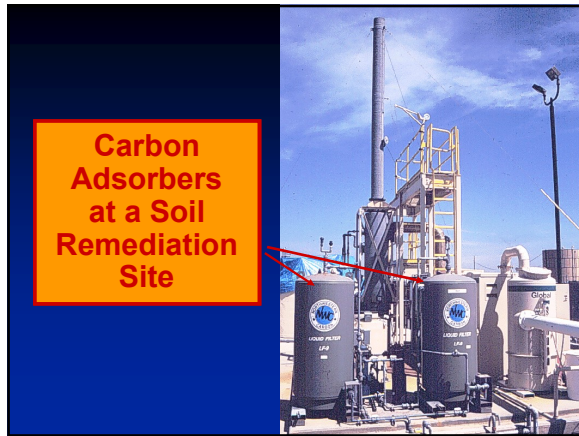
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# Introduction to Control Devices




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**Monitoring Approach – VOC**

Indicator	Vacuum	Carbon bed I/M	LDAR
Approach	Pressure transducer	Daily insp. And annual sample	Monthly leak check w. portable analyzer
Indicator range	<2.5 min @ - 27.5" Hg, shutdown	Failure to conduct, corrective action and reporting	> 10K ppm, corrective action and reporting

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# Introduction to Control Devices

Monitoring Approach – VOC			
Indicator	Vacuum	Carbon bed I/M	LDAR
Measuring location	Pump suction line	Visual, bed sample	Handheld monitor
QA/QC	Annual cal.	Training	Method 21
Frequency	Continuous during cycle	Daily and annual	Monthly

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VOC Control via Incineration in Oxidizers

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**Combustion Considerations**

Remember the 3 T's of Combustion

- Residence Time
- Temperature
- Turbulence (mixing)
- Increase 3T's = more NO<sub>x</sub>
- Decrease 3T's = more CO and uncontrolled pollutant

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# Introduction to Control Devices



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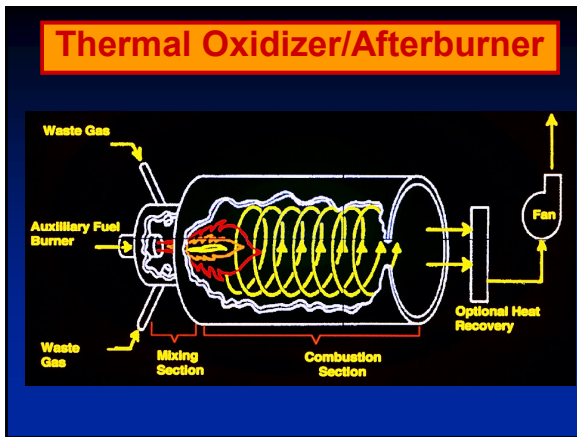
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**Thermal Oxidizer**

- General description
  - VOC gas (& organic HAP) gets oxidized to  $H_2O$  and  $CO_2$
  - Higher operating temperatures (~ 1400°F to 1800°F)
  - Typically requires auxiliary fuel (natural gas or propane)

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# Introduction to Control Devices

### Thermal Oxidizer

- Good combustion requires
  - Adequate temperature
  - Turbulent mixing of waste gas with oxygen
  - Sufficient time for reactions to occur
  - Enough O<sub>2</sub> to completely combust waste gas

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### Thermal Oxidizer

- Only temperature and O<sub>2</sub> can be controlled after construction
  - Waste gas has to be heated to autoignition temperature
  - Common design relies on 0.2 to 2 seconds residence time, 2 to 3 length to diameter ratio, and gas velocity of 10 to 50 feet per second

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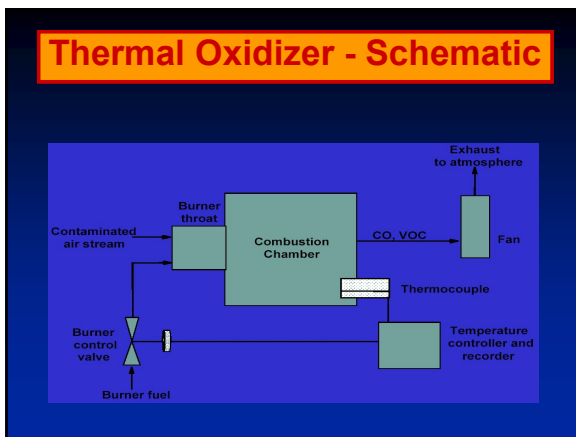
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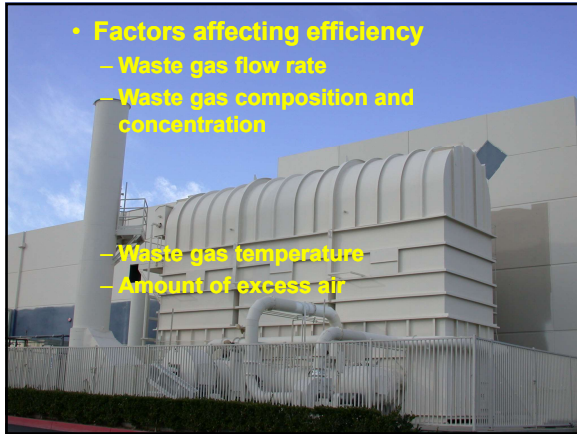
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# Introduction to Control Devices



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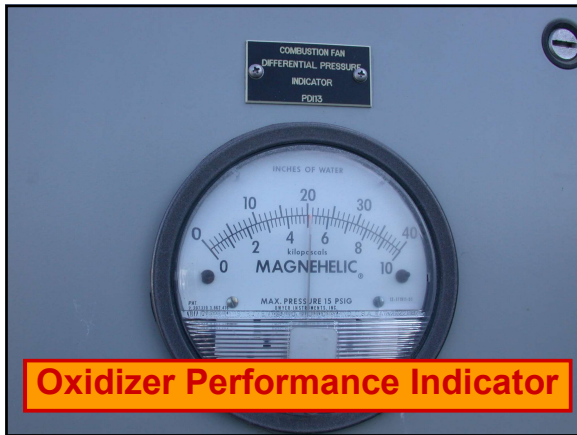
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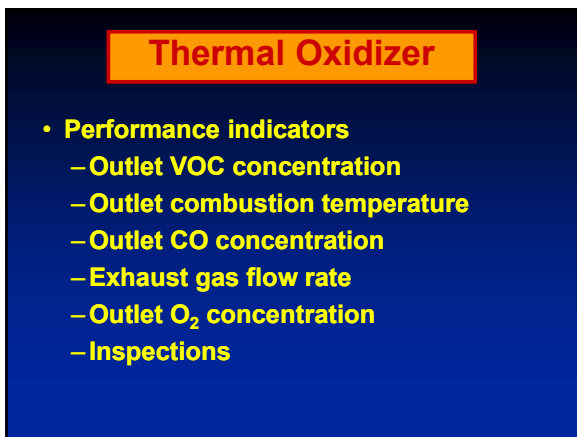
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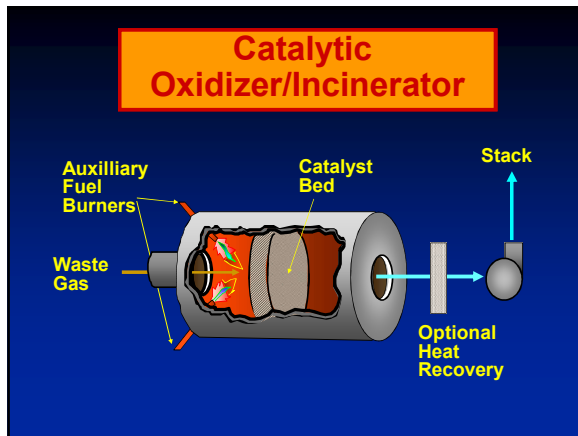
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# Introduction to Control Devices



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**Catalytic Oxidizer/Incinerator**

- General description
  - VOC gas (& organic HAP) gets oxidized to H<sub>2</sub>O and CO<sub>2</sub>
  - Catalyst causes reaction to occur faster and at lower temperatures
  - Saves auxiliary fuel

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**Catalytic Oxidizer/Incinerator**

- General description (continued)
  - Catalysts allow lower operation temperatures (~ 600°F to 800°F)
  - Catalyst bed generally lasts from 2 to 5 years
    - Thermal aging, poisoning, and masking are concerns

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# Introduction to Control Devices

### Catalytic Oxidizer/Incinerator

- General description (continued)
  - Excess air is added to assist combustion
  - Residence time and mixing are fixed during design
  - Only temperature and oxygen can be controlled after construction

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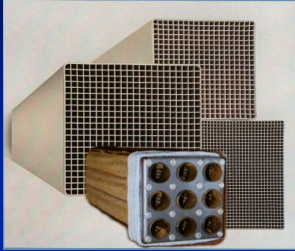
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### Catalytic Oxidizer Incinerator Examples



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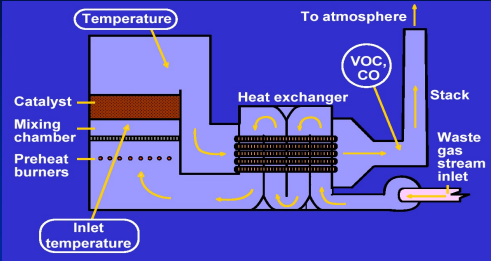
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### Catalytic Oxidizer Incinerator Schematic



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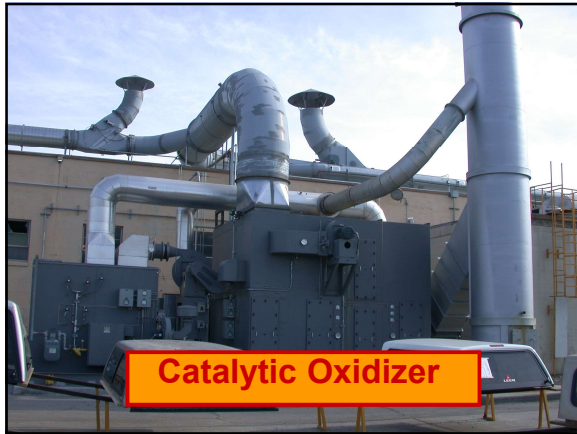
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# Introduction to Control Devices



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**Catalytic Oxidizer/Incinerator**

- **Factors affecting efficiency**
  - Pollutant concentration
  - Flow rate
  - Operating temperature
  - Excess air
  - Waste stream contaminants
    - Metals, sulfur, halogens, plastics

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**Catalytic Oxidizer/Incinerator**

- **Performance Indicators**
  - Outlet VOC concentration
  - Catalyst bed inlet temperature
  - Catalyst activity
  - Outlet CO concentration
  - Temperature rise across catalyst bed
  - Exhaust gas flow rate

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# Introduction to Control Devices

**Catalytic Oxidizer/Incinerator**

- Performance Indicators (continued)
  - Catalyst bed outlet temperature
  - Fan current
  - Outlet O<sub>2</sub> or CO<sub>2</sub> concentration
  - Pressure differential across catalyst bed

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**Catalytic Oxidizer – Monitoring Approach**

- Key Factors to Consider When Monitoring a Catalytic Oxidizer:
  - Catalyst bed operating temperature (inlet & outlet)
  - Catalyst activity (life) (core sampling & testing)
  - Periodic Inspection
  - Annual performance testing

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**Catalytic vs. Thermal for VOC Control**

Catalytic	Thermal
Lower Operating Temp. & Lower Fuel Usage	Higher Operating Temp. & Higher Fuel Usage
Higher Capital & Maintenance Costs	Lower Capital & Maintenance Costs
Catalyst Fouling & Poisoning	No Catalyst Involved Here

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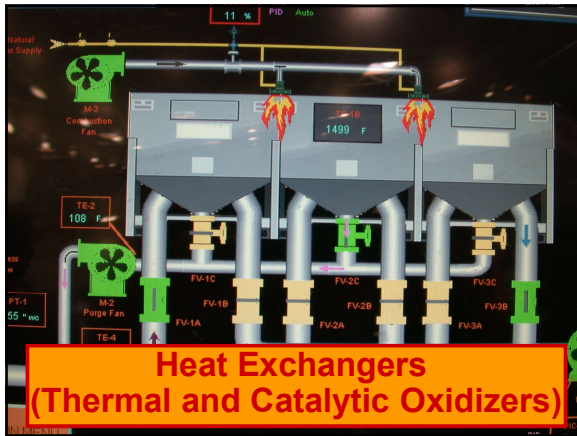
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# Introduction to Control Devices



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### Thermal & Catalytic Oxidizer Heat Exchangers

There are two basic types of heat exchangers used for thermal or catalytic oxidizers

- Metal Heat Exchangers or “recuperative heat exchangers”
- Ceramic Bed Heat Exchangers or “regenerative heat exchangers”

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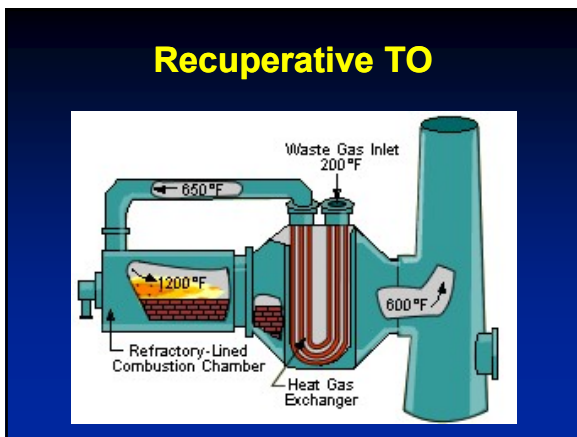
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# Introduction to Control Devices

## TO with Recuperative Heat Exchangers

- Thermal efficiency range of 30% to 70%
- Shell & tube or plate-type
- Usually constructed of alloy steel
- Welded systems have very low leakage rates when new
- Susceptible to cross-leakage as heat exchanger ages
- Not typically used with acid gases
- Susceptible to thermal shock on startup and shutdown

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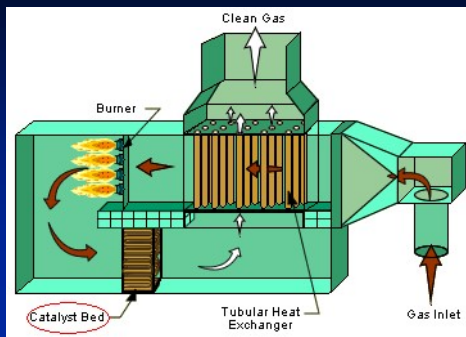
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## Catalytic Recuperative



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## Recuperative TO – Monitoring Approach

- Key Factors to Consider When Monitoring a Recuperative TO:
  - Annual inspection and/or testing of heat exchanger to assess leakage per manufacturer's recommendations.

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# Introduction to Control Devices

## Regenerative Thermal Oxidizers



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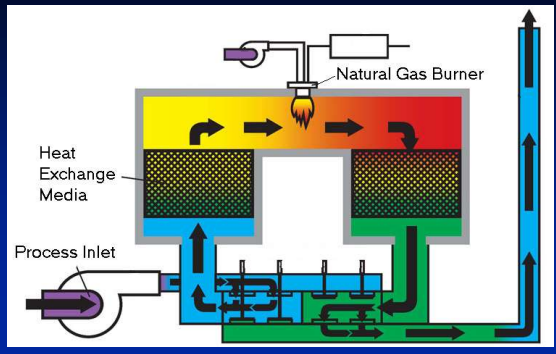
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## Regenerative TO



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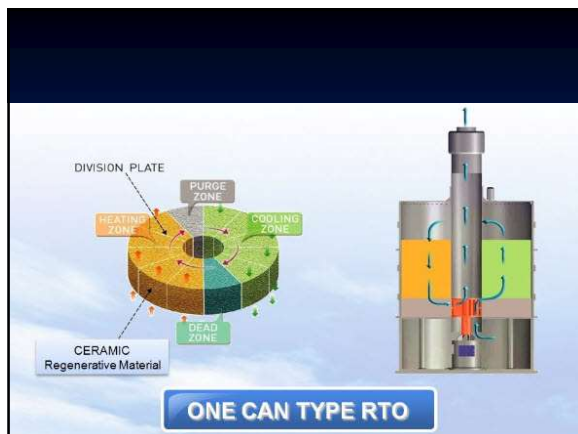
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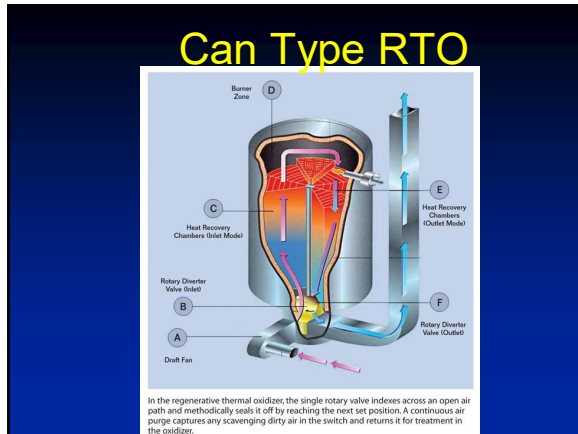
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# Introduction to Control Devices



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- ## Regenerative Thermal Oxidizer (RTO)
- ◆ Thermal efficiency range of 80% to 95%
  - ◆ Can be random packing or structured
  - ◆ Extremely tolerant of very high temperatures
  - ◆ Highly resistant to thermal shock
  - ◆ Can resist corrosion by many acid gases
  - ◆ May be susceptible to fouling or plugging
  - ◆ Subject to cross-leakage because of geometry
  - ◆ May be used with catalysts (RCOs)

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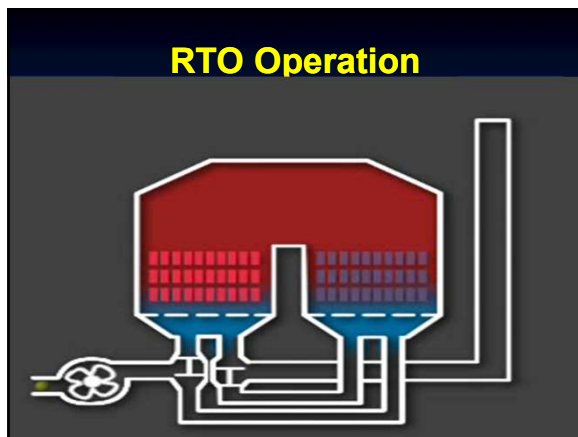
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# Introduction to Control Devices

## Regenerative Thermal Oxidizer Monitoring Approach

- Key Factors to Consider When Monitoring a Regenerative TO:
  - Assessment of proper closure of valves: Annual inspection/testing
  - Annual documentation of valve timing control system parameters

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## Heat Exchange Problems

- Any cracks or leaks in a recuperative HX will bleed emissions into the clean side
- Uncoordinated valves in a regenerative HX will transfer emissions into the clean air.
- A regenerative HX usually burps some emissions into the clean air each time the valves switch the flow.

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## Compliance Issues?



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# Introduction to Control Devices



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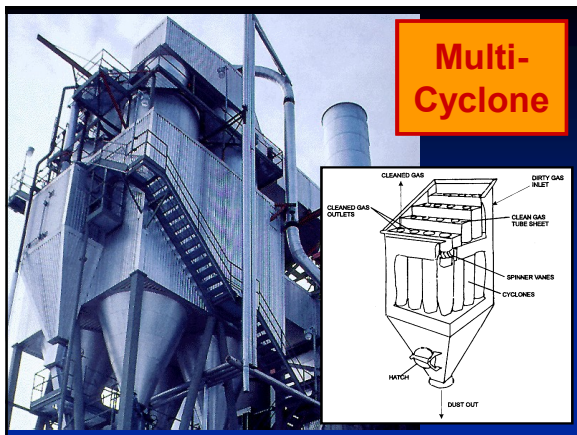
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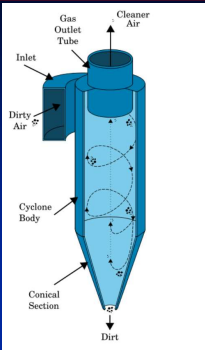
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# Introduction to Control Devices

**PM Control Techniques - Cyclone**



The diagram shows a vertical cyclone separator. Dirty air enters from the side through an inlet tube. The air spirals downwards in the cyclone body. Cleaner air exits from the top through a gas outlet tube. Dirt is collected at the bottom in a conical section.

**How a Cyclone Works**

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**PM Control Techniques - Cyclone**

- **General description**
  - Particles hit wall sides and fall out
  - Often used as precleaners
    - Especially effective for particles larger than 20 microns
  - Inexpensive to build and operate
  - Can be combined in series or parallel

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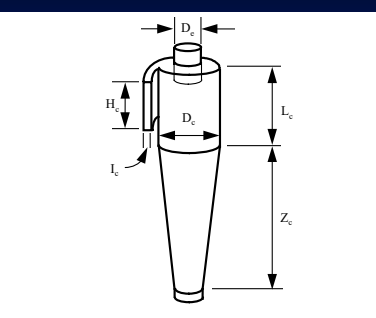
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**Cyclone - Classification**



The diagram shows a cyclone separator with dimensions labeled:  $D_c$  (diameter),  $L_c$  (height of the cylindrical section),  $Z_c$  (height of the conical section),  $H_c$  (height of the inlet tube), and  $L_c$  (height of the inlet tube).

**1D-2D vs. 1D-3D**

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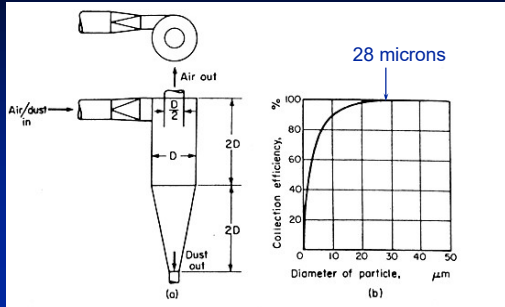
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# Introduction to Control Devices

## Cyclone – Control Efficiency



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## Cyclone – Control Efficiency

- Conventional Cyclones
  - 30-90% for  $\text{PM}_{10}$
  - 0-40% for  $\text{PM}_{2.5}$
- High Efficiency Single Cyclones
  - 60-95% for  $\text{PM}_{10}$
  - 20-70% for  $\text{PM}_{2.5}$
- Multi-Cyclones
  - 80-95% for  $\text{PM}_5$

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## Cyclone – Failure Modes

- Failure Modes
  - Inlet and outlet plugging
  - Air leakage
    - Component erosion
    - Acid gas corrosion

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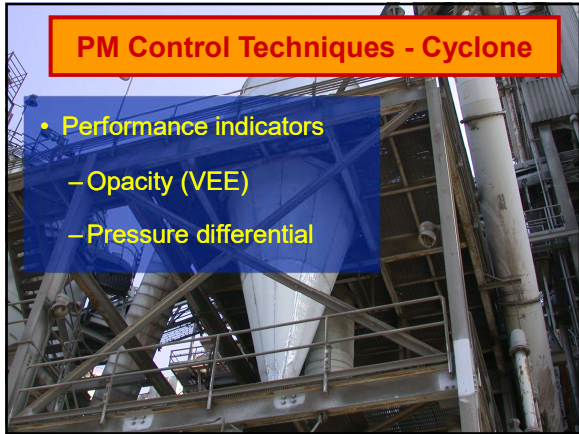
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# Introduction to Control Devices



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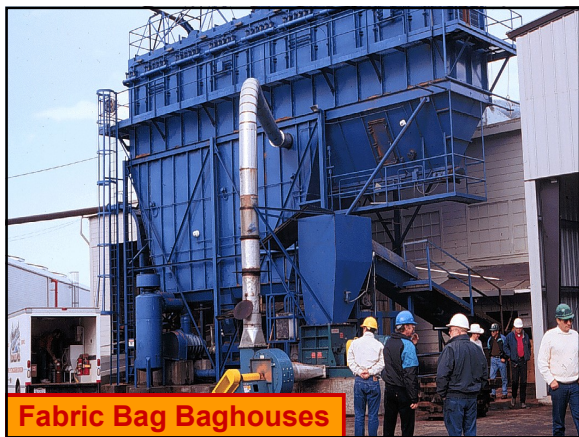
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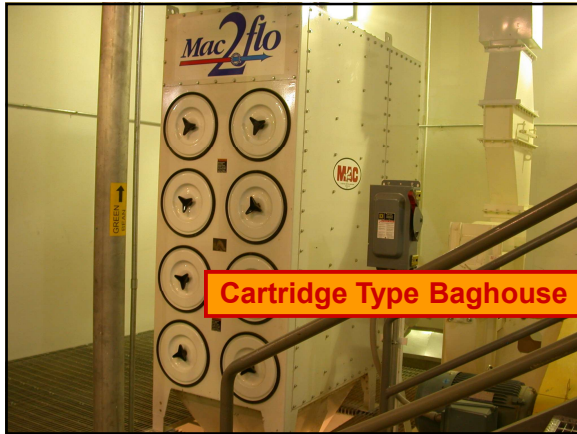
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# Introduction to Control Devices



**Cartridge Type Baghouse**

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**PM Control Techniques – Baghouse**

- **General description**
  - Generic name - dust collectors
  - Particles trapped on filter media, then removed
  - Either interior or exterior filtration systems
  - Forced Draft or Induced Draft fan
  - Require a cleaning mechanism

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**PM Control Techniques – Baghouse**

Forced Draft vs. Induced Draft		
Fan Type	Pros	Cons
<b>Forced</b>	<ul style="list-style-type: none"> <li>• Smaller motor</li> <li>• Less expensive</li> <li>• Easy to identify leaks</li> </ul>	<ul style="list-style-type: none"> <li>• Fan Blade Erosion</li> </ul>
<b>Induced</b>	<ul style="list-style-type: none"> <li>• Fan on clean side</li> <li>• Particulate Contained</li> </ul>	<ul style="list-style-type: none"> <li>• Larger motor</li> <li>• More expensive</li> <li>• Harder to identify leaks</li> </ul>

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# Introduction to Control Devices

## PM Control Techniques – Baghouse

### • Cleaning Mechanisms

#### 4 Types

- Mechanical Shaker (off-line)
- Reverse air (low pressure, long time, off line)
- Pulse jet (60 to 120 psi air, on line)
- Sonic horn (150 to 550 Hz @ 120 to 140 dB, on line) – rarely used alone

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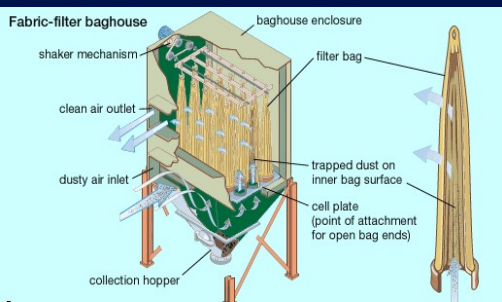
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## Baghouse Cleaning Methods

### Mechanical Shaker



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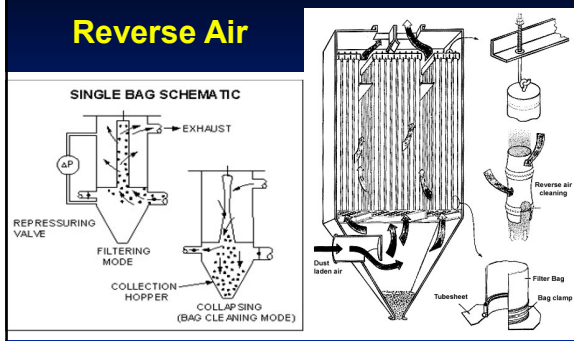
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## Baghouse Cleaning Methods

### Reverse Air



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# Introduction to Control Devices



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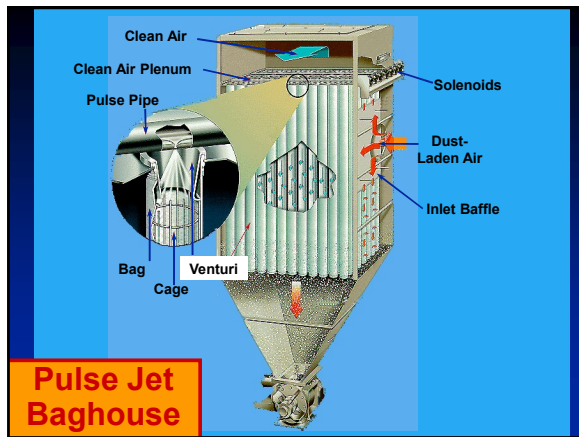
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# Introduction to Control Devices



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**Control Efficiency - Baghouse**

- Conventional Baghouses
  - 95% - 99.9% for  $PM_{10}$
  - 95% - 99% for  $PM_{2.5}$
- High Efficiency Particle Air (HEPA)
  - 99.97% for  $PM_{0.3}$
- Ultra Low Penetration Air (ULPA)
  - 99.9995% for  $PM_{0.12}$

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**Baghouse Design Considerations**

- Pressure Drop
- Air-To-Cloth Ratio
- Collection Efficiency
- Fabric Type
- Cleaning
- Temperature Control
- Space and Cost

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# Introduction to Control Devices

**Causes of Failure - Baghouse**

- Bag
  - Abrasion
  - High temperature
  - Chemical attack
  - Concretion of particulate
- Plenum
  - Abrasion
  - Chemical attack
  - Corrosion
- Outer Wall
  - Abrasion
  - Chemical attack
  - Corrosion
  - Physical Damage



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**Baghouse – Performance Indicators**

- Performance indicators
  - Outlet opacity (VEE)
  - Pressure differential
  - Outlet PM concentration (COMS)
  - Bag leak detectors
  - Exhaust gas flow rate
  - Cleaning mechanism operation
  - Inspections and maintenance

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**Monitoring Equipment**

- Magnehelic or Manometer ( $\Delta P$ )
- Continuous Opacity Monitoring Systems (COMS)
- Tribo Electric Sensors

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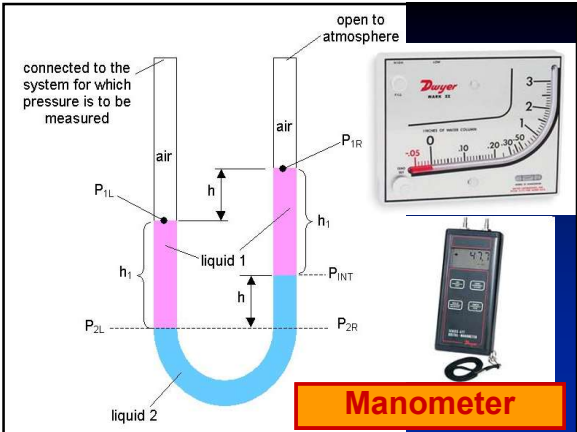
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# Introduction to Control Devices




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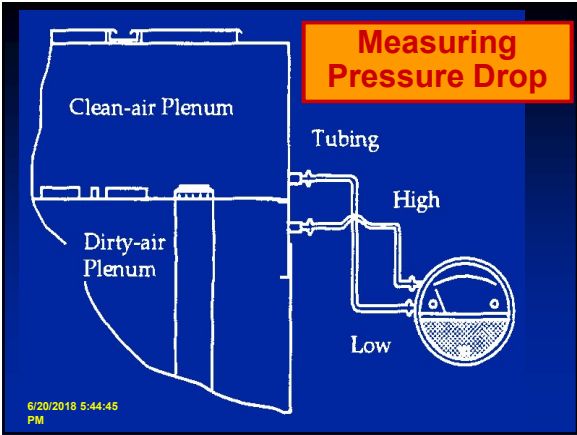
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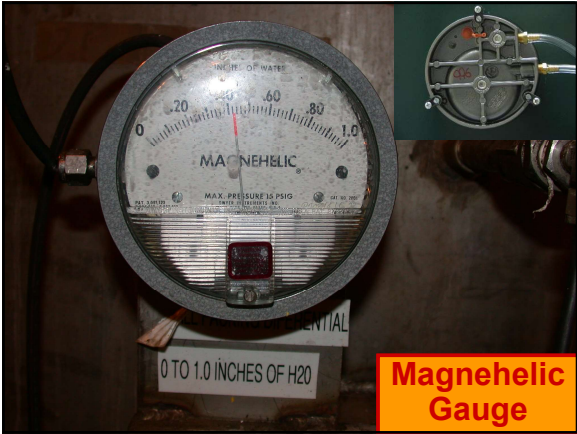
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# Introduction to Control Devices

## Baghouse Pressure Drop

- $\Delta P$  shows air flow – it's in operation
- $\Delta P$  may fluctuate 10% as a function of the bag cleaning cycle.
- Continued rise in  $\Delta P$  will result from bags that become permanently plugged (blinded).
- High  $\Delta P$  will lead to premature bag failure.
- Daily/weekly record of  $\Delta P$  can be a useful monitoring tool

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## PM COMS



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## Opacity

- Not very sensitive – shows a gross failure.
- Baseline (new) bag house opacity is probably  $\ll 1\%$
- Emissions must increase about 10x to be visible.
- Opacity useful where particulate emissions limit is high.

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# Introduction to Control Devices

## Trieboelectric Sensors

The diagram illustrates the components of a triboelectric sensor system. It includes a triboflow sensor, triboelectric electronics, a special cable, and a digital display. The display shows a scale of 'x SE 2' and a reading of '4208'.

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## Baghouse Monitoring

- Normal baghouse emissions are very low.
  - Opacity sensors (COM) aren't very good below 1-2%, so they don't detect initial problems.
  - Opacity will show a major particulate emissions increase.
  - COM or Method 9 may be OK for loose emission limits.

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## Tribo Electric Sensors

- Tribo electric sensors (TES) work well at very low particle concentrations (very sensitive).
- TES detects micro amp current from particles hitting a metal probe.
- TES is simple and inexpensive.
- TES is an effective monitor when a small to moderate increase in emissions is of concern.

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# Introduction to Control Devices

## BH Monitoring Summary

- Use TES for sensitive indication of changes in particulate emissions
- Opacity will indicate large increases in particulate emissions.
- An increasing pressure drop is indicative of long term problems

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## Baghouse : Secondary Containment



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## Let's Discuss Electrostatic Precipitators (ESP)



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# Introduction to Control Devices



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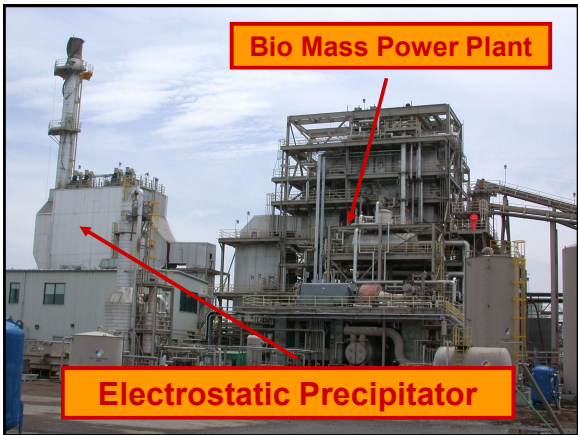
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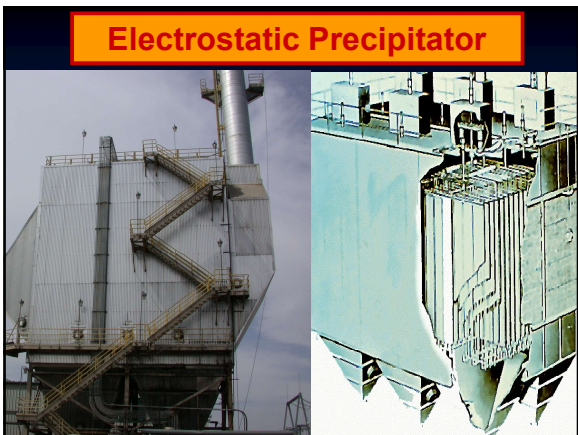
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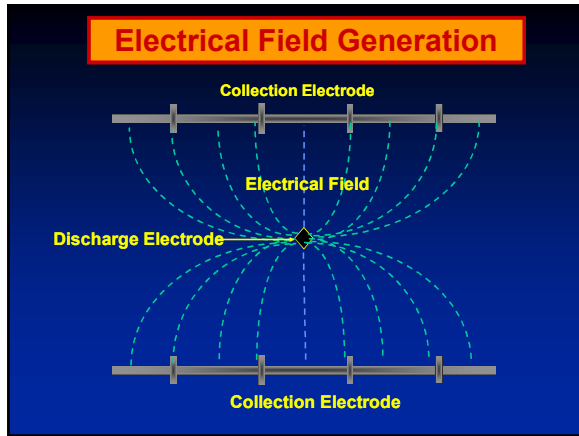
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# Introduction to Control Devices



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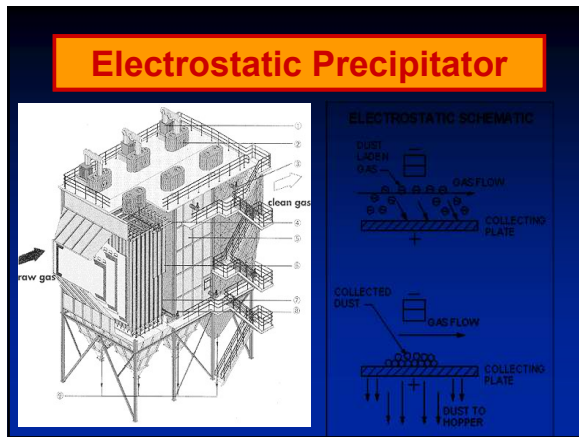
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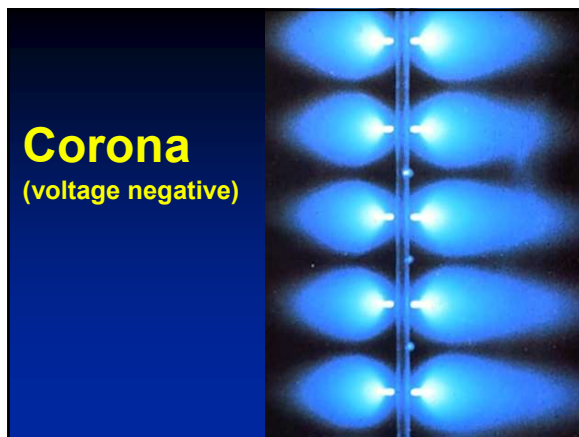
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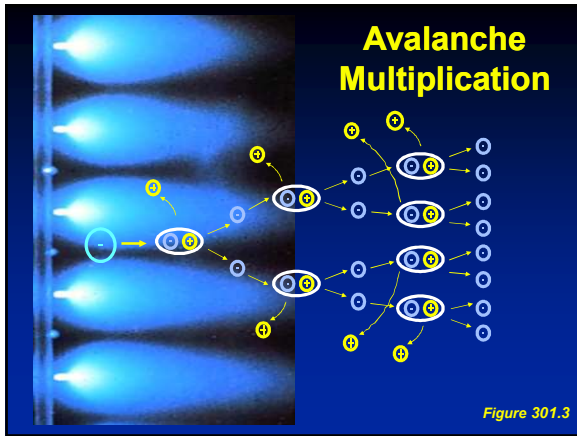
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# Introduction to Control Devices



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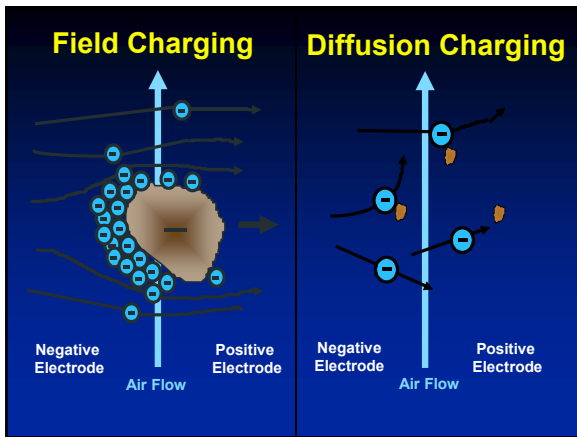
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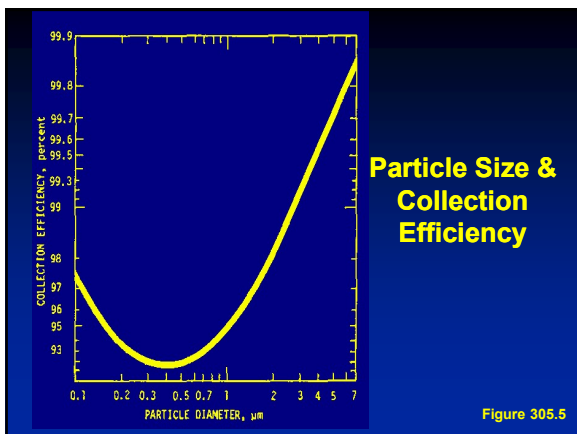
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# Introduction to Control Devices

## Electrostatic Precipitator

- **General Description**
  - **Two types**
    - **Dry type use mechanical action to clean plates**
    - **Wet type use water to prequench and to rinse plates**

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## Electrostatic Precipitator

- **General Description**
  - **High voltages are required**
    - **20,000 – 100,000 VDC**
  - **Multiple sections (fields) may be used**
  - **They usually can meet emission target with one field out of service or operating at reduced power**

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## Electrostatic Precipitator

- **General Description**
  - **High airflow rates**
    - **200,000 – 1,000,000 scfm**
  - **High temperatures**
    - **Up to 1,300 °F**
  - **Pollutant Loading**
    - **1 – 50 grains/scfm**

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# Introduction to Control Devices

## Electrostatic Precipitator

- A high voltage field creates a corona (current)
  - Particles are charged by electrons in the corona
  - The DC field draws charged particles to the plate
- Dust layers on the plates are cleared by mechanical rapping. Dust falls into the hoppers.
- Several fields in the direction of flow
  - Voltage/current to each is separately controlled
  - The first field collects most of the dust (75%)
  - Not much dust left in the last field

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## Mechanical Tumbling Hammer Rappers



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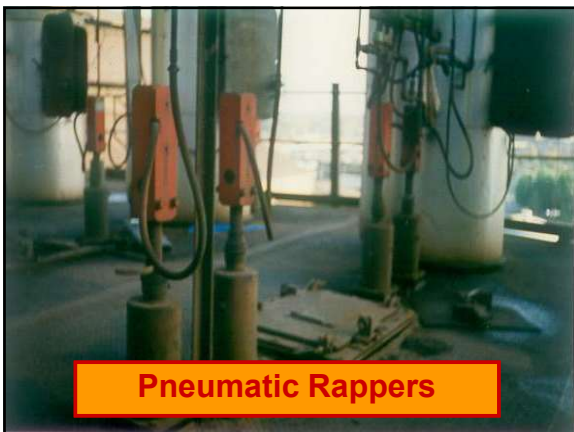
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## Pneumatic Rappers



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# Introduction to Control Devices



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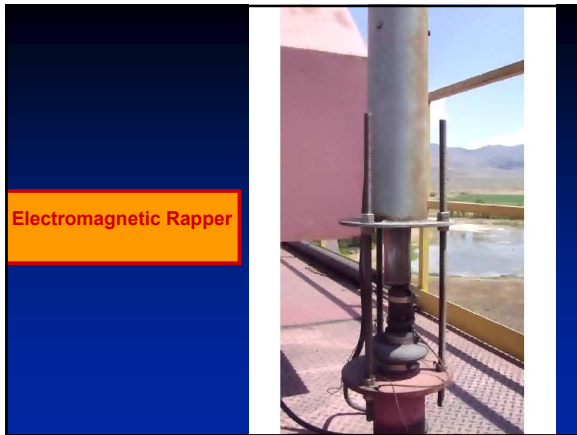
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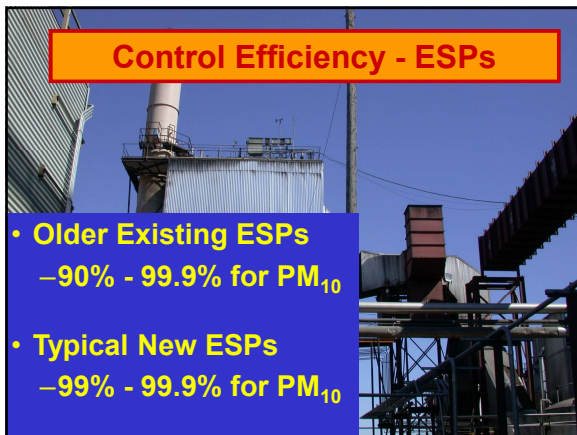
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# Introduction to Control Devices

## ESPs: Design Factors Affecting Performance

- Specific Collection Area
- Aspect Ratio
- Collection Plate Spacing
- Sectionalization
- Power Requirements/Spark Rate

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## Electrostatic Precipitator

- Factors affecting efficiency
  - Gas temperature, humidity, flow rate
  - Particle resistivity
  - Fly ash/Fuel composition
  - Plate length
  - Surface area

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## Electrostatic Precipitator

- Factors affecting efficiency
  - ESP is sensitive to gas flow rate
  - Flow monitoring may be appropriate
  - An ESP won't work well if the velocity distribution is not uniform.
- ESP internal factors
  - Dust layer thickness & electrical resistance.
  - Changes in geometry (damage)
  - Air leaks, condensation

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# Introduction to Control Devices

## Electrostatic Precipitator

Baseline operating and emission data is needed to establish:

- Emissions level and control capability at max gas flow.
  - Does it work as intended?
  - Typical secondary current and voltage levels
- Operating margin - number of fields and power required to meet emission requirements.
- Normal operating temperature.

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## Electrostatic Precipitator

### • Performance indicators

- Outlet opacity (VEE)
- Pressure differential
- Outlet PM concentration (COMS)
- Secondary corona power (current & voltage)
- Spark rate
- Primary power (current & voltage)

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## ESP: Performance Indicators



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# Introduction to Control Devices

## Electrostatic Precipitator

- Performance indicators (cont.)
  - Inlet gas temperature
  - Gas flow rate
  - Rapper operation
  - Fields in operation
  - Inlet water flow rate (wet type)
  - Flush water solids content (wet type)

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## Summary of ESP Monitoring

- Obtain convincing baseline emissions data
  - Linked to flow rate, power levels and type of fuel
- Key monitoring parameters
  - Opacity
  - Electrical power levels (Secondary I & V)
- Secondary parameters
  - Temperature
  - Fuel composition
  - Inspection & routine maintenance

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# Introduction to Control Devices



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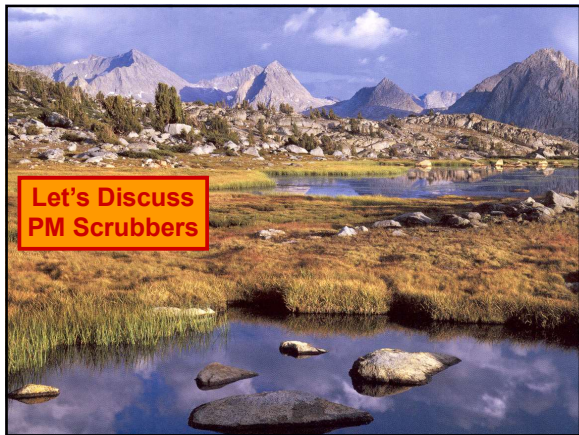
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**Control Techniques – Wet Scrubber**

- **General description**
  - Particles (and gases) get trapped in liquids
    - Inertial impaction and diffusion
  - Liquids must contact pollutants and dirty liquids must be removed from exhaust gas
  - Four types
    - Spray; venturi or orifice; spray rotors; and moving bed or packed towers

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# Introduction to Control Devices



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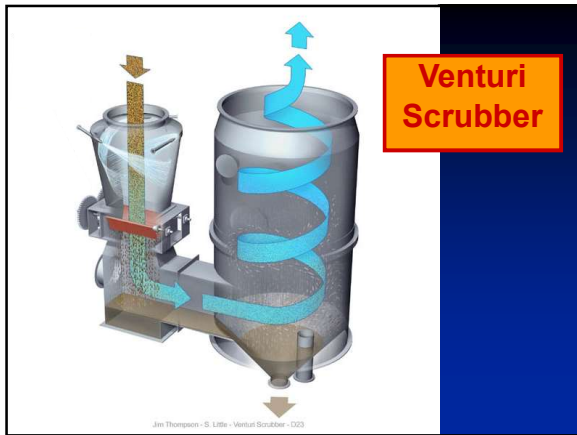
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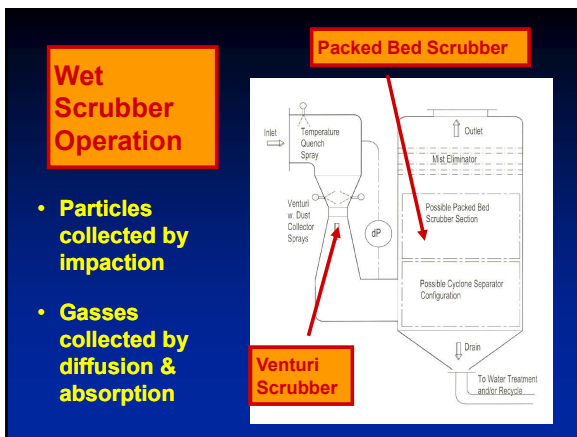
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# Introduction to Control Devices

## Venturi Scrubbers

- Control Efficiency
  - 70 - 99% for  $PM_{10}$
- Moderate airflow rates
  - 500 – 100,000 scfm
- Moderate temperatures
  - Up to 750 °F
- Pollutant Loading
  - 0.1 – 50 grains/scfm

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## Scrubber Control Efficiency

- Factors affecting efficiency
  - Gas and liquid flow rate
  - Condensation of aerosols
  - Poor liquid distribution
  - High dissolved solids content in liquid
  - Nozzle erosion or pluggage
  - Re-entrainment
  - Scaling

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## Scrubber Monitoring

- Venturi pressure drop ( $\Delta P$ )
  - The higher the  $\Delta P$  the smaller the collected particles
  - Some venturis have adjustable vanes
- Water flow rate (gallons/min)
  - Flow below a critical level will degrade performance.
- Water cleanliness – evaporated residue & mist carryover.

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# Introduction to Control Devices

## Scrubber Performance

- **Performance indicators**
  - Pressure differential
  - Liquid flow rate
  - Gas flow rate
  - Scrubber outlet gas temperature
  - Makeup / blowdown rates
  - Scrubber liquid solids content (PM)

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## Scrubber Performance

- **Performance indicators (continued)**
  - Scrubber inlet gas and process exhaust gas temperature (PM)
  - Scrubber liquid pH (Acid gas)
  - Neutralizing chemical feed rate (Acid gas)
  - Scrubber liquid specific gravity (Acid gas)

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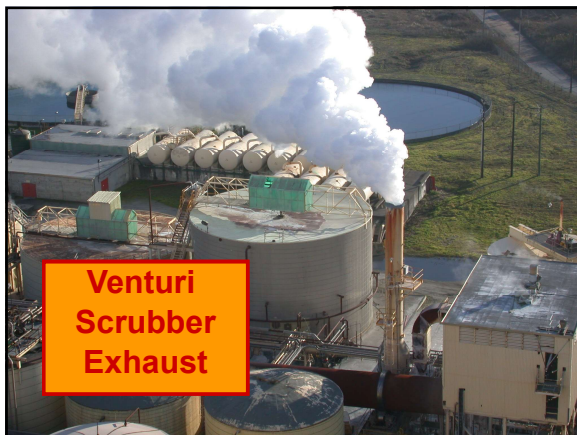
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# Introduction to Control Devices



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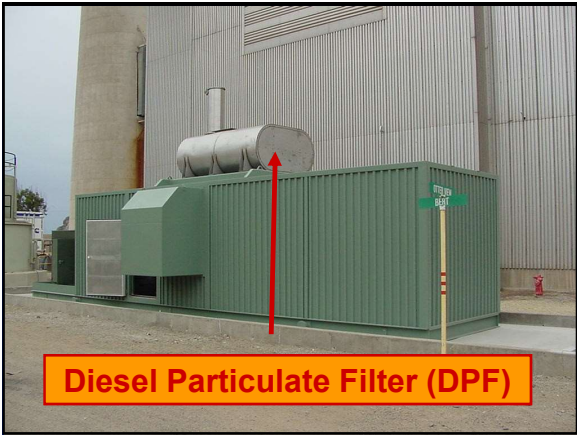
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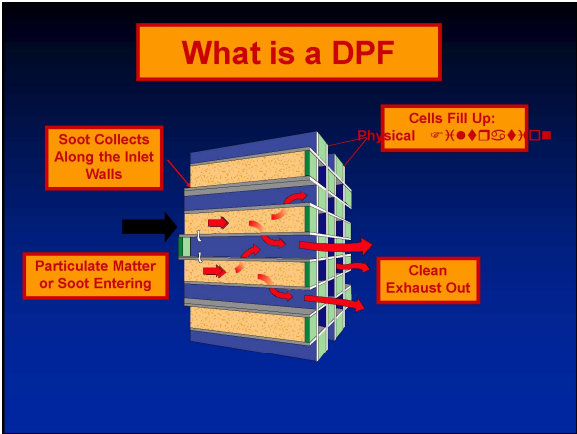
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# Introduction to Control Devices



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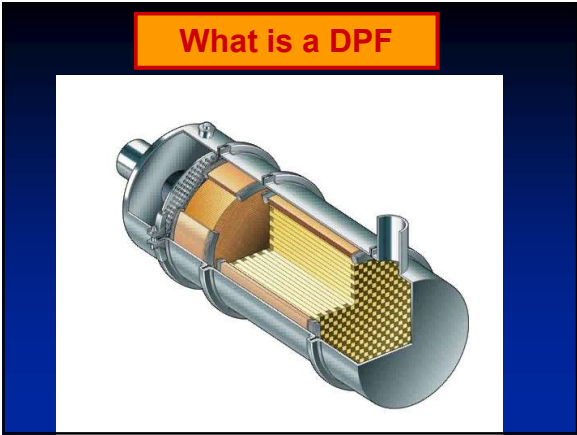
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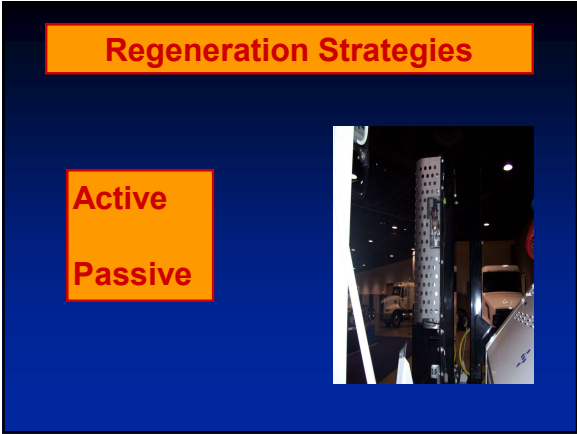
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# Introduction to Control Devices

**Diesel Particulate Filter (DPF)**

- ◆ High temperature regeneration (600-650 °C)  
 $C + O_2 \rightarrow CO_2$
- ◆ Catalytic regeneration (~250 °C)
- ◆ Oxidize NO to  $NO_2 \rightarrow$  adsorbs  $\rightarrow$  reduces regeneration temperature
- ◆ Fuel-borne catalyst
- ◆ Ceramic coatings
- ◆ Engine adjustments necessary
- ◆ Total PM efficiency > 90%

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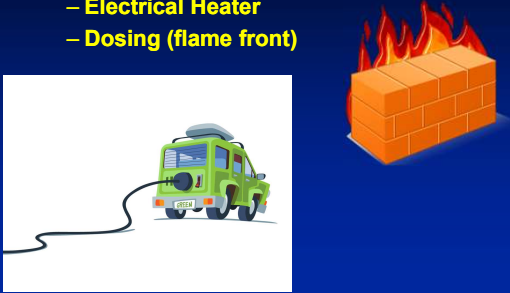
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**Active Regeneration**

- Achieving 550°C
  - Electrical Heater
  - Dosing (flame front)



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**Electrical Heater**

- Uses a heating element similar to an electric stove
- Performed while vehicle is offline



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# Introduction to Control Devices



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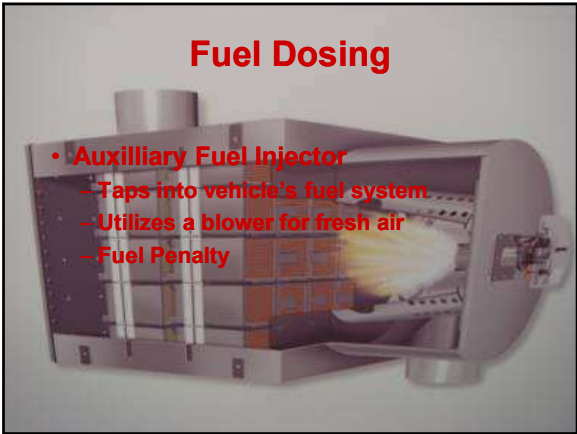
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# Introduction to Control Devices



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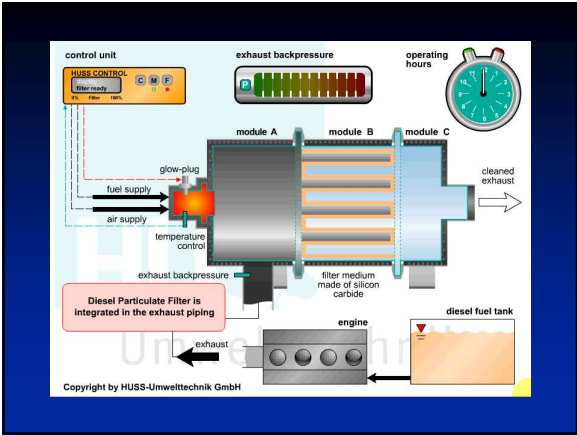
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# Introduction to Control Devices

## Active Regeneration (O<sub>2</sub> @ 550° C)

- **Electrical**
  - Online Electrical (Rypos)
  - Offline/Off-board
  - Offline/On-board
- **Fuel Dosing**
  - Flame front using auxiliary injection and vehicle's fuel supply
  - Air intake



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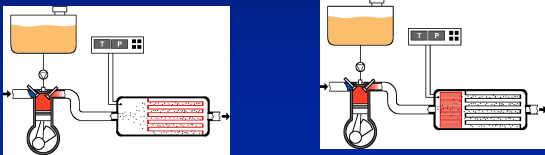
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## Passive Regeneration

- NO<sub>2</sub> oxidizes soot @ around 250° C
- NO<sub>2</sub> generation
  - Diesel Oxidation Catalysts
  - Catalyzed Filters
  - Fuel Born Catalysts (FBC)?



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## Soot & Ash

Definition of Soot  
Soot is a byproduct of incomplete combustion



- Soot Cleaning**
- DPF Collects the soot
  - Elevated exhaust temperatures convert the soot to vapor

Definition of Ash  
Ash is Noncombustible residue of a lubricating oil or fuel



- Ash Cleaning**
- DPF Collects the ash
  - Ash is removed using a special service tool

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# Introduction to Control Devices

**Aftertreatment Regeneration Device (ARD)**

- **What is ARD?**
  - ARD is the device that increases exhaust gas temperature to enable regenerate the DPF
- **What are the benefits of the CRS System?**
  - Regenerates under all conditions

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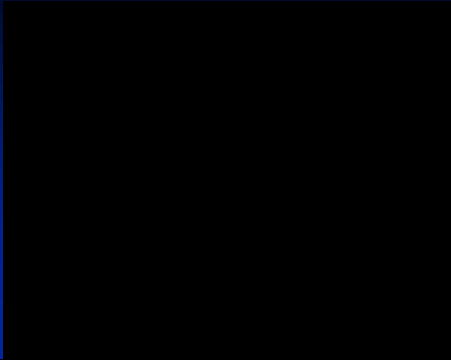
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**Comical Relief**



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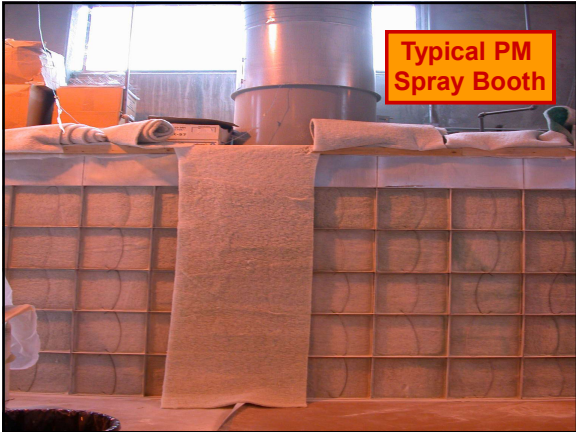
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**Typical PM Spray Booth**

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# Introduction to Control Devices



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# Introduction to Control Devices



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# Introduction to Control Devices



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**3-way Catalyst:  
Non-Selective Catalytic Reduction**

- Rich burn/NG fired engine
- $2\text{CO} + 2\text{NO} \rightarrow 2\text{CO}_2 + \text{N}_2$
- $\text{NO} + \text{HC} + \text{O}_2 \rightarrow \text{N}_2 + \text{CO}_2 + \text{H}_2\text{O}$
- 98% control for  $\text{NO}_x$  & CO

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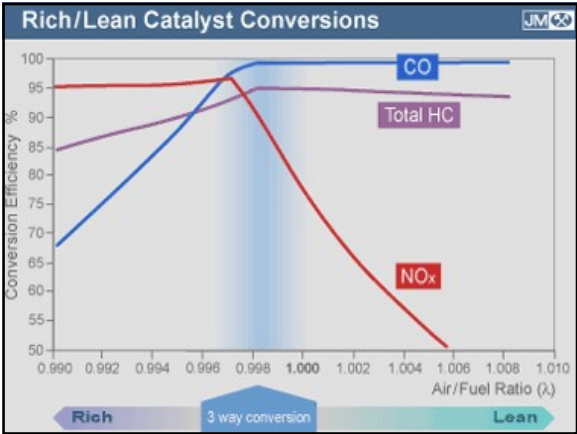
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# Introduction to Control Devices



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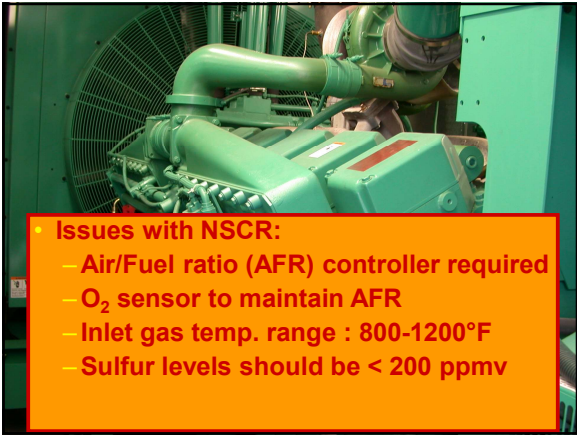
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# Introduction to Control Devices

Comparison of NOx Control Technologies – Gas-Fired Boilers			
Technology	Approx. Reduction	Approx. lbs/MMBTU	Approx. ppmv @ 3% O2
Standard burners	Base case	0.14	120
Low NOx burners	60%	0.06	45
Ultra Low NOx Burners – 1 <sup>st</sup> gen.	80%	0.03	25
Ultra Low NOx Burners – 2 <sup>nd</sup> gen.	95%	0.007	6
FGR	55%	0.025	20
Compu- NOx w/ FGR	90%	0.015	12
SNCR	40%	0.033 - 0.085	27 - 70
Catalytic Scrubbing	70%	0.017 - 0.044	14 - 36
SCR	90 – 95%	0.006 - 0.015	5 - 12

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